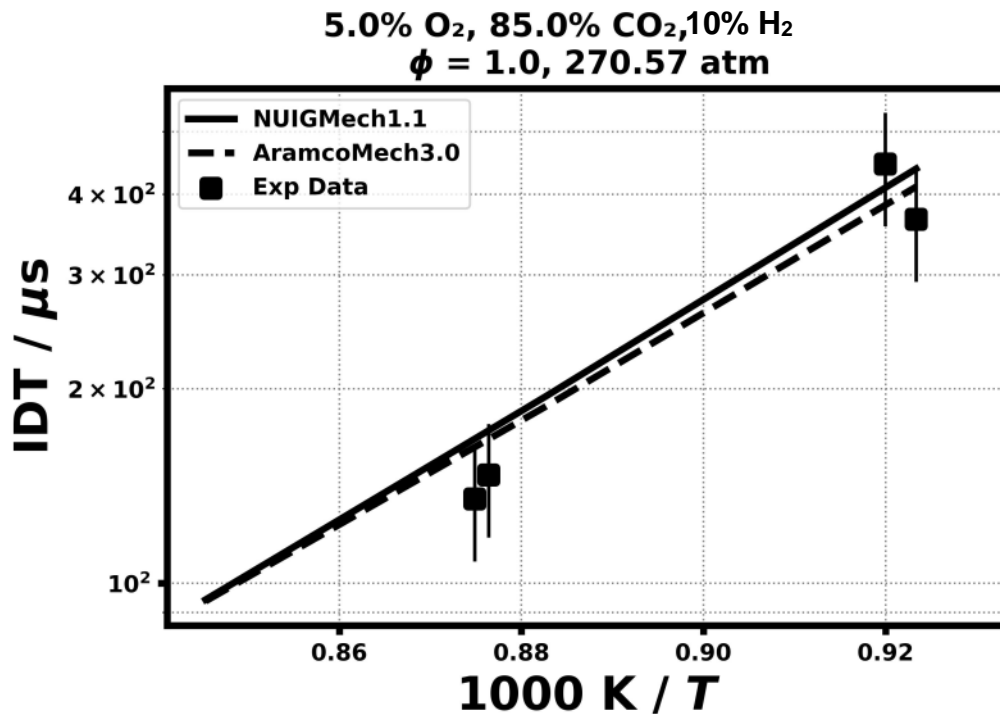
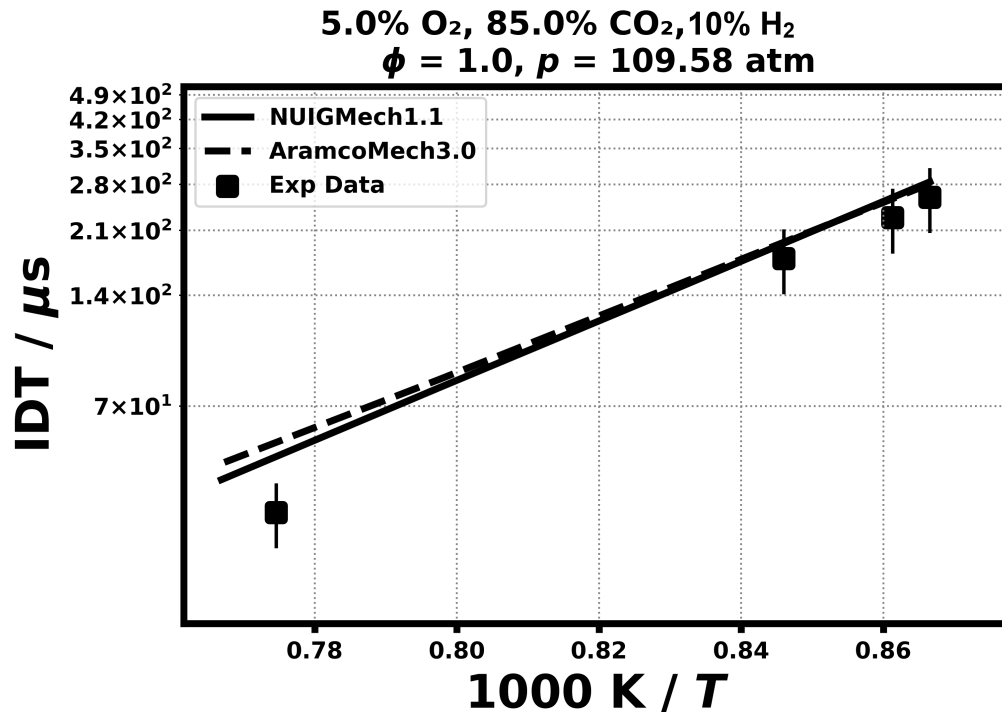


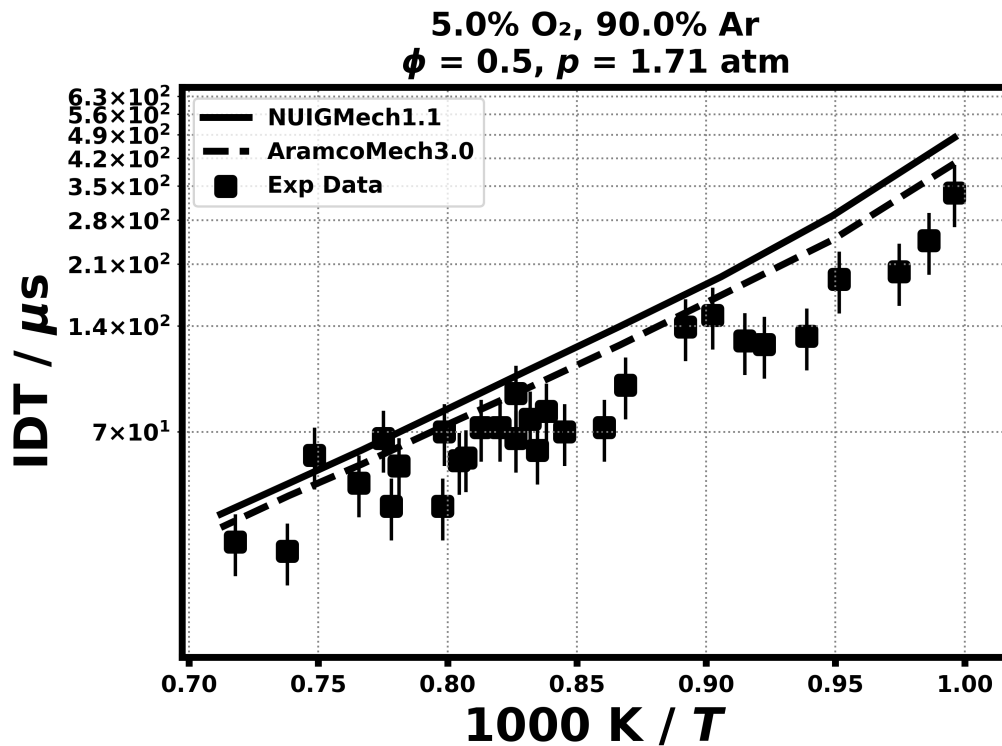
1. Validation for H₂

Shock tube ignition delay time

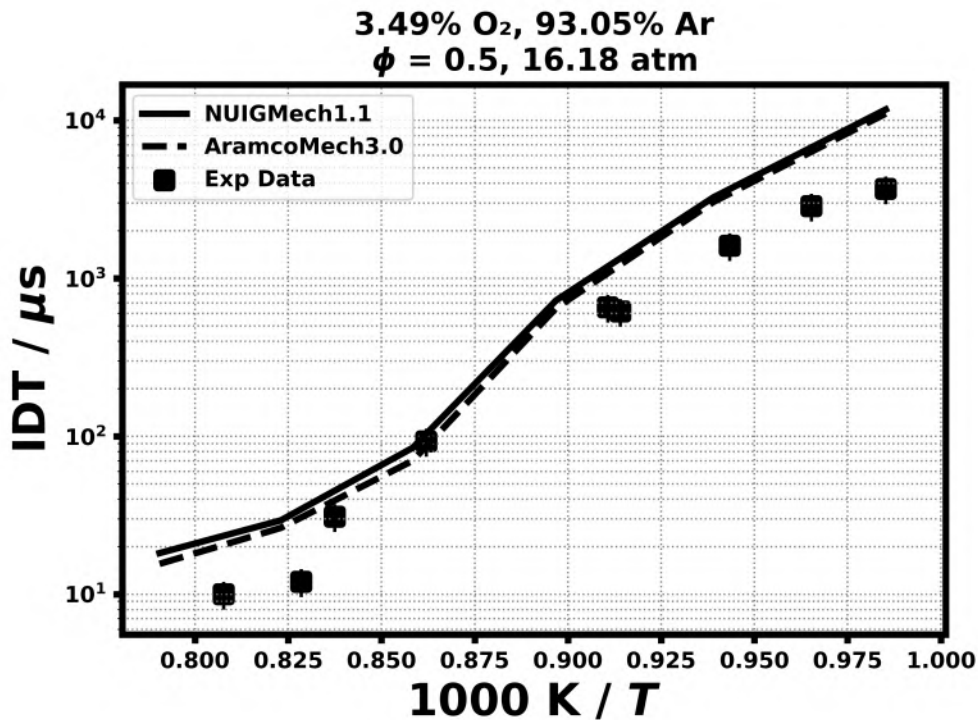
1.1) Shao, J., Choudhary, R., Davidson, D. F., Hanson, R. K., Barak, S., & Vasu, S. Proceedings of the Combustion Institute, 37(4) (2019) 4555-4562.



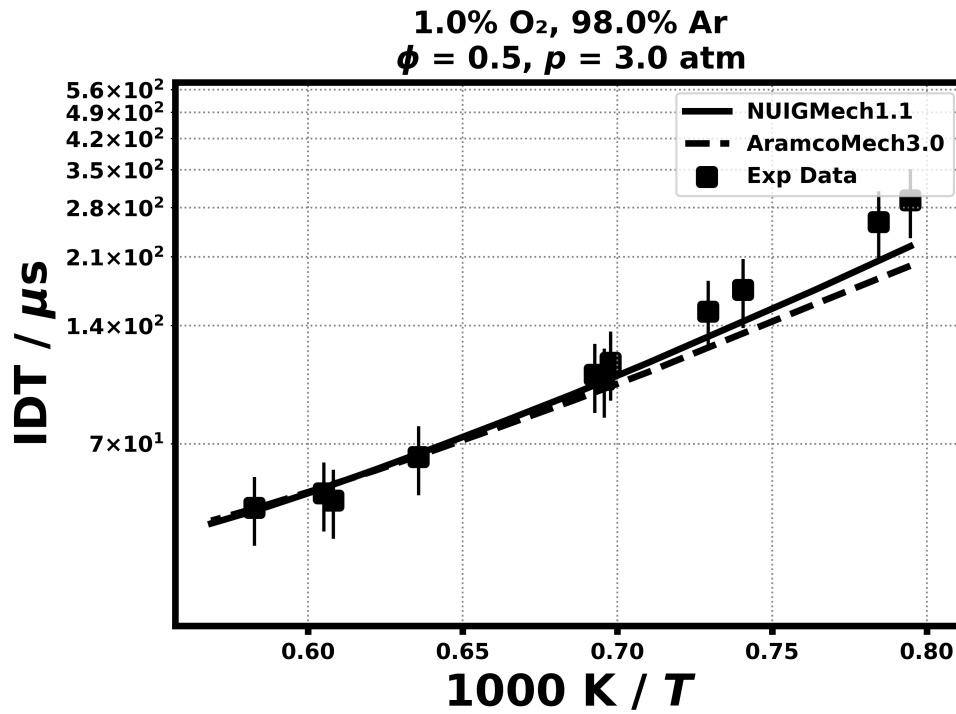
1.2) Cheng, R. K., & Oppenheim, A. K., Combustion and flame 58(2) (1984) 125-139.



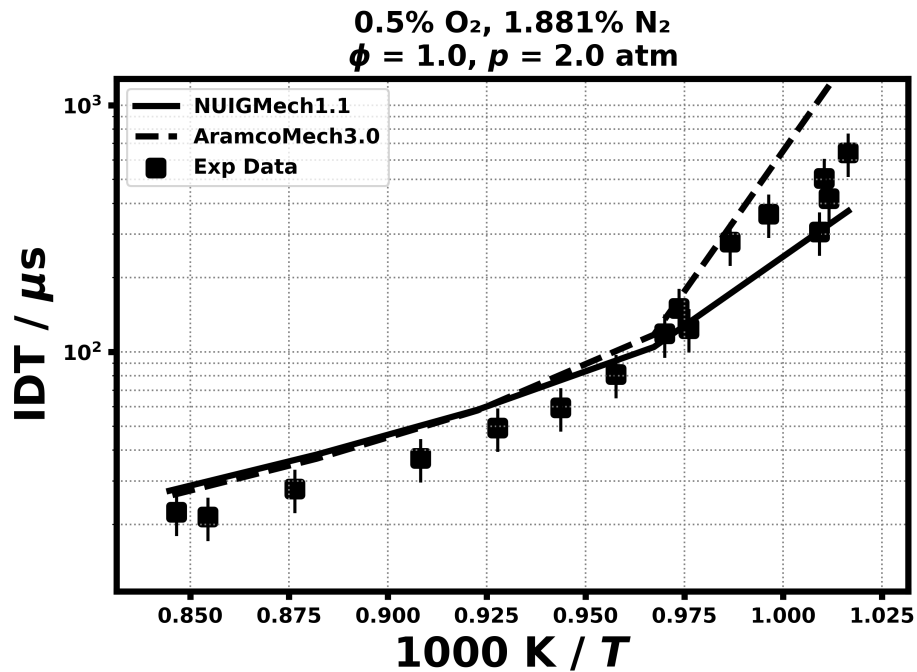
1.3) Herzler, J., & Naumann, C., Proceedings of the combustion institute, 32(1) (2009) 213-220.



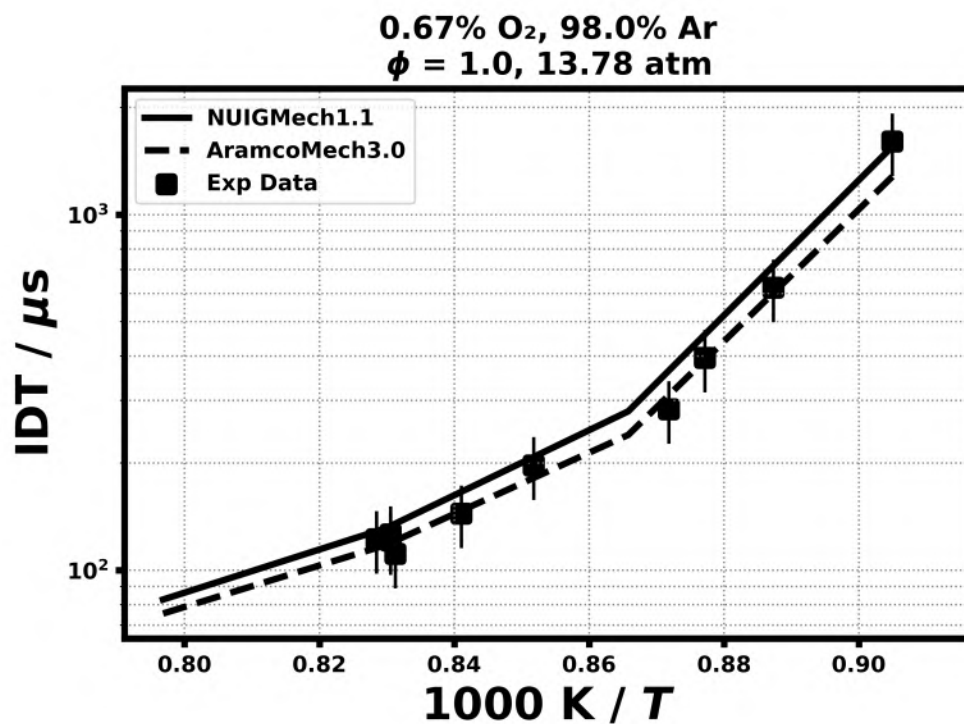
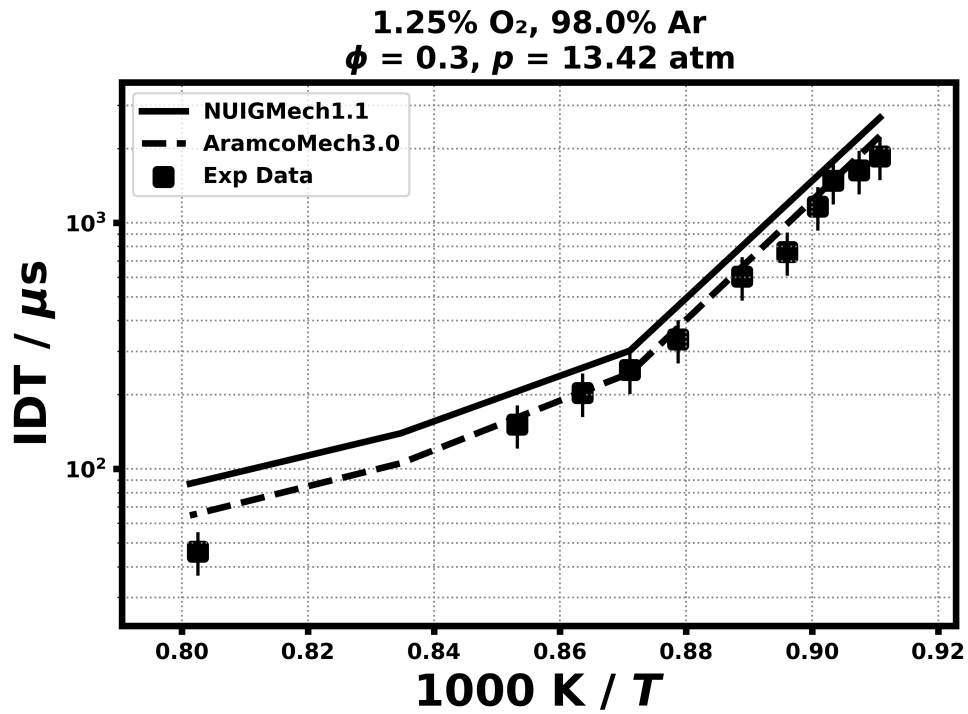
1.4) Hidaka, Y., Sato, K., Henmi, Y., Tanaka, H., & Inami, K., Combustion and flame, 118(3) (1999) 340-358.



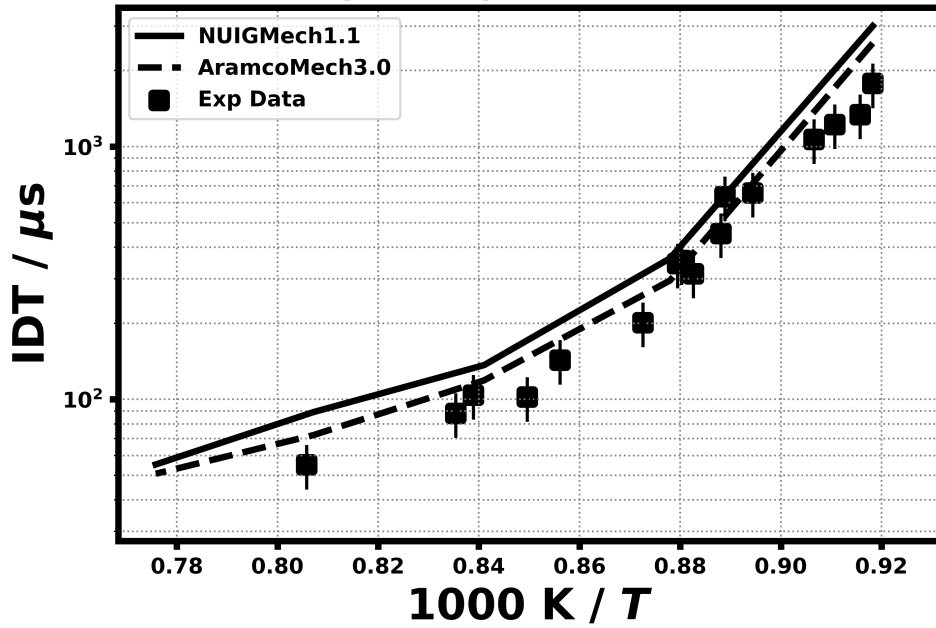
1.5) M. W. Slack, Combustion and Flame, 28 (1977) 241-249.



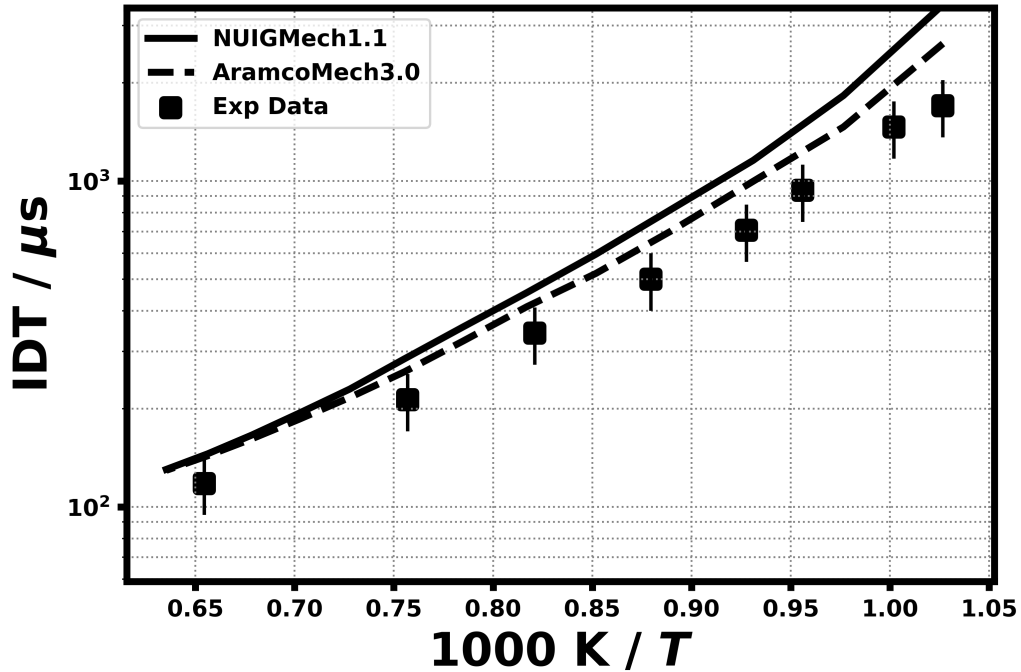
1.6) Kéromnès, A., Metcalfe, W. K., Heufer, K. A., Donohoe, N., Das, A. K., Sung, C. J., & Krejci, M. C., Combustion and Flame, 160(6), (2013) 995-1011.



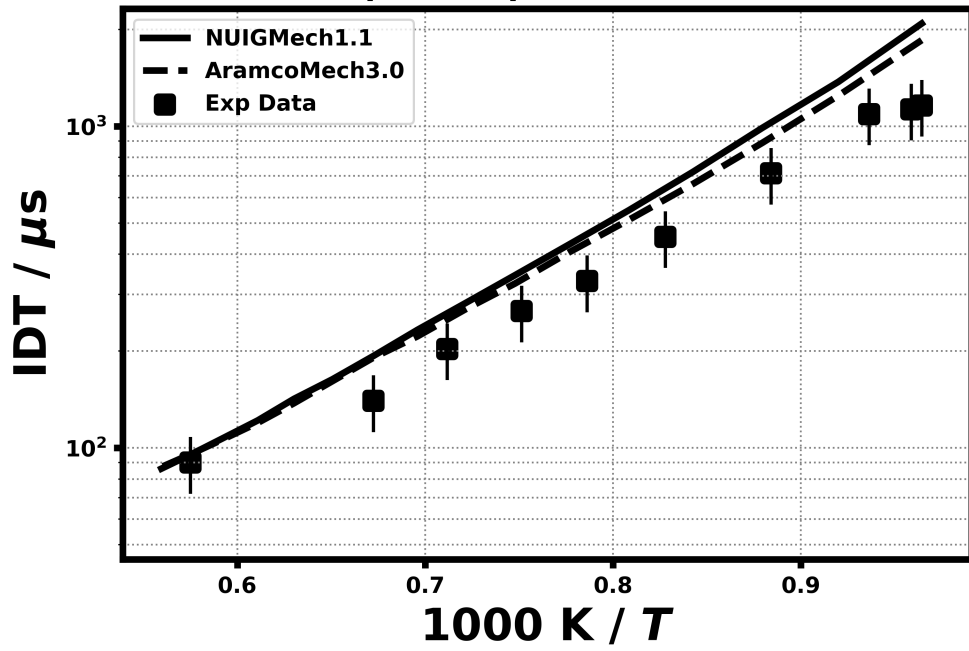
1.0% O₂, 98.0% Ar
 $\phi = 0.5, p = 13.29 \text{ atm}$



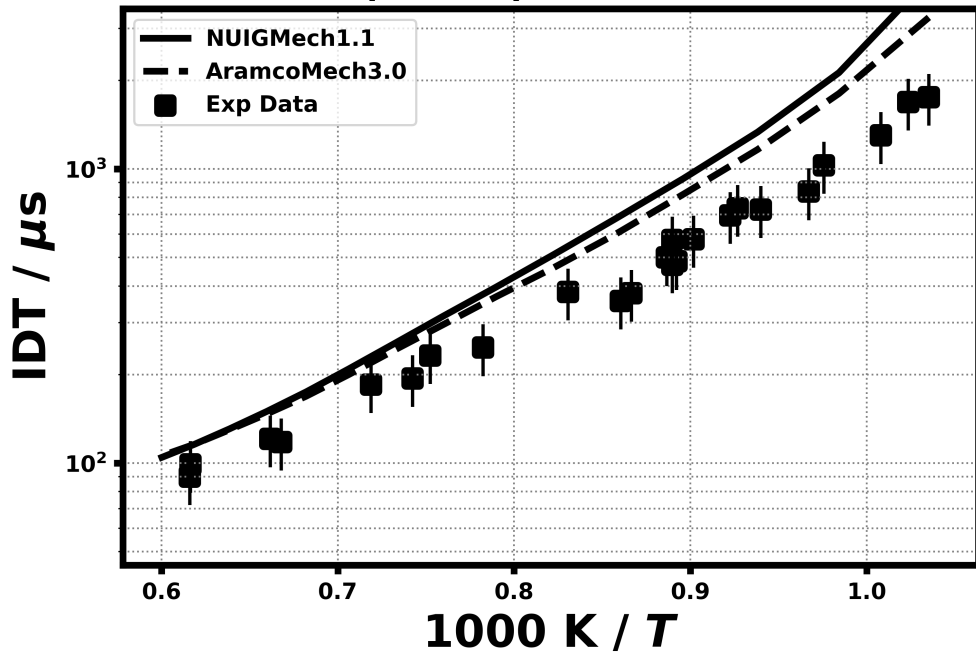
1.25% O₂, 98.0% Ar
 $\phi = 0.3, p = 1.67 \text{ atm}$



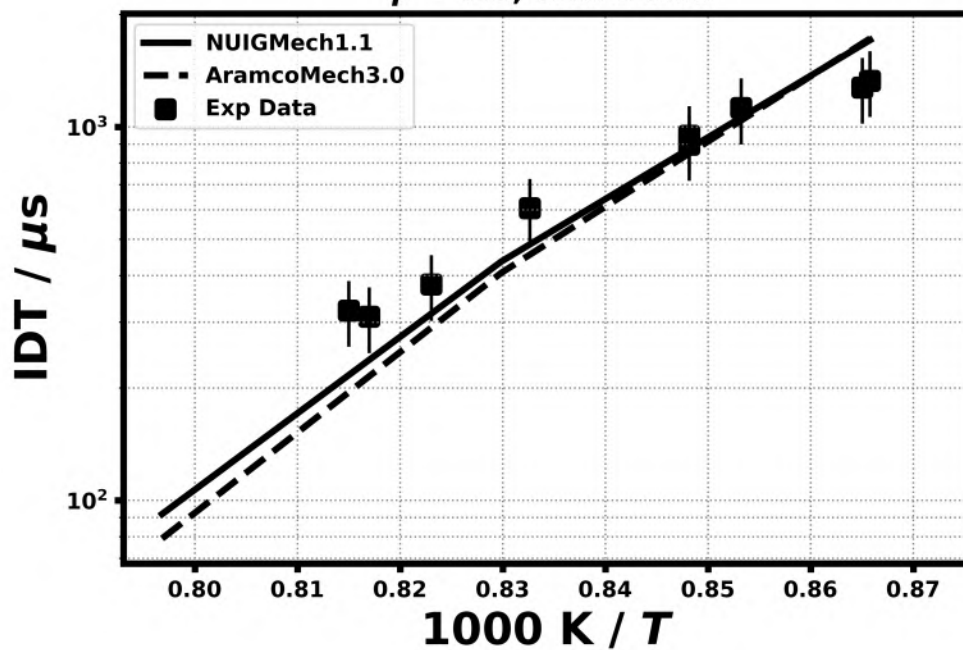
0.67% O₂, 98.0% Ar
 $\phi = 1.0, p = 1.67$ atm



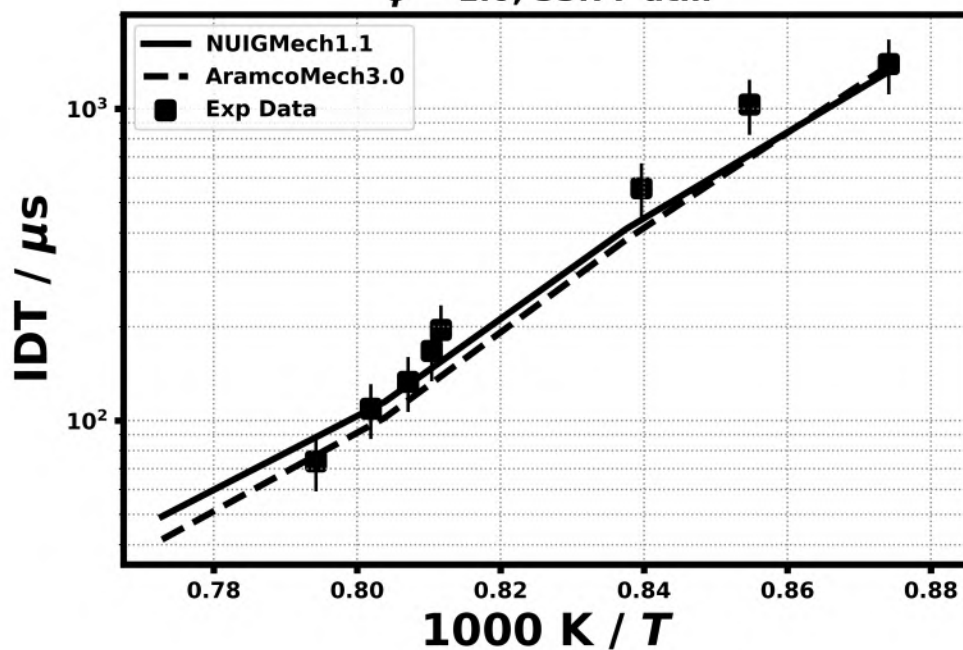
1.0% O₂, 98.0% Ar
 $\phi = 0.5, p = 1.62$ atm

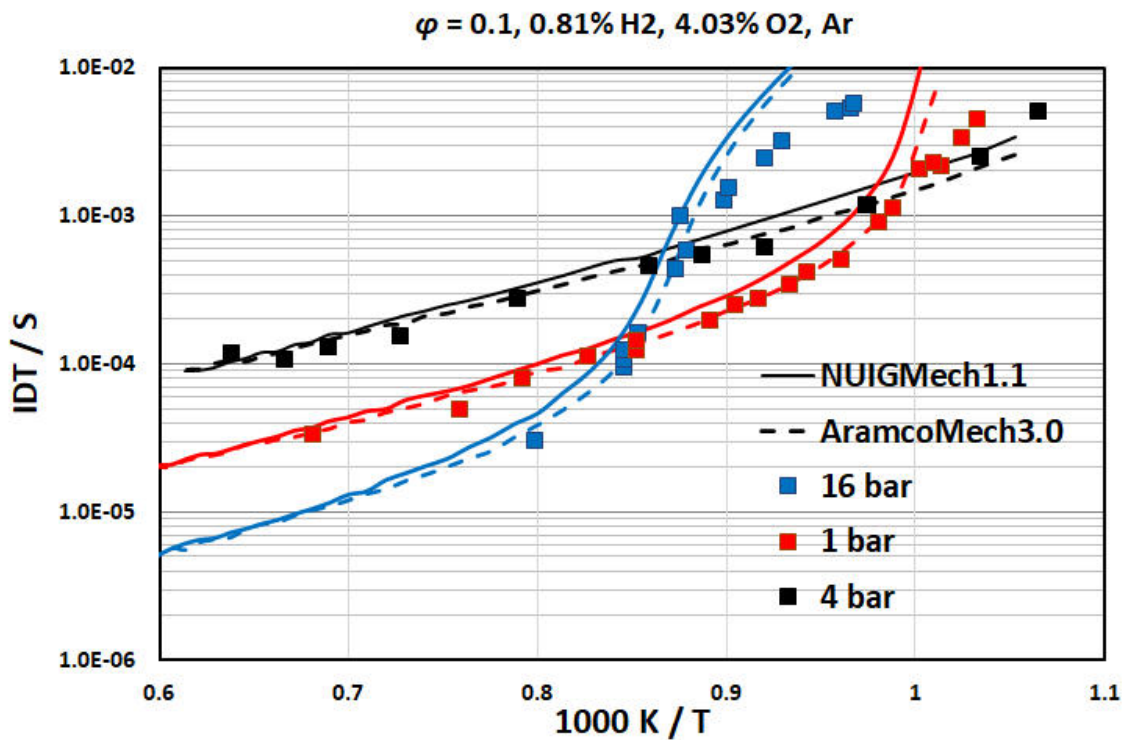


1.25% O₂, 98.0% Ar
 $\phi = 0.3, 32.71 \text{ atm}$



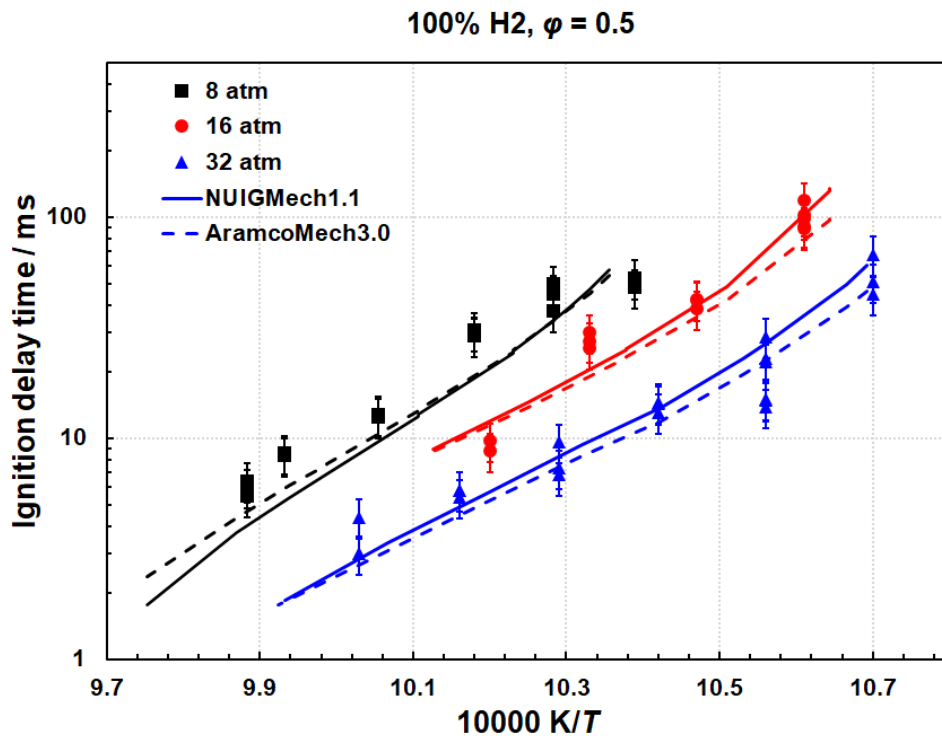
0.67% O₂, 98.0% Ar
 $\phi = 1.0, 33.77 \text{ atm}$





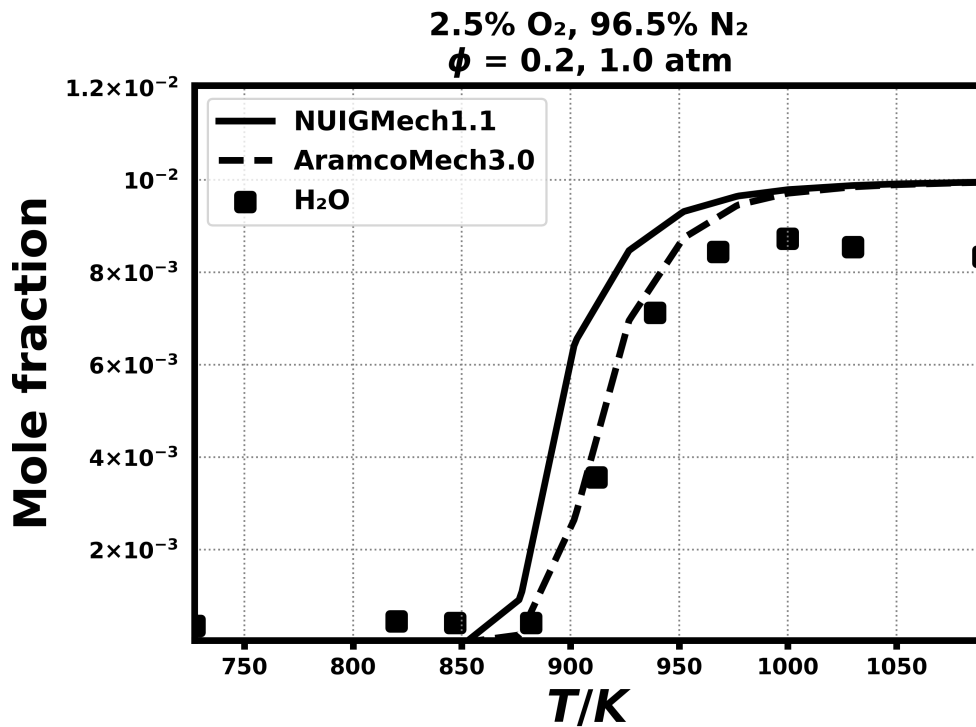
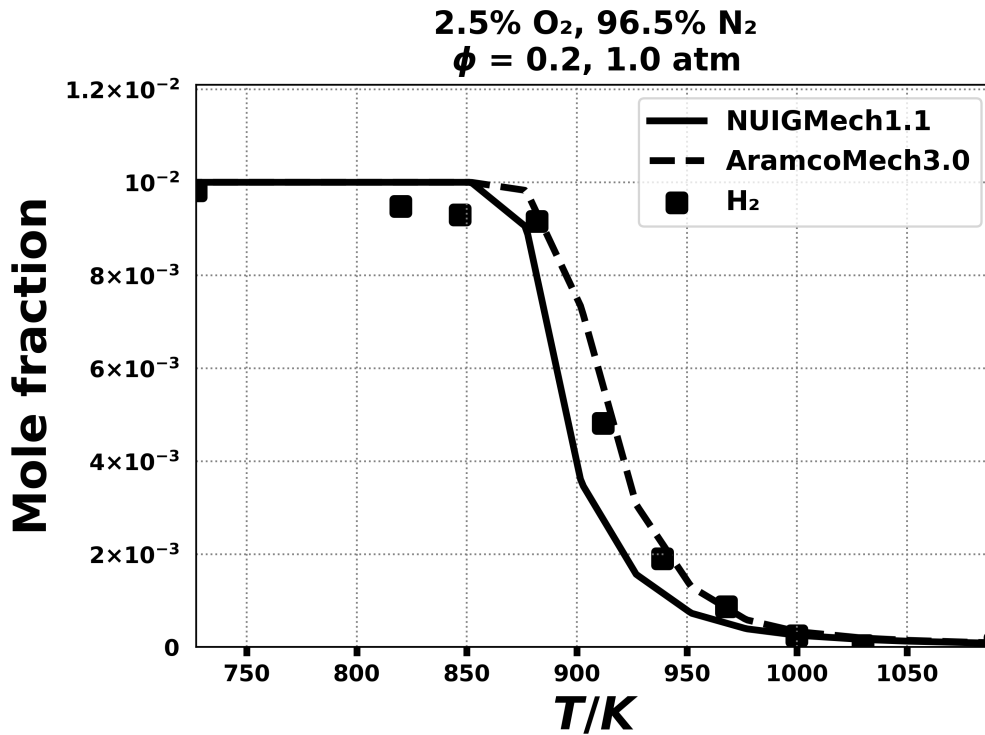
RCM Ignition delay time

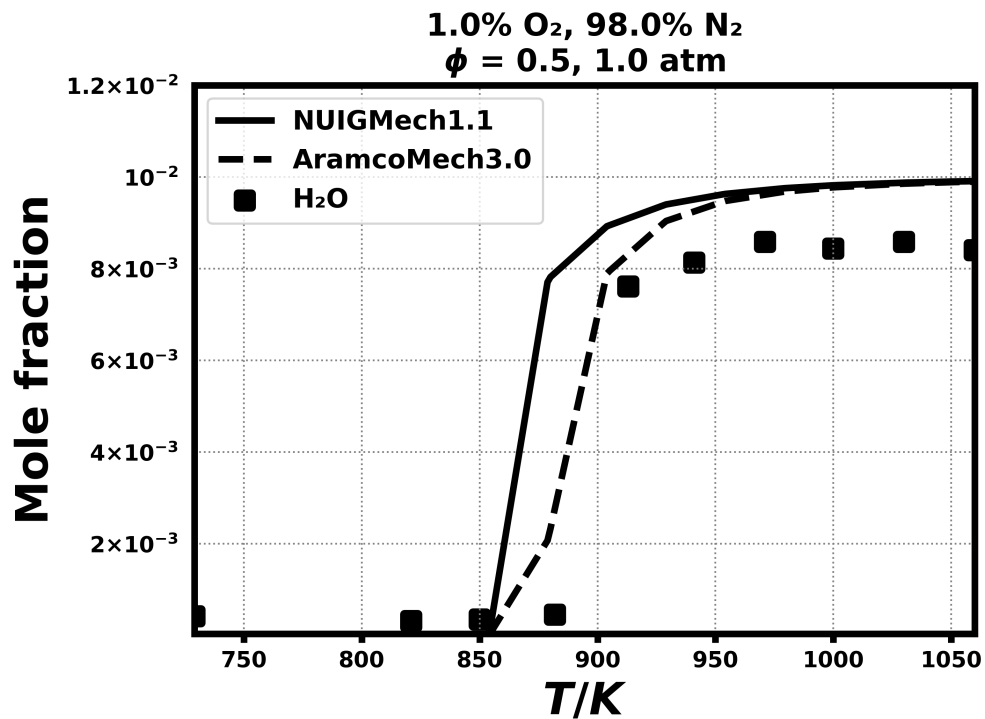
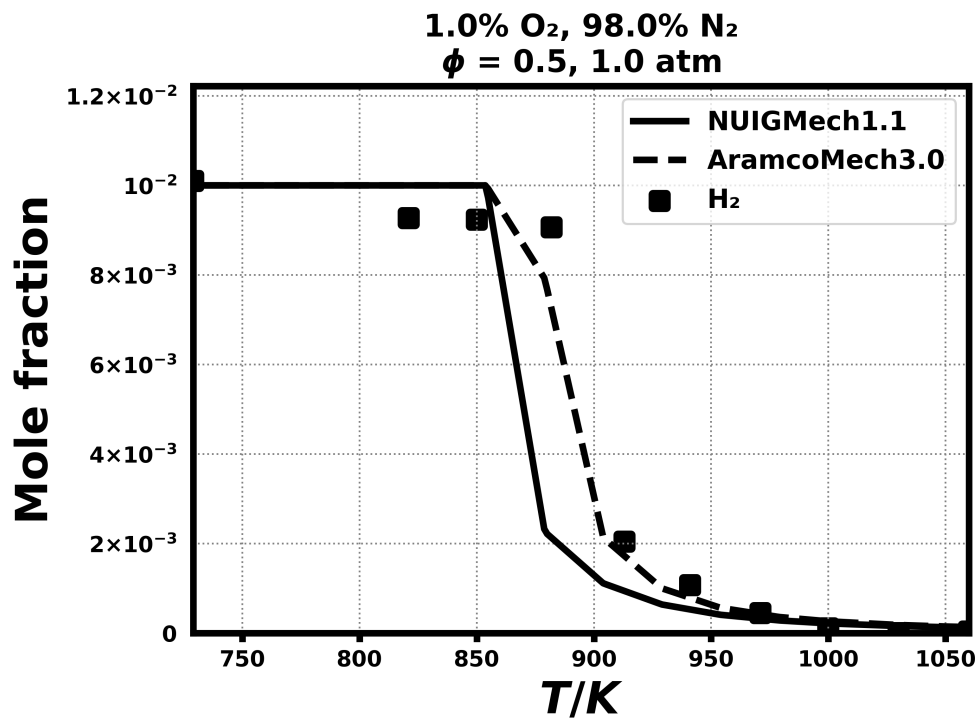
1.7) Kéromnès, A., Metcalfe, W. K., Heufer, K. A., Donohoe, N., Das, A. K., Sung, C. J., & Krejci, M. C., *Combustion and Flame*, 160(6), (2013) 995-1011.

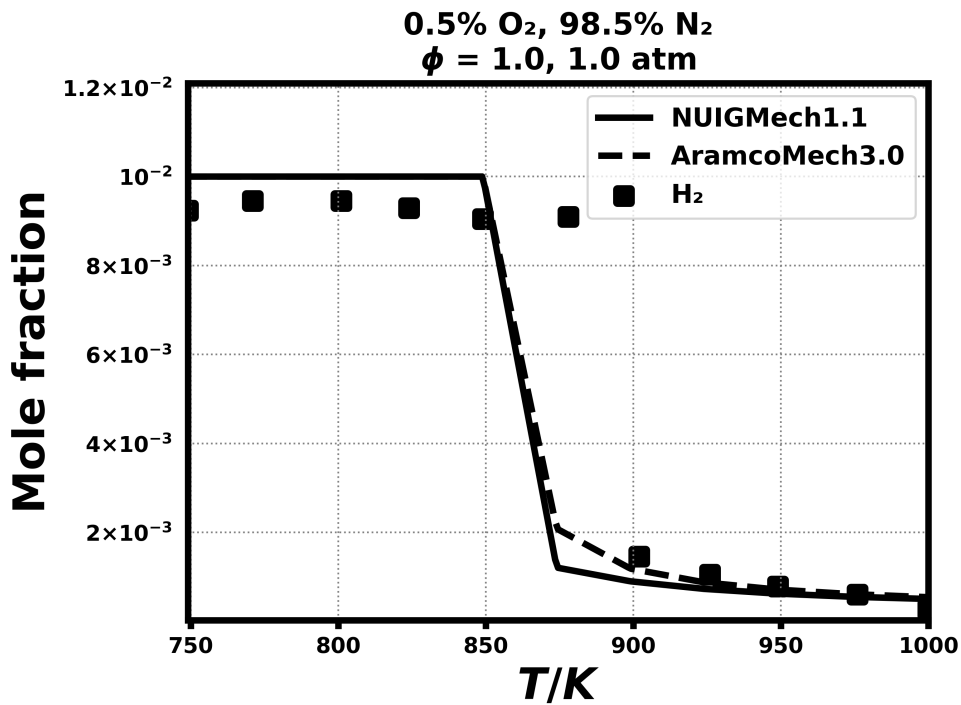


Speciation in Jet-stirred reactor

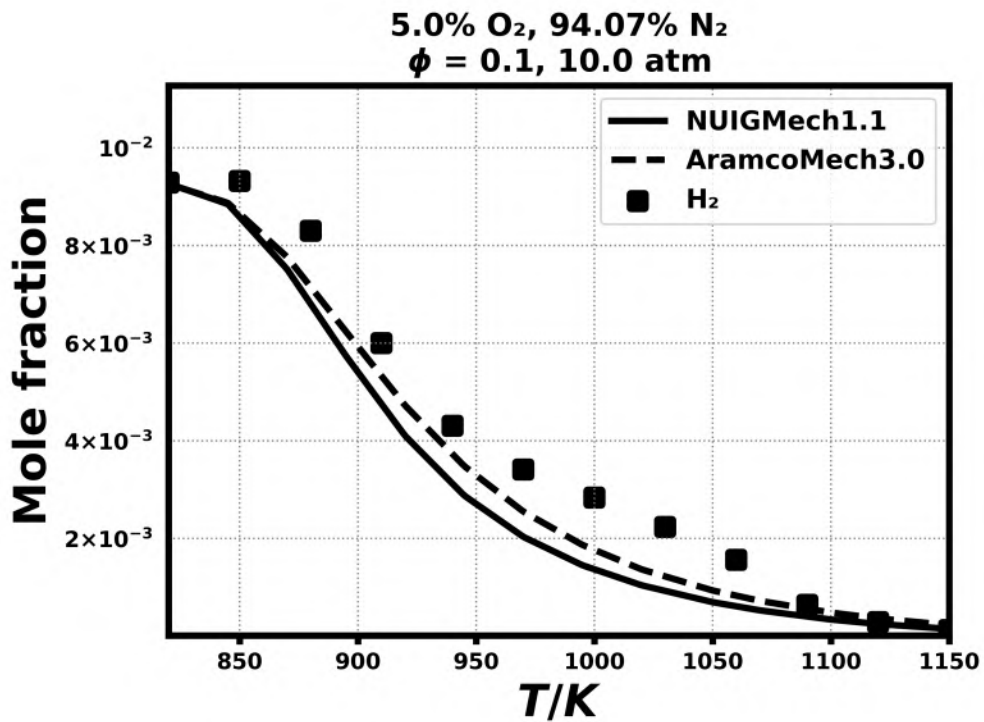
1.8) G. Dayma et al., Combustion science and technology, 178(10-11), 1999-2024.

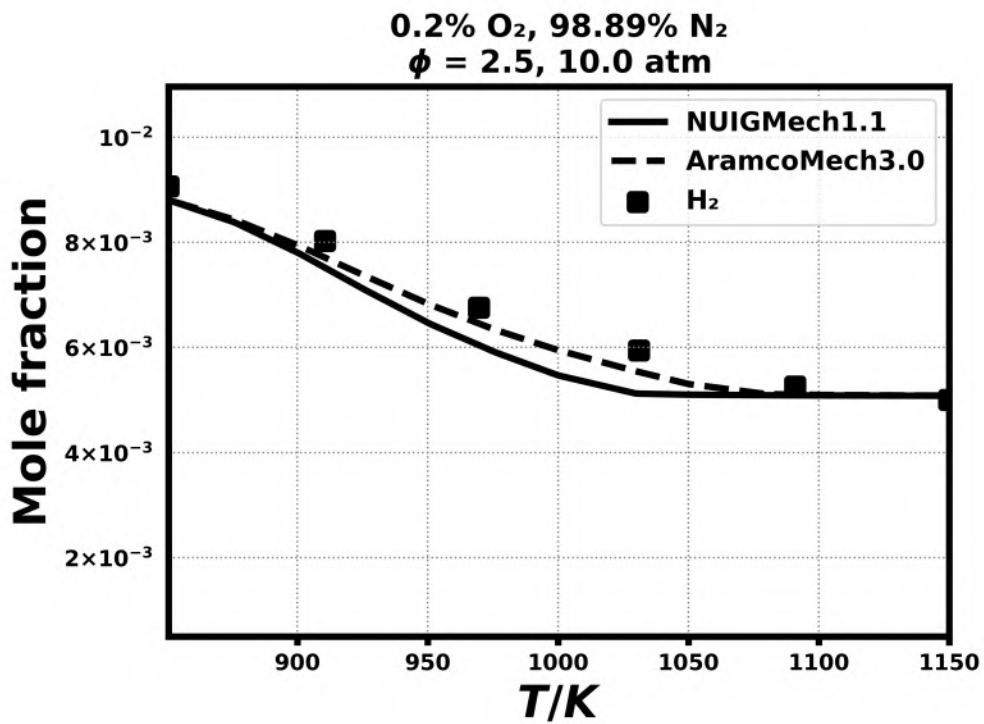
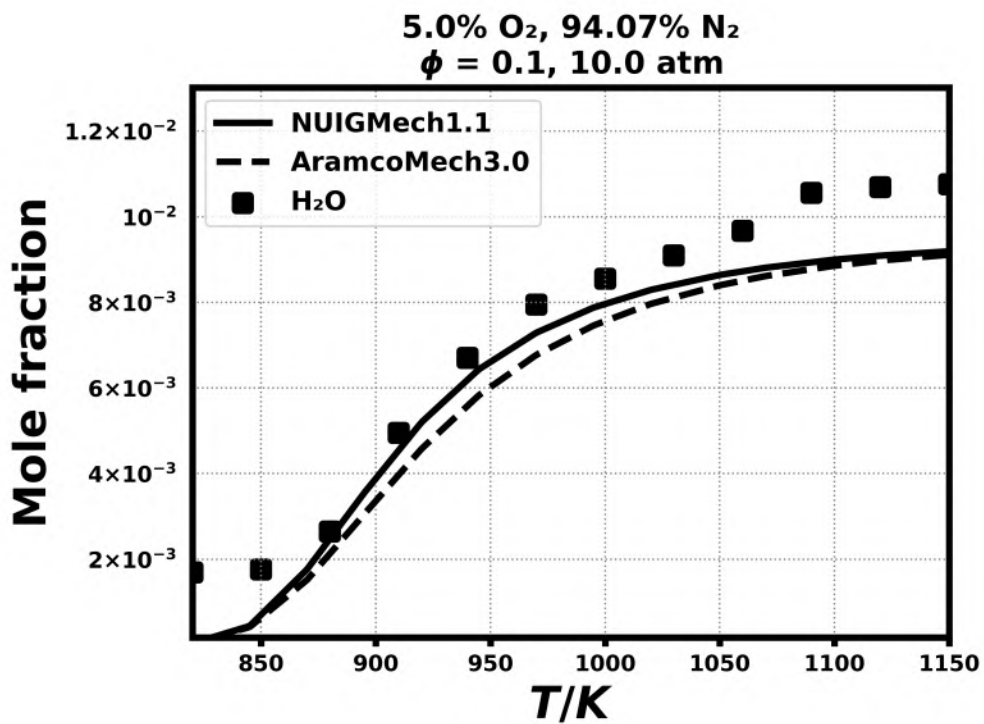




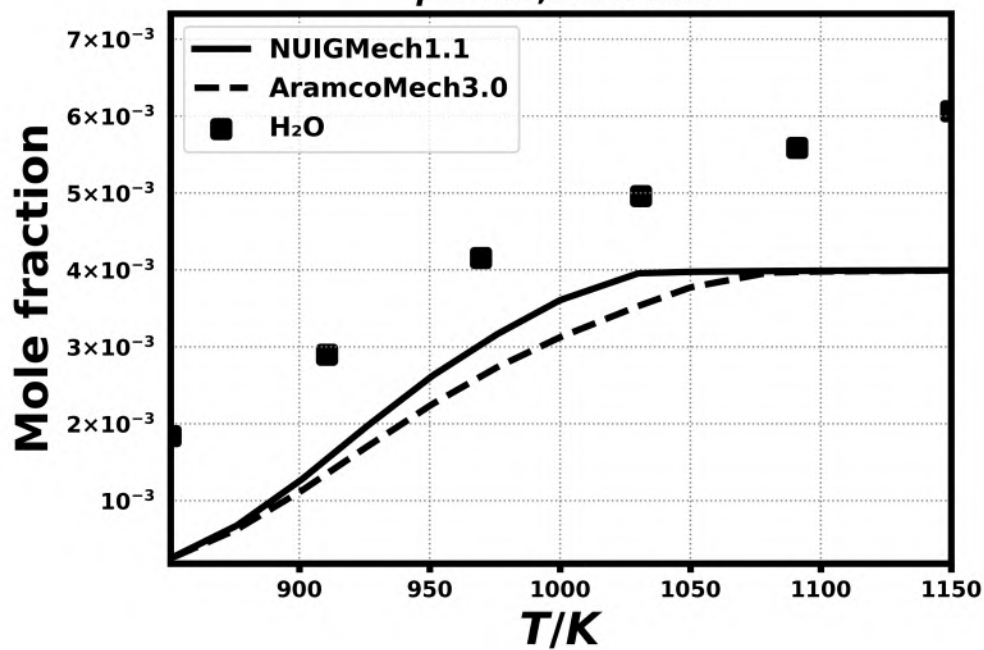


1.9) Le Cong, T., & Dagaut, P., Energy & Fuels, 23(2) (2009) 725-734.

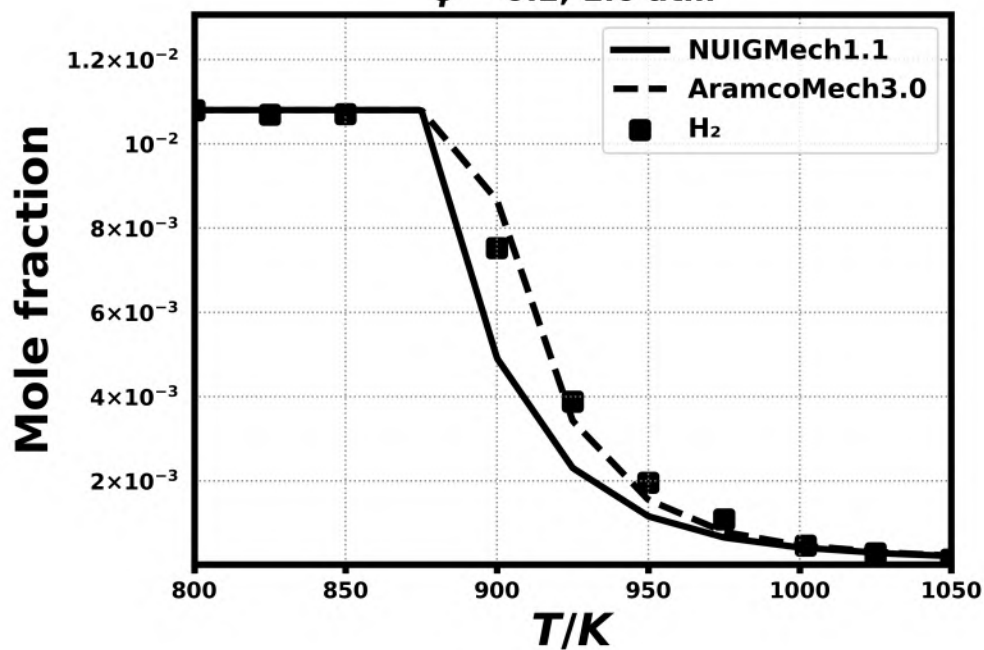




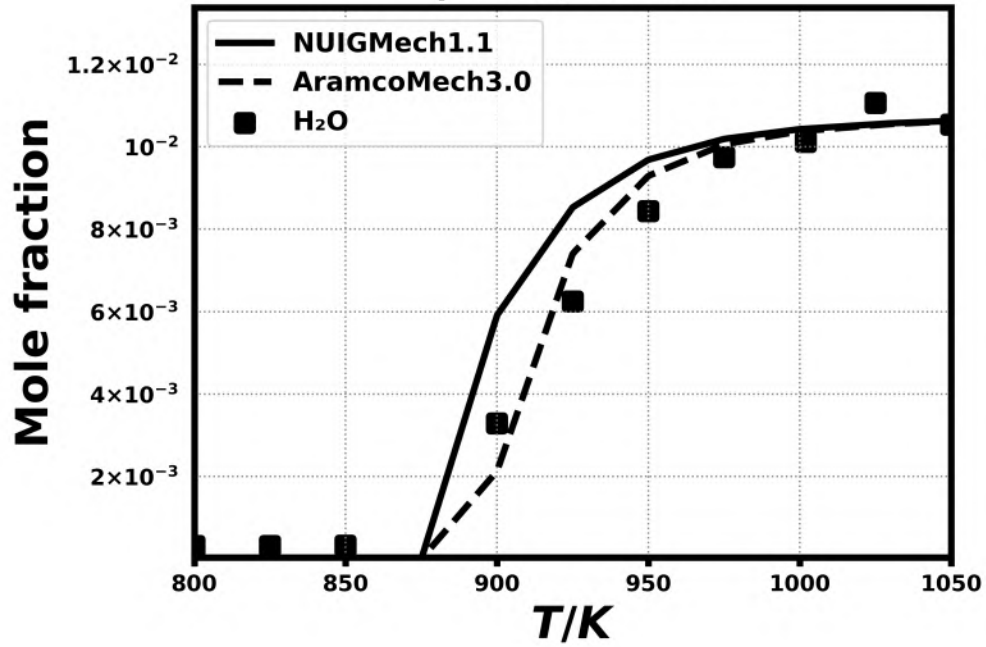
0.2% O₂, 98.89% N₂
 $\phi = 2.5, 10.0 \text{ atm}$



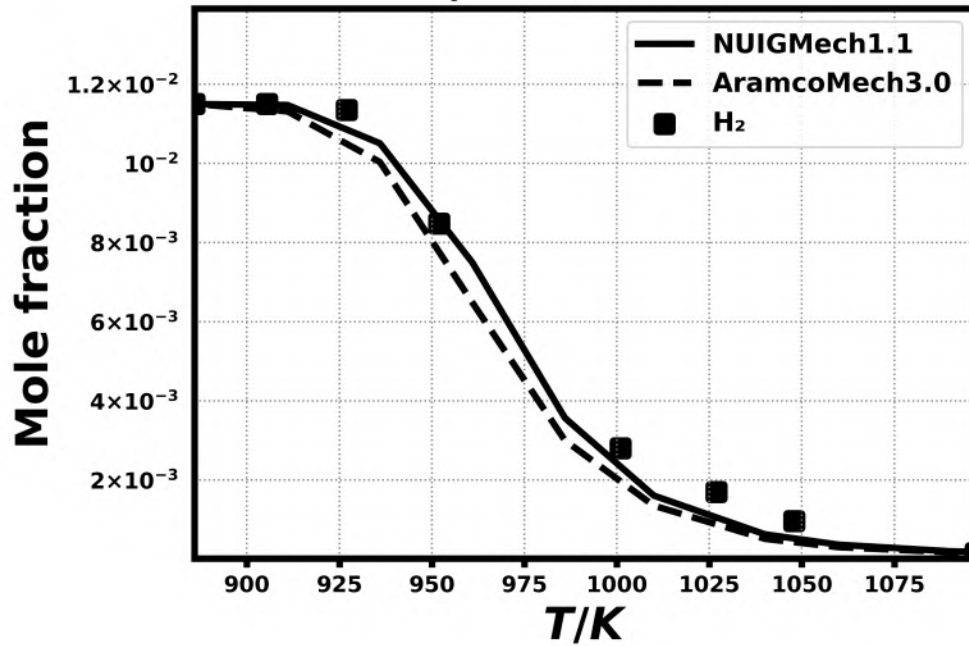
2.5% O₂, 96.42% N₂
 $\phi = 0.2, 1.0 \text{ atm}$

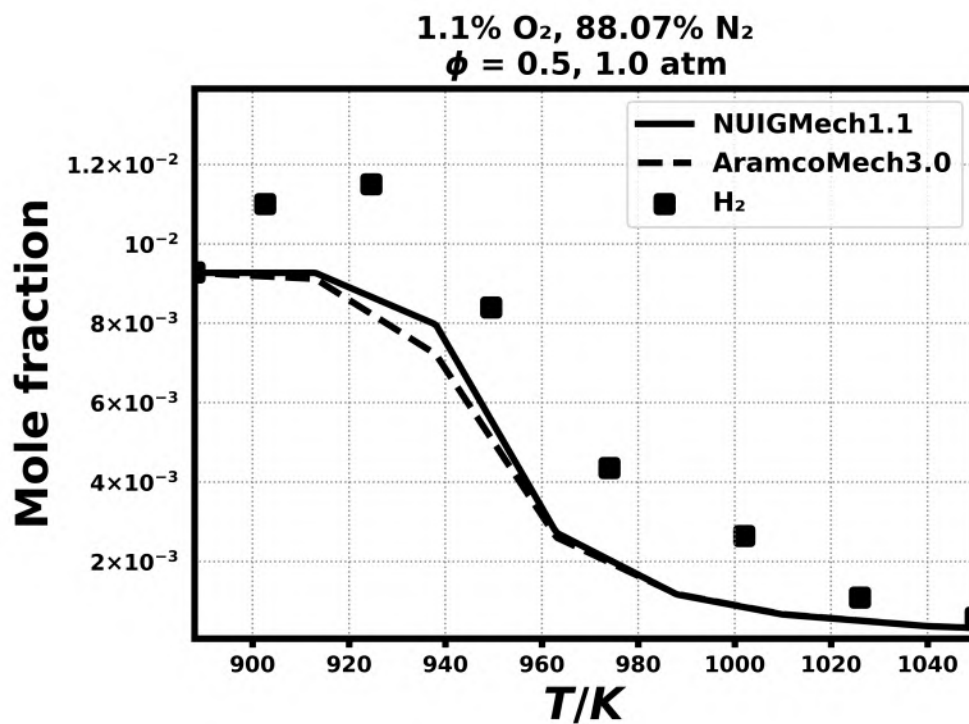
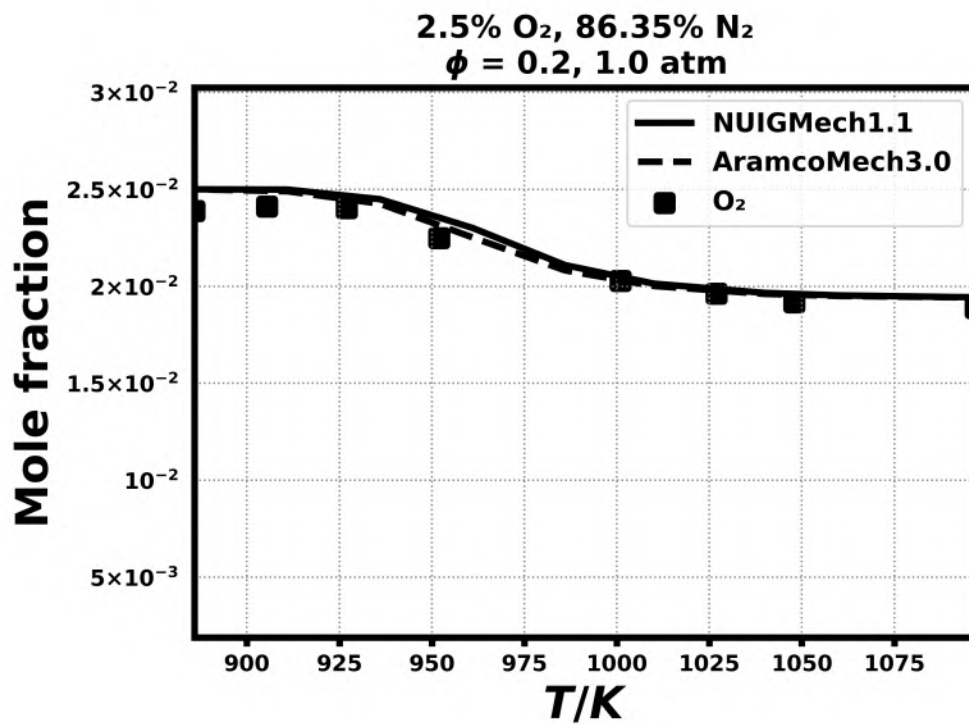


2.5% O₂, 96.42% N₂
 $\phi = 0.2, 1.0 \text{ atm}$

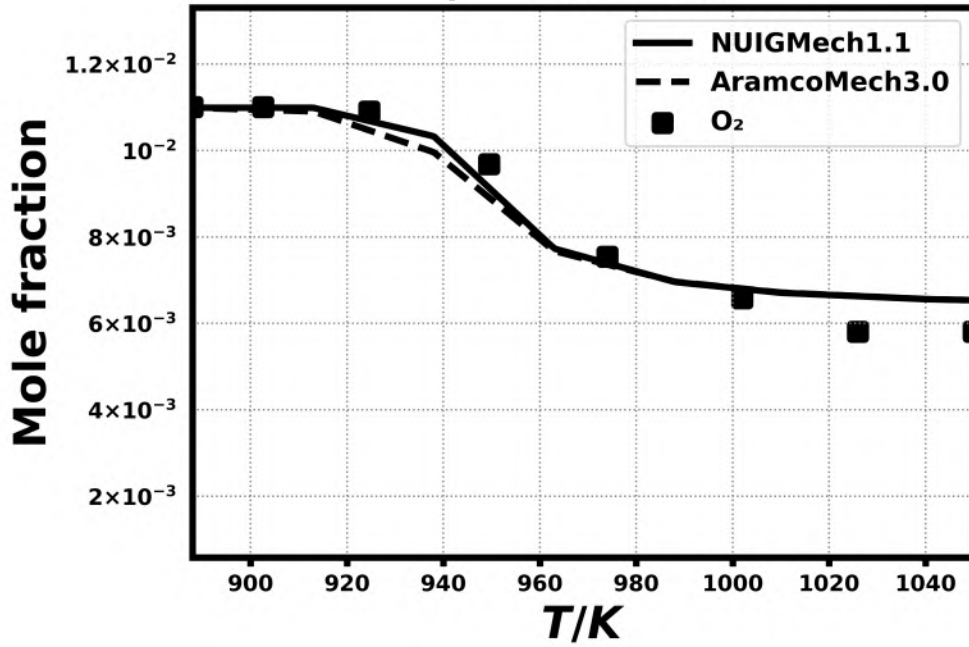


2.5% O₂, 86.35% N₂
 $\phi = 0.2, 1.0 \text{ atm}$

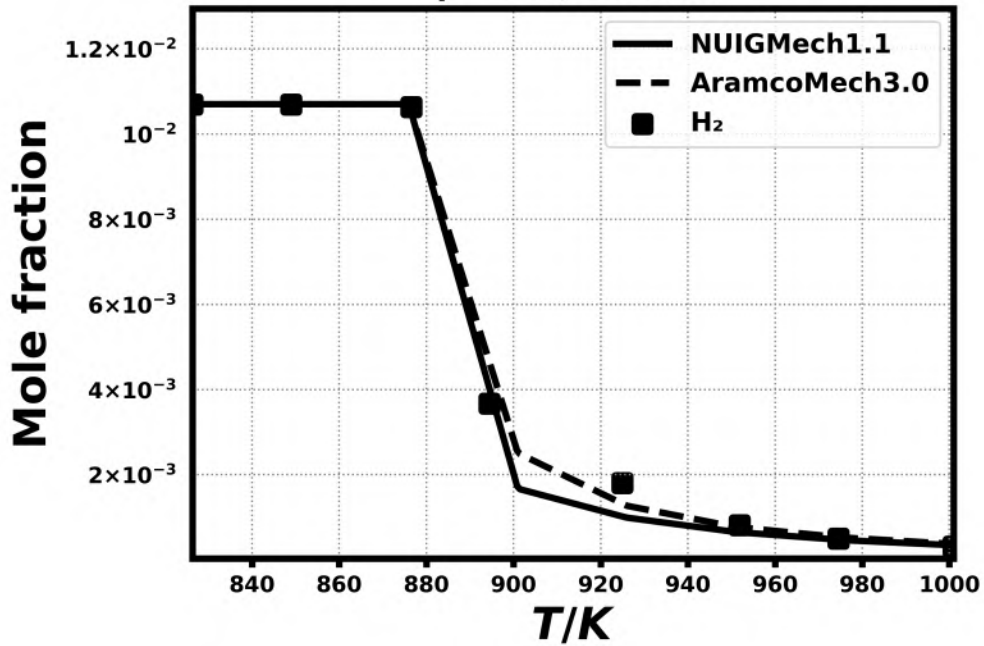




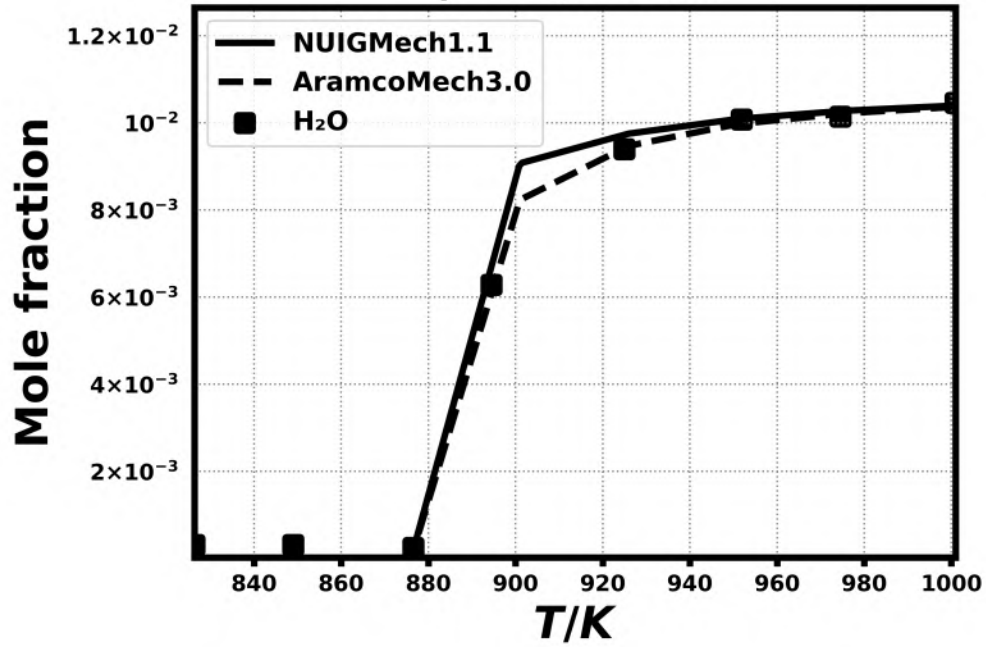
1.1% O₂, 88.07% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



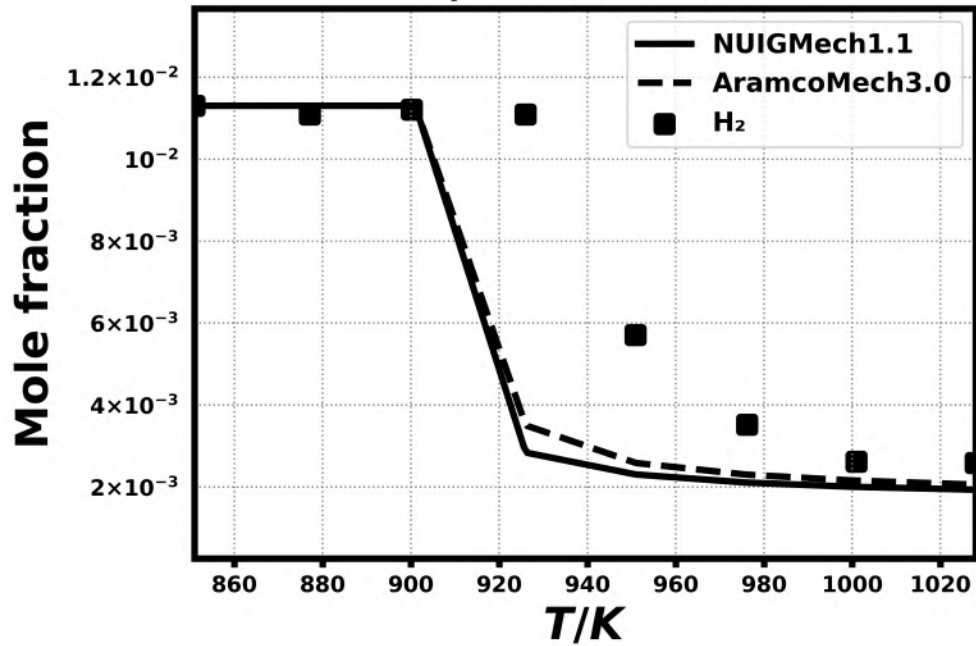
1.0% O₂, 97.93% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



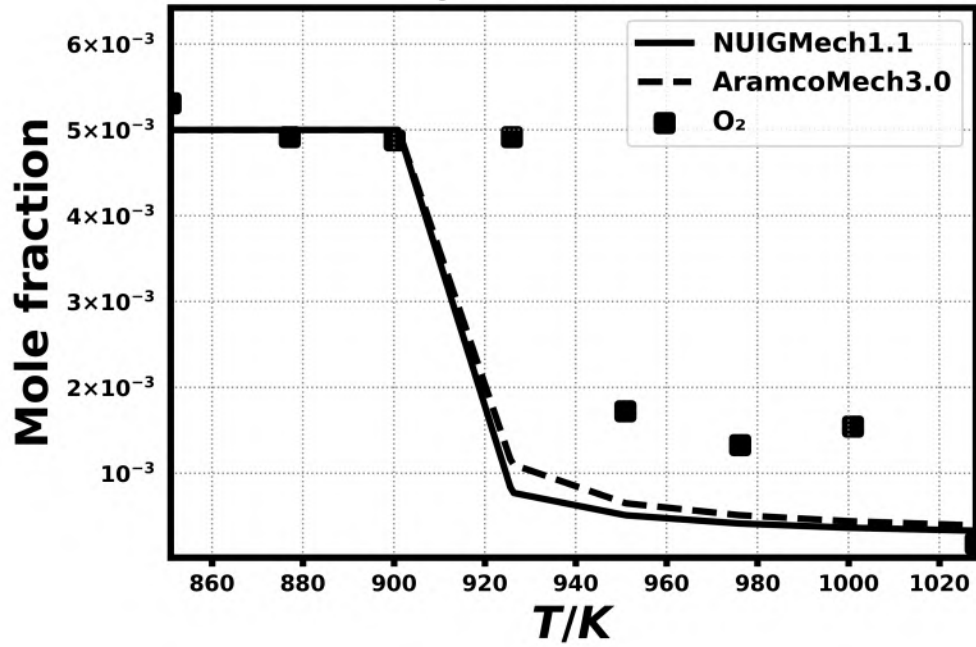
1.0% O₂, 97.93% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



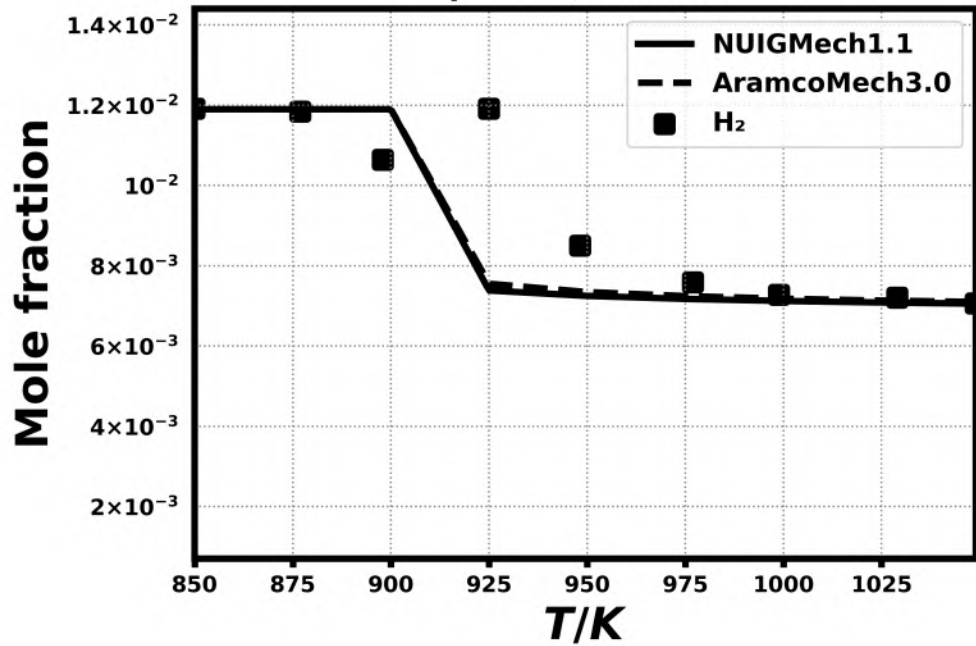
0.5% O₂, 88.37% N₂
 $\phi = 1.0, 1.0 \text{ atm}$



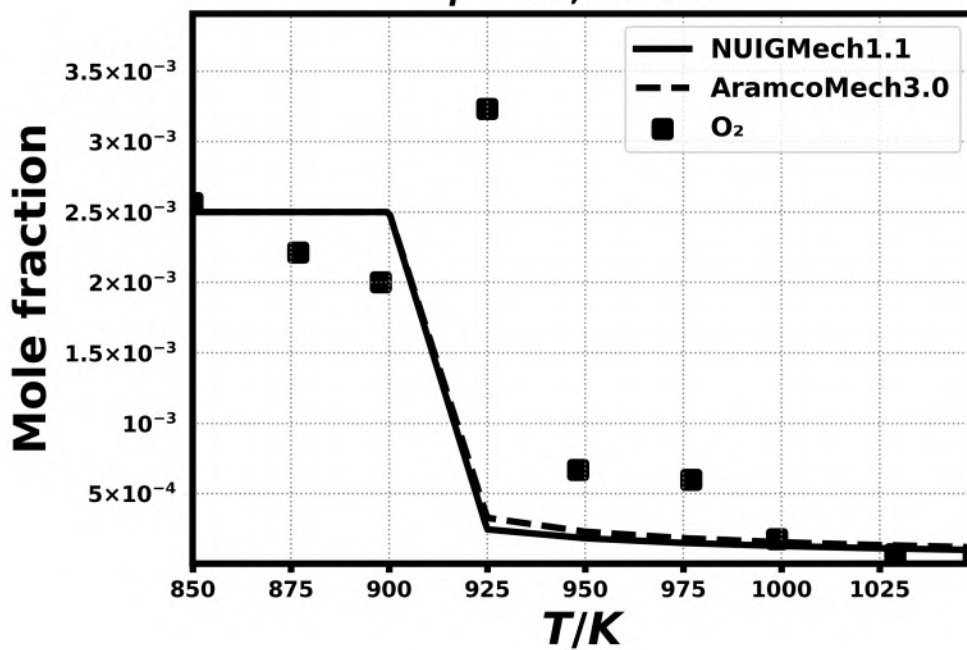
0.5% O₂, 88.37% N₂
 $\phi = 1.0, 1.0 \text{ atm}$



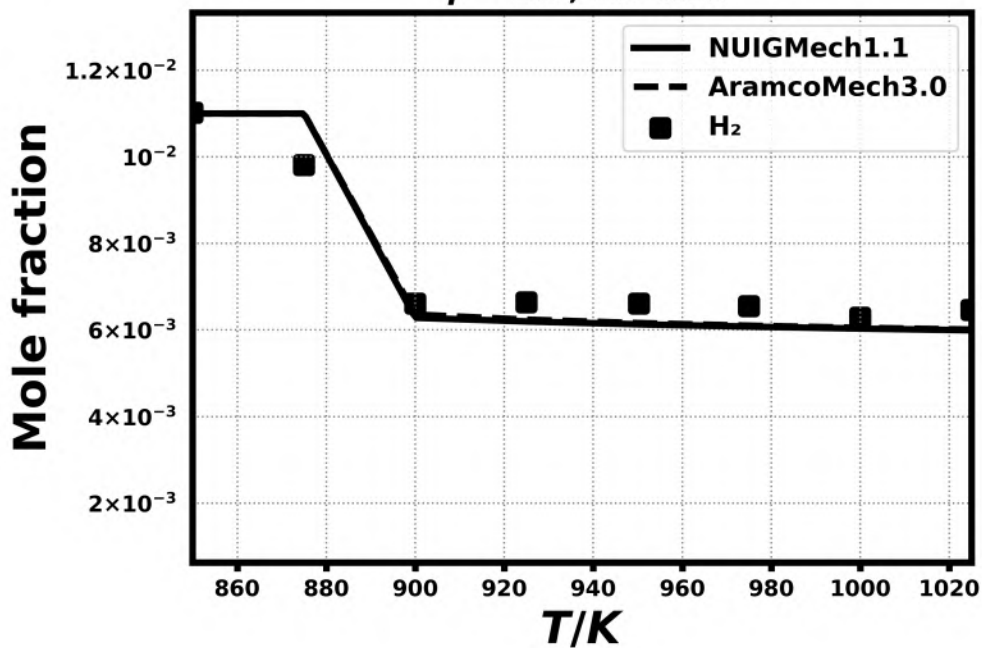
0.25% O₂, 88.56% N₂
 $\phi = 2.0, 1.0 \text{ atm}$

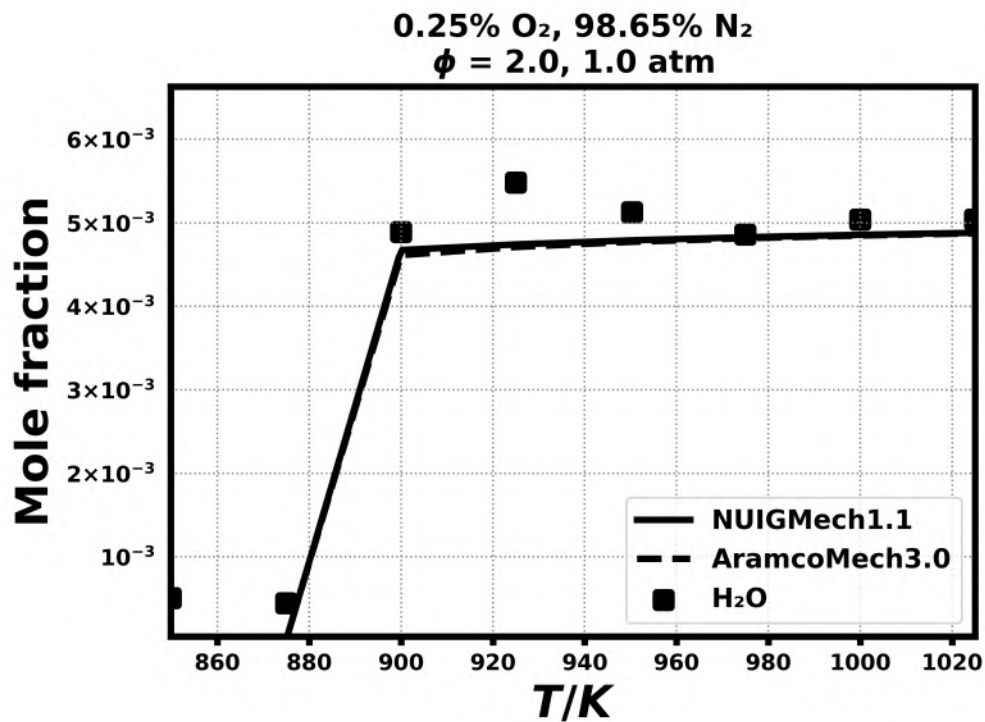


0.25% O₂, 88.56% N₂
 $\phi = 2.0, 1.0 \text{ atm}$



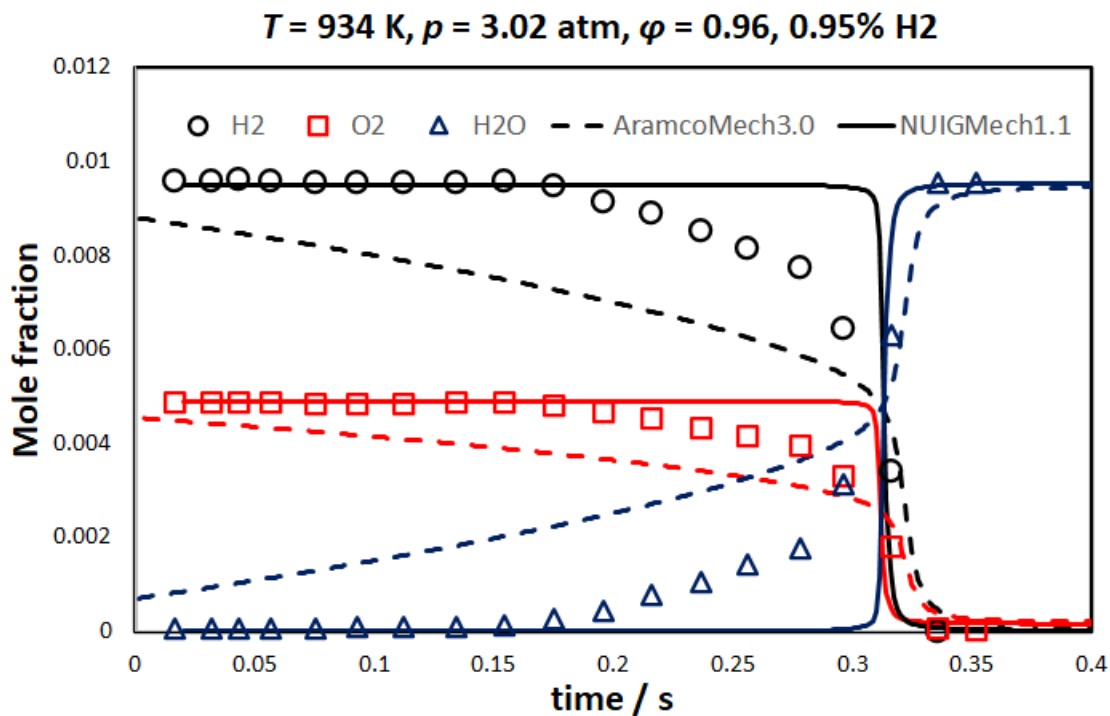
0.25% O₂, 98.65% N₂
 $\phi = 2.0, 1.0 \text{ atm}$



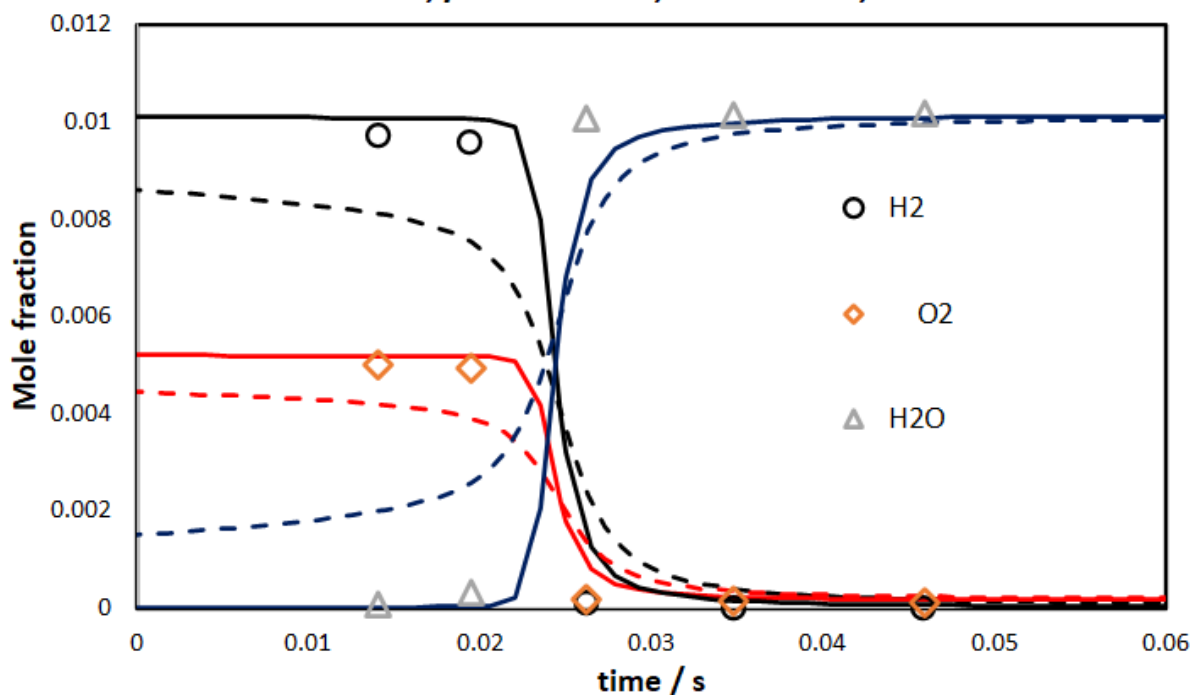


Speciation in Flow reactor

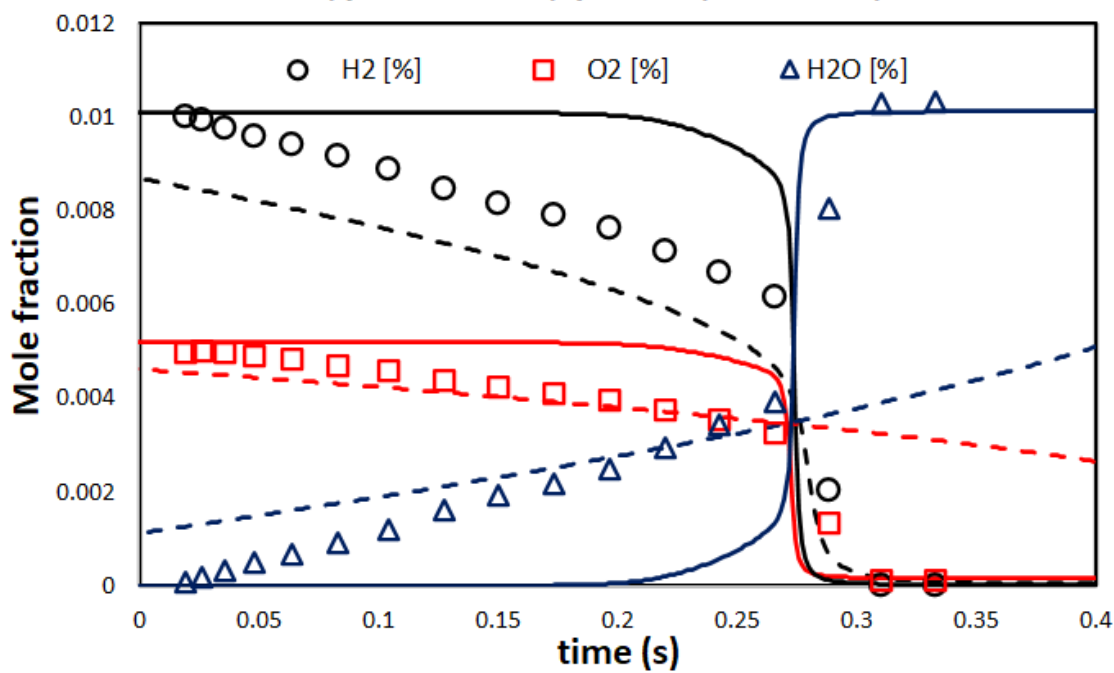
1.10) Mueller, M. A., T. J. Kim, R. A. Yetter, and F. L. Dryer, " International Journal of Chemical Kinetics 31, 2 (1999) 113-125.



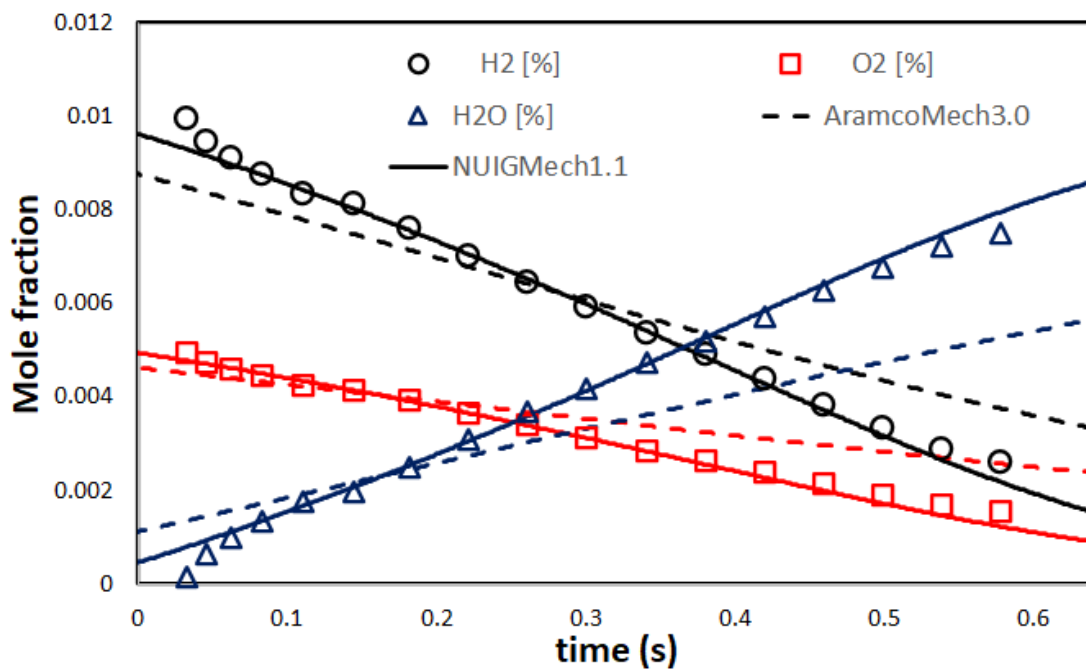
$T = 935 \text{ K}, p = 2.55 \text{ atm}, \text{H}_2 = 1.01\%, \text{O}_2 = 0.52\%$



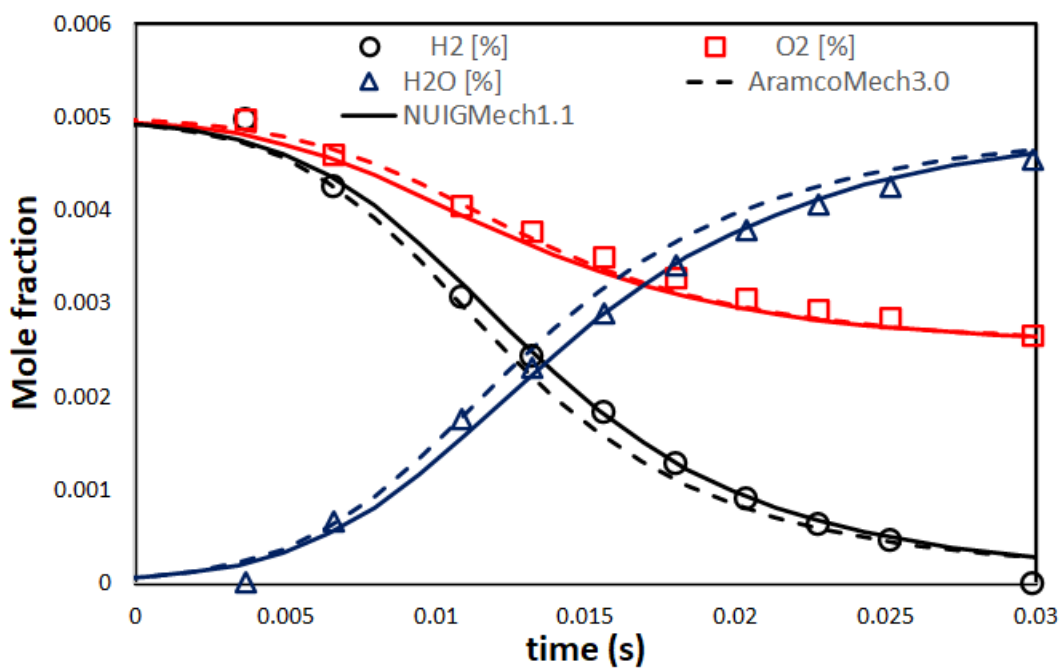
$T = 933 \text{ K}, p = 3.44 \text{ atm}, \varphi = 0.96, 1.01\% \text{ H}_2, 0.52\% \text{ O}_2$

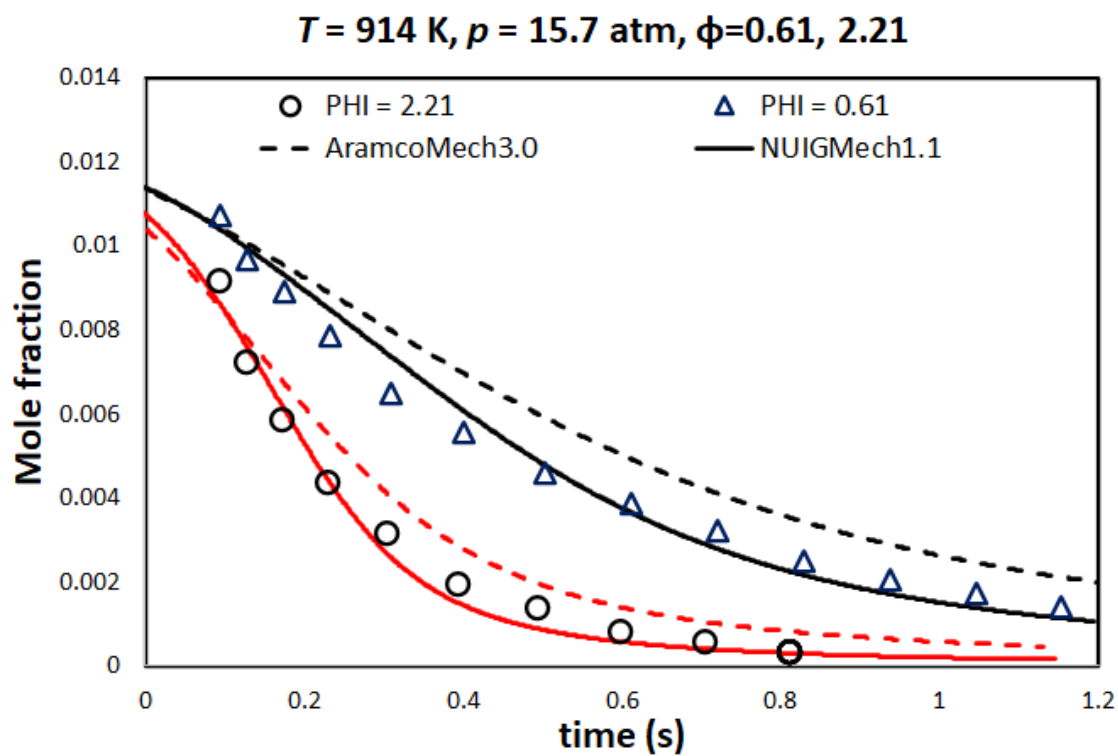


$T = 934 \text{ K}, p = 6.0 \text{ atm}, 1.01\% \text{ H}_2, 0.52\% \text{ O}_2$

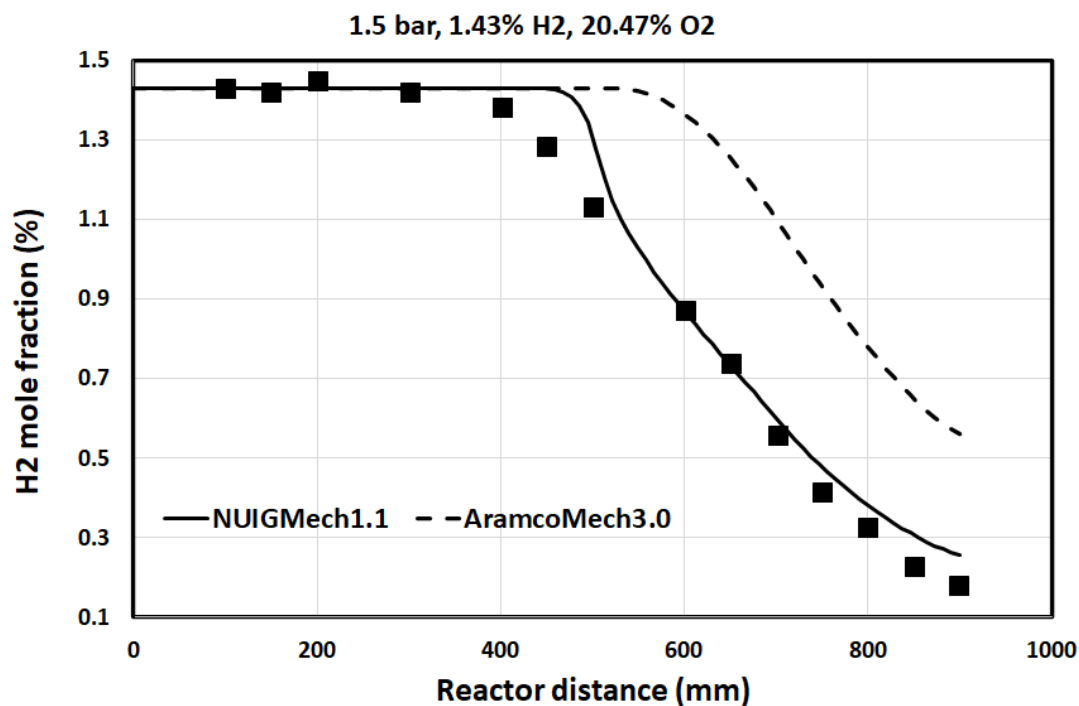


$T = 880 \text{ K}, p = 0.3 \text{ atm}, 0.5\% \text{ H}_2, 0.5\% \text{ O}_2$

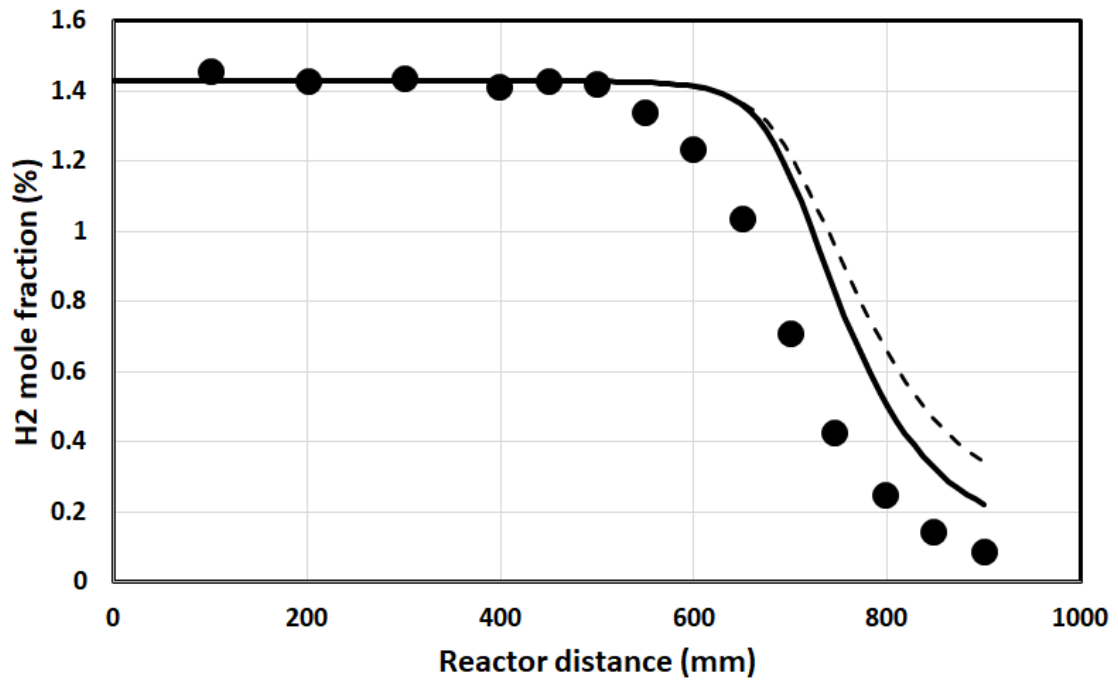




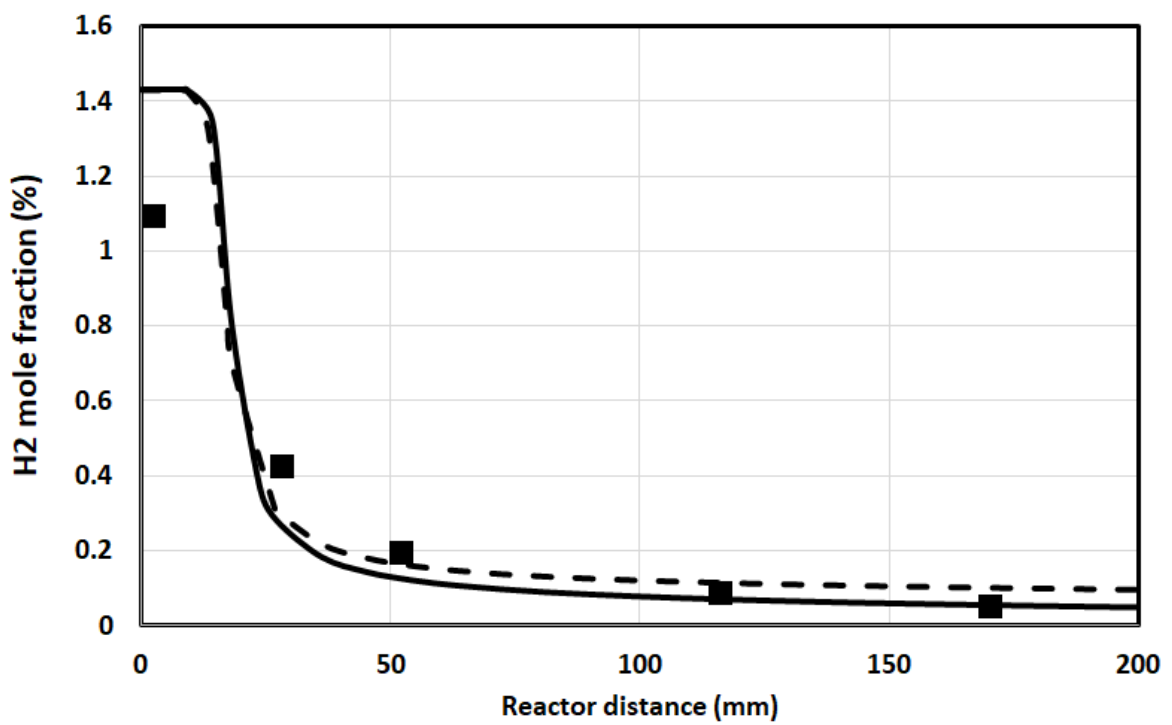
1.11) Lu, Zhewen, Junqiu Jiang, Yi Yang, Joshua Lacey, and Michael J. Brear., Proceedings of the Combustion Institute (2020).



8.0 bar, 1.43% H₂, 20.47% O₂



1.0 bar, 1.43% H₂, 20.47% O₂



Laminar flame speed

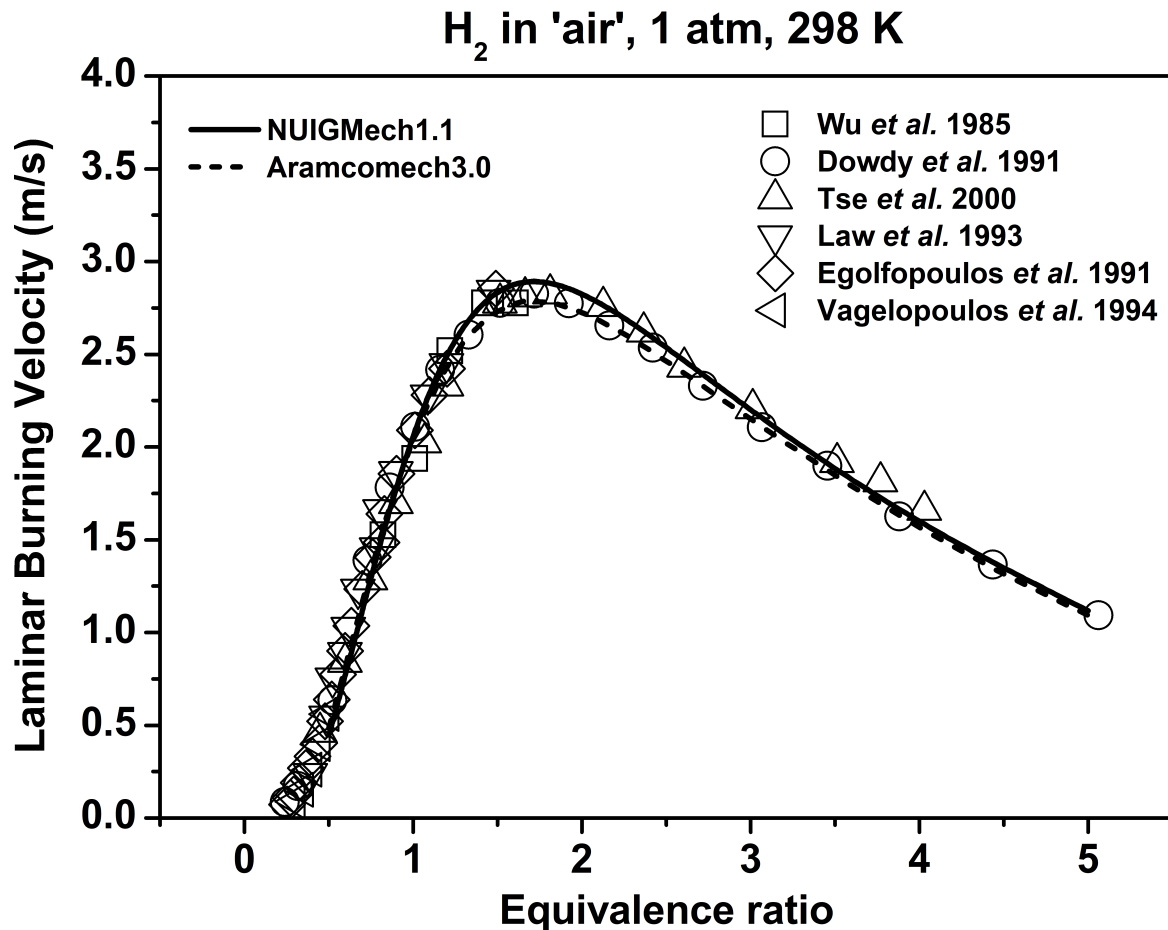
1.11) Tse, S. D., Zhu, D. L., & Law, C. K., Proceedings of the combustion institute, 28 (2) (2000) 1793-1800.

1.12) Wu, C., & Law, C. K., Symposium (International) on Combustion, 20 (1985, January) 1941-1949

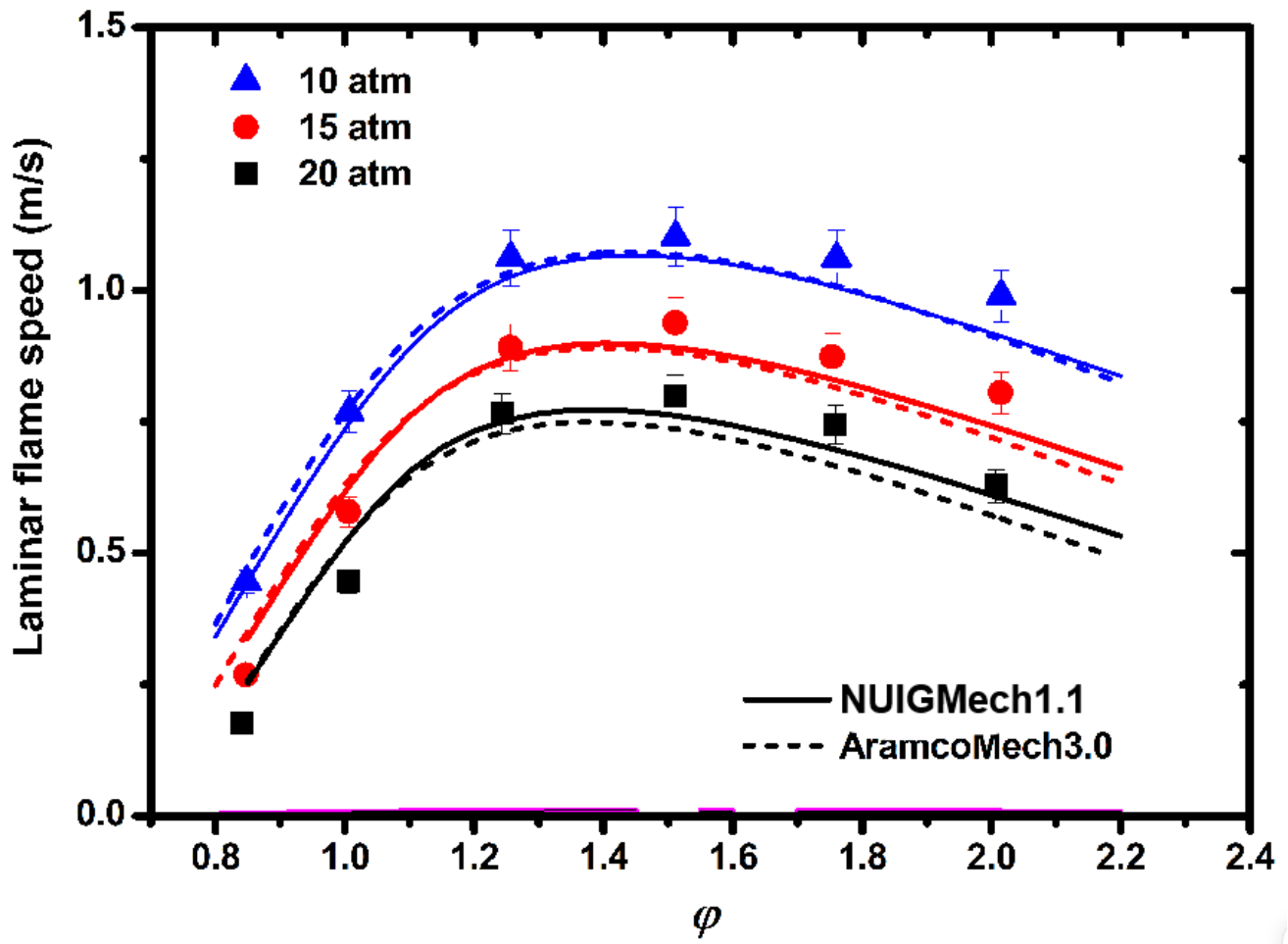
1.13) Dowdy, D. R., Smith, D. B., Taylor, S. C., & Williams, A., In Symposium (International) on Combustion, 23 (1991, January) 325-332

1.14) Egolfopoulos, F. N., & Law, C. K., In Symposium (international) on combustion, 23, (1991, January) 333-340

1.15) Vagelopoulos, C. M., Egolfopoulos, F. N., & Law, C. K., In Symposium (international) on combustion, 25 (1994, January) 1341-1347



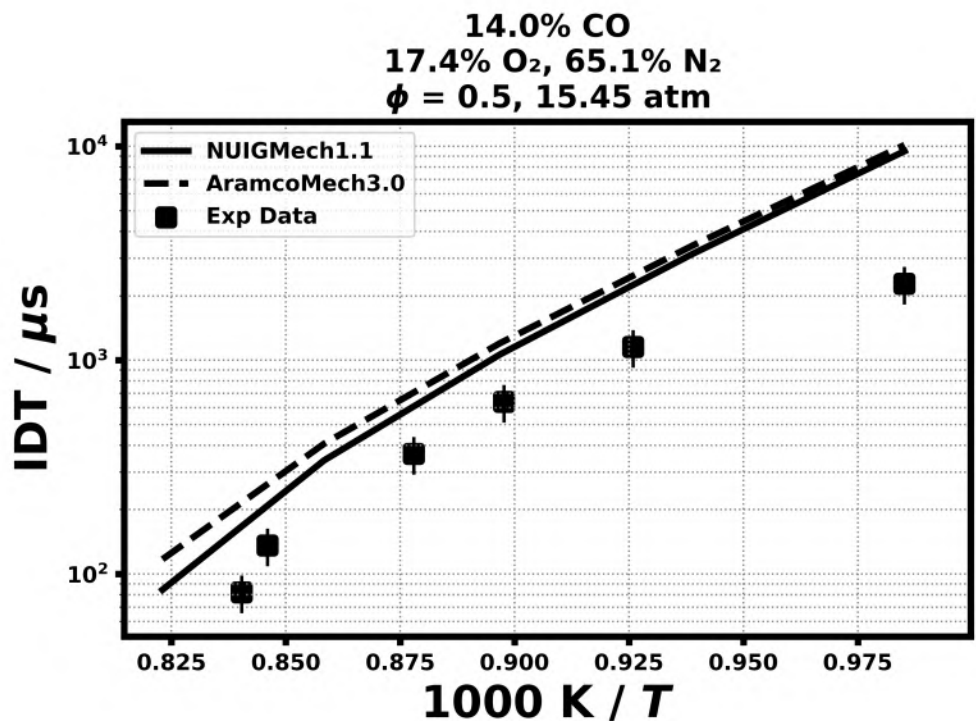
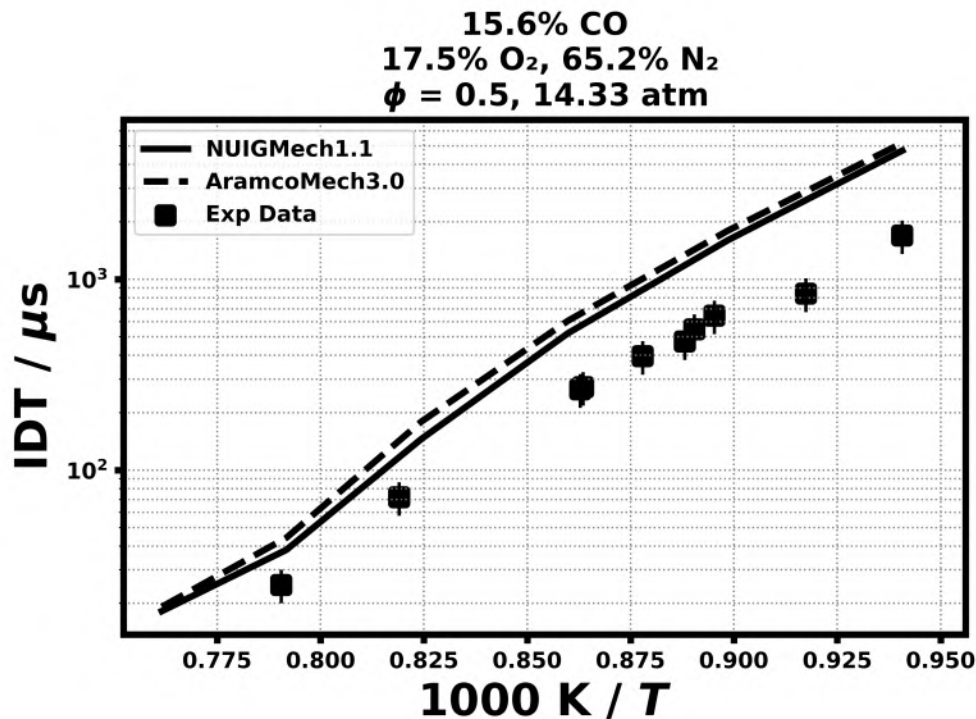
H₂, 1:11.5 O₂:He, 298 K



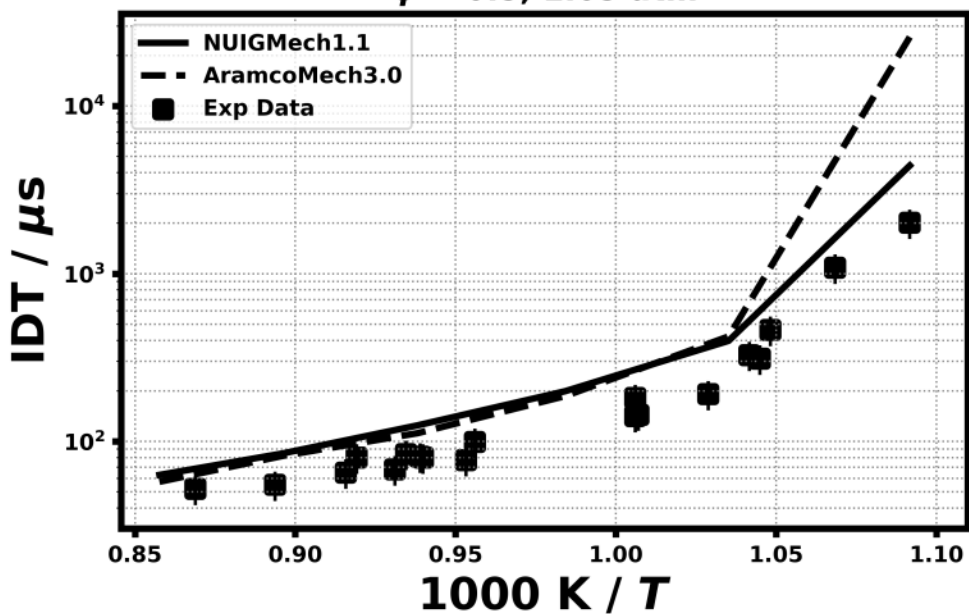
2. Validation for H₂/CO:

Shock tube ignition delay time

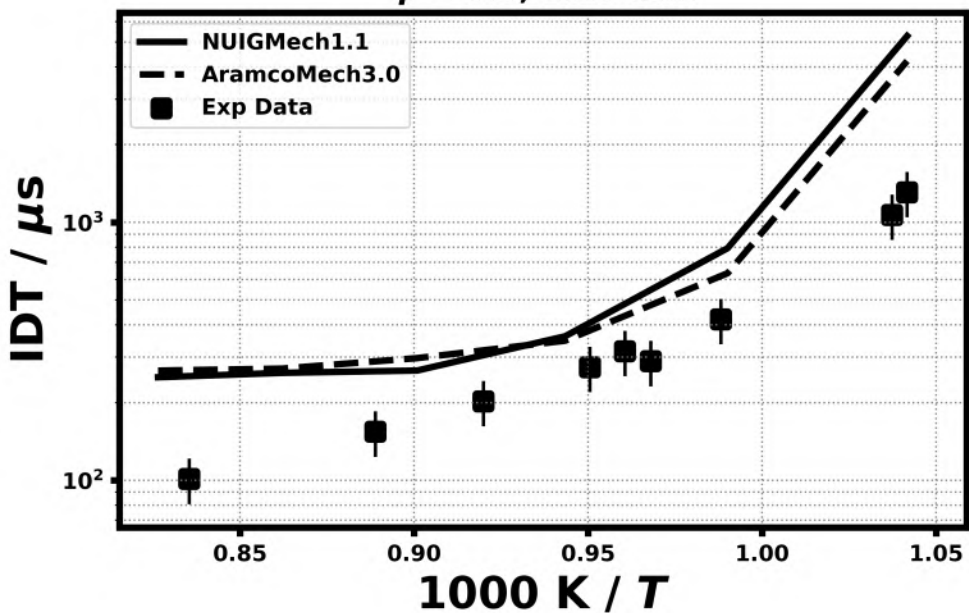
2.1) Kalitan, D. M., Mertens, J. D., Crofton, M. W., & Petersen, E. L., Journal of propulsion and power, 23(6) (2007) 1291-1301.

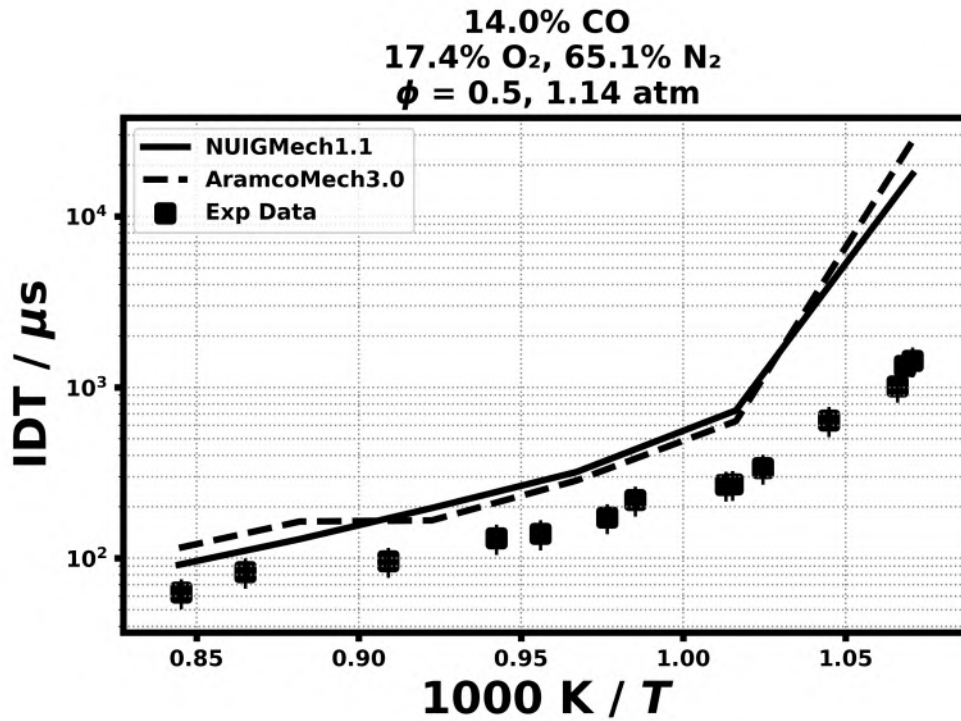


3.5% CO
17.4% O₂, 65.2% N₂
 $\phi = 0.5, 1.05 \text{ atm}$



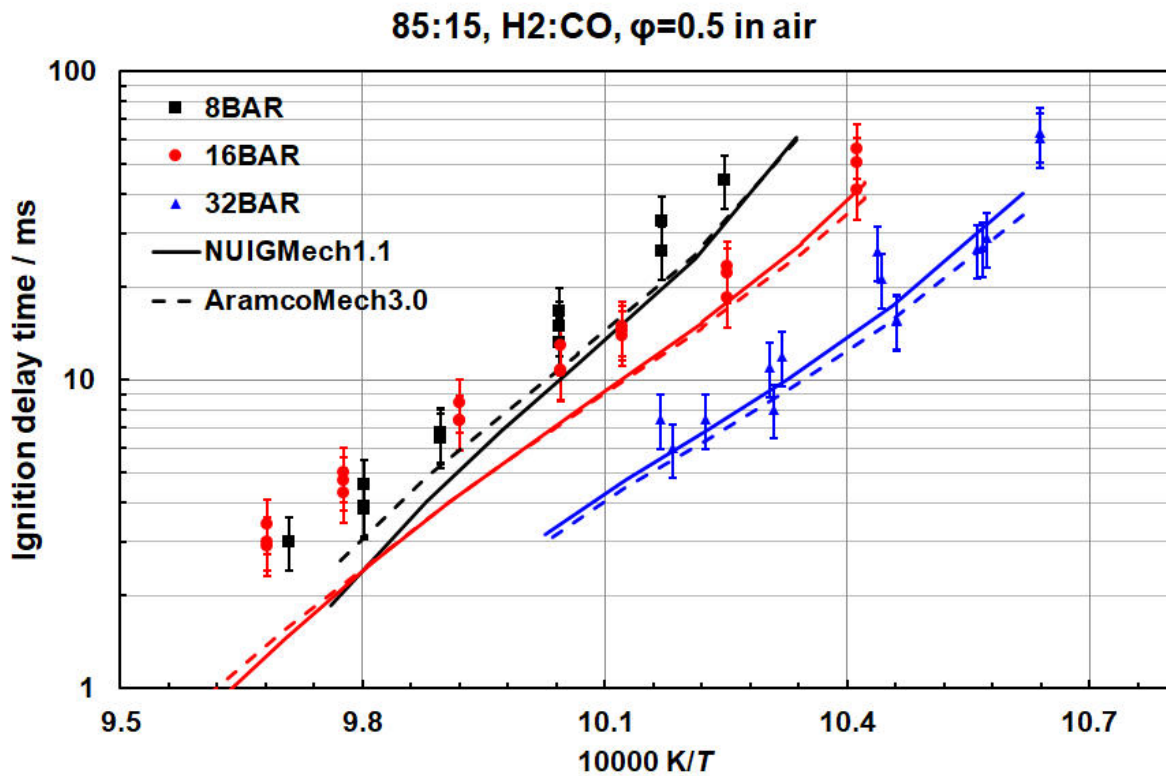
15.6% CO
17.5% O₂, 65.2% N₂
 $\phi = 0.5, 1.13 \text{ atm}$



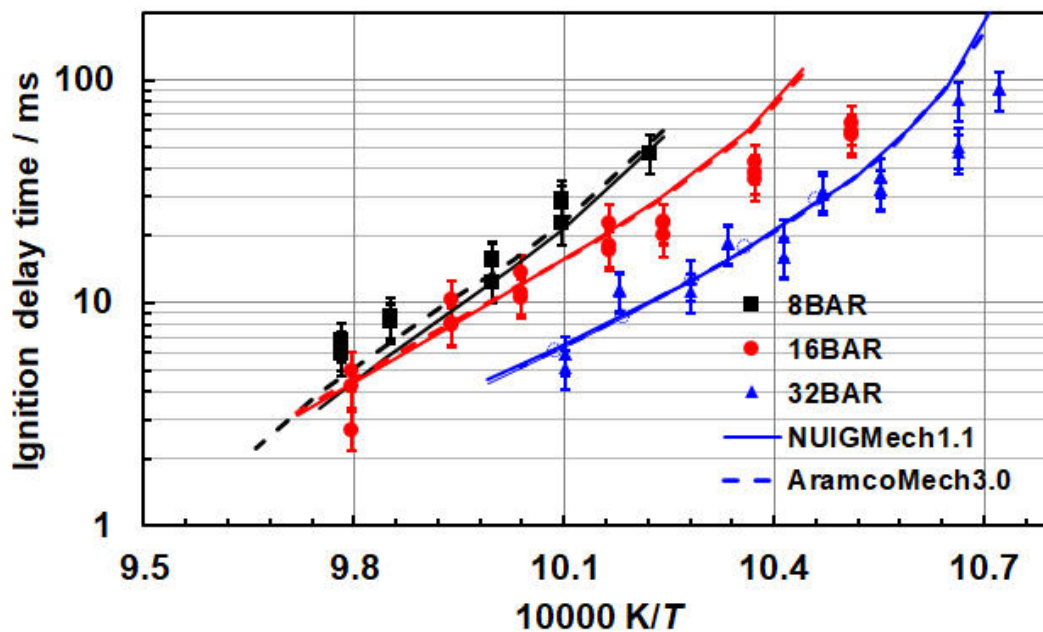


RCM Ignition delay time

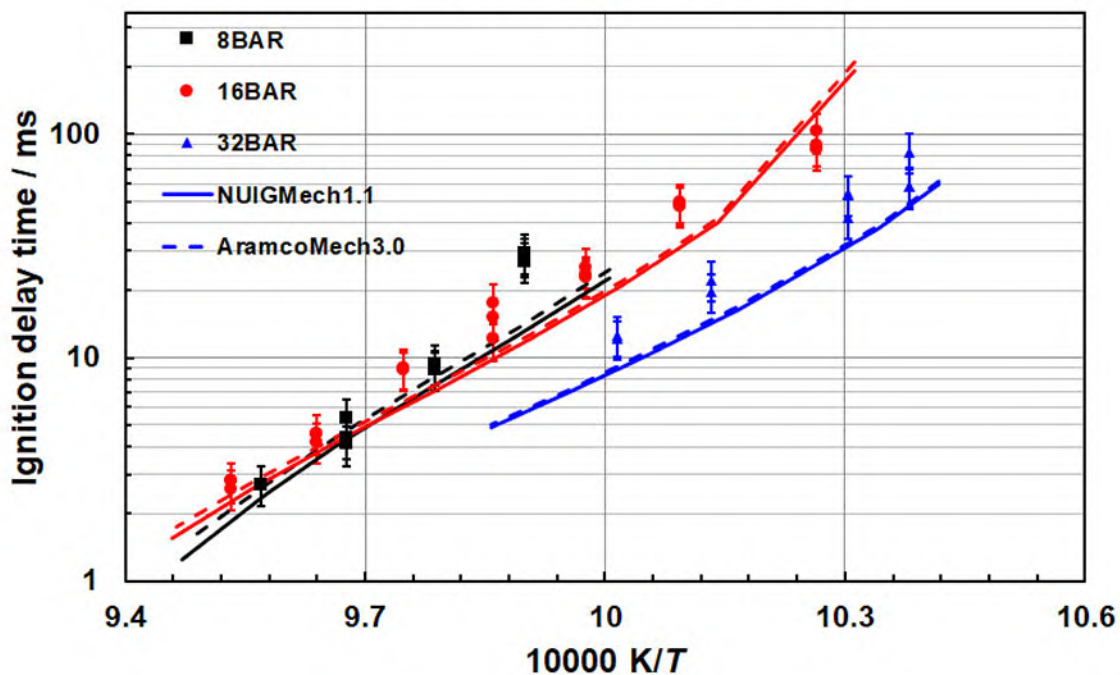
2.1) Kéromnès, A., Metcalfe, W. K., Heufer, K. A., Donohoe, N., Das, A. K., Sung, C. J., & Krejci, M. C., *Combustion and Flame*, 160(6), (2013) 995-1011.



50:50, H₂:CO, $\phi = 0.5$ in air

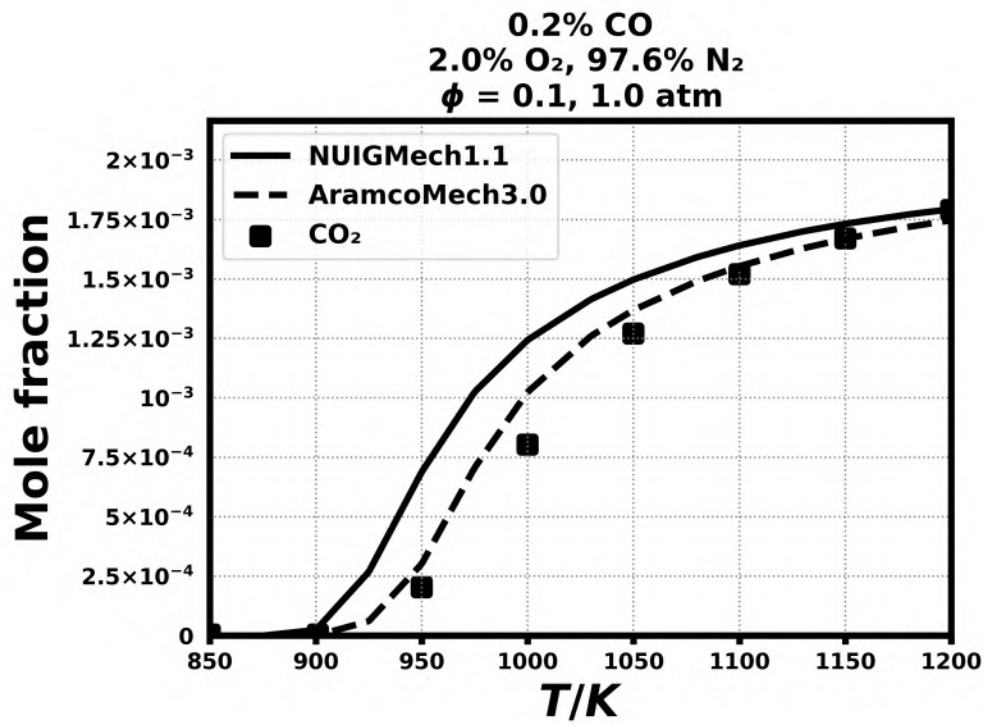
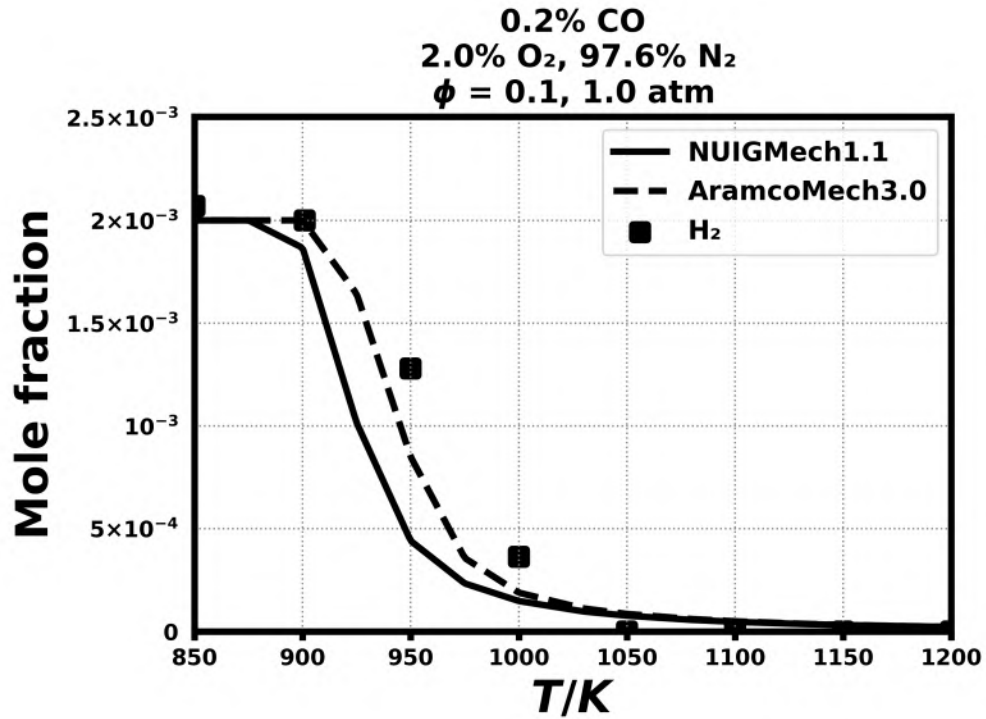


25:75, H₂:CO, $\phi = 0.5$ in air

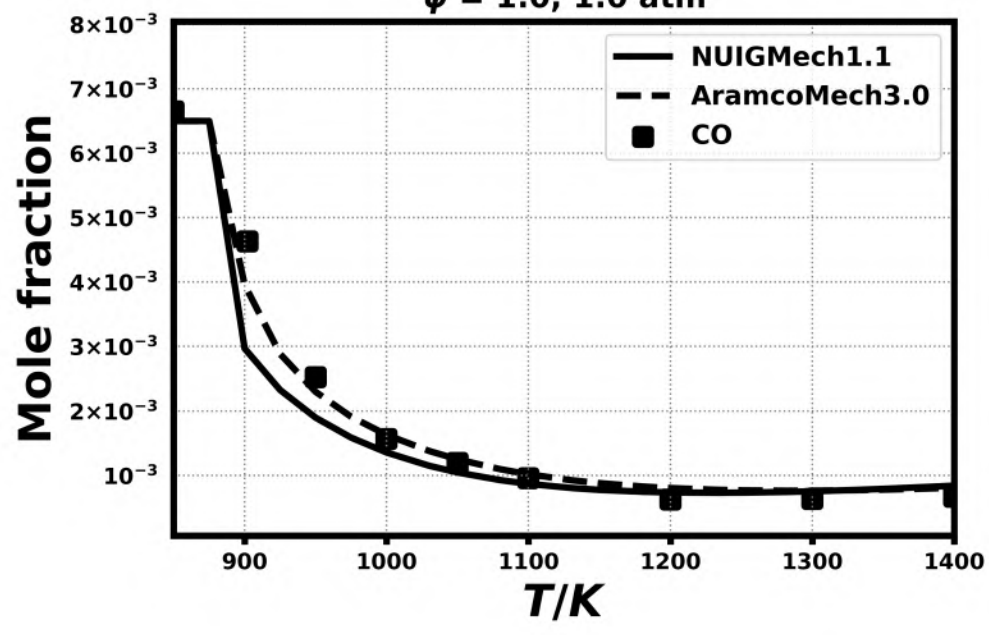


Speciation in Jet-stirred reactor

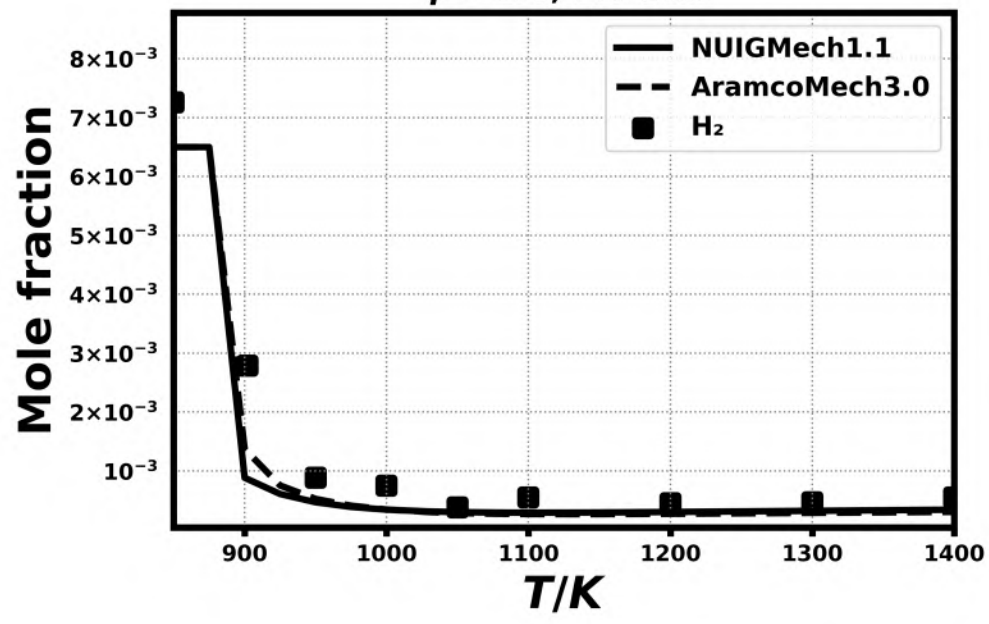
2.2) Le Cong, T., & Dagaut, P., *Energy & Fuels*, 23(2) (2009) 725-734.



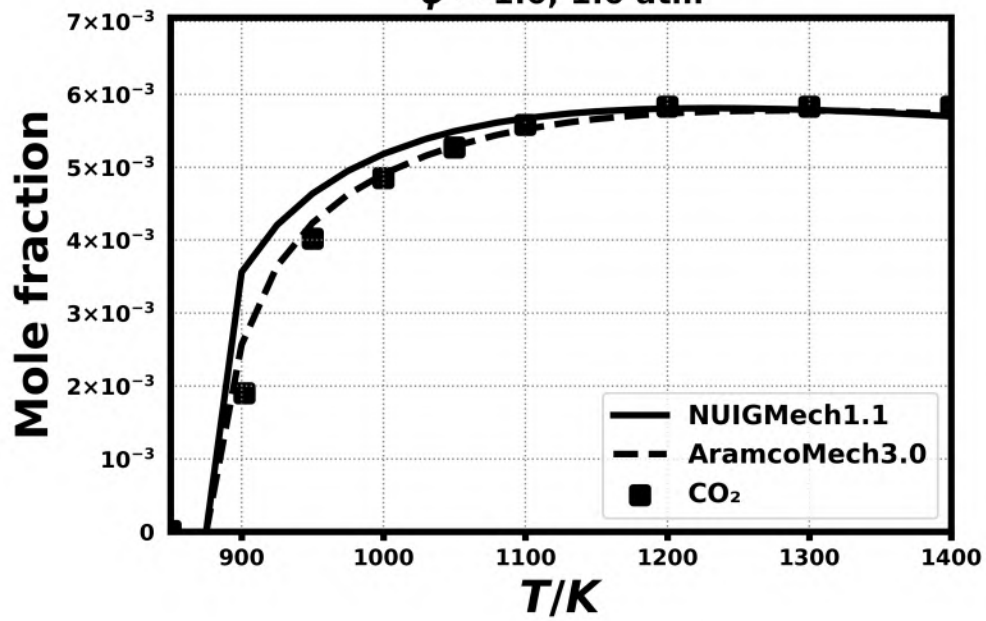
0.65% CO
0.65% O₂, 98.05% N₂
 $\phi = 1.0, 1.0 \text{ atm}$



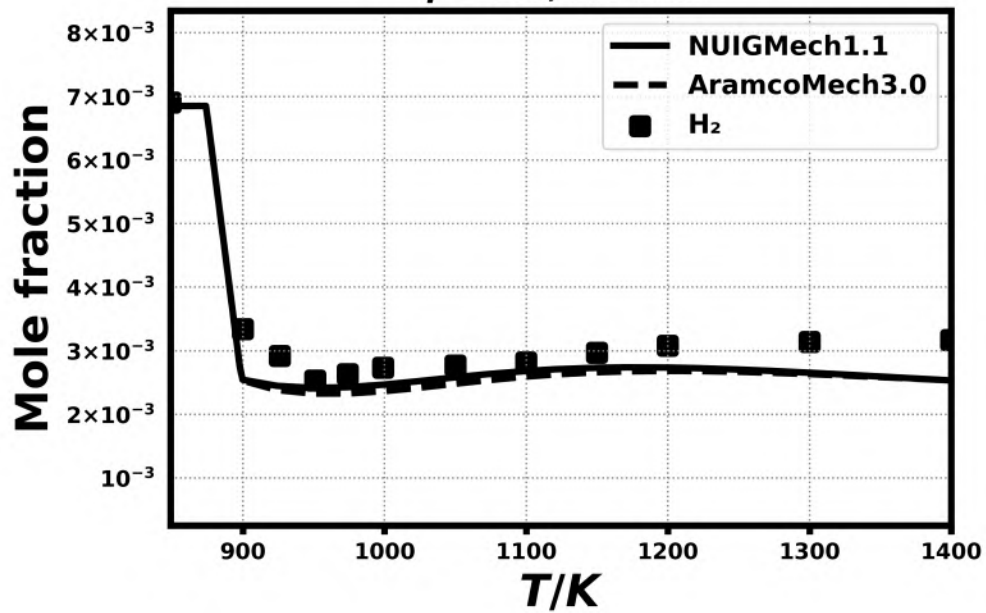
0.65% CO
0.65% O₂, 98.05% N₂
 $\phi = 1.0, 1.0 \text{ atm}$



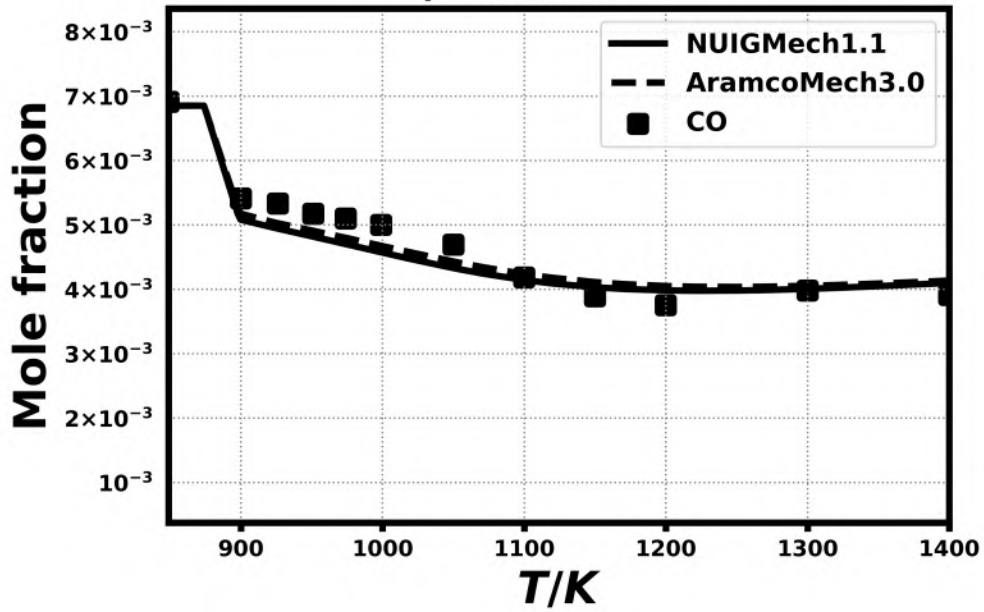
0.65% CO
0.65% O₂, 98.05% N₂
 $\phi = 1.0, 1.0 \text{ atm}$



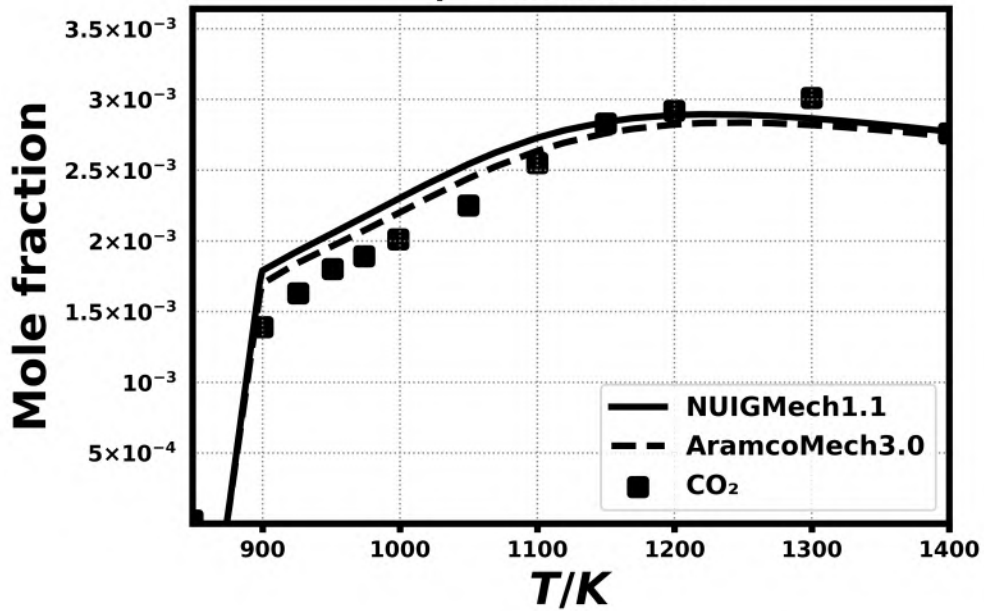
0.685% CO
0.343% O₂, 98.287% N₂
 $\phi = 2.0, 1.0 \text{ atm}$



0.685% CO
0.343% O₂, 98.287% N₂
 $\phi = 2.0, 1.0 \text{ atm}$

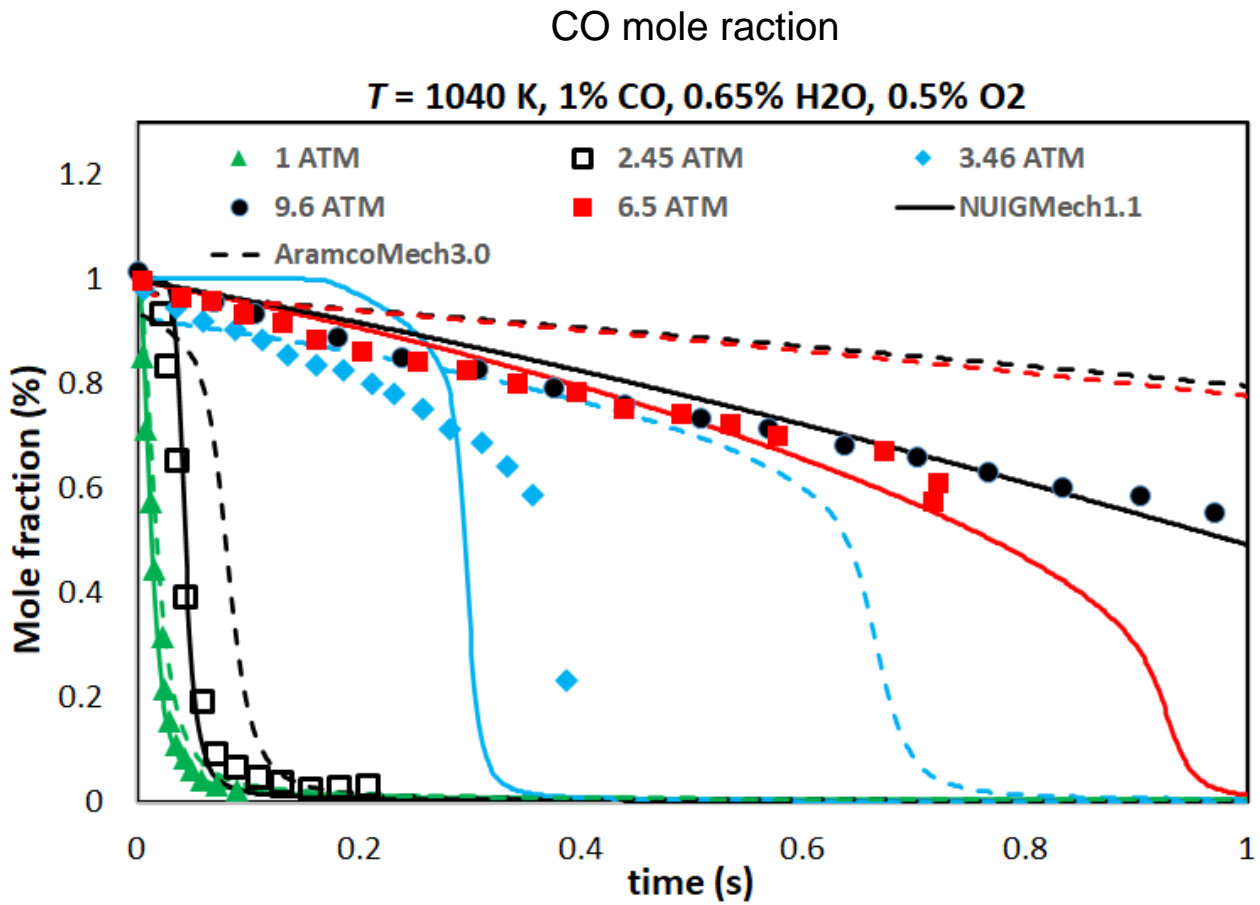


0.685% CO
0.343% O₂, 98.287% N₂
 $\phi = 2.0, 1.0 \text{ atm}$



Speciation in Flow reactor

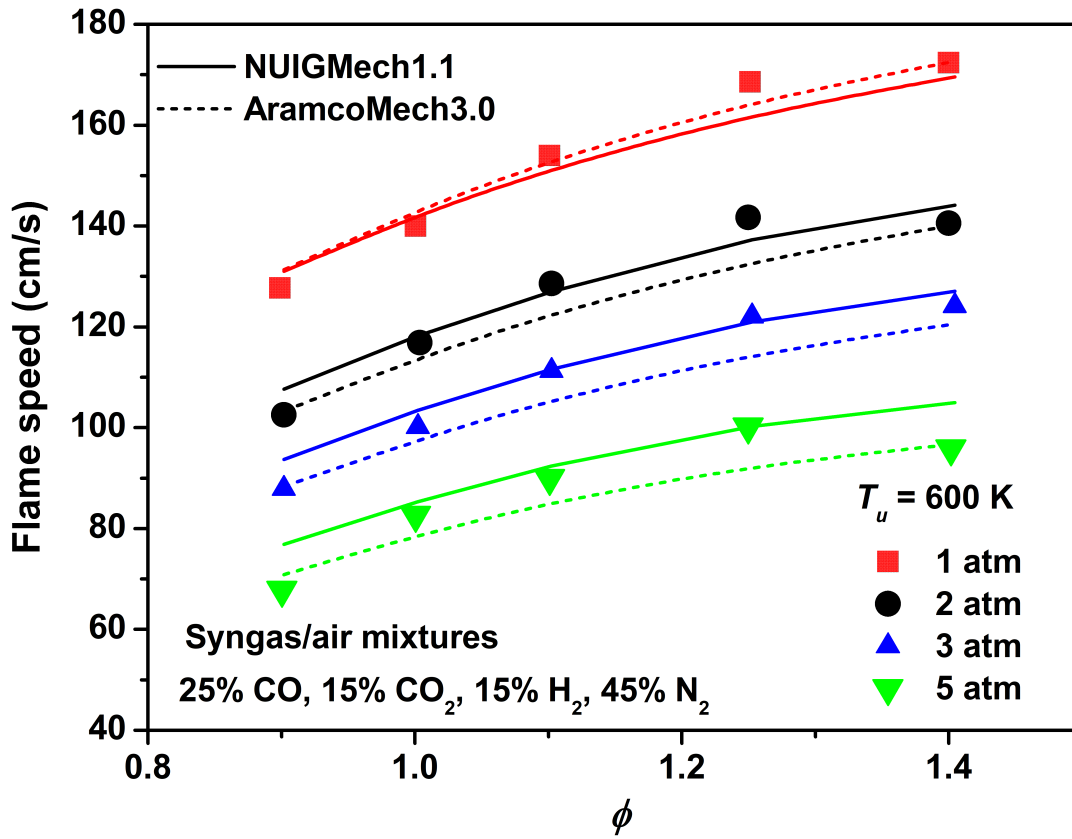
T.J. Kim, R.A. Yetter, F.L. Dryer, Twenty-Fifth Symposium (International) on Combust. (1994) 759–766.



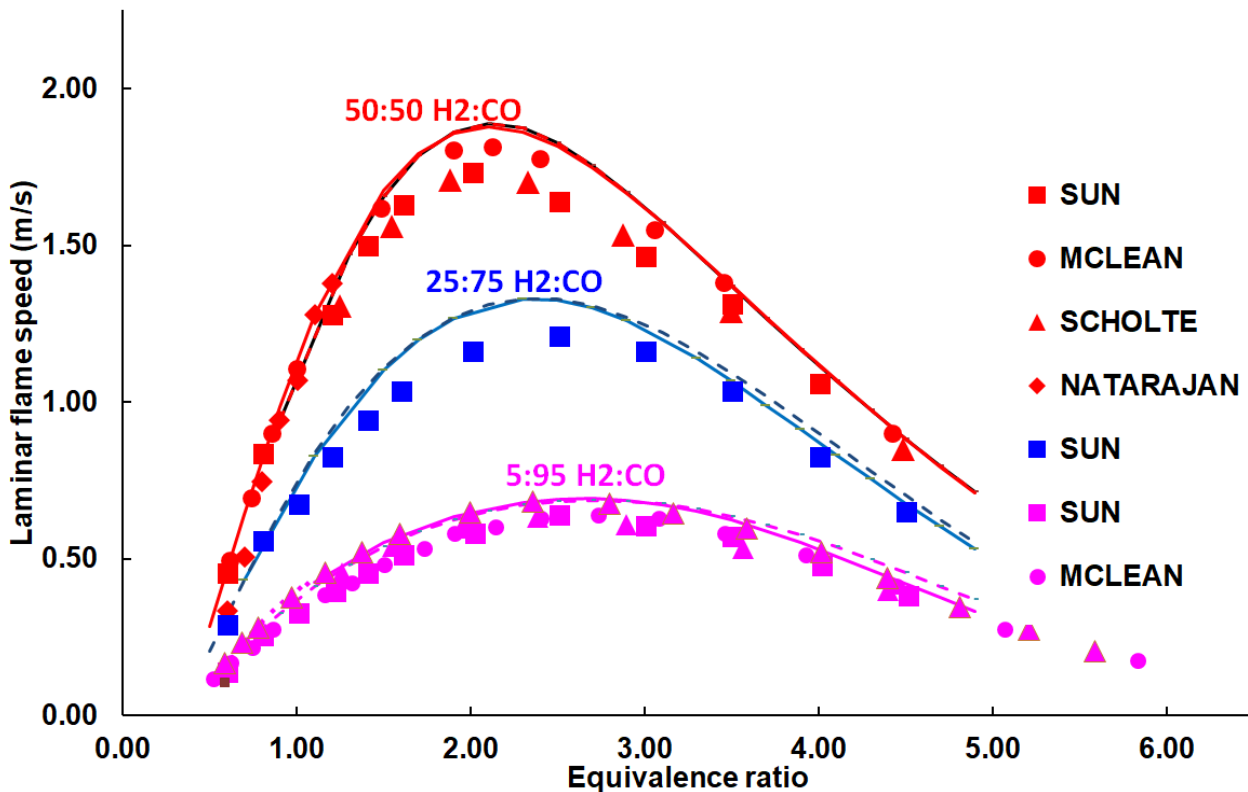
Laminar flame speed

2.4) Varghese, Robin John, and Sudarshan Kumar., Fuel 279 (2020) 118475.

2.5) H. Sun, S.I. Yang, G. Jomaas, C.K. Law, Proc. Combust. Inst. 31 (2007) 439–446



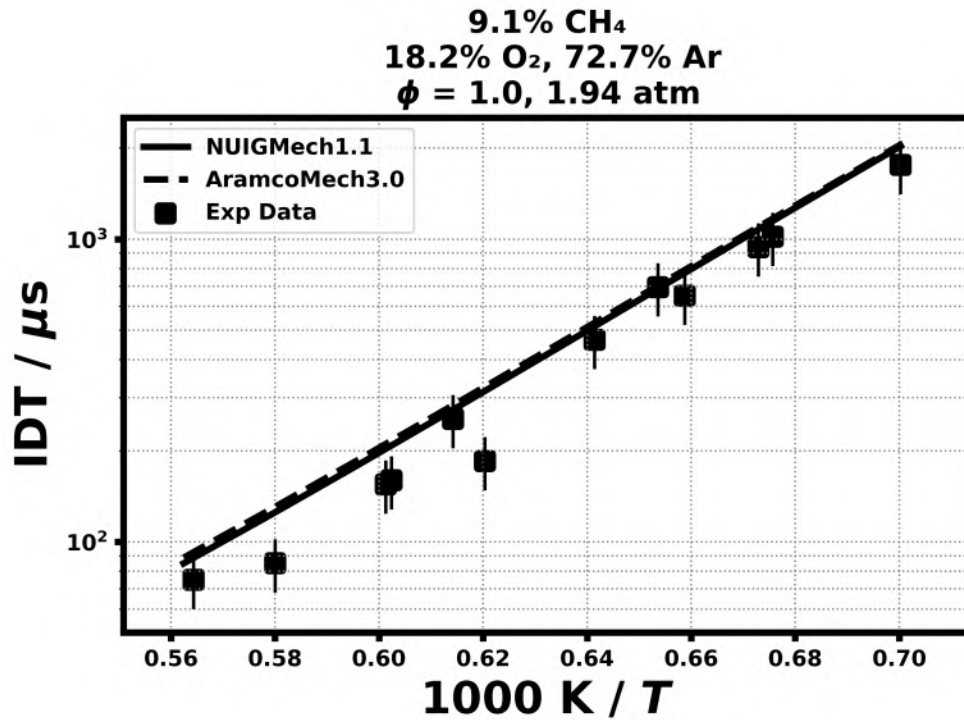
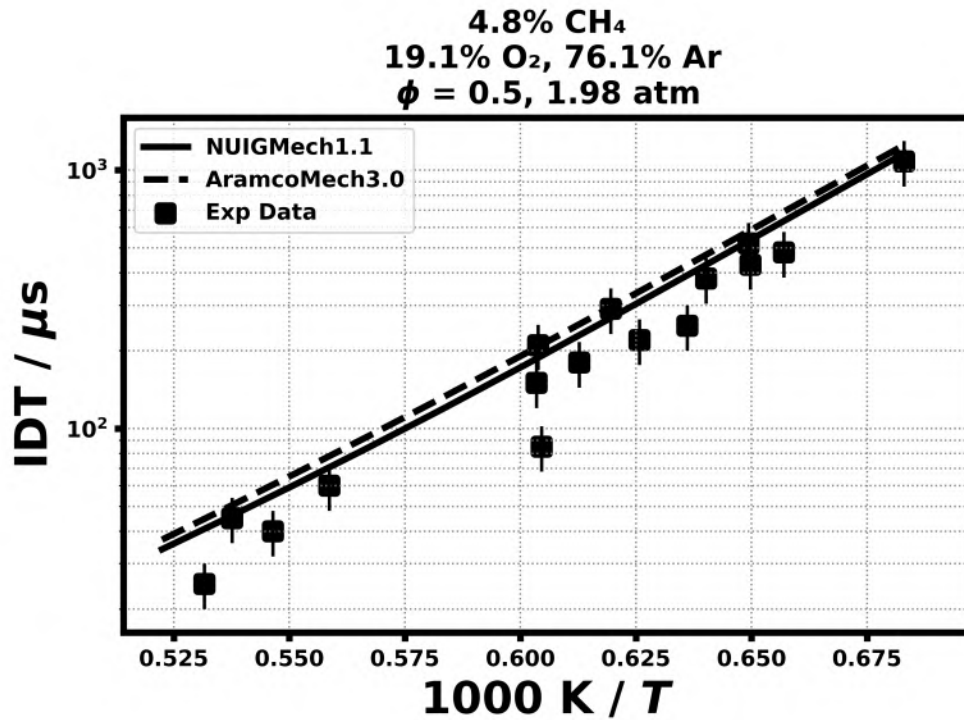
H₂/CO in 'air', 1atm, 298K



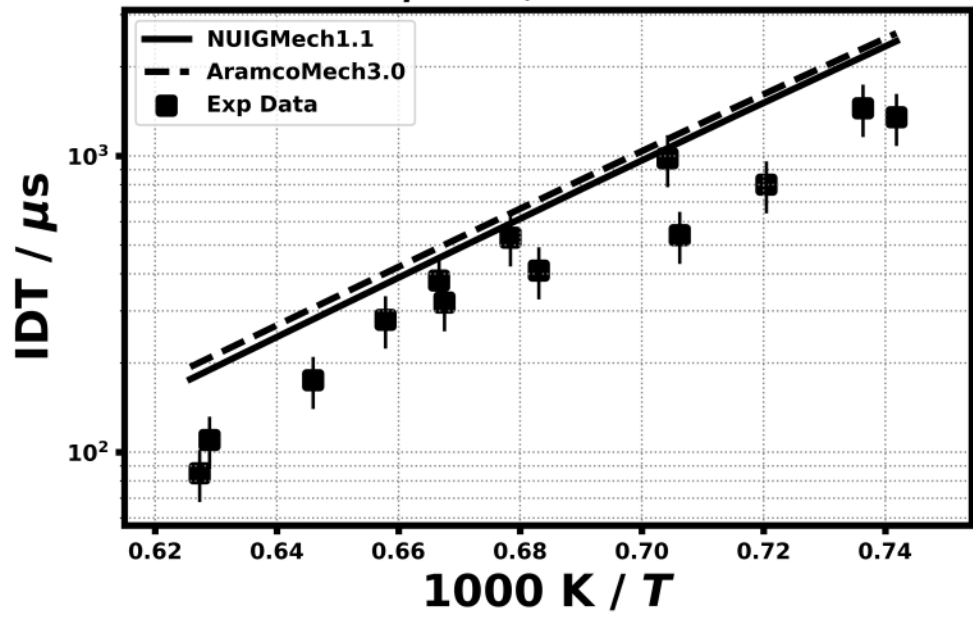
3. Validation for CH₄

Shock tube ignition delay time

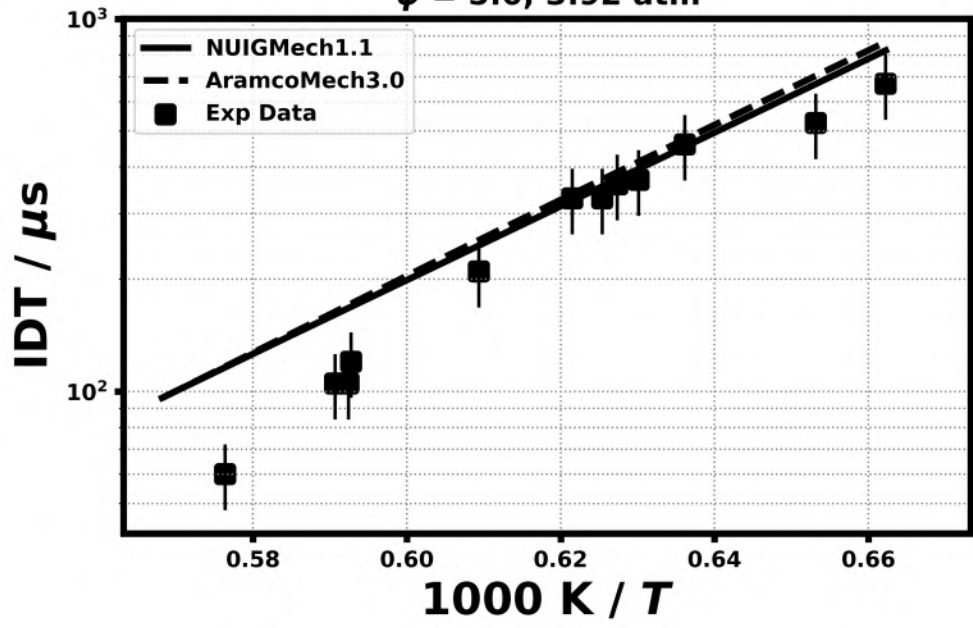
3.1) D.J. Seery. and C.T. Bowman., *Combustion and Flame*, 14 1 (1970) 37-47.



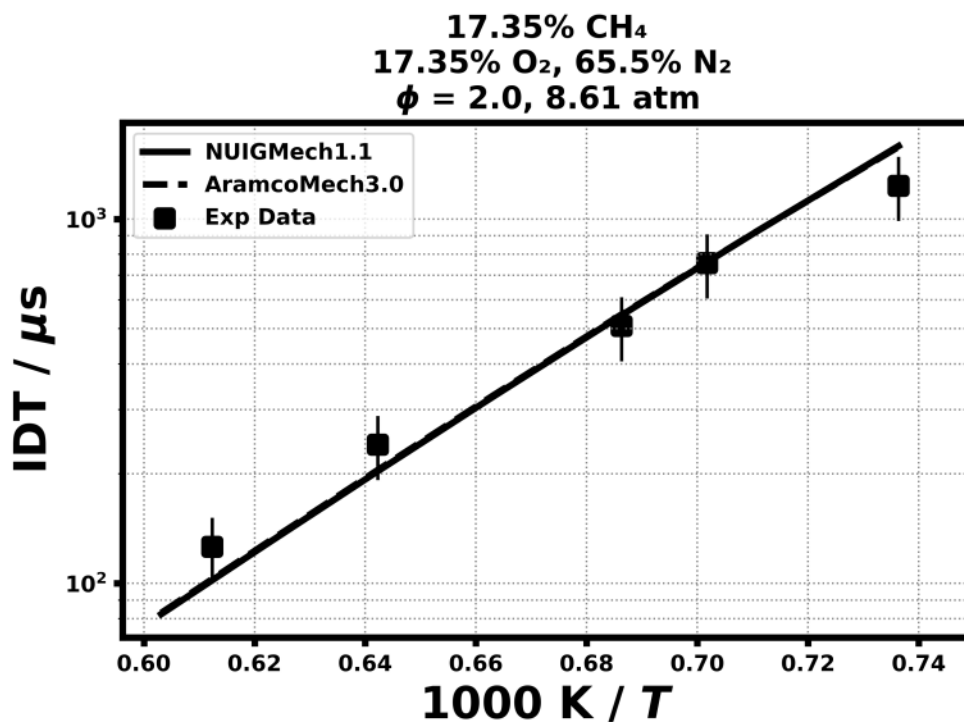
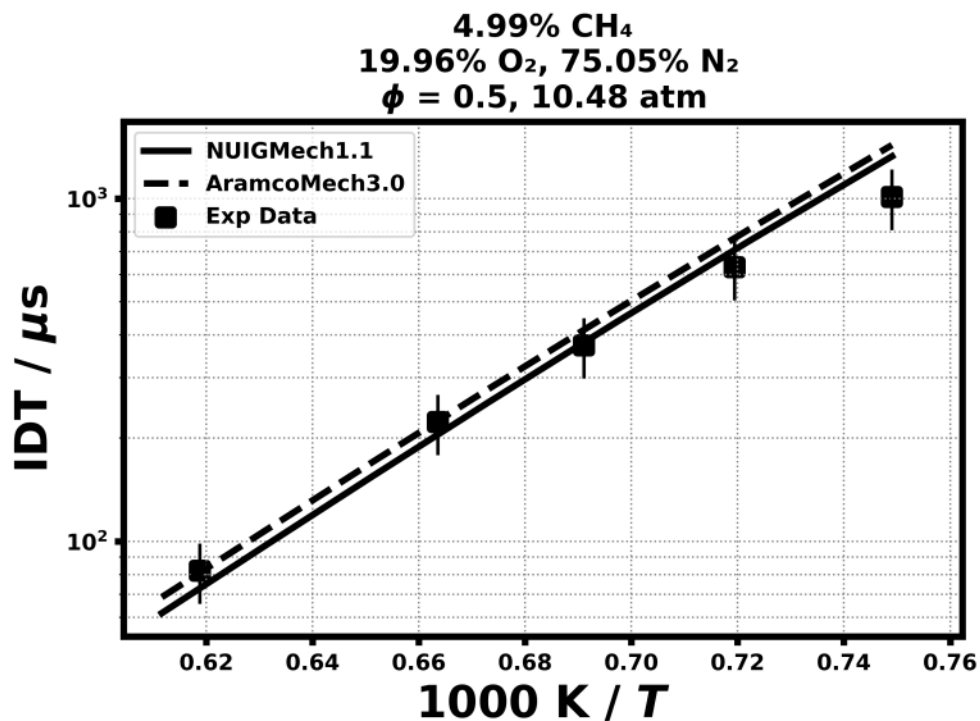
4.8% CH₄
19.1% O₂, 76.1% Ar
 $\phi = 0.5$, 3.92 atm



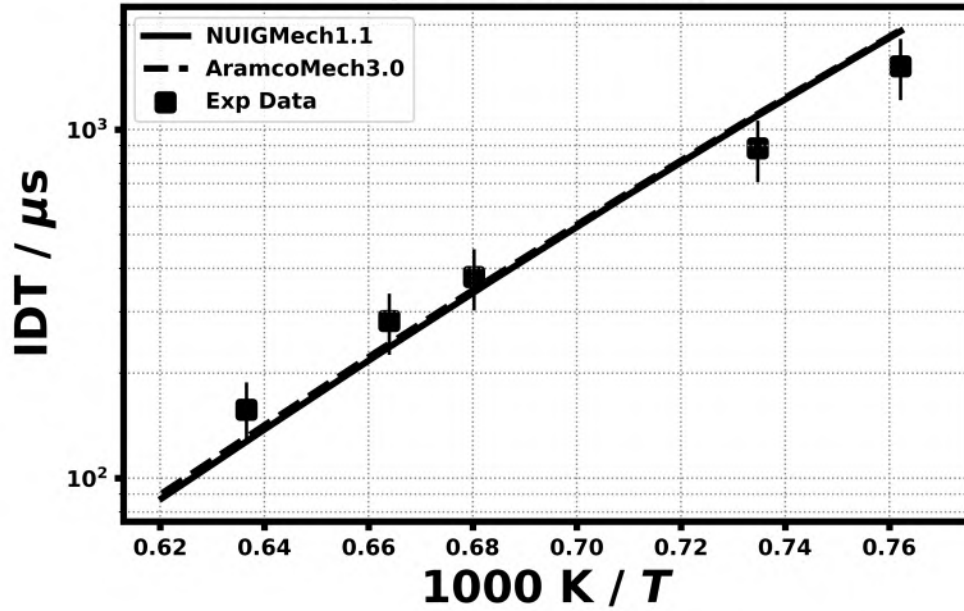
33.3% CH₄
13.3% O₂, 53.4% Ar
 $\phi = 5.0$, 3.92 atm



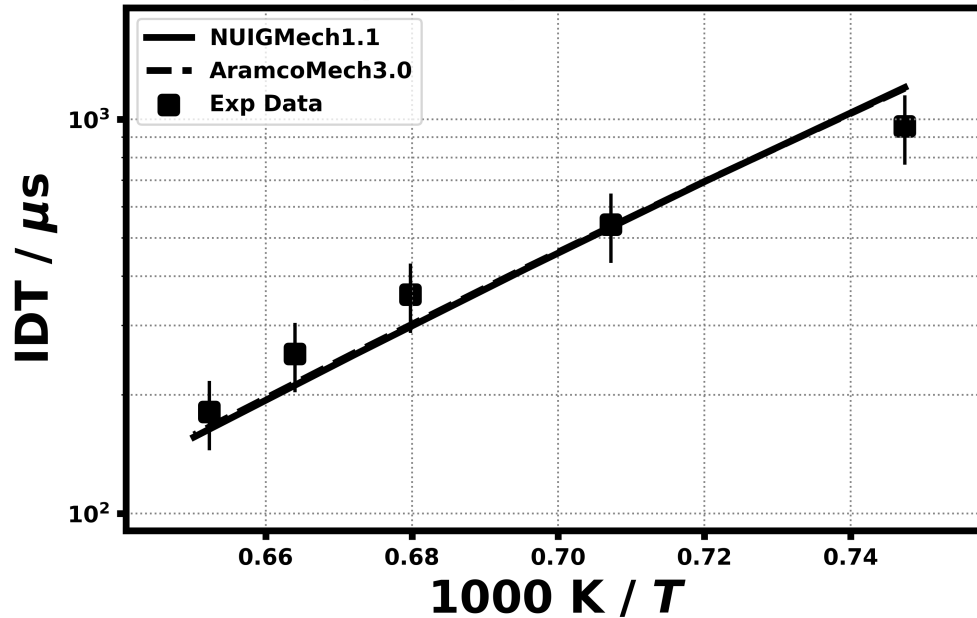
3.2) Burke, U., Somers, K. P., O'Toole, P., Zinner, C. M., Marquet, N., Bourque, G., ... & Curran, H. J., Combustion and flame, 162(2) (2015) 315-330.



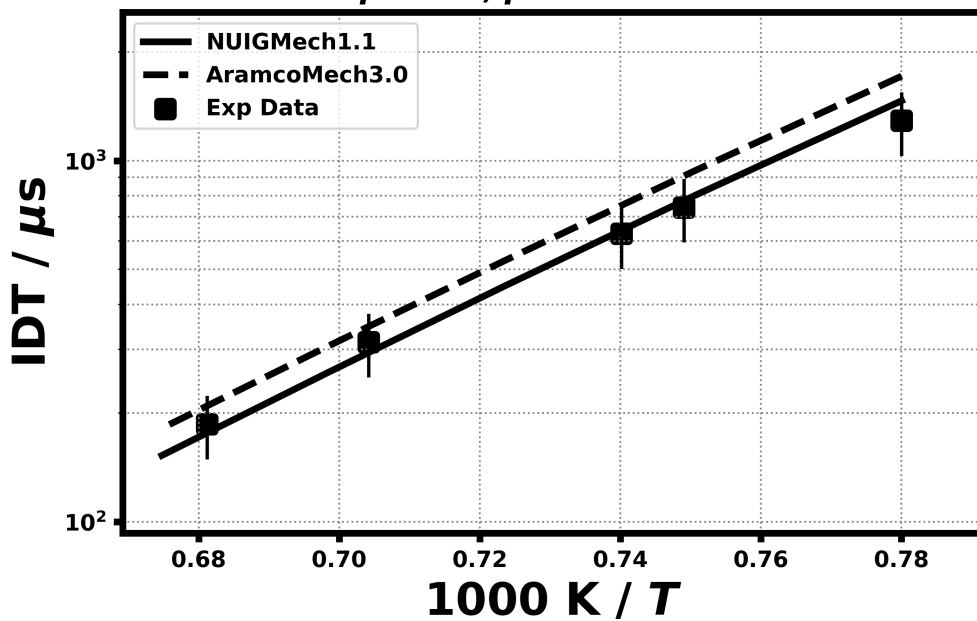
9.51% CH₄
18.91% O₂, 71.57% N₂
 $\phi = 1.0, 10.61 \text{ atm}$



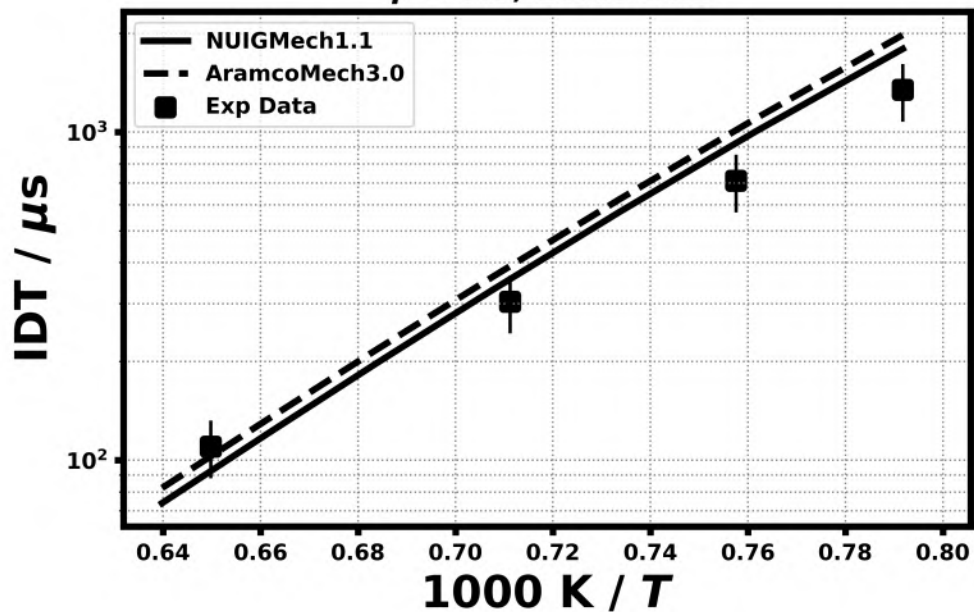
17.35% CH₄
17.35% O₂, 65.5% N₂
 $\phi = 2.0, p = 14.15 \text{ atm}$



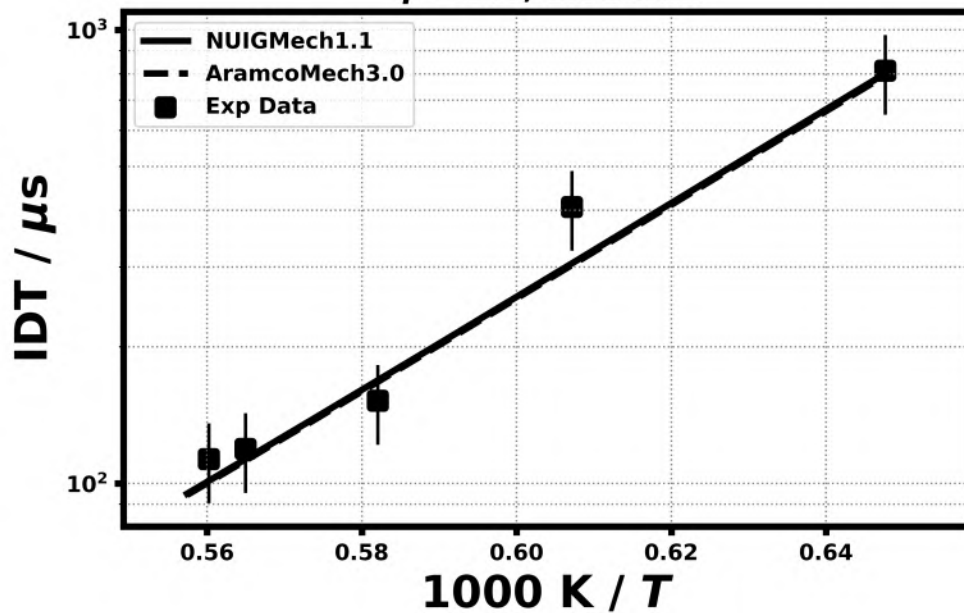
3.05% CH₄
20.37% O₂, 76.58% N₂
 $\phi = 0.3, p = 18.02 \text{ atm}$



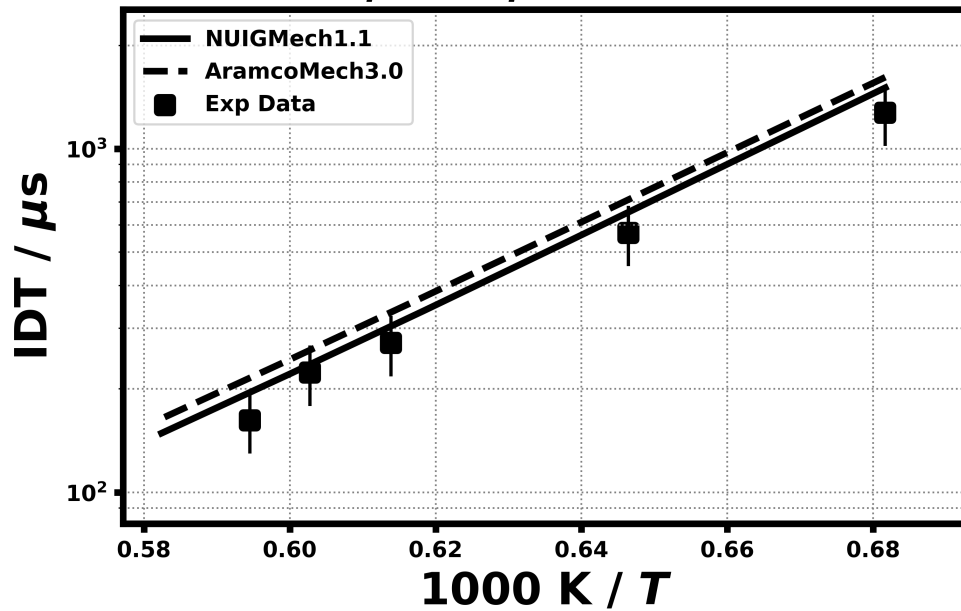
4.99% CH₄
19.96% O₂, 75.05% N₂
 $\phi = 0.5, 18.43 \text{ atm}$



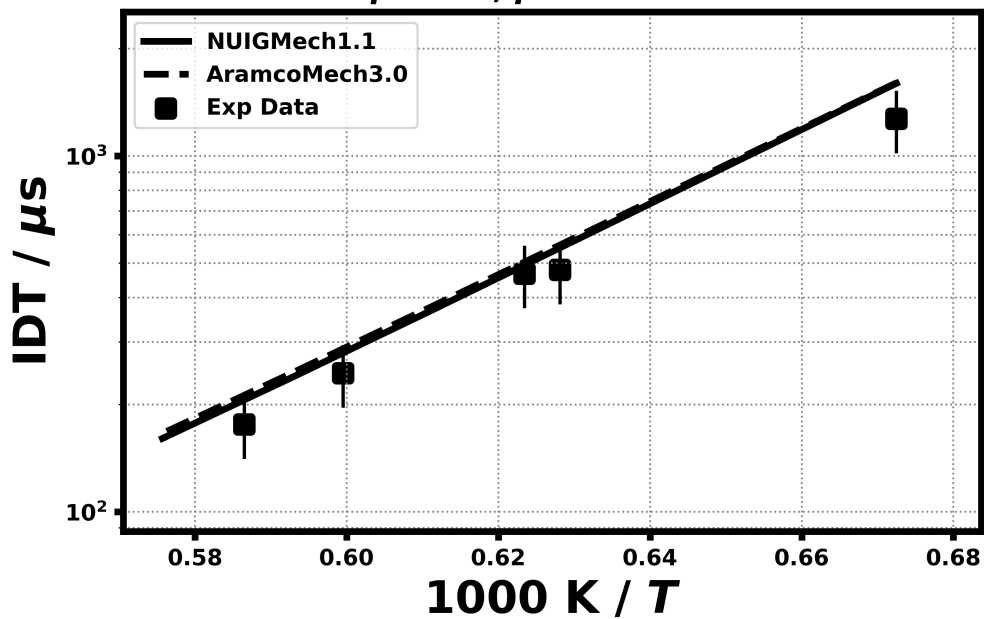
17.35% CH₄
17.35% O₂, 65.5% N₂
 $\phi = 2.0, 1.92 \text{ atm}$



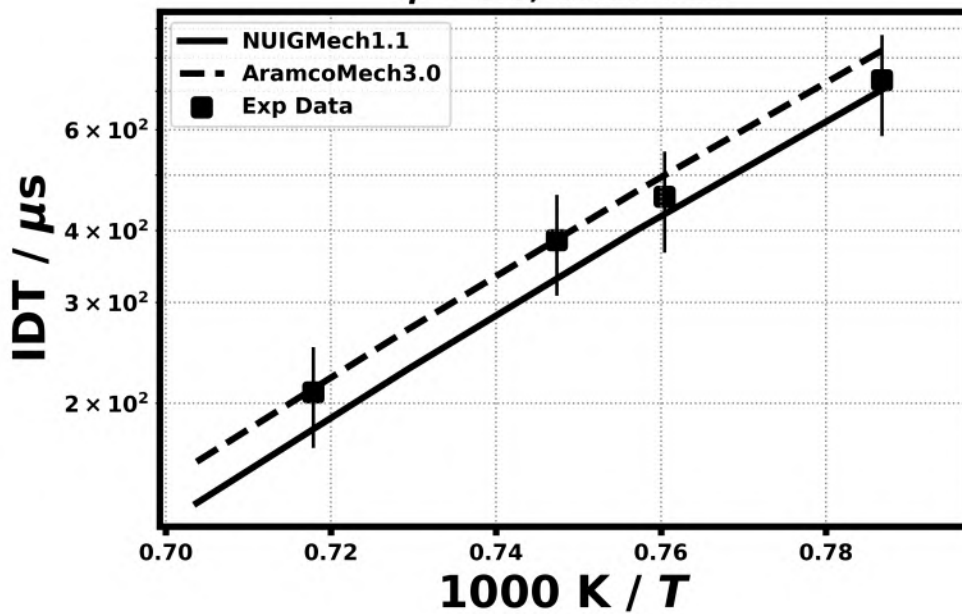
4.99% CH₄
19.96% O₂, 75.05% N₂
 $\phi = 0.5, p = 1.56 \text{ atm}$



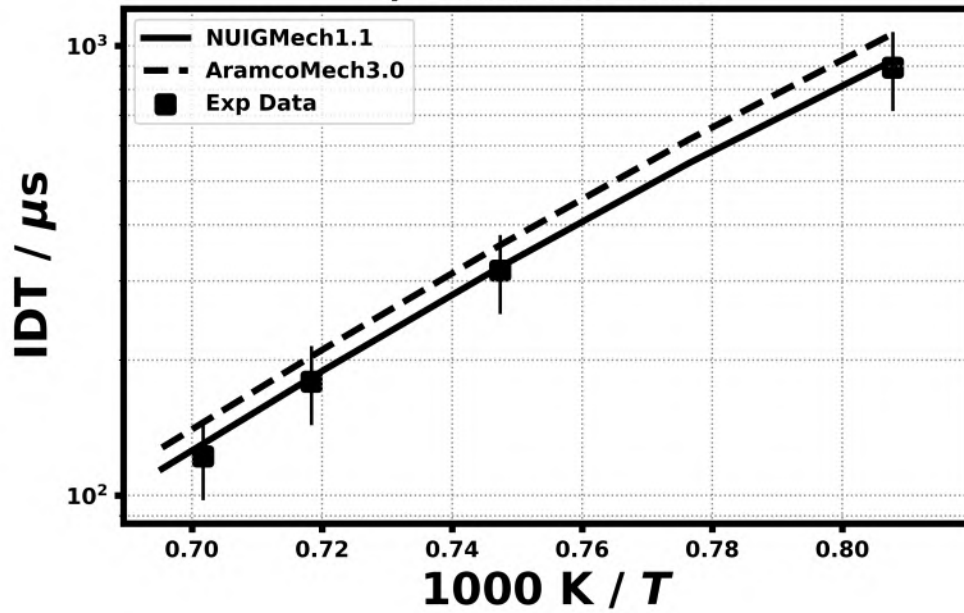
9.51% CH₄
18.91% O₂, 71.57% N₂
 $\phi = 1.0, p = 1.38 \text{ atm}$



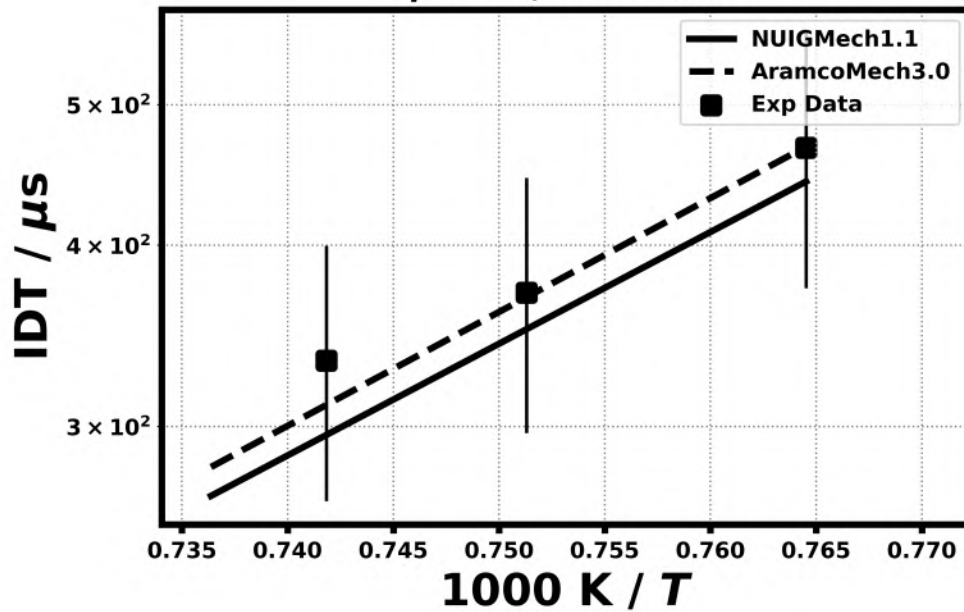
3.05% CH₄
20.37% O₂, 76.58% N₂
 $\phi = 0.3, 46.17 \text{ atm}$

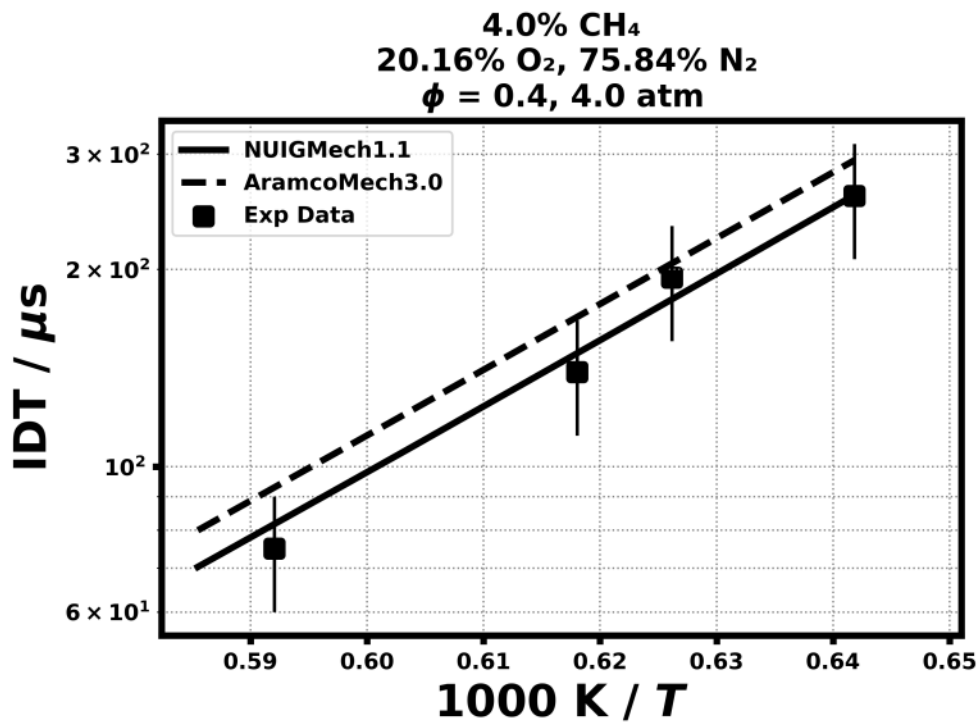


4.99% CH₄
19.96% O₂, 75.05% N₂
 $\phi = 0.5$, 43.95 atm

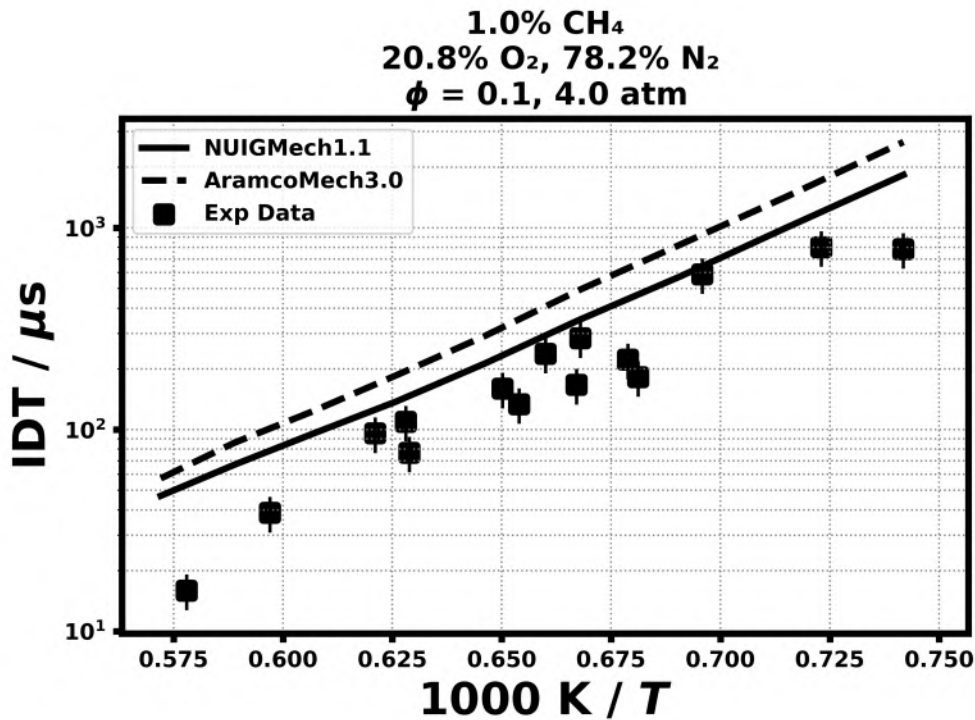


9.51% CH₄
18.91% O₂, 71.57% N₂
 $\phi = 1.0$, 42.67 atm

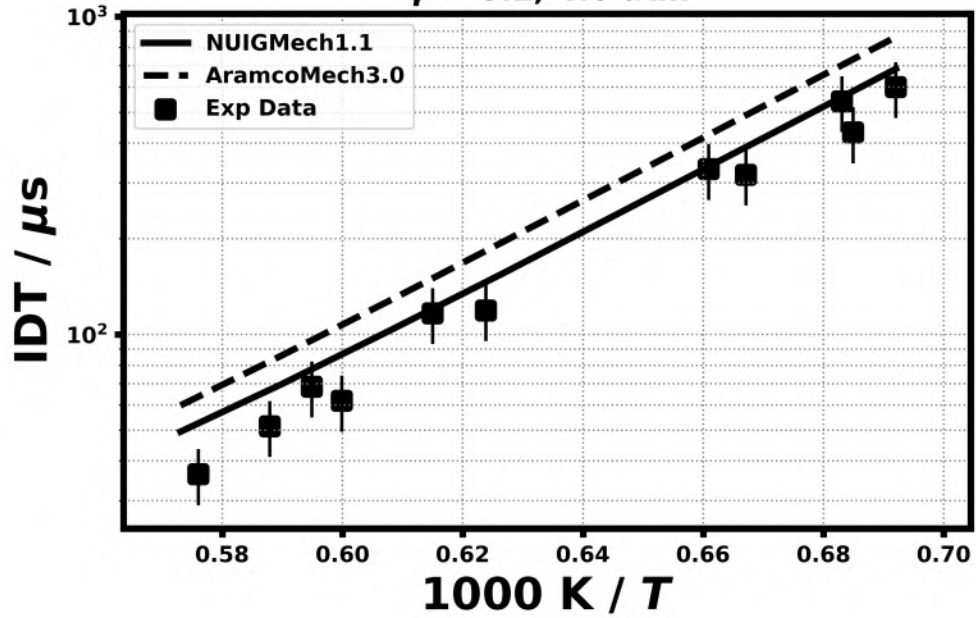




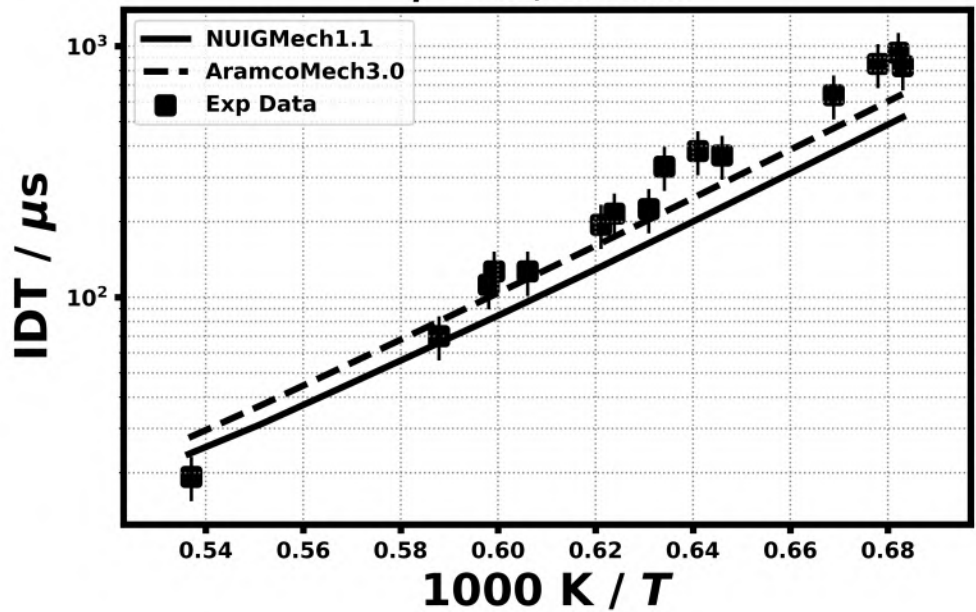
3.3) Eubank, C. S., M. J. Rabinowitz, W. C. Gardiner Jr, and R. E. Zellner, In Symposium (International) on Combustion, 18 (1981) 1767-1774.



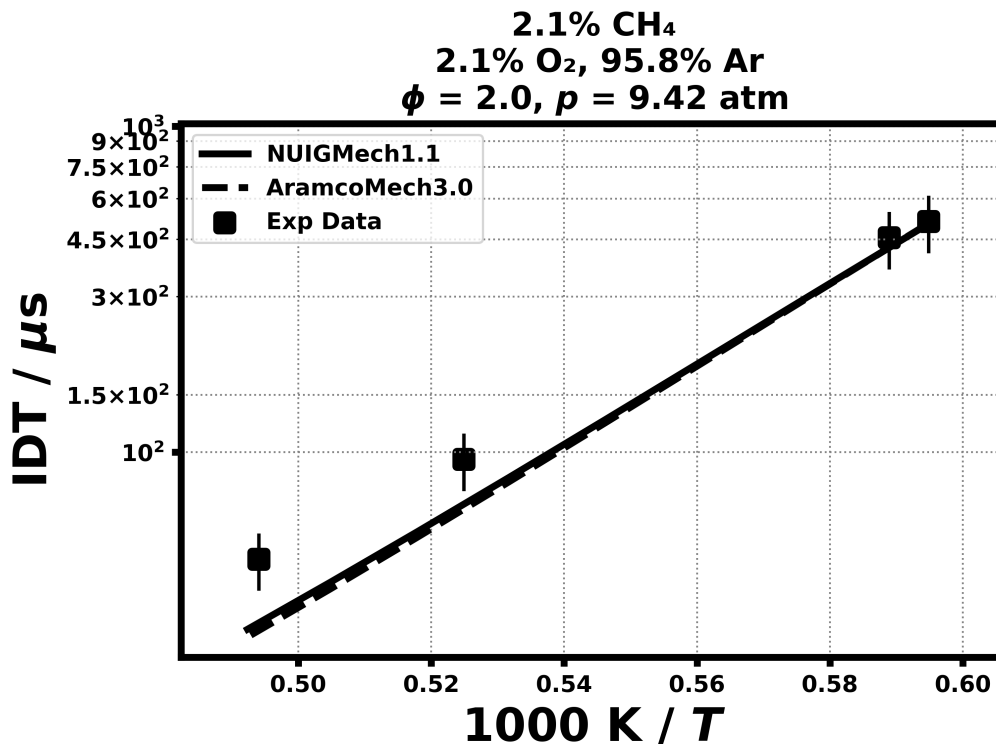
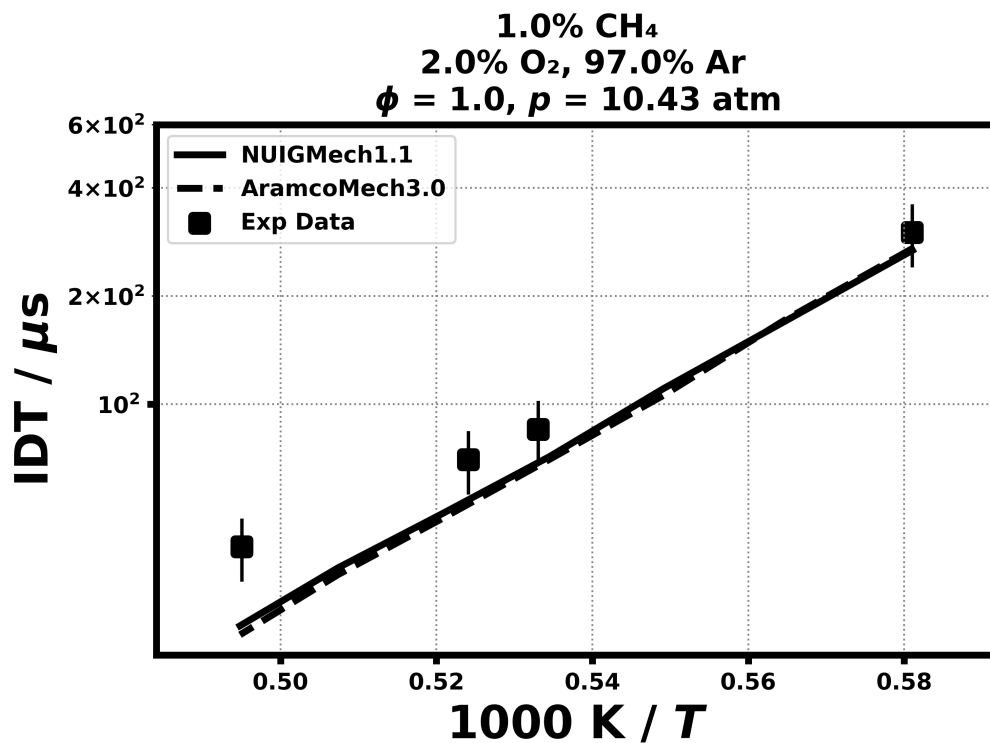
2.0% CH₄
20.6% O₂, 77.4% N₂
 $\phi = 0.2, 4.0 \text{ atm}$



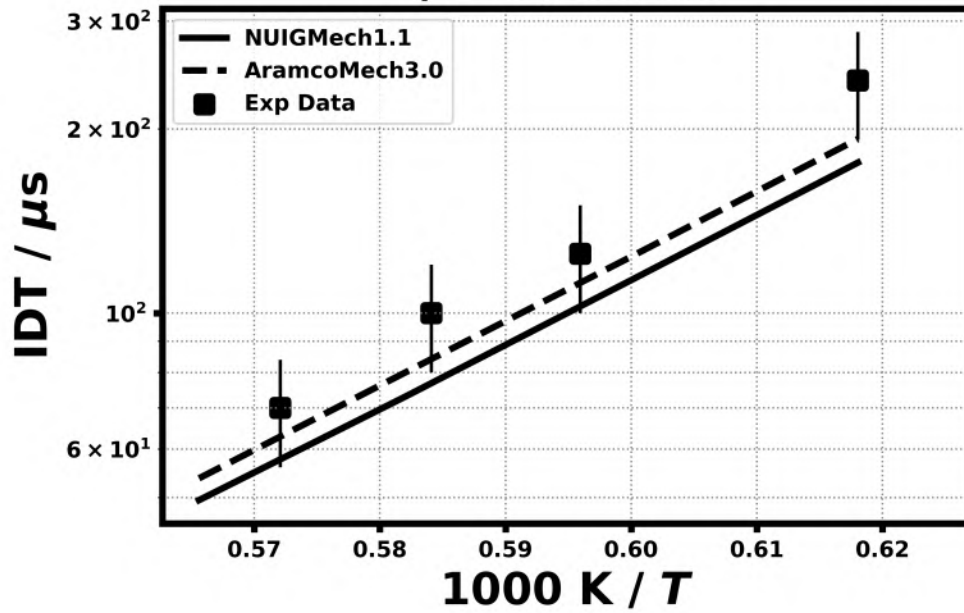
2.0% CH₄
20.0% O₂, 78.0% Ar
 $\phi = 0.2, 4.0 \text{ atm}$



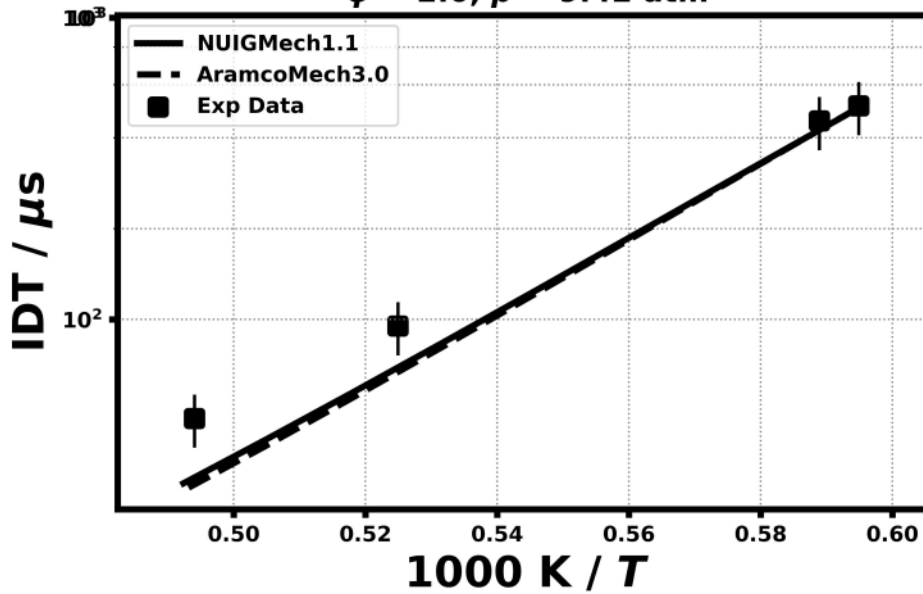
3.4) Lifshitz, A., Scheller, K., Burcat, A., & Skinner, G. B. Combustion and Flame, 16(3) (1971) 311-321.



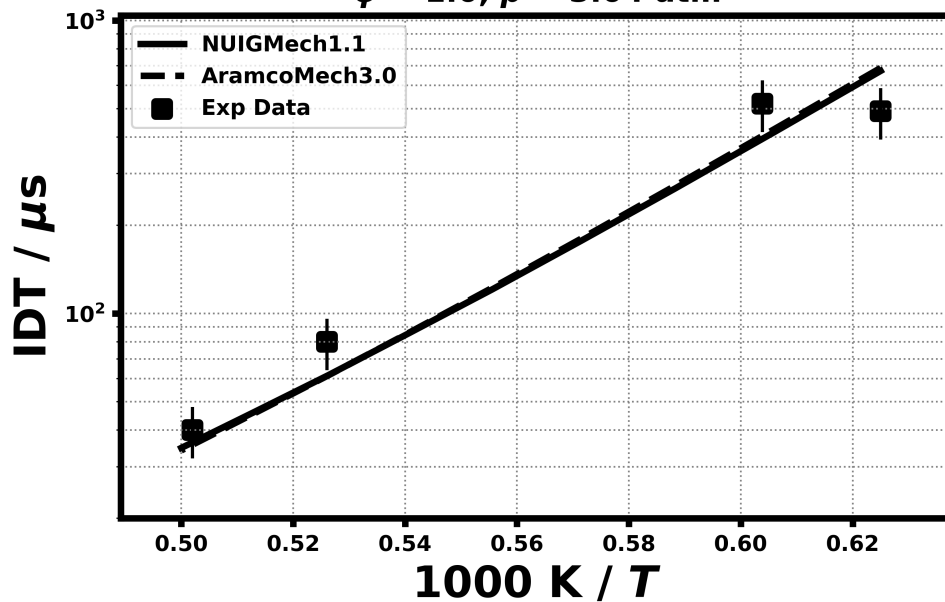
2.0% CH₄
8.0% O₂, 90.0% Ar
 $\phi = 0.5, 9.21 \text{ atm}$



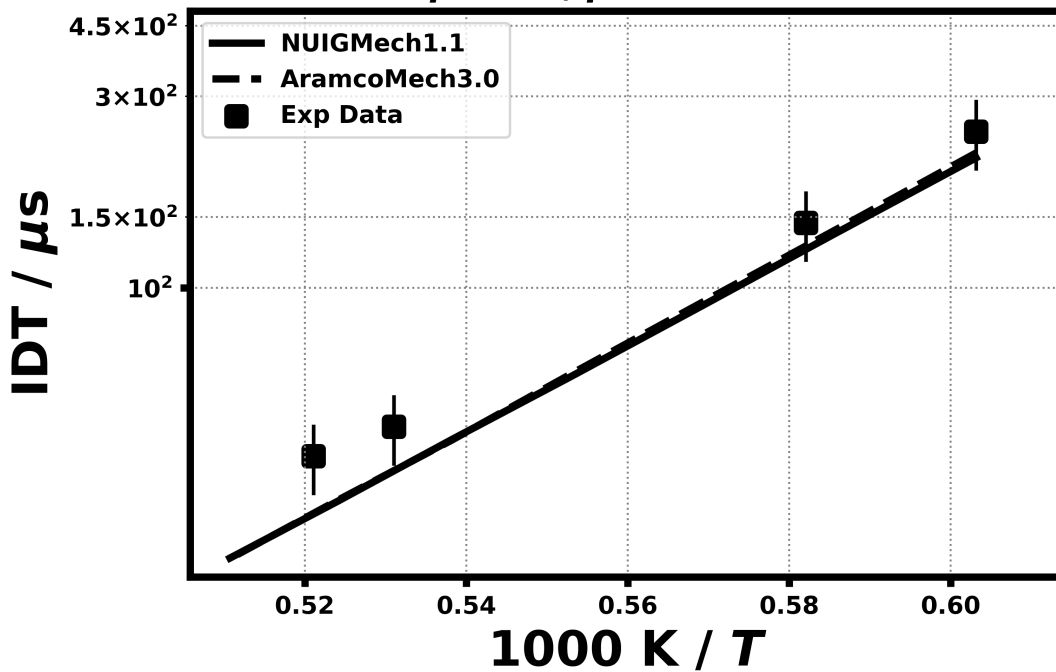
2.1% CH₄
2.1% O₂, 95.8% Ar
 $\phi = 2.0, p = 9.42 \text{ atm}$



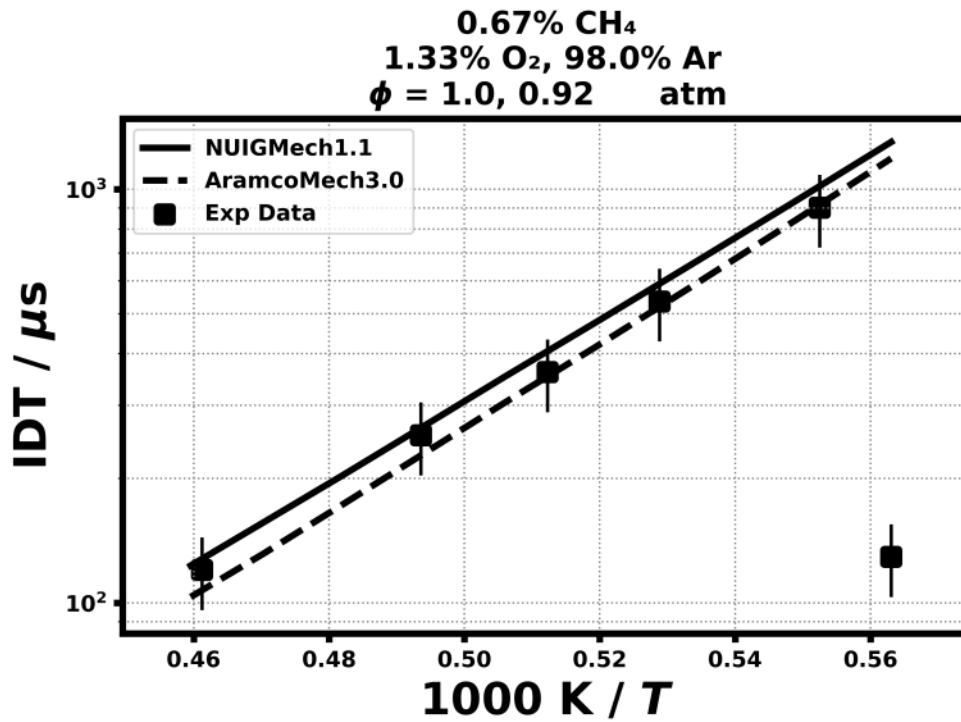
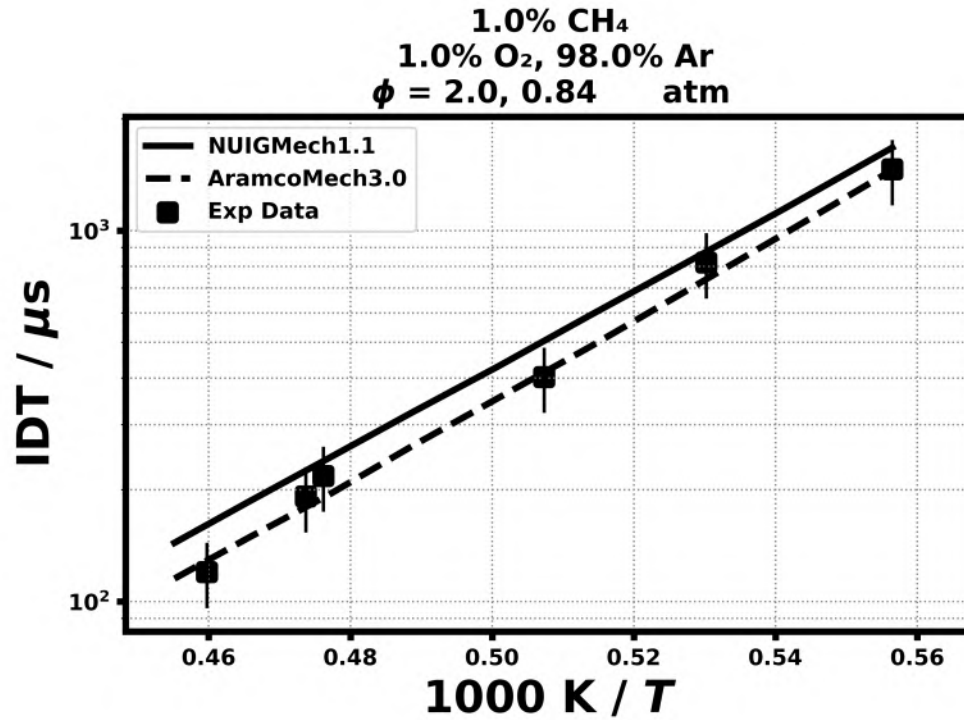
3.5% CH₄
7.0% O₂, 89.5% Ar
 $\phi = 1.0, p = 3.04 \text{ atm}$



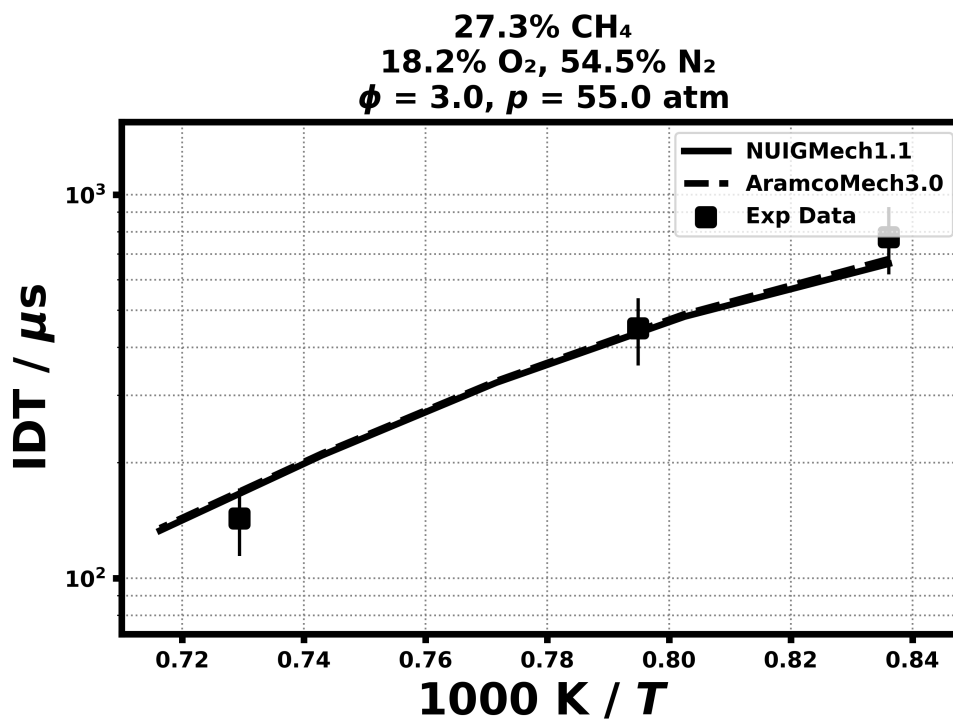
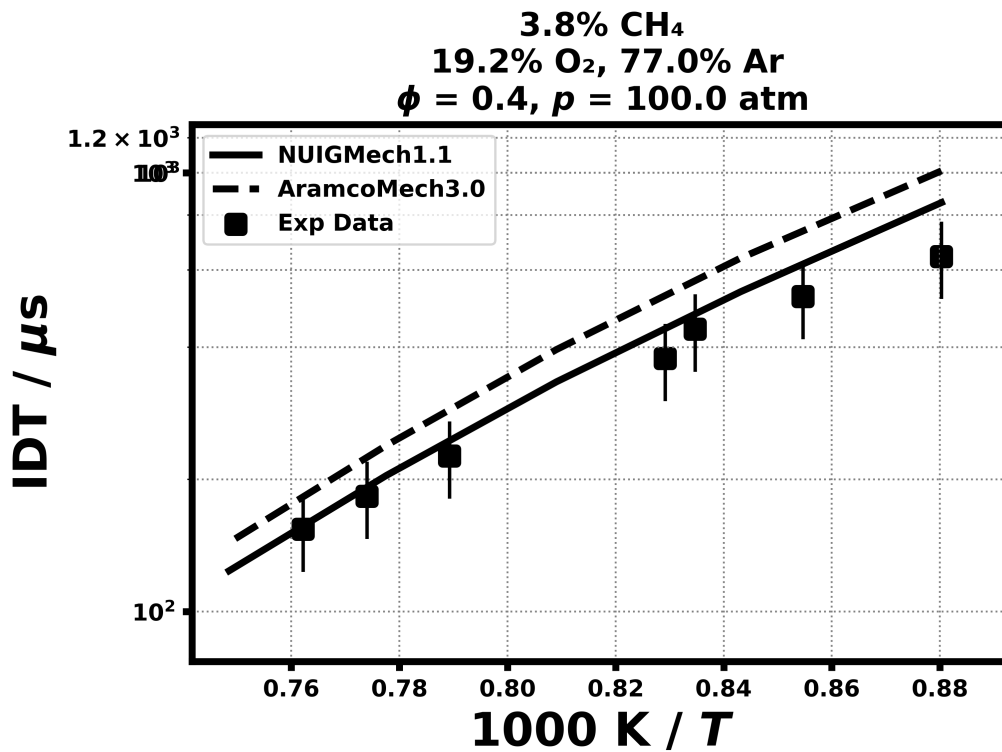
6.7% CH₄
6.7% O₂, 86.6% Ar
 $\phi = 2.0, p = 8.68 \text{ atm}$



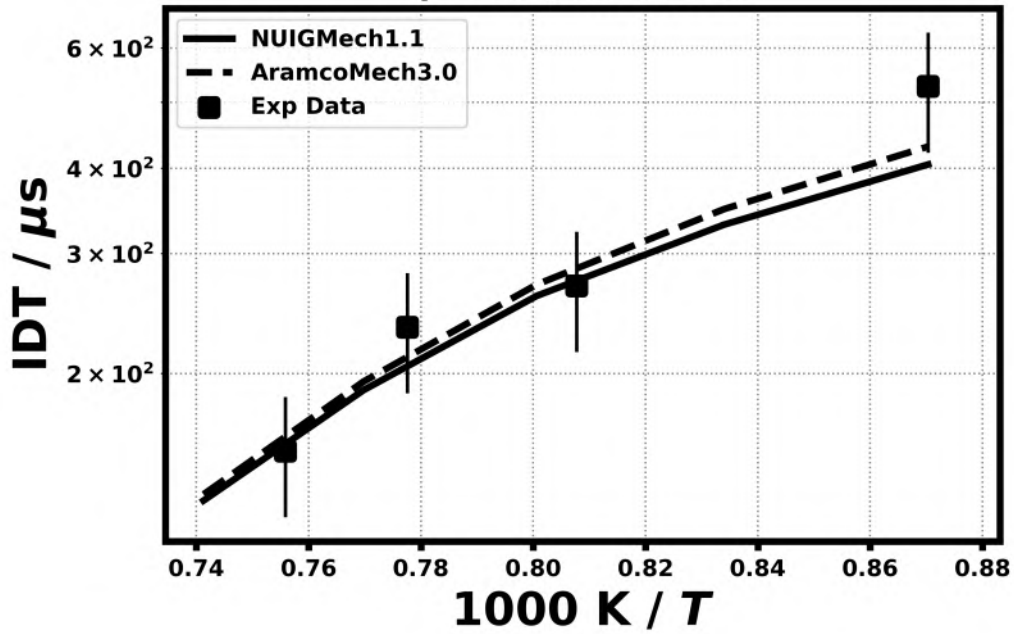
3.5) Mathieu, O., Goulier, J., Gourmel, F., Mannan, M. S., Chaumeix, N., & Petersen, E. L., Proceedings of the Combustion Institute, 35(3) (2015) 2731-2739.



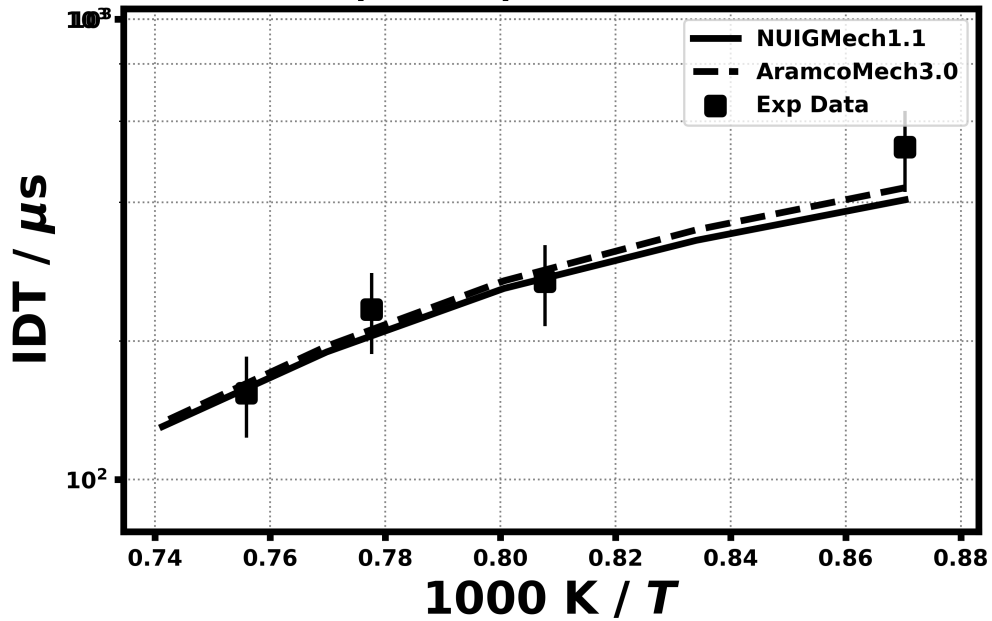
3.6) Petersen, E. L., Davidson, D. F., & Hanson, R. K. Journal of Propulsion and Power, 15(1) (1999) 82-91.



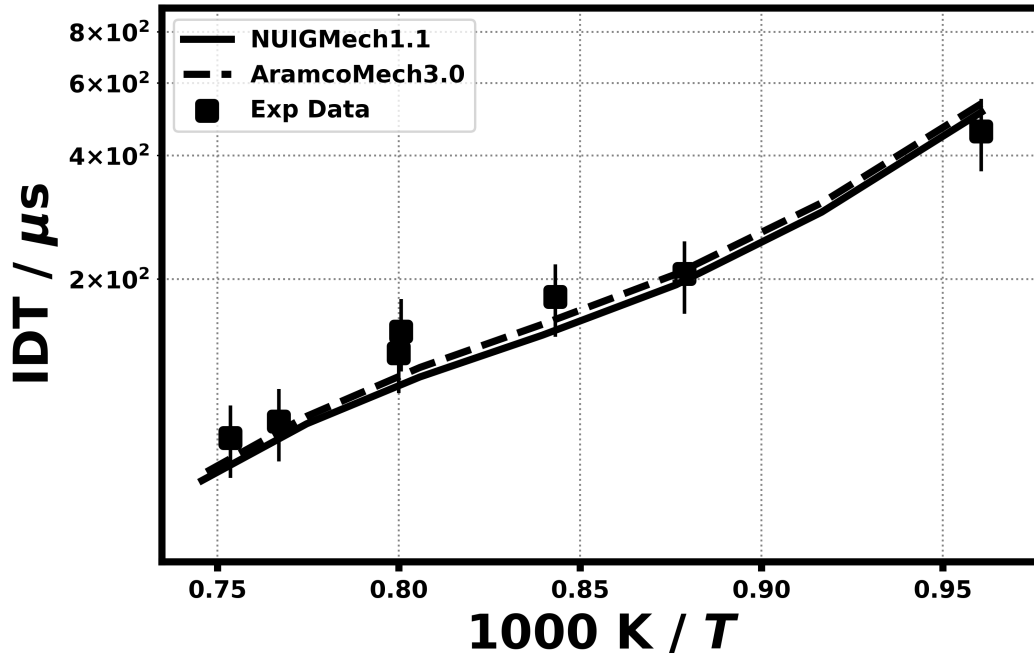
20.0% CH₄
13.3% O₂, 66.7% N₂
 $\phi = 3.0, 115.0 \text{ atm}$



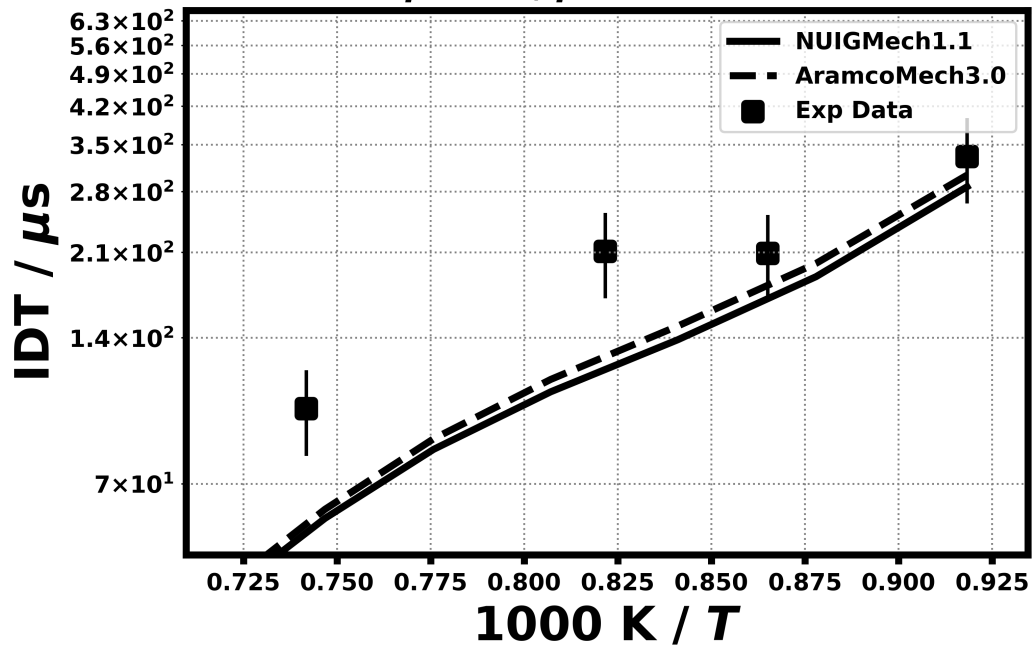
20.0% CH₄
13.3% O₂, 66.7% N₂
 $\phi = 3.0, p = 115.0 \text{ atm}$



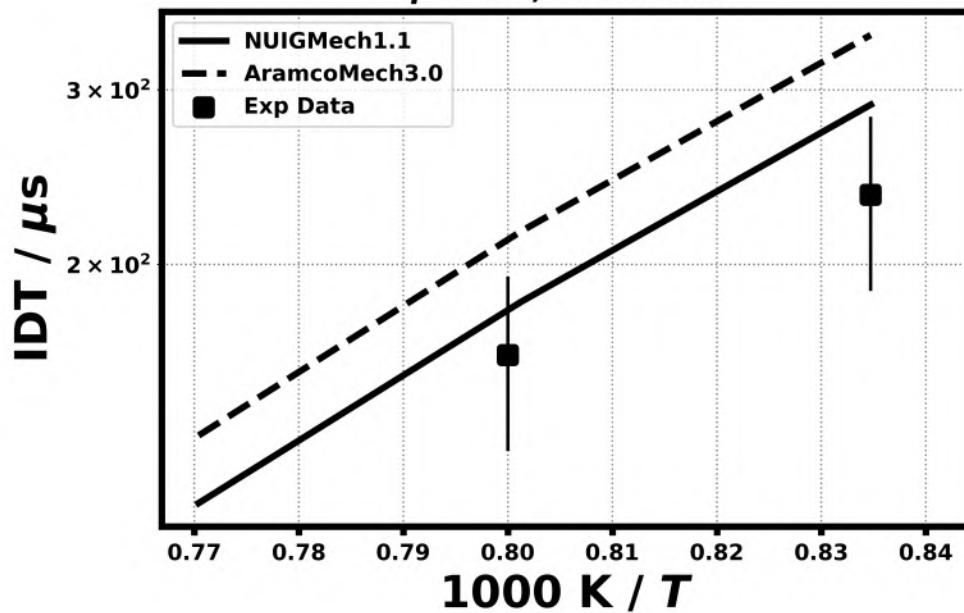
27.3% CH₄
18.2% O₂, 54.5% N₂
 $\phi = 3.0, p = 130.0$ atm



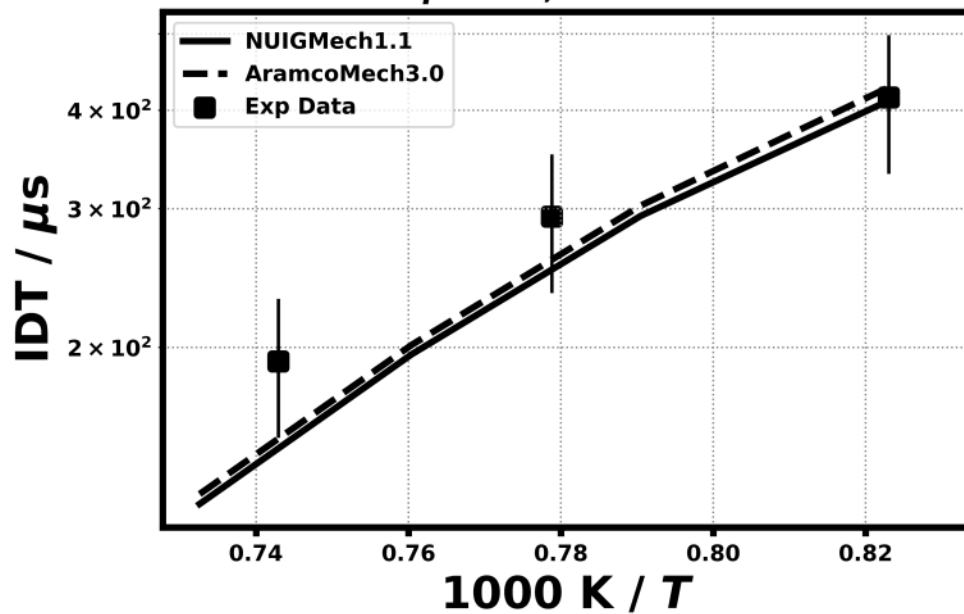
27.3% CH₄
18.2% O₂, 54.5% Ar
 $\phi = 3.0, p = 130.0$ atm



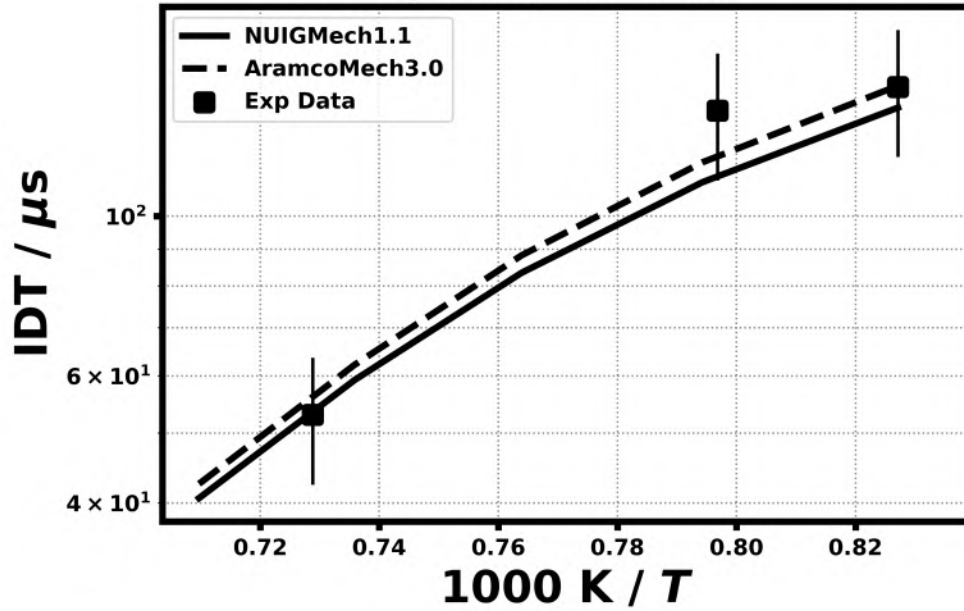
3.8% CH₄
19.2% O₂, 77.0% Ar
 $\phi = 0.4$, 150.0 atm



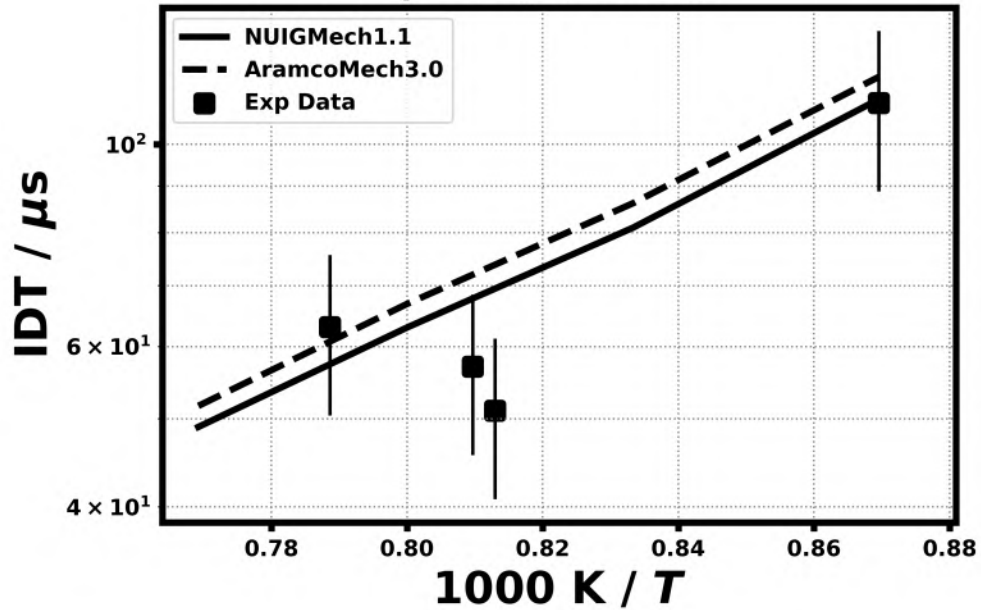
27.3% CH₄
18.2% O₂, 54.5% Ar
 $\phi = 3.0$, 65.0 atm



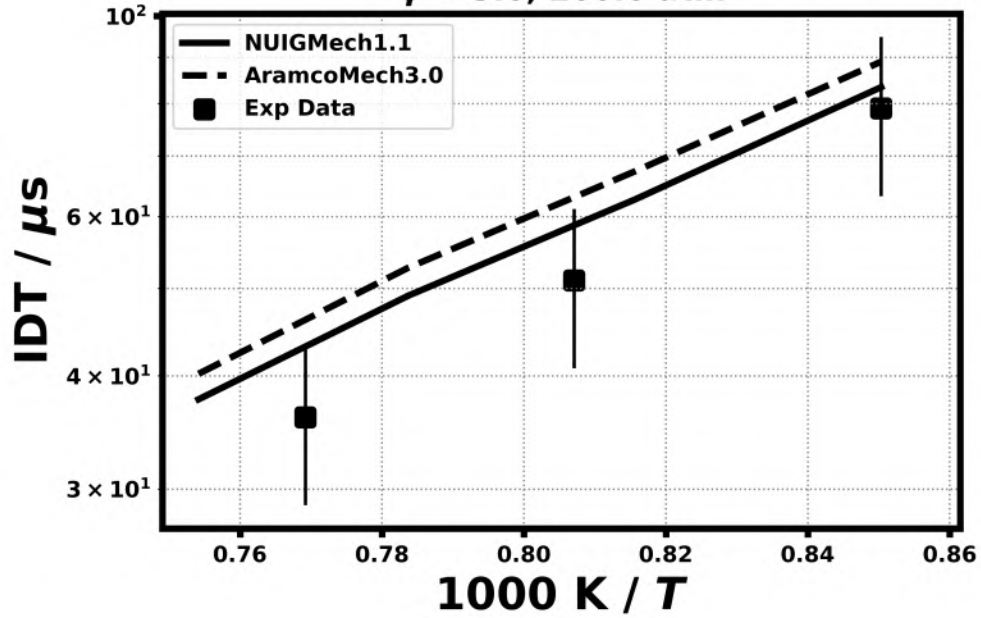
20.0% CH₄
13.3% O₂, 66.7% Ar
 $\phi = 3.0, 170.0 \text{ atm}$



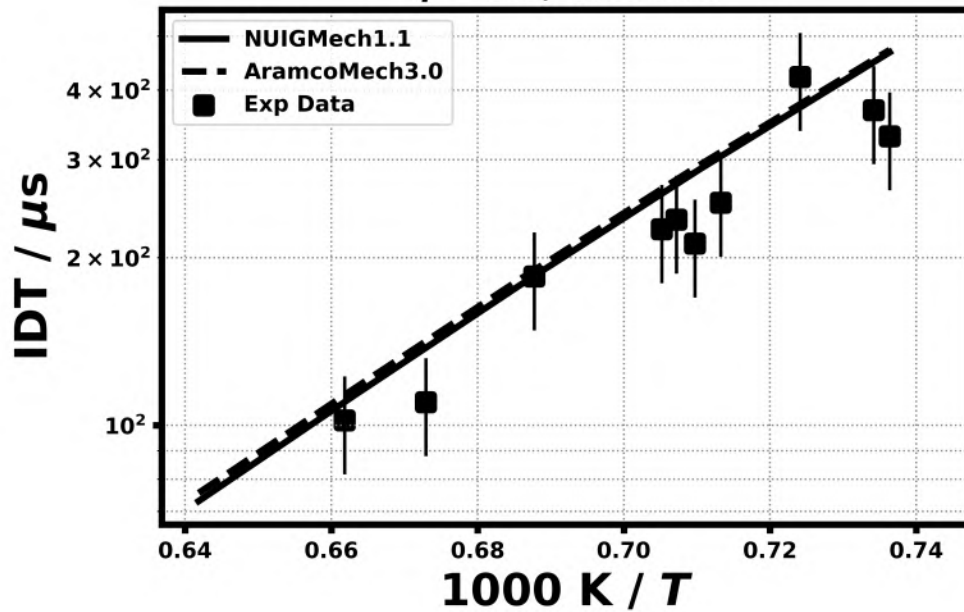
27.3% CH₄
18.2% O₂, 54.5% N₂
 $\phi = 3.0, 180.0 \text{ atm}$



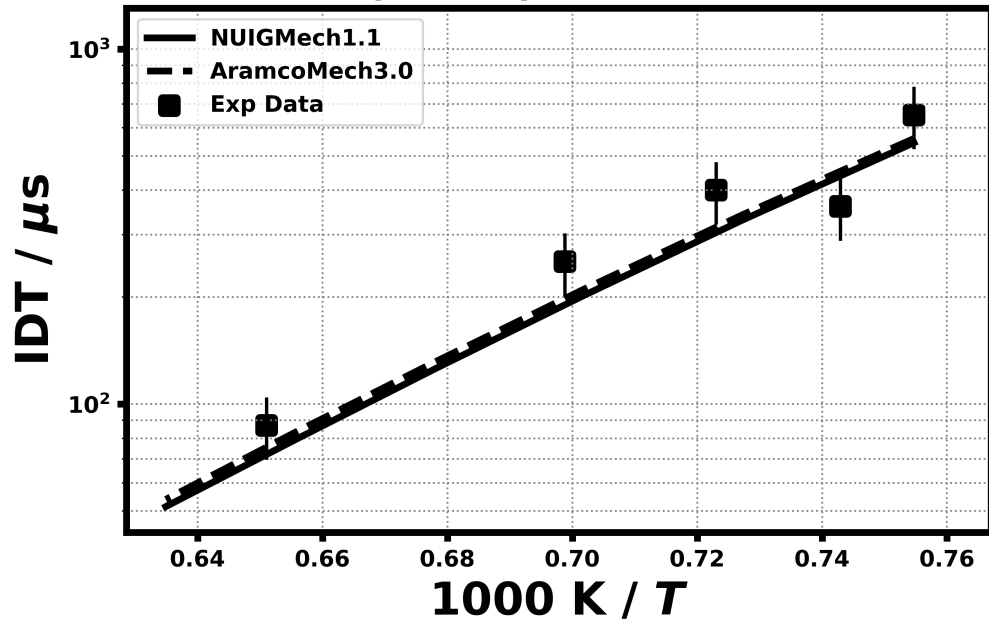
20.0% CH₄
13.3% O₂, 66.7% Ar
 $\phi = 3.0, 260.0 \text{ atm}$



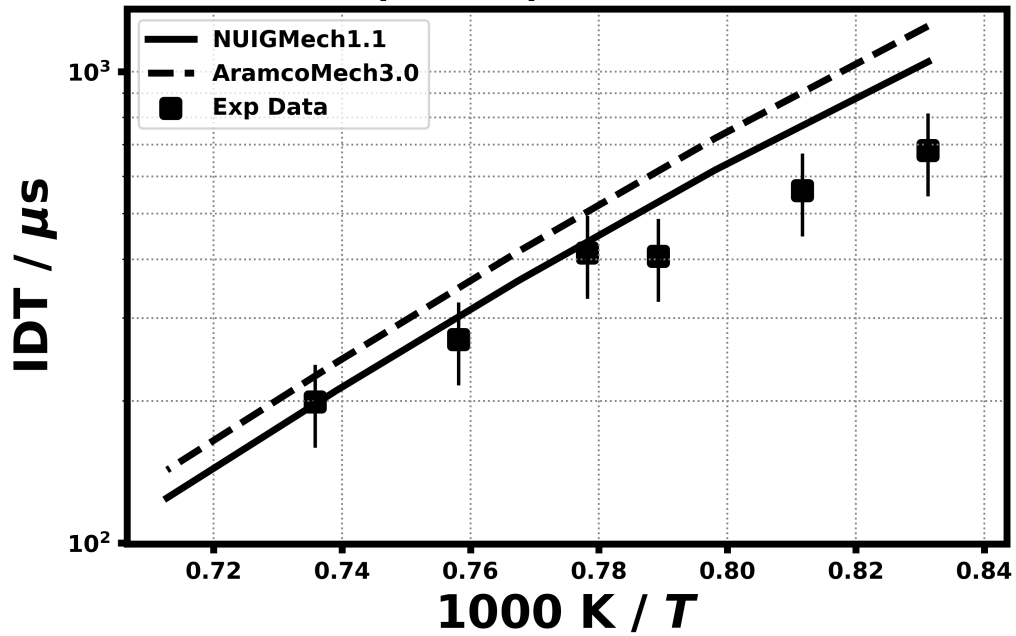
20.0% CH₄
13.3% O₂, 66.7% N₂
 $\phi = 3.0, 40.0 \text{ atm}$



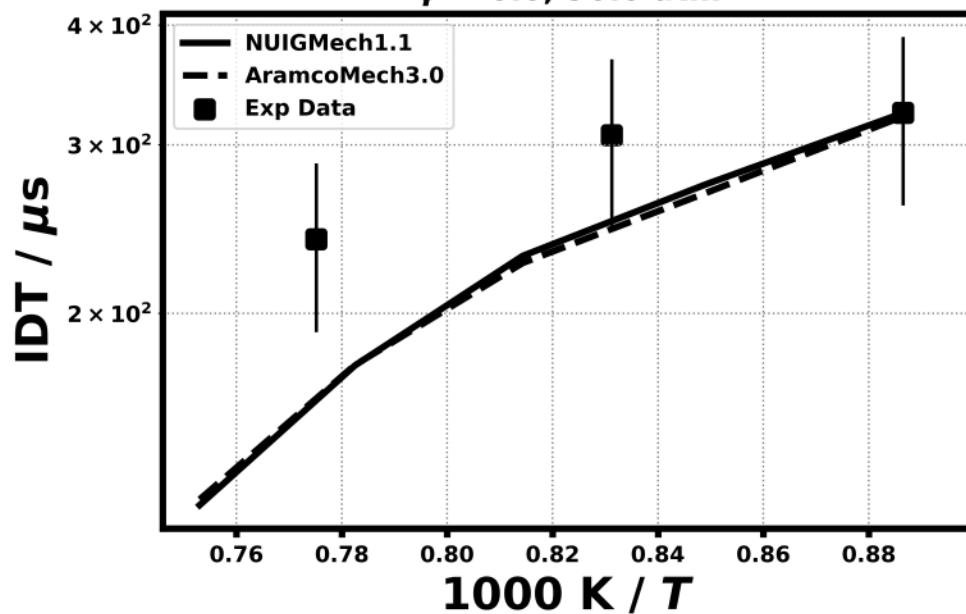
20.0% CH₄
13.3% O₂, 66.7% Ar
 $\phi = 3.0, p = 40.0$ atm



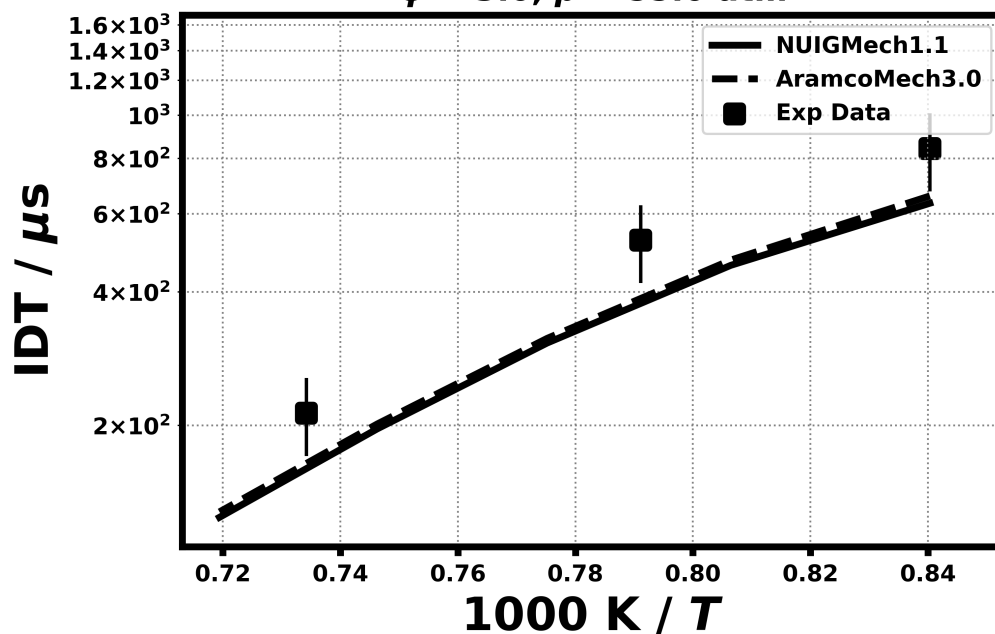
3.8% CH₄
19.2% O₂, 77.0% Ar
 $\phi = 0.4, p = 50.0$ atm



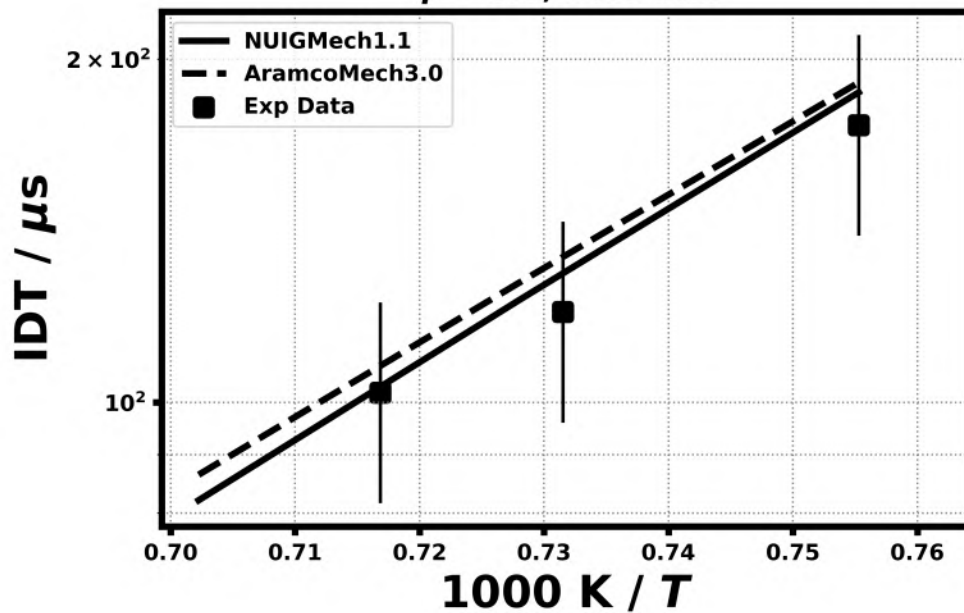
50.0% CH₄
16.7% O₂
 $\phi = 6.0, 90.0 \text{ atm}$



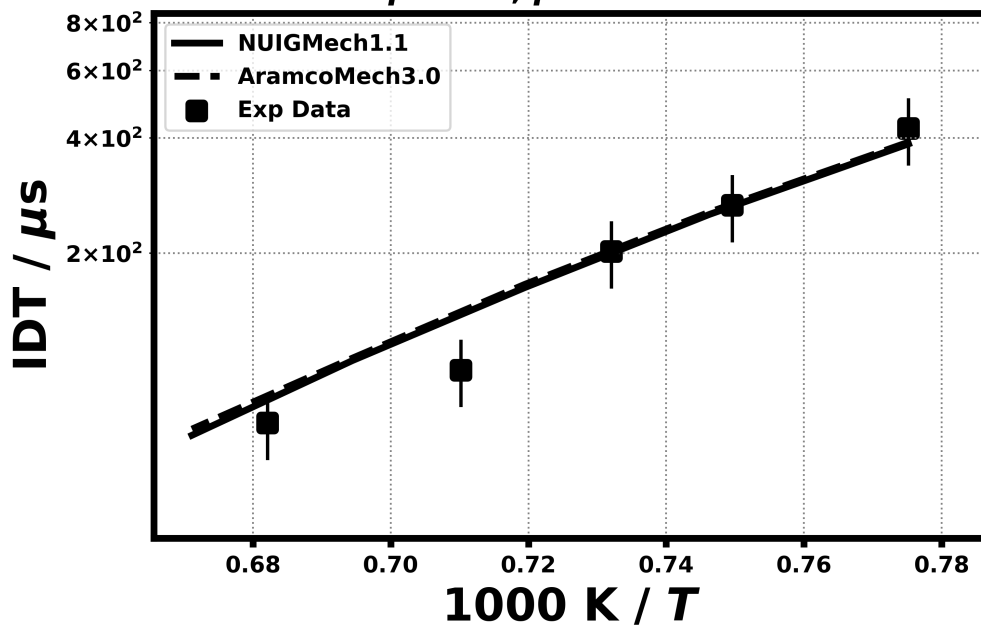
27.3% CH₄
18.2% O₂, 54.5% Ar
 $\phi = 3.0, p = 55.0 \text{ atm}$



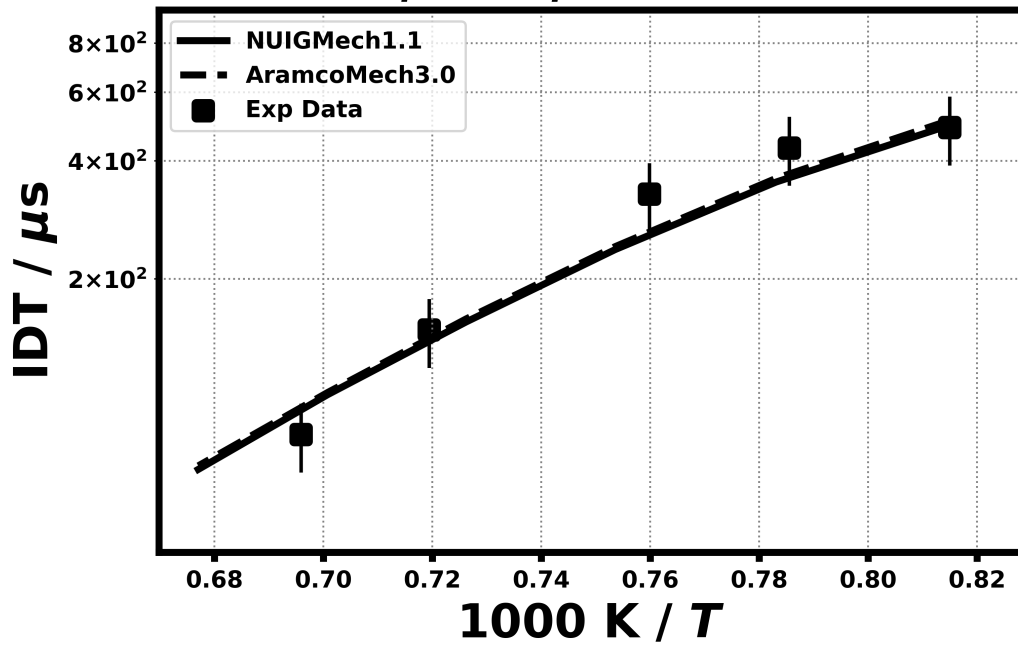
50.0% CH₄
16.7% O₂
 $\phi = 6.0, 70.0 \text{ atm}$



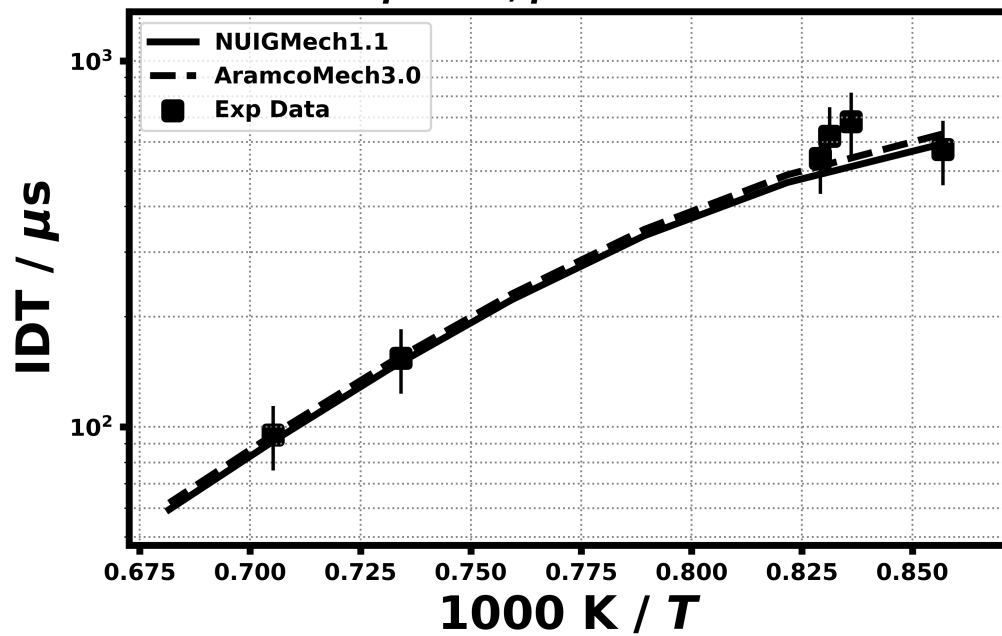
20.0% CH₄
13.3% O₂, 66.7% N₂
 $\phi = 3.0, p = 75.0 \text{ atm}$



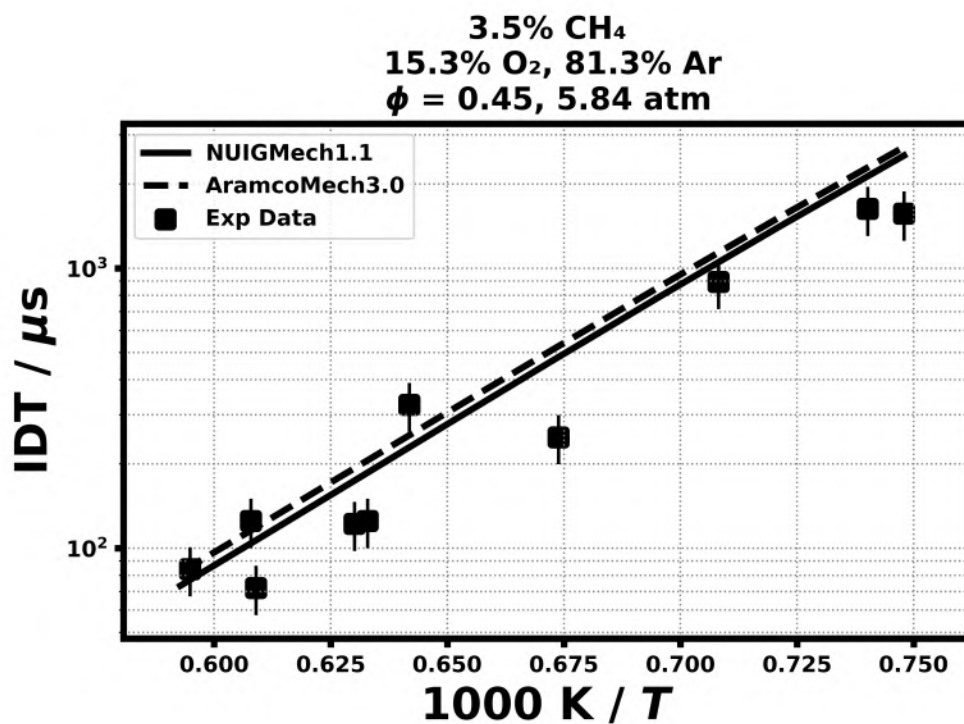
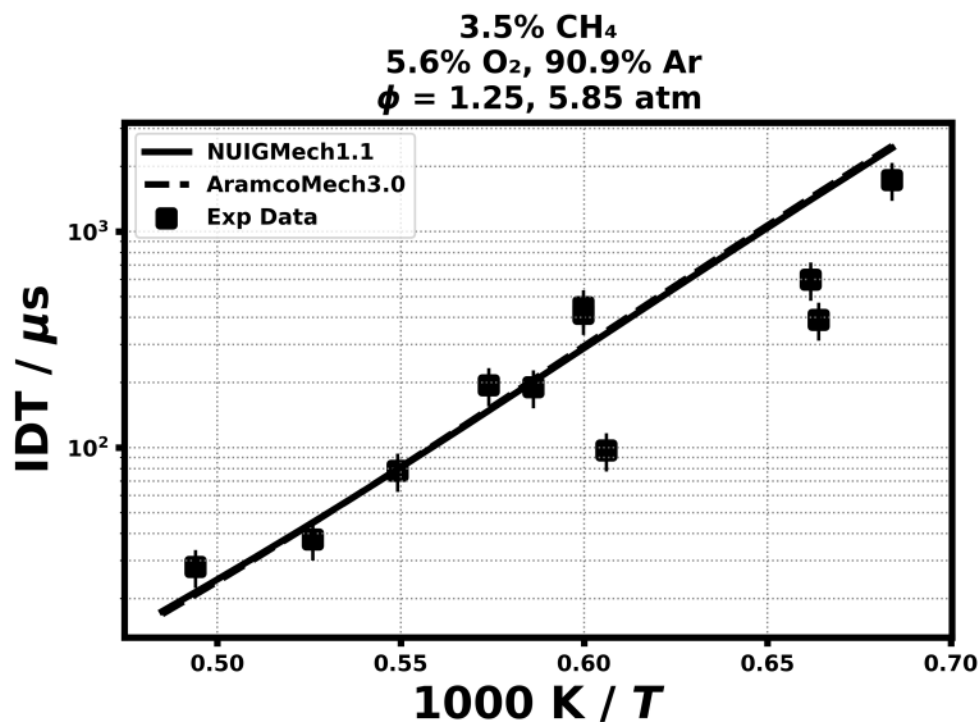
20.0% CH₄
13.3% O₂, 66.7% N₂
 $\phi = 3.0, p = 85.0$ atm



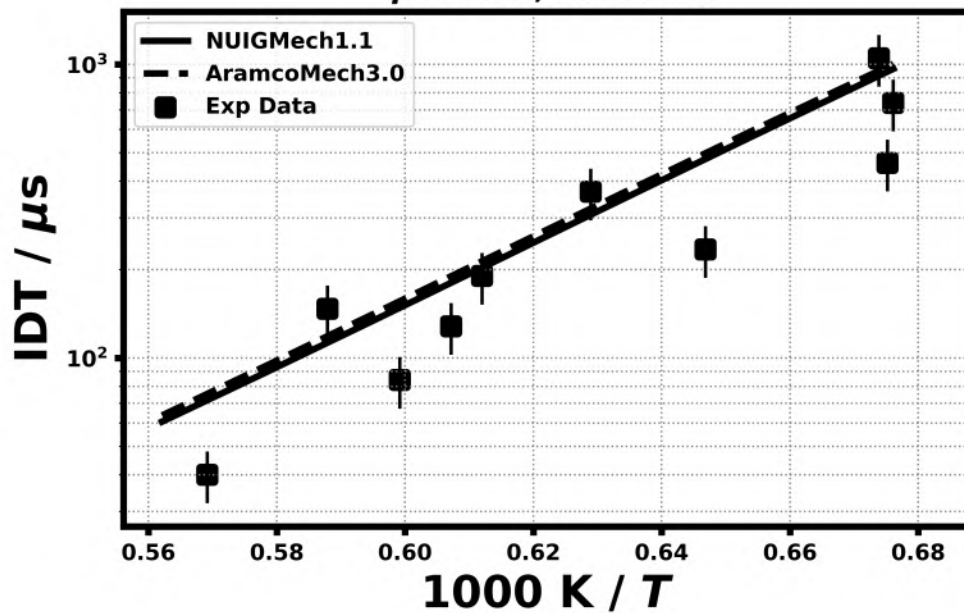
20.0% CH₄
13.3% O₂, 66.7% Ar
 $\phi = 3.0, p = 85.0$ atm



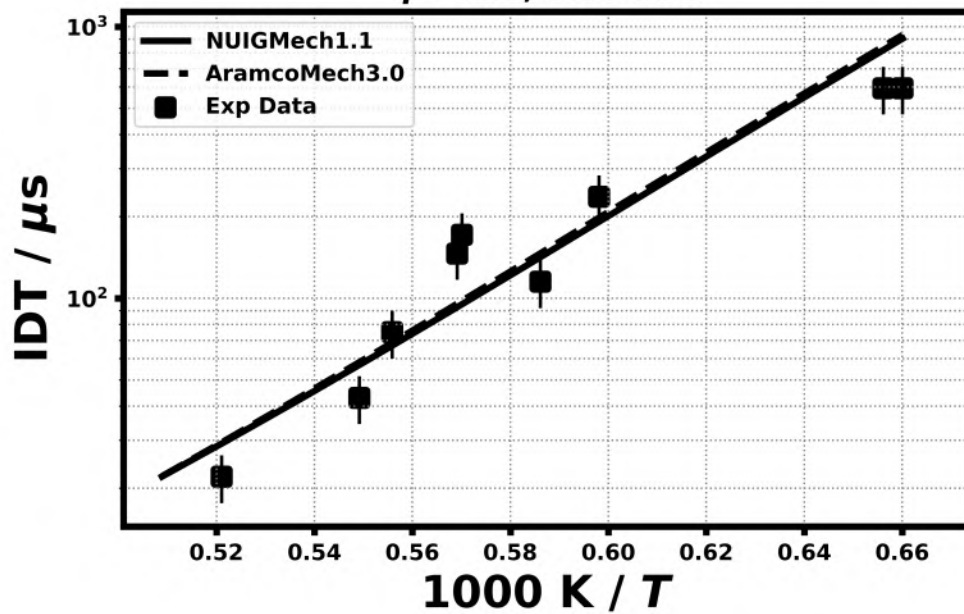
3.8) Spadaccini, L. J., & Colket Iii, M. B., Progress in energy and combustion science, 20 (5) (1994) 431-460.



3.5% CH₄
9.3% O₂, 87.2% Ar
 $\phi = 0.75$, 6.01 atm

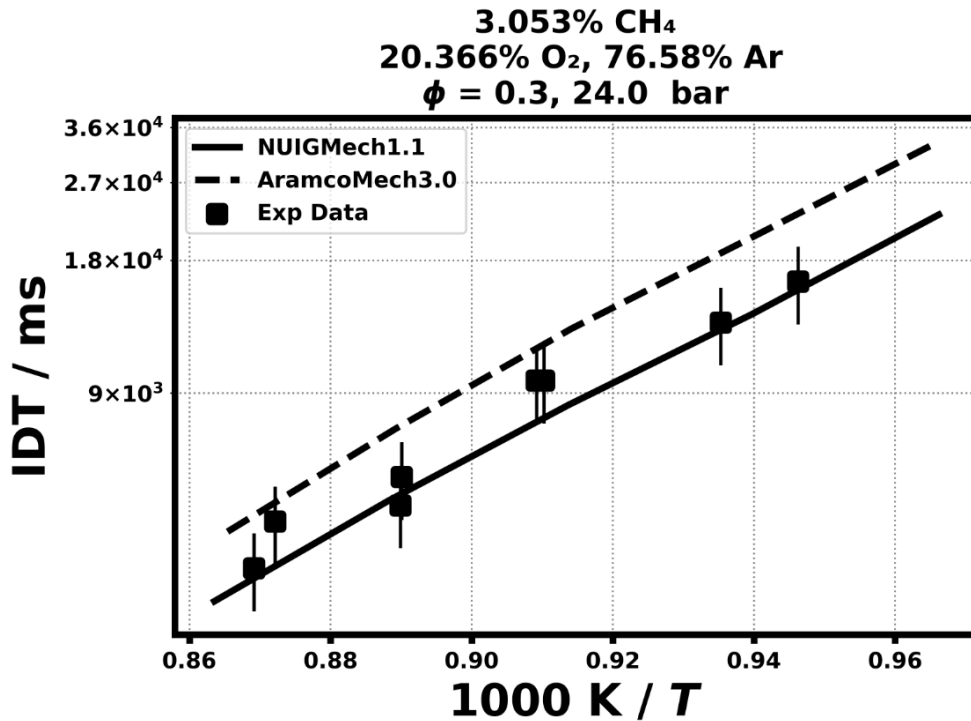
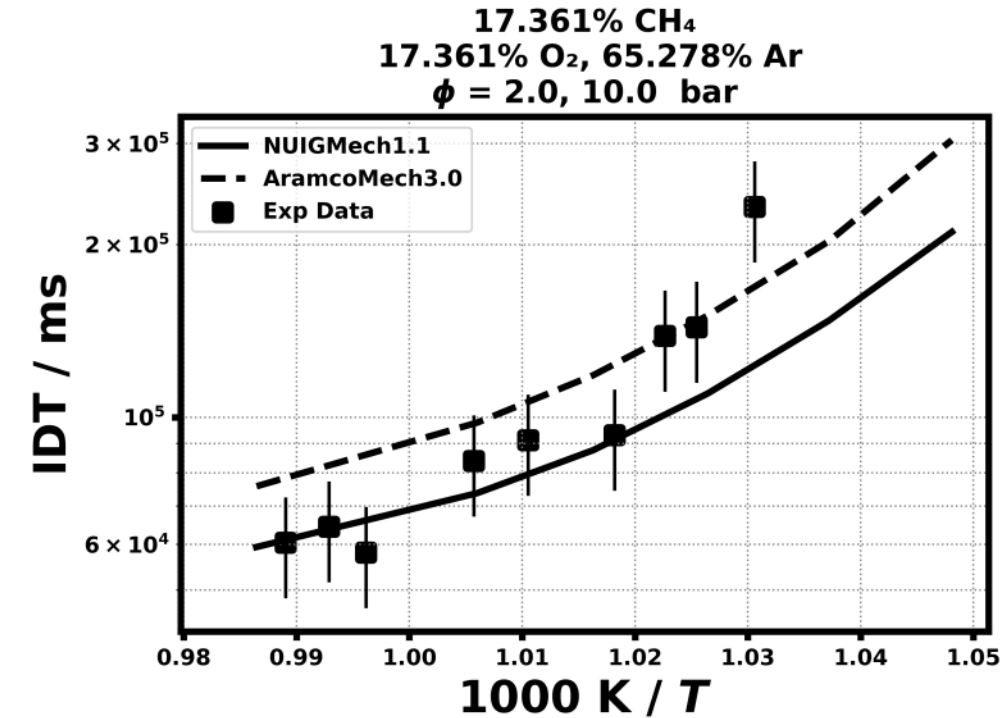


3.5% CH₄
7.0% O₂, 89.5% Ar
 $\phi = 1.0$, 6.53 atm

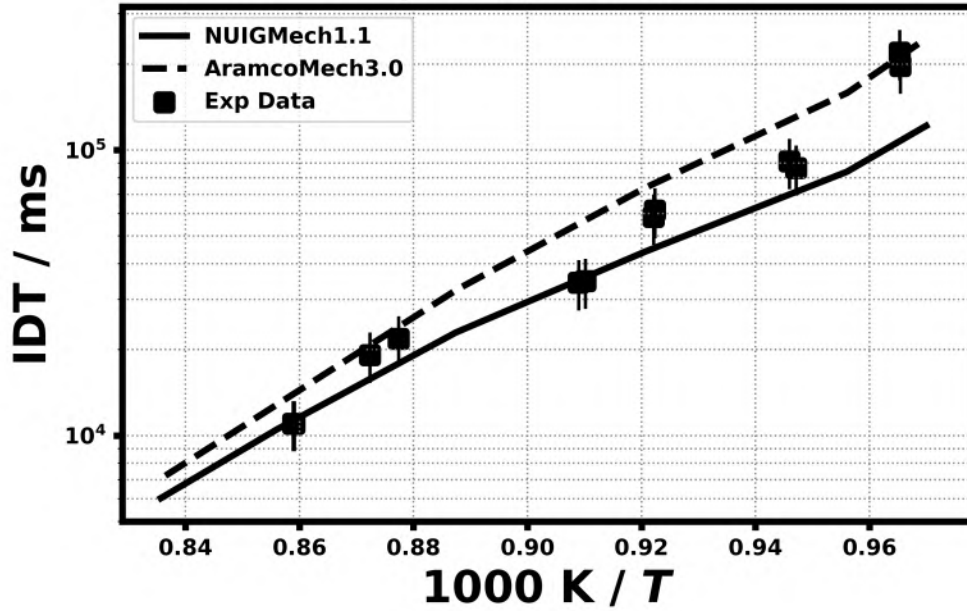


RCM Ignition delay time

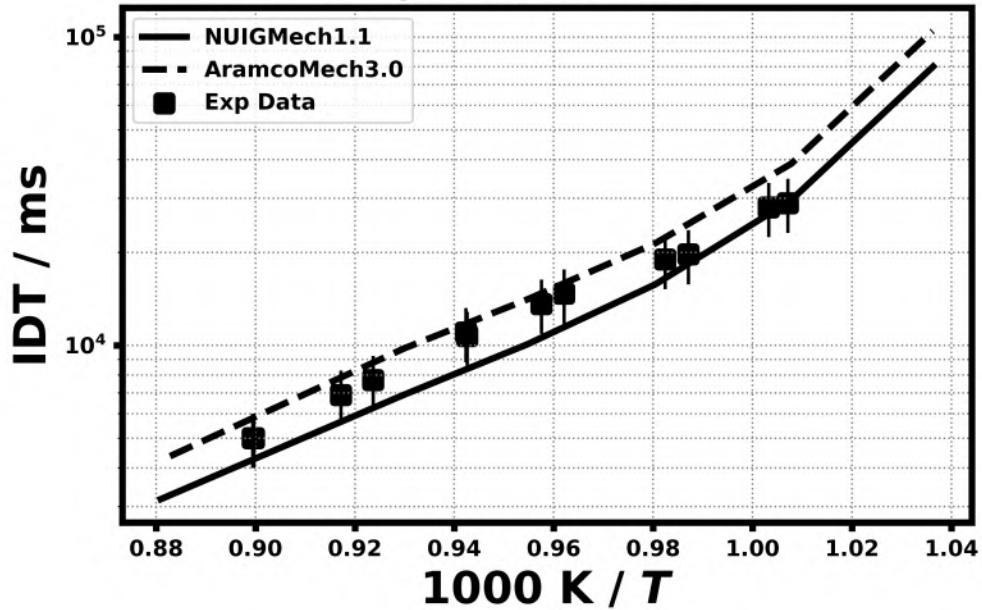
3.10) Burke, U., Somers, K. P., O'Toole, P., Zinner, C. M., Marquet, N., Bourque, G., & Curran, H. J., *Combustion and flame*, 162(2) (2015) 315-330.



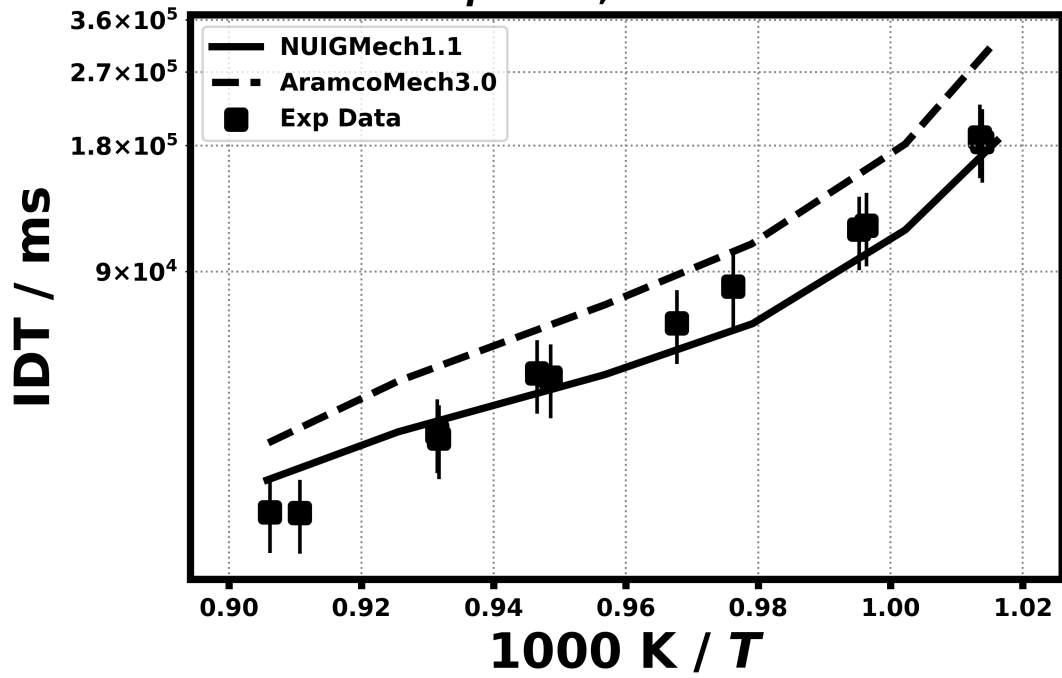
4.988% CH₄
19.95% O₂, 75.062% Ar
 $\phi = 0.5, 10.0 \text{ bar}$



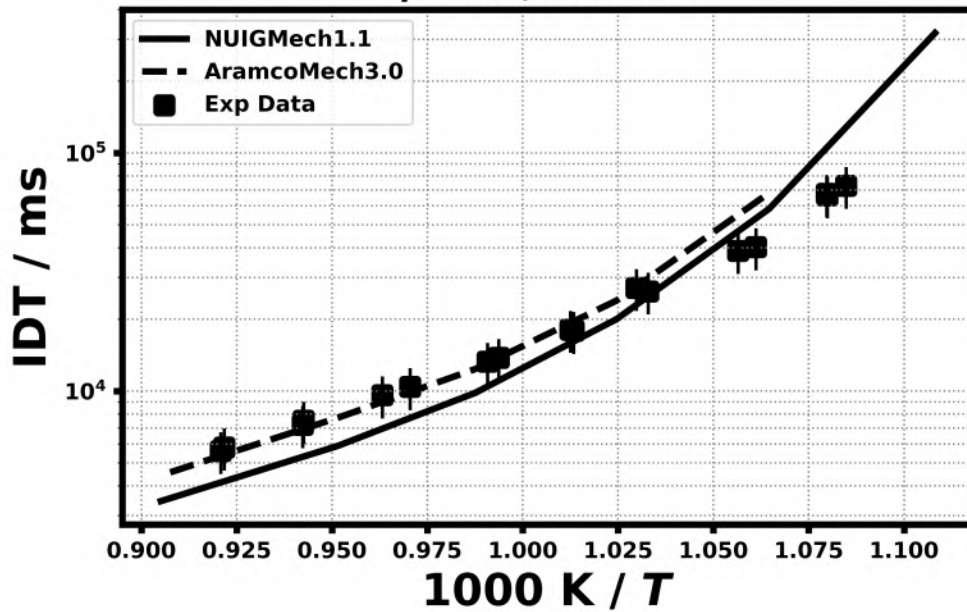
4.988% CH₄
19.95% O₂, 75.062% Ar
 $\phi = 0.5, 24.0 \text{ bar}$



9.506% CH₄
19.011% O₂, 71.483% Ar
 $\phi = 1.0, 10.0$ bar

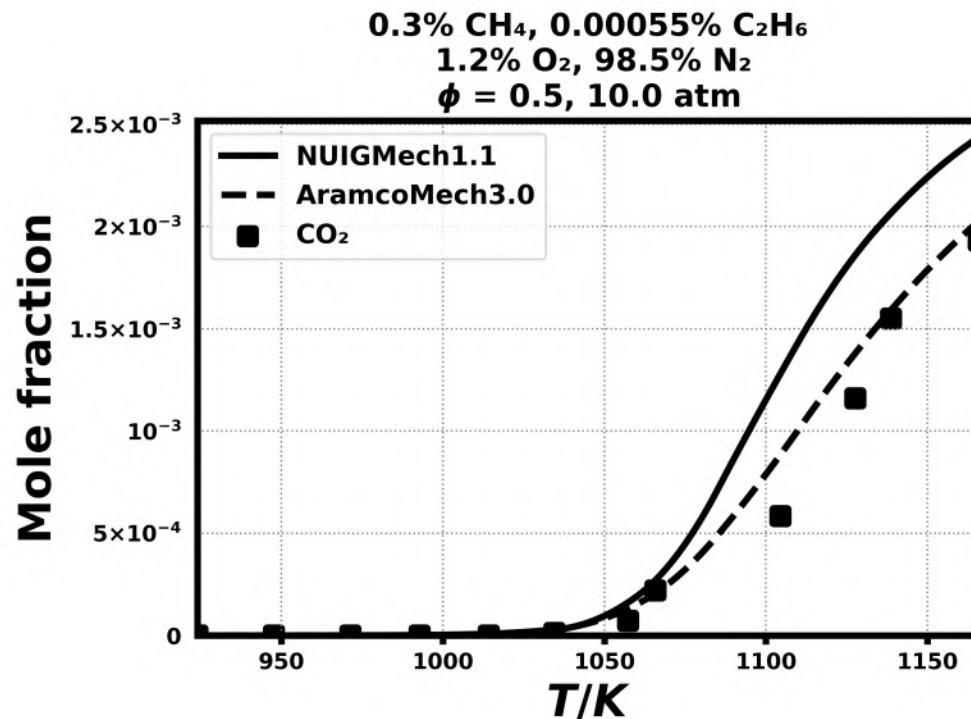
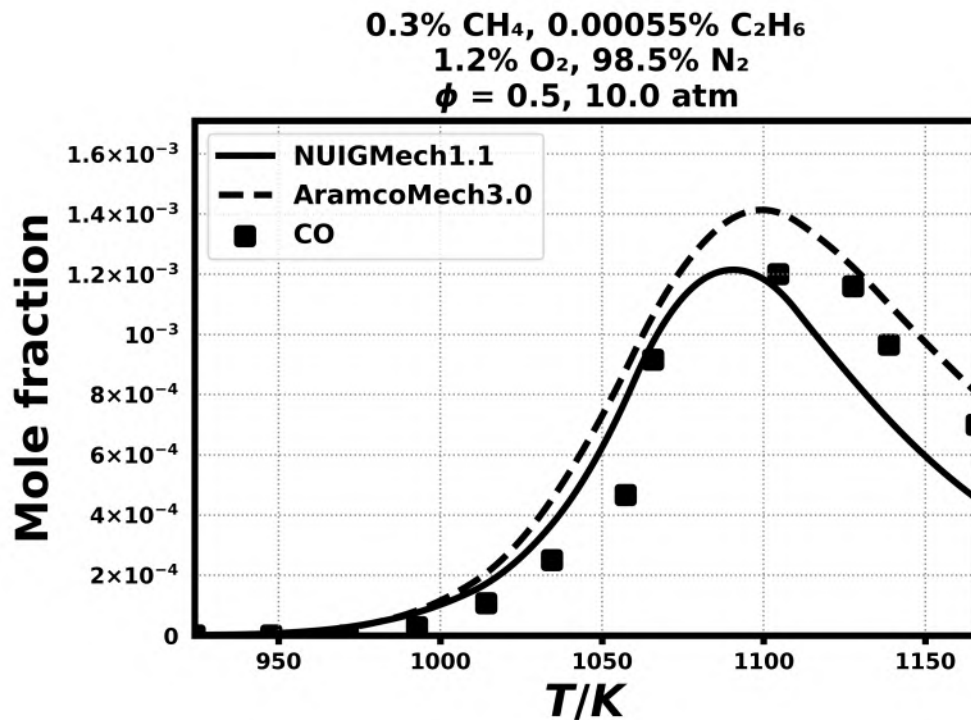


9.506% CH₄
19.011% O₂, 71.483% Ar
 $\phi = 1.0, 24.0$ bar

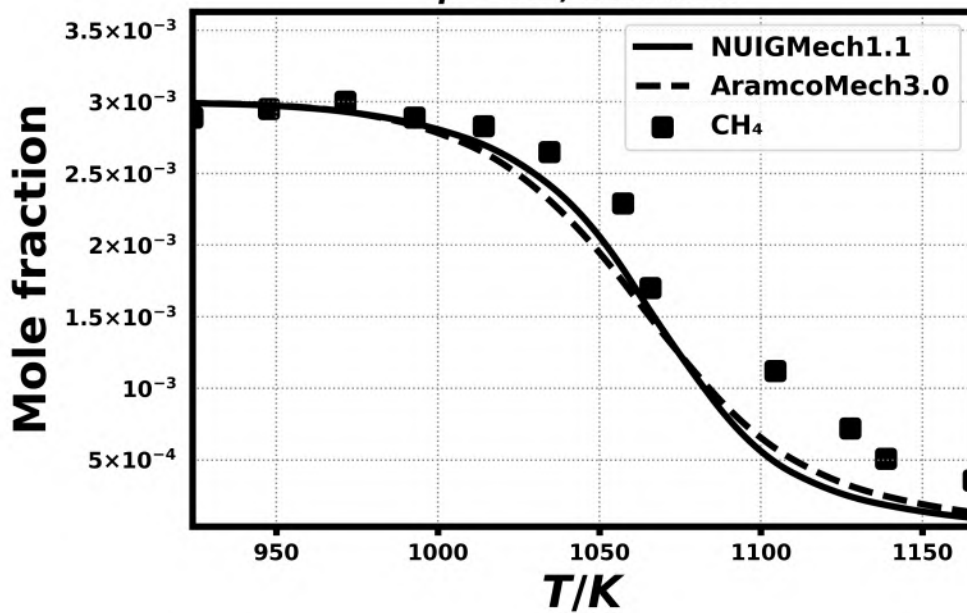


Speciation in Jet-stirred reactor

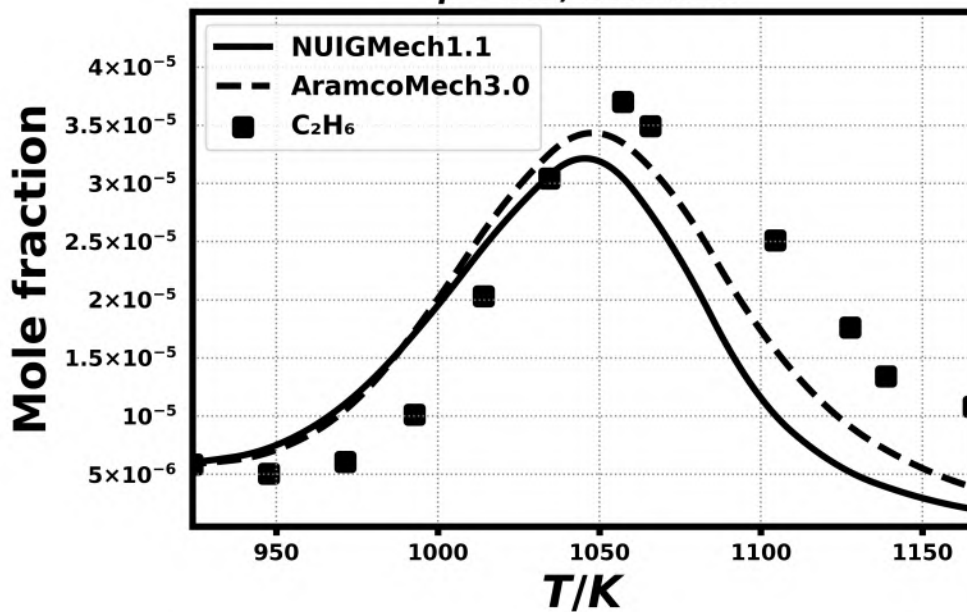
3.11) Dagaut, P., Boettner, J. C., & Cathonnet, M., Combustion science and technology, 77(1-3) (1991) 127-148.



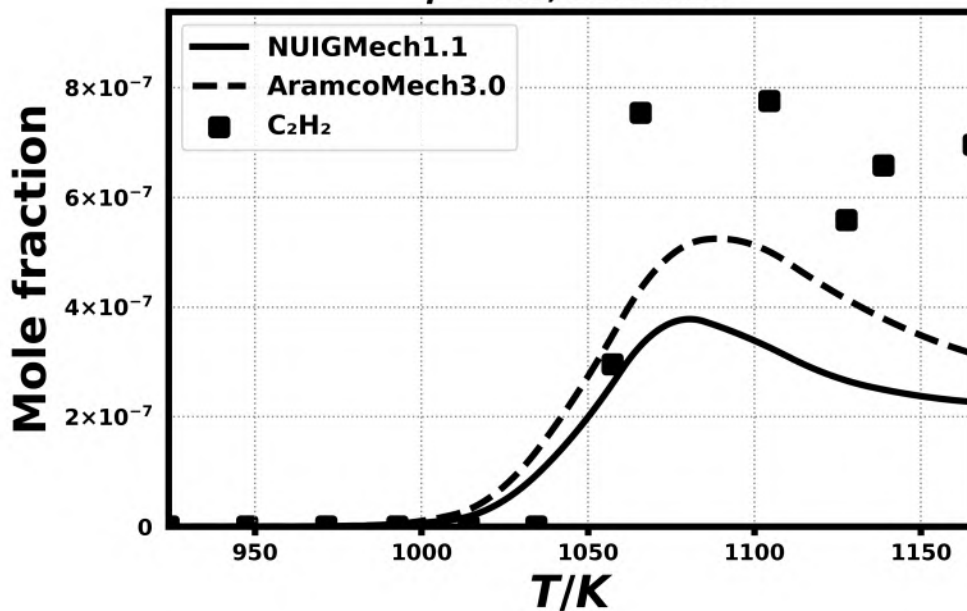
0.3% CH₄, 0.00055% C₂H₆
1.2% O₂, 98.5% N₂
 $\phi = 0.5, 10.0 \text{ atm}$



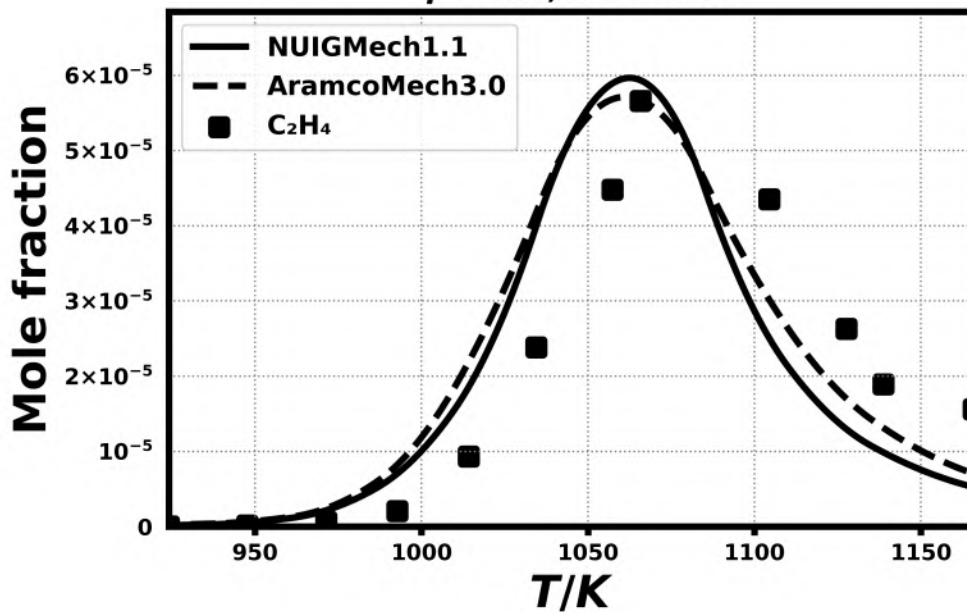
0.3% CH₄, 0.00055% C₂H₆
1.2% O₂, 98.5% N₂
 $\phi = 0.5, 10.0 \text{ atm}$



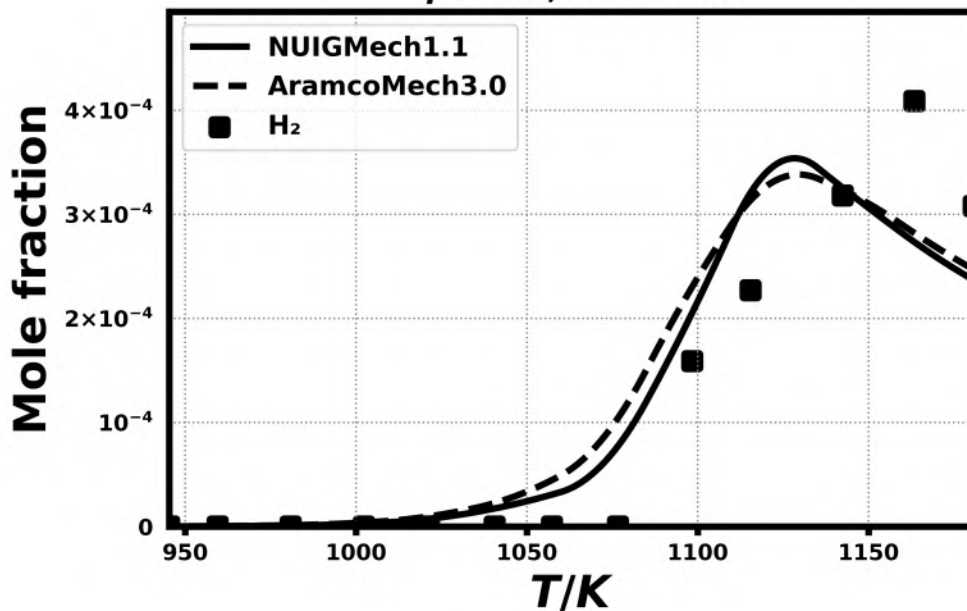
0.3% CH₄, 0.00055% C₂H₆
1.2% O₂, 98.5% N₂
 $\phi = 0.5, 10.0 \text{ atm}$



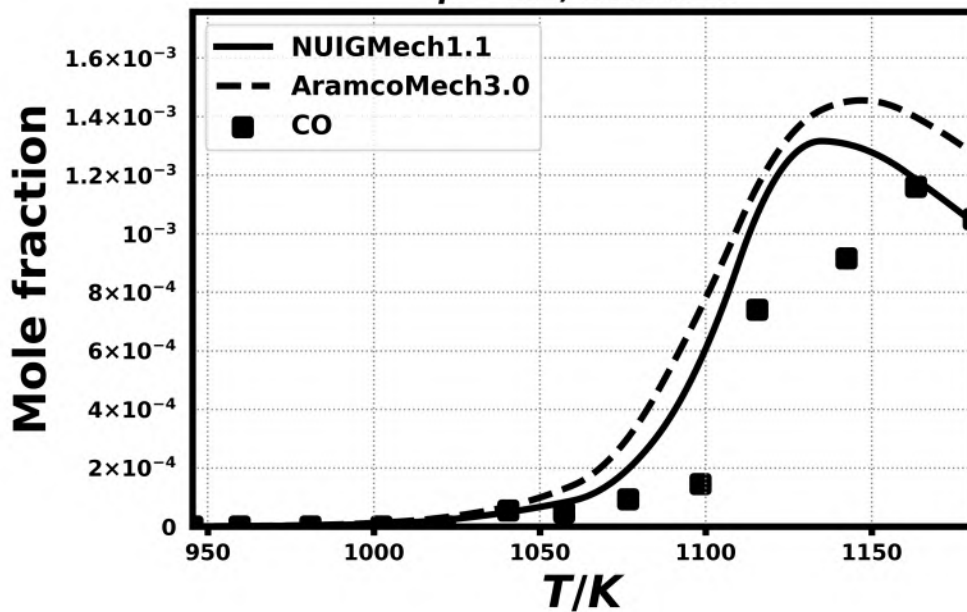
0.3% CH₄, 0.00055% C₂H₆
1.2% O₂, 98.5% N₂
 $\phi = 0.5, 10.0 \text{ atm}$



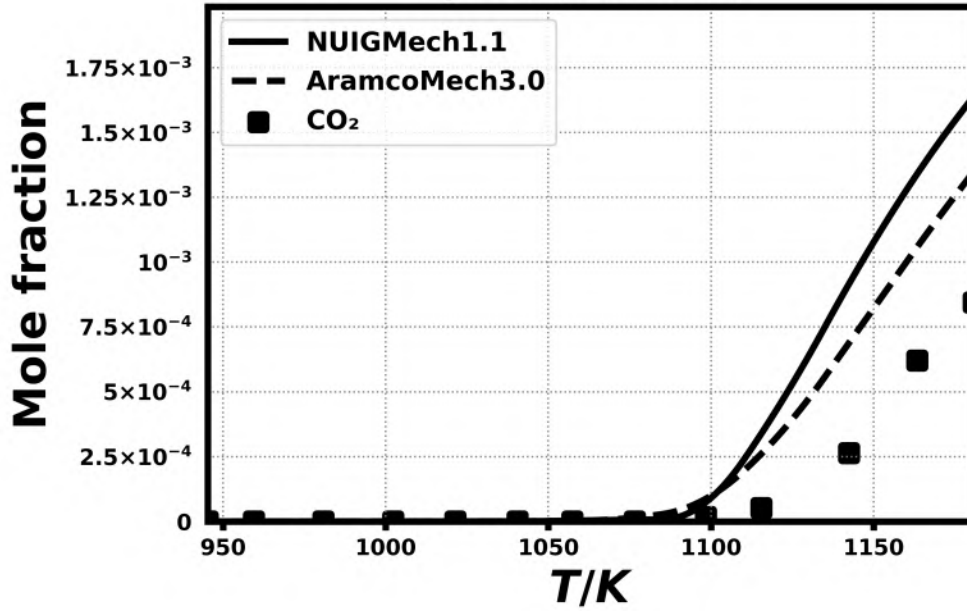
0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0$, 10.0 atm



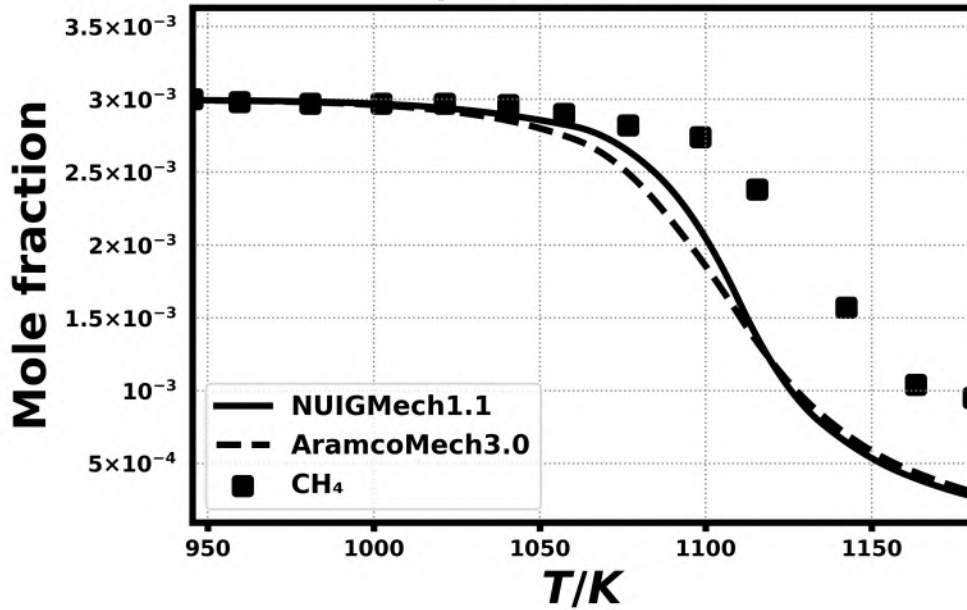
0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0$, 10.0 atm



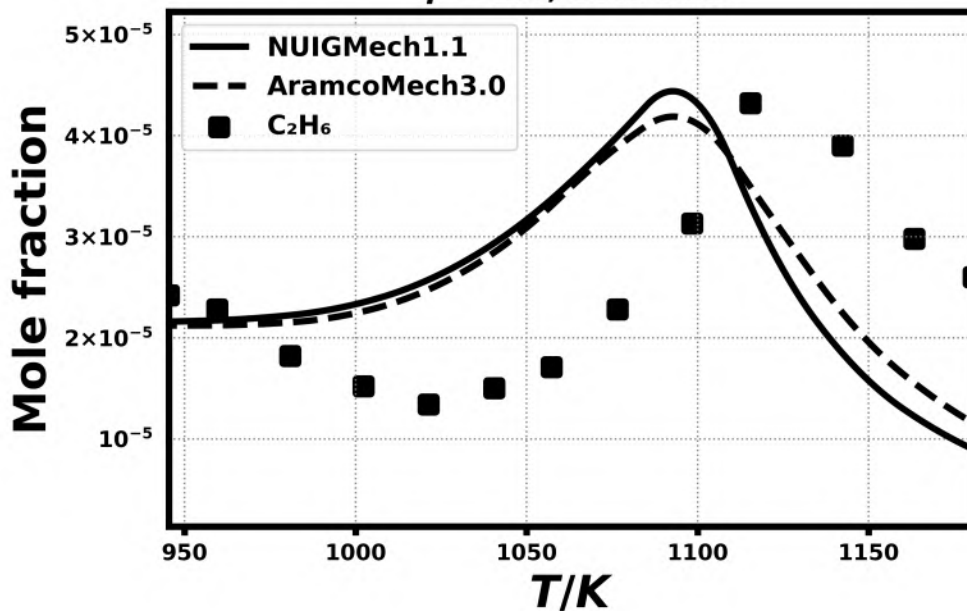
0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0, 10.0 \text{ atm}$



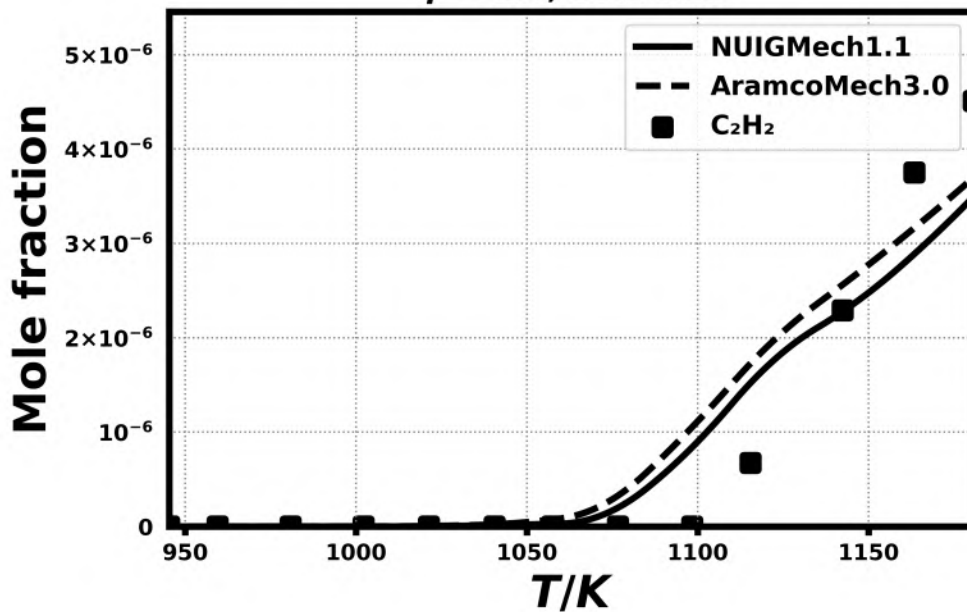
0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0, 10.0 \text{ atm}$

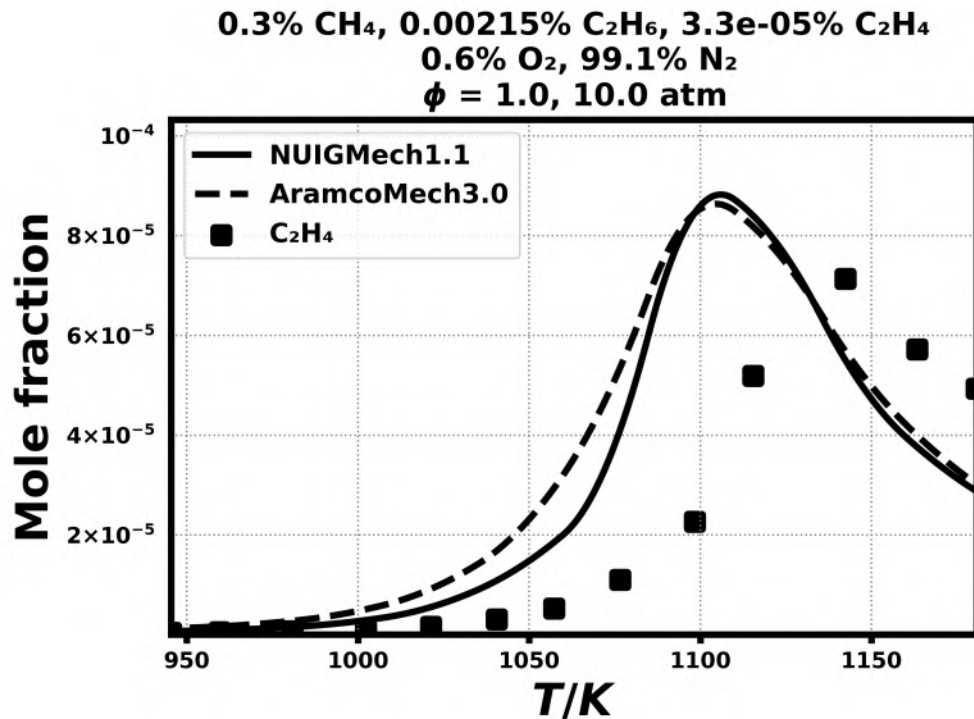


0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0, 10.0 \text{ atm}$



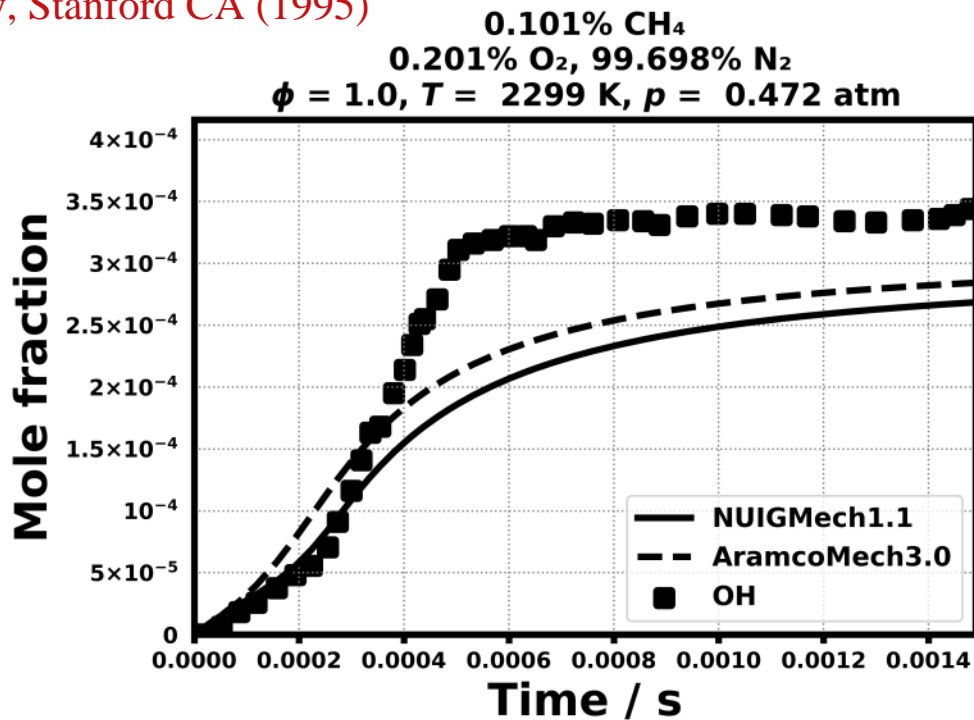
0.3% CH₄, 0.00215% C₂H₆, 3.3e-05% C₂H₄
0.6% O₂, 99.1% N₂
 $\phi = 1.0, 10.0 \text{ atm}$

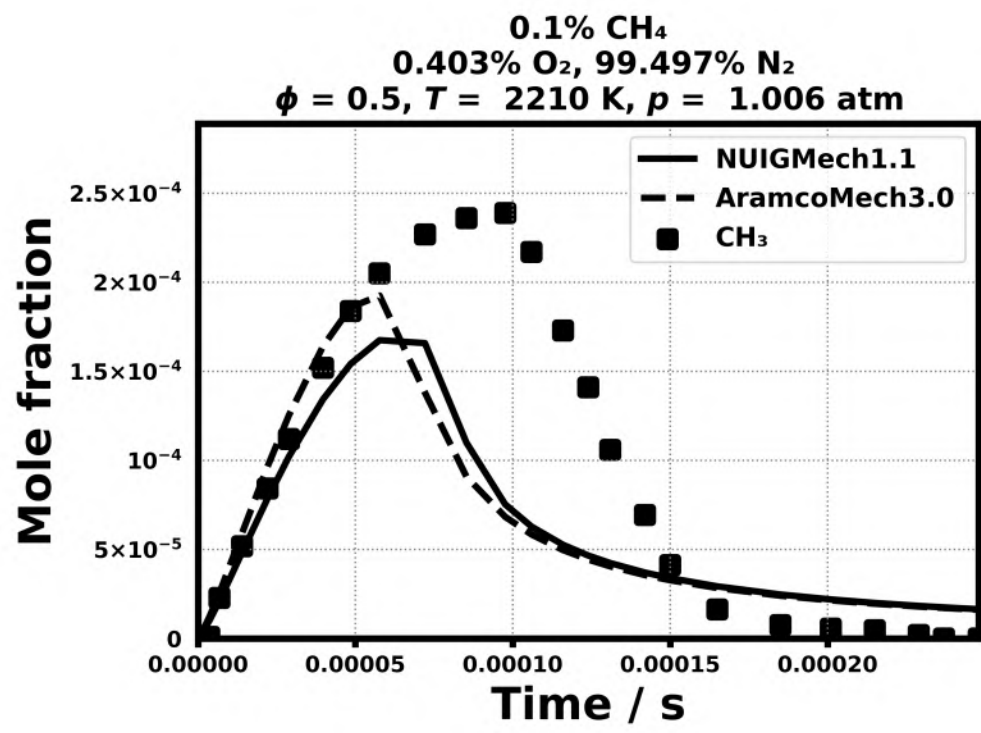
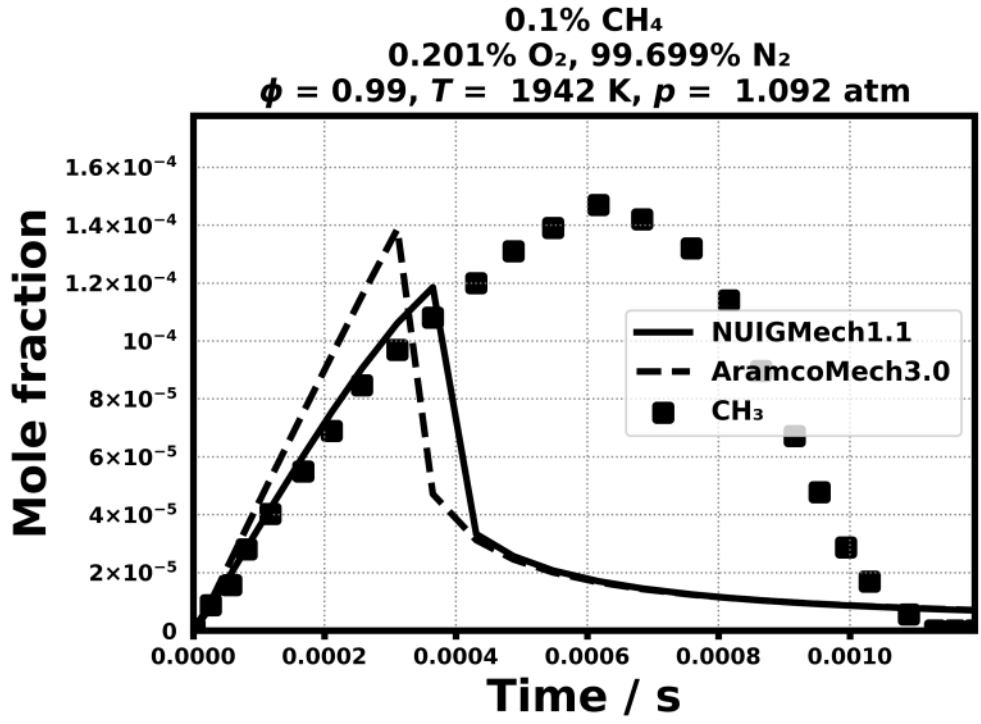




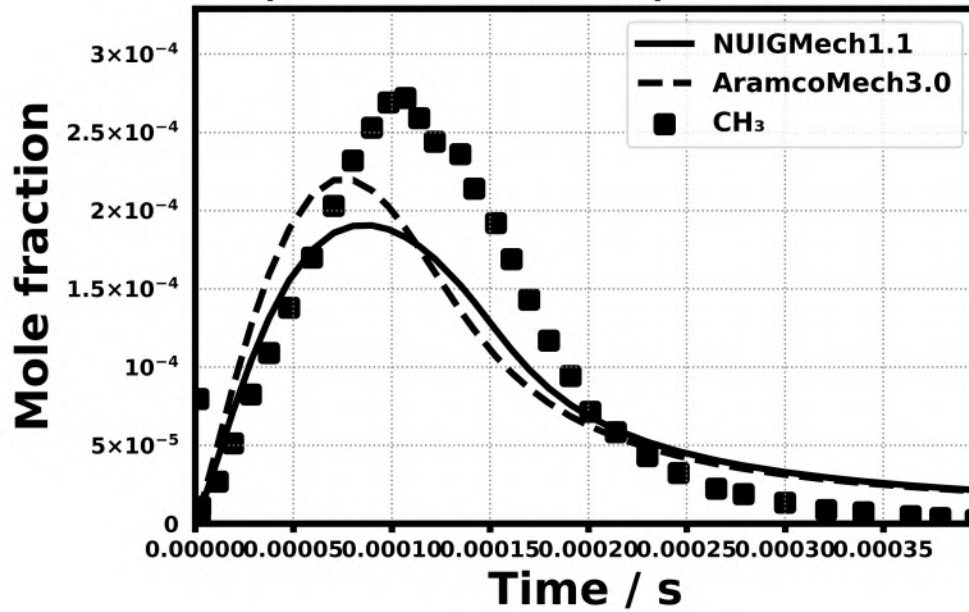
Speciation in Flow reactor

3.12) E.J.Chang, M. ENG. Thesis, Mechanical Engineering Department, Stanford University, Stanford CA (1995)

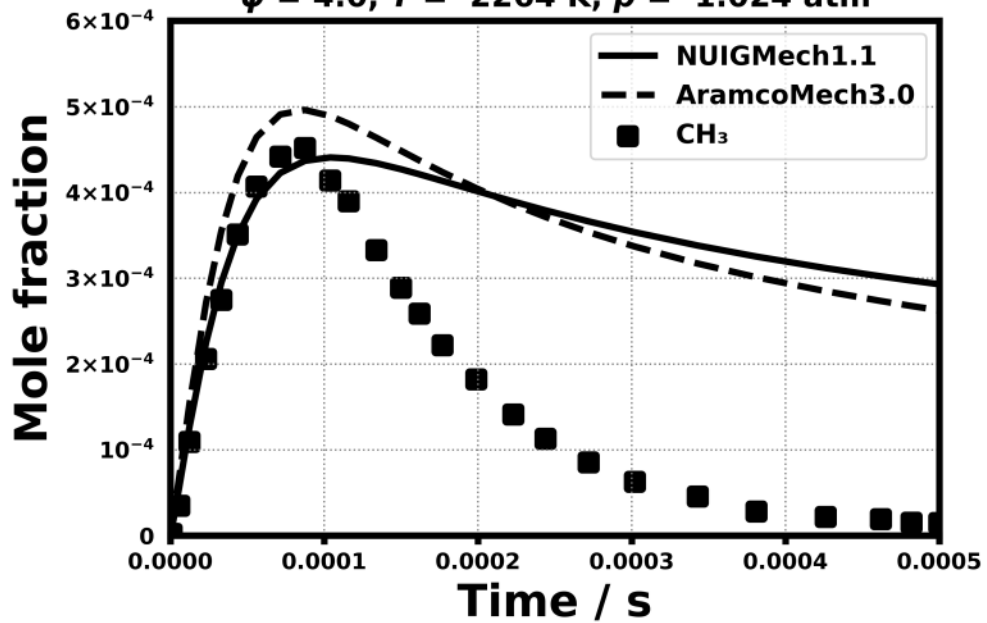




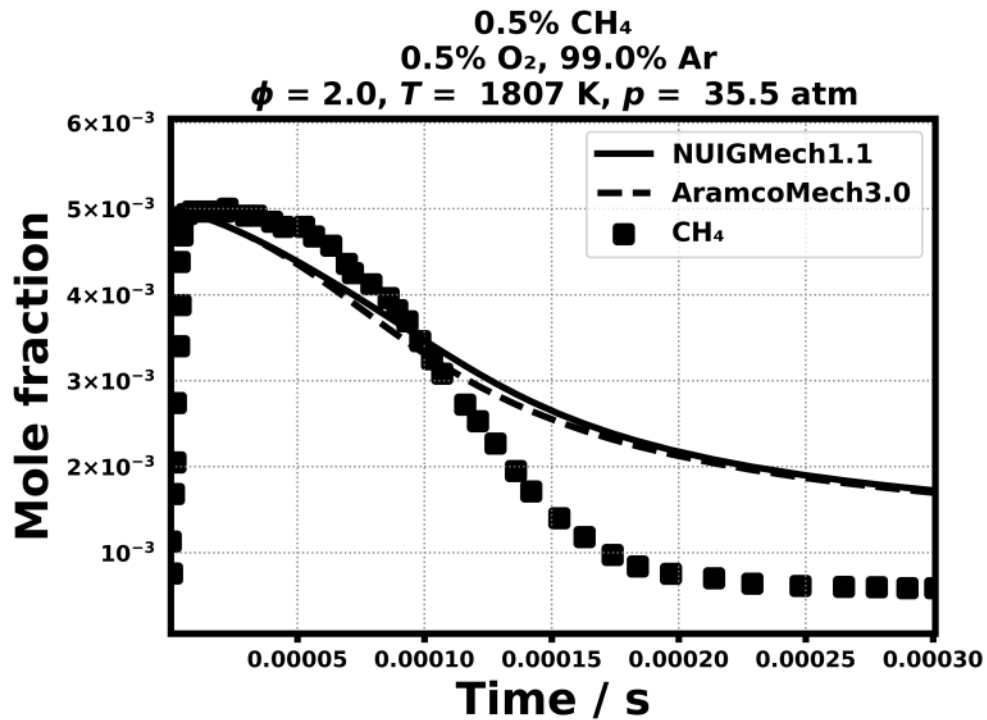
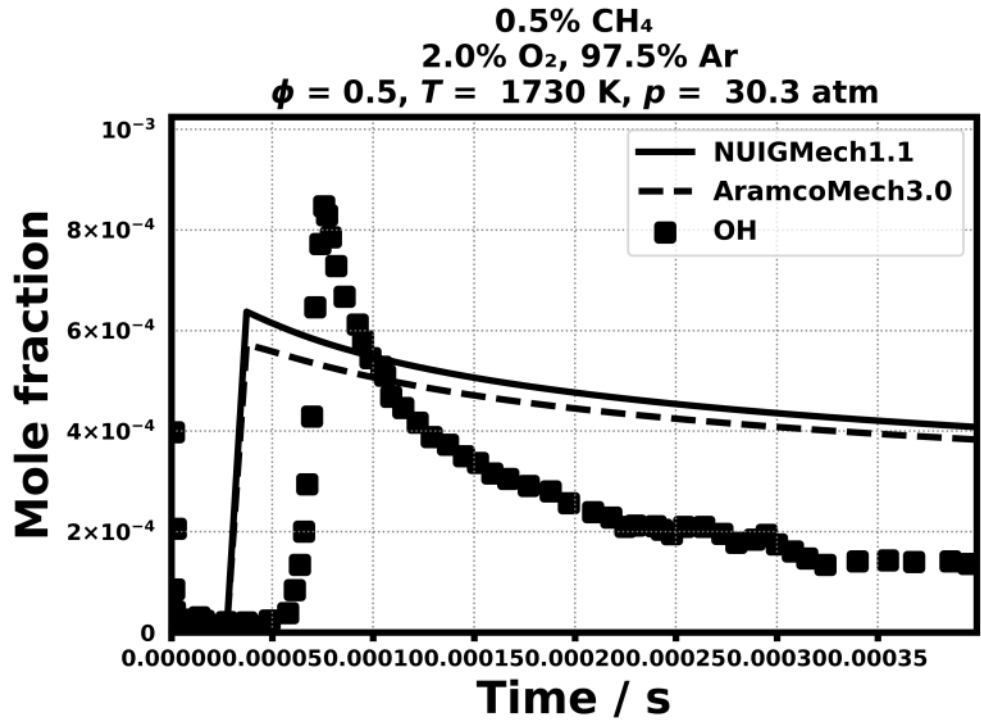
0.099% CH₄
0.202% O₂, 99.698% N₂
 $\phi = 0.98, T = 2224 \text{ K}, p = 1.039 \text{ atm}$



0.201% CH₄
0.1% O₂, 99.699% N₂
 $\phi = 4.0, T = 2264 \text{ K}, p = 1.024 \text{ atm}$



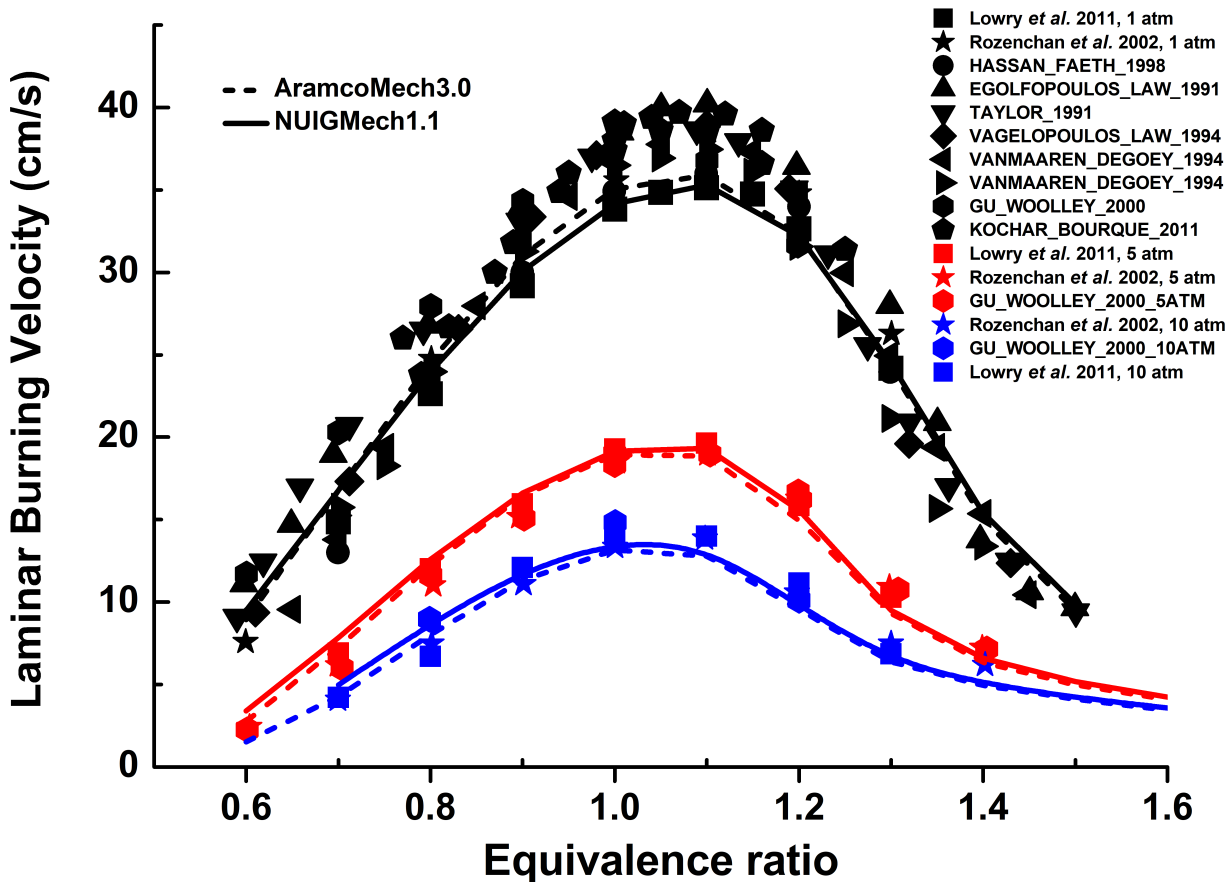
3.13) Petersen, E. L., Röhrig, M., Davidson, D. F., Hanson, R. K., & Bowman, C. T., In Symposium (International) on Combustion, 26 (1996, January) 799-806



Laminar flame speed

- 3.14) Gu, X. J., Haq, M. Z., Lawes, M., & Woolley, R., *Combustion and flame*, 121(1-2) (2000) 41-58.
- 3.15) Rozenchan, G., Zhu, D. L., Law, C. K., & Tse, S. D., *Proceedings of the Combustion Institute*, 29(2) (2002) 1461-1470.
- 3.16) Hassan, M. I., Aung, K. T., & Faeth, G. M., *Combustion and flame*, 115(4) (1998) 539-550..
- 3.17) Zhu, D. L., Egolfopoulos, F. N., & Law, C. K., In *Symposium (International) on Combustion*, 22, (1989, January) 1537-1545
- 3.18) Lowry, W., de Vries, J., Krejci, M., Petersen, E., Serinyel, Z., Metcalfe, W., Bourque, G., *Jornal of Engineering for Gas Turbines and Power*, 133(9) (2011).
- 3.19) Egolfopoulos, F. N., Zhu, D. L., & Law, C. K., In *Symposium (International) on Combustion*, 23 (1991, January) 471-478.
- 3.20) S. Taylor, PhD diss., University of Leeds, 1991.
- 3.21) Vagelopoulos, C. M., Egolfopoulos, F. N., & Law, C. K., In *Symposium (international) on combustion*, 25 (1994, January) 1341-1347
- 3.22) Van Maaren, A., Thung, D. S., & DE GOEY, L. R. H., *Combustion Science and Technology*, 96(4-6) (1994) 327-344.
- 3.23) U. Kochar, *Turbo Expo: Power for Land, Sea, and Air* (Vol. 54624 (2011) 129-140.

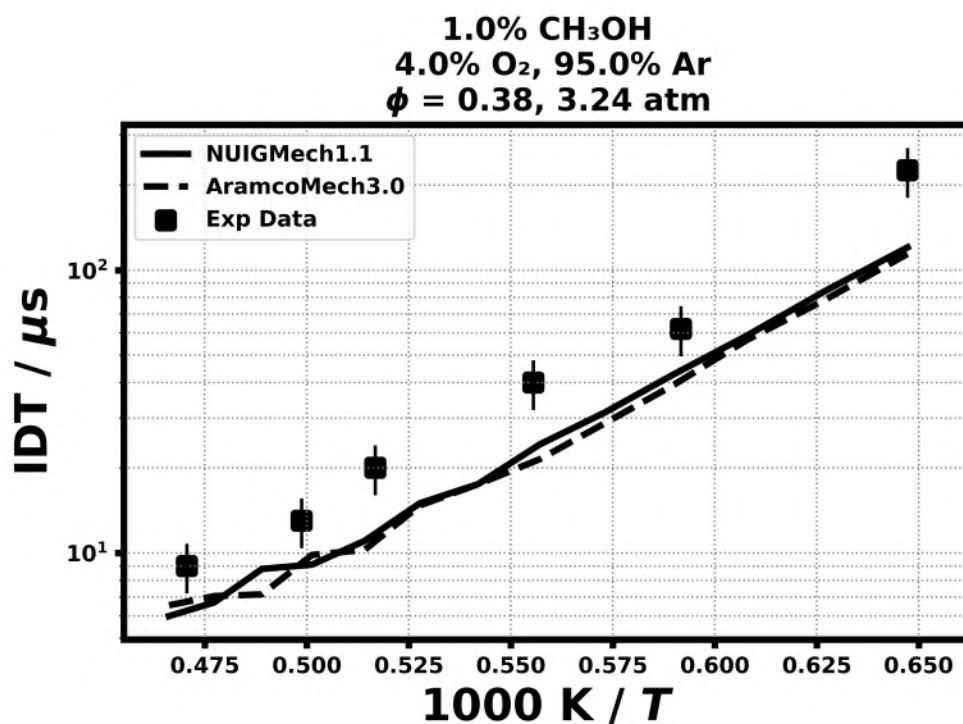
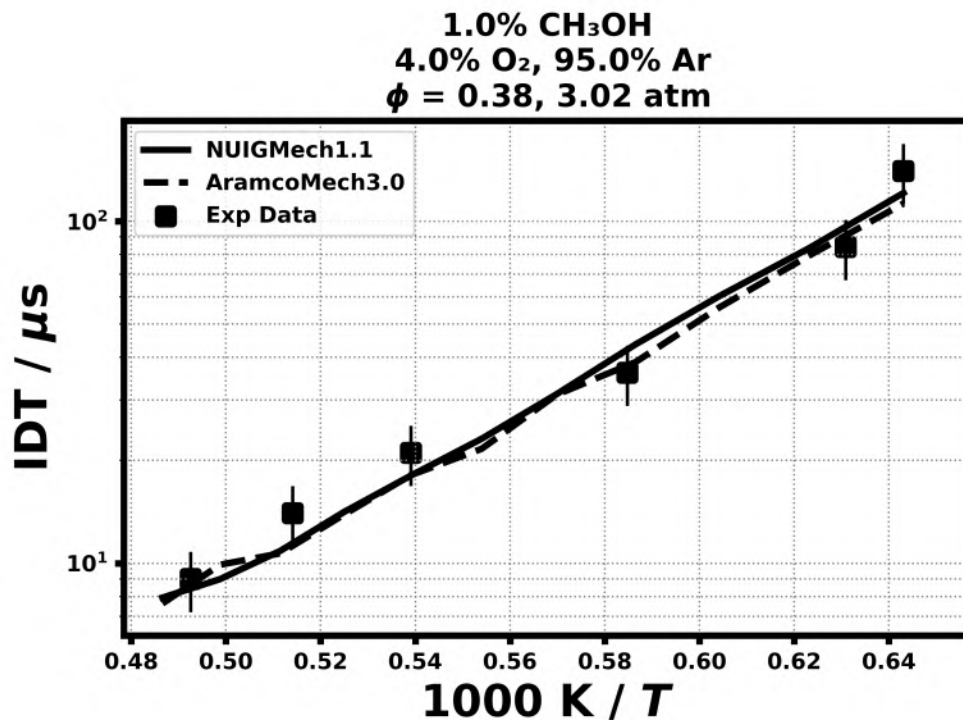
CH₄ in 'air', T_u = 298 K, p = 1 / 5 / 10 atm



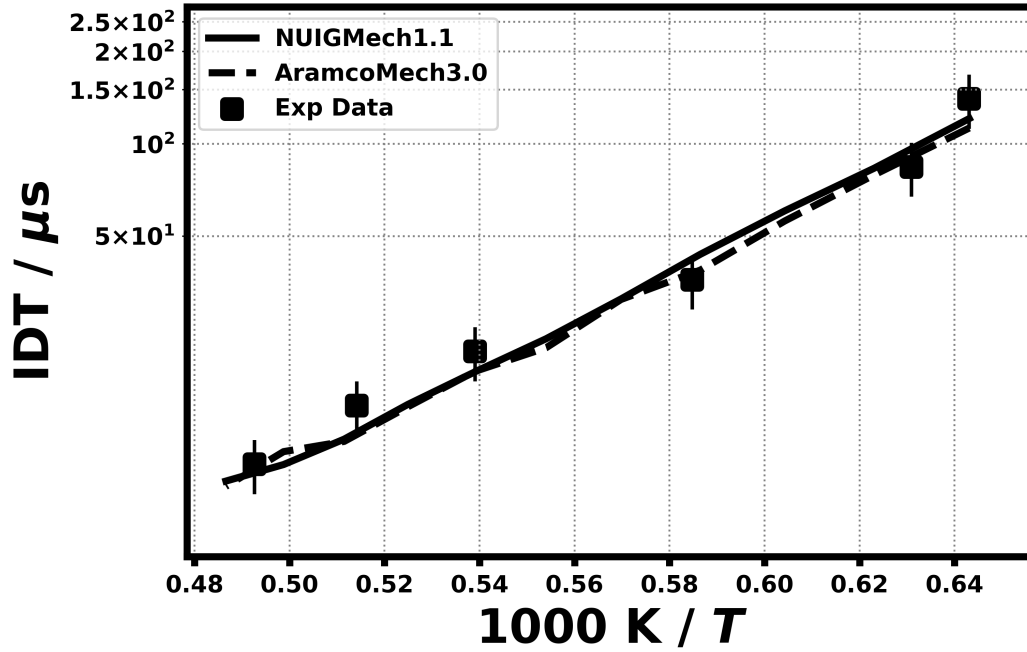
4. Validation for CH₃OH

Shock tube ignition delay time

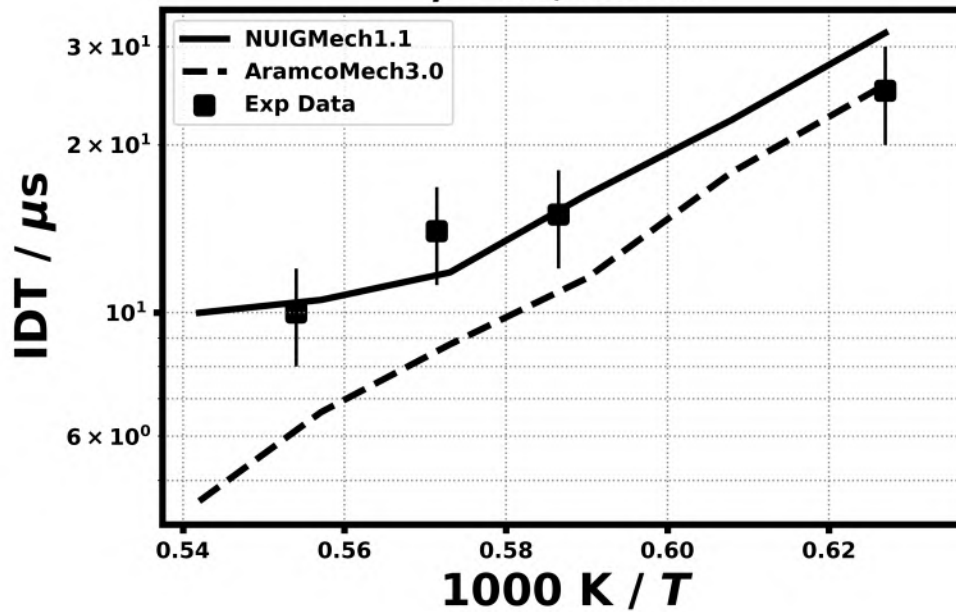
4.1) Bowman, C. T., *Combustion and Flame* 25 (1975) 343-354.



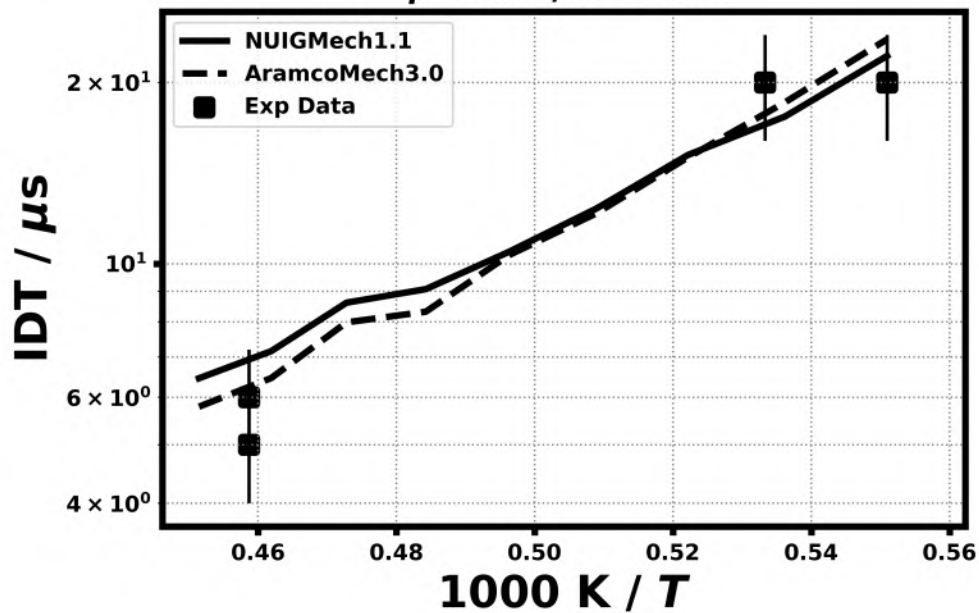
1.0% CH₃OH
4.0% O₂, 95.0% Ar
 $\phi = 0.38, p = 3.02 \text{ atm}$



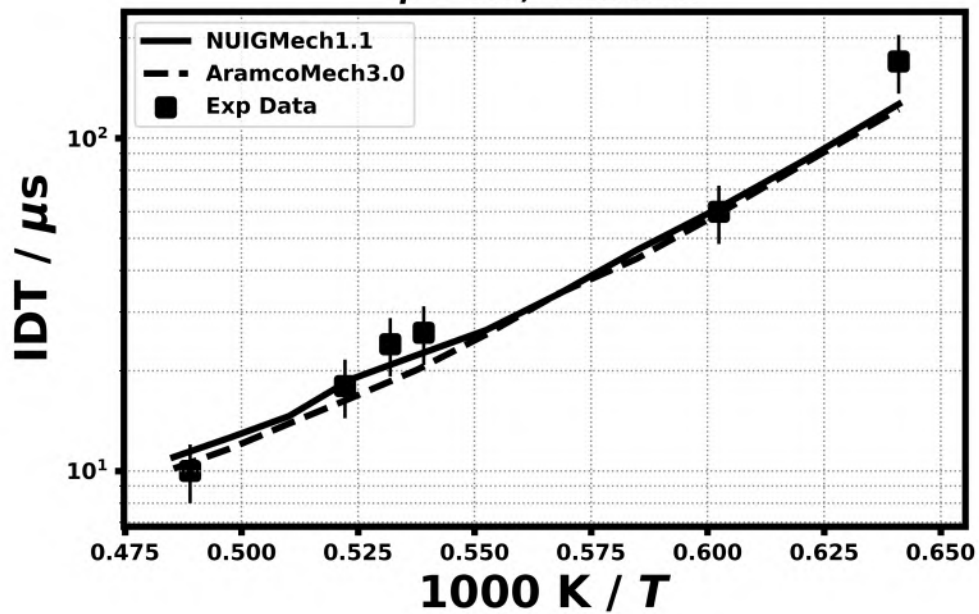
4.0% CH₃OH
1.0% O₂, 95.0% Ar
 $\phi = 6.0, 3.0 \text{ atm}$



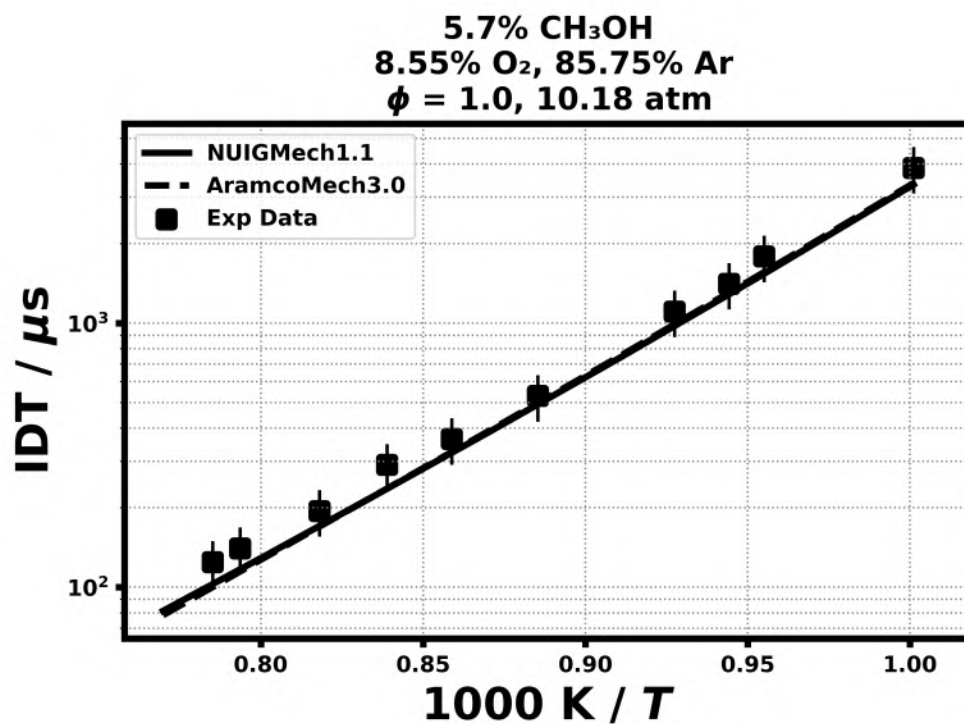
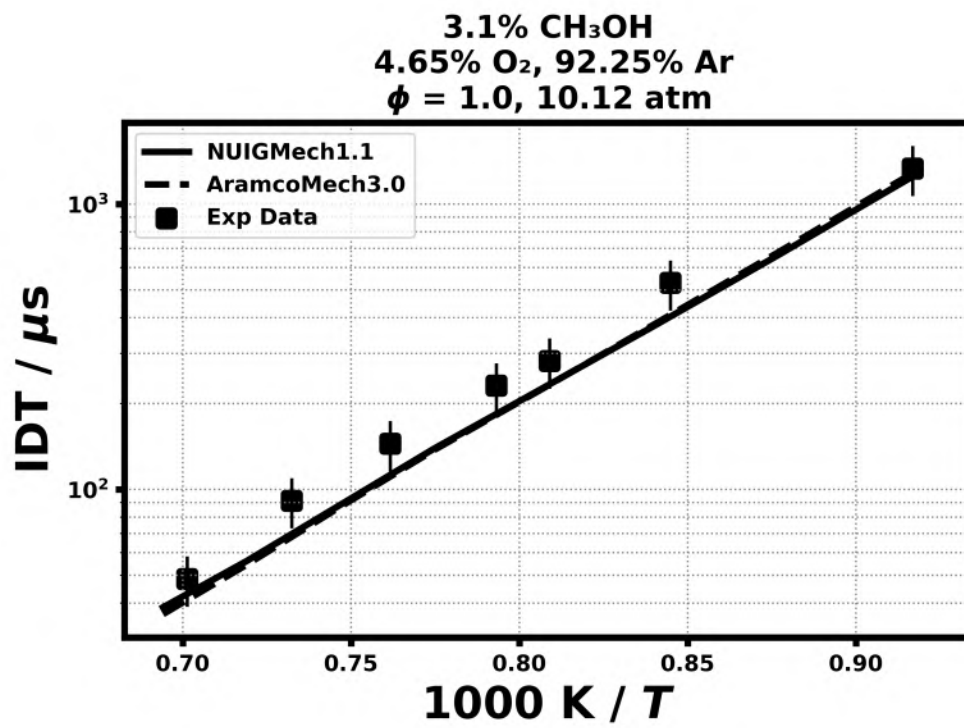
0.75% CH₃OH
1.5% O₂, 97.75% Ar
 $\phi = 0.75$, 4.56 atm



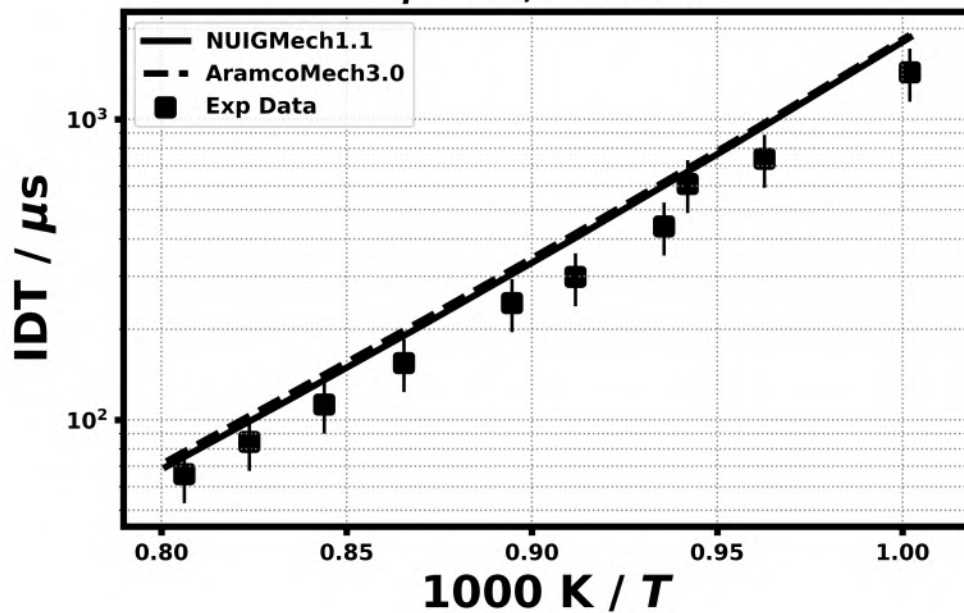
0.75% CH₃OH
1.25% O₂, 98.0% Ar
 $\phi = 0.9$, 4.21 atm



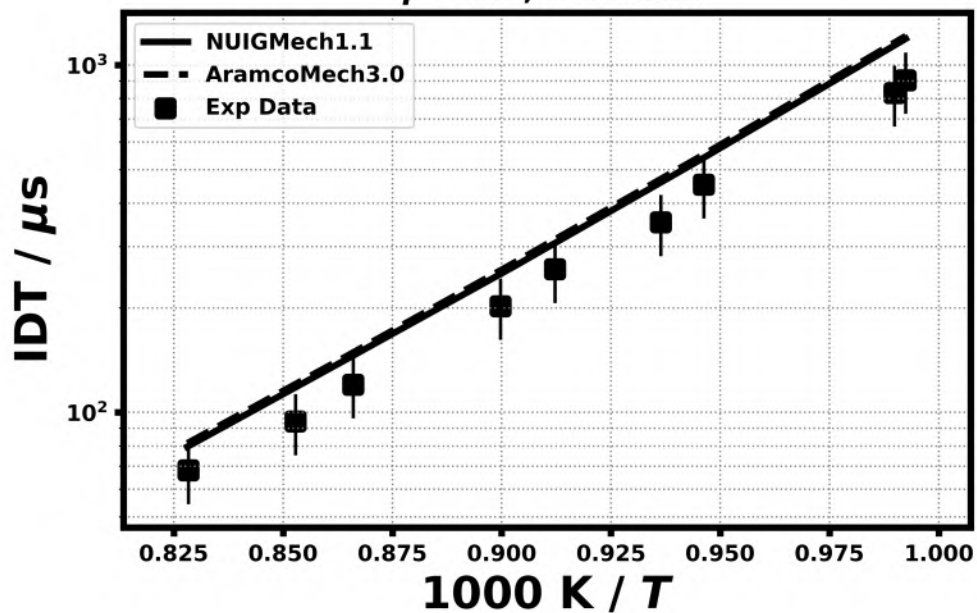
4.2) Burke, U., Metcalfe, W. K., Burke, S. M., Heufer, K. A., Dagaut, P., & Curran, H. J. Combustion and Flame, 165 (2016) 125-136.



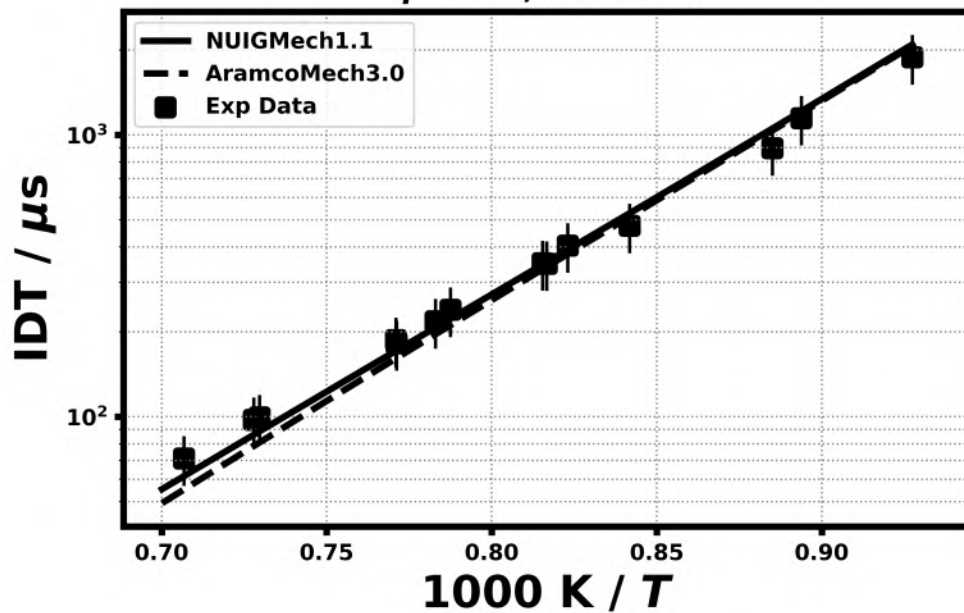
5.0% CH₃OH
15.0% O₂, 80.0% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



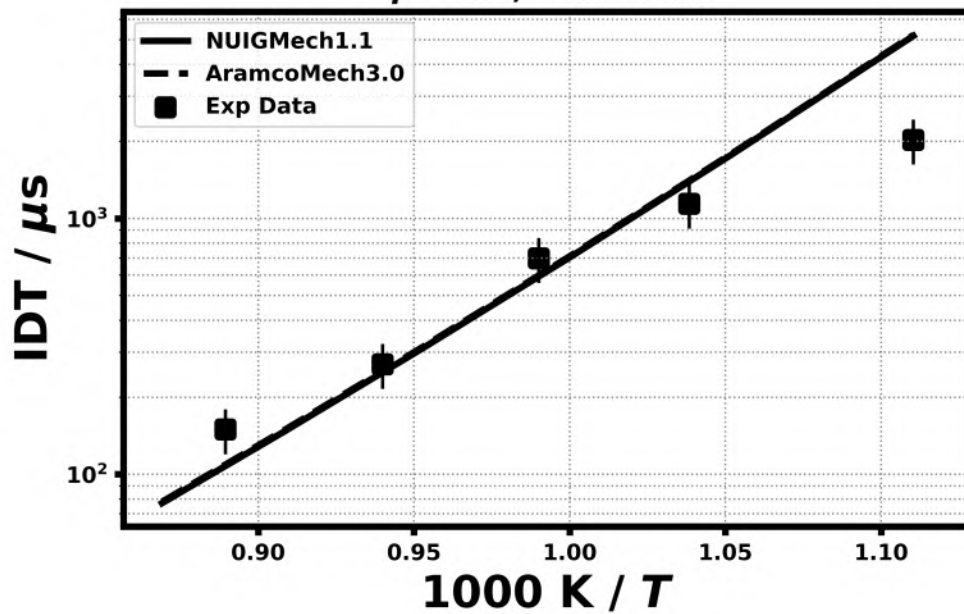
8.0% CH₃OH
12.0% O₂, 80.0% N₂
 $\phi = 1.0, 20.0 \text{ atm}$

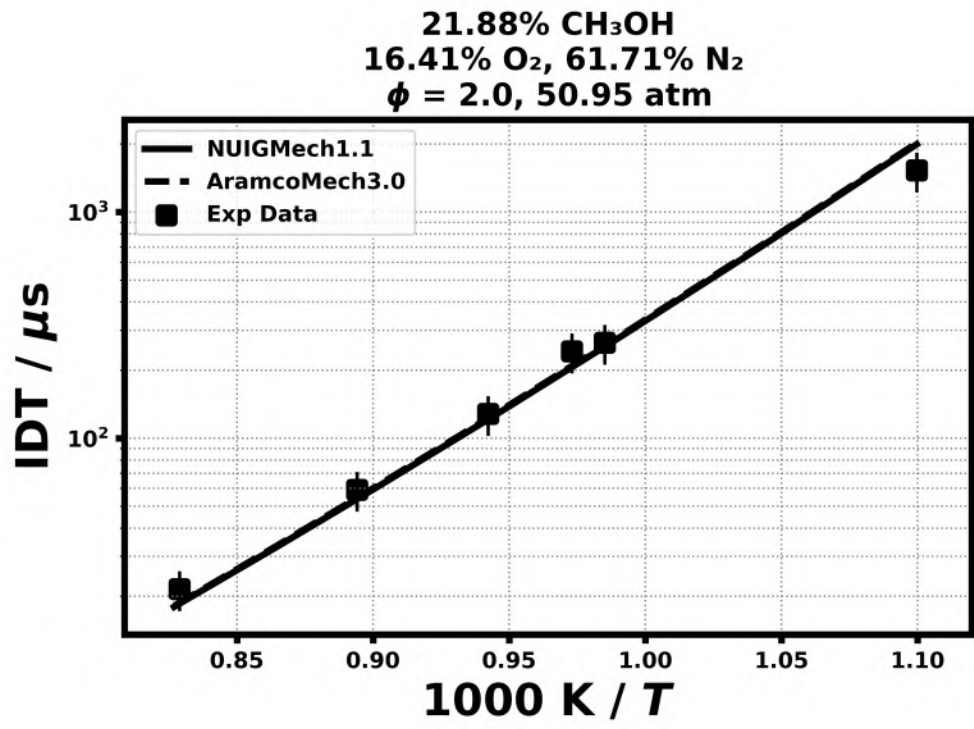
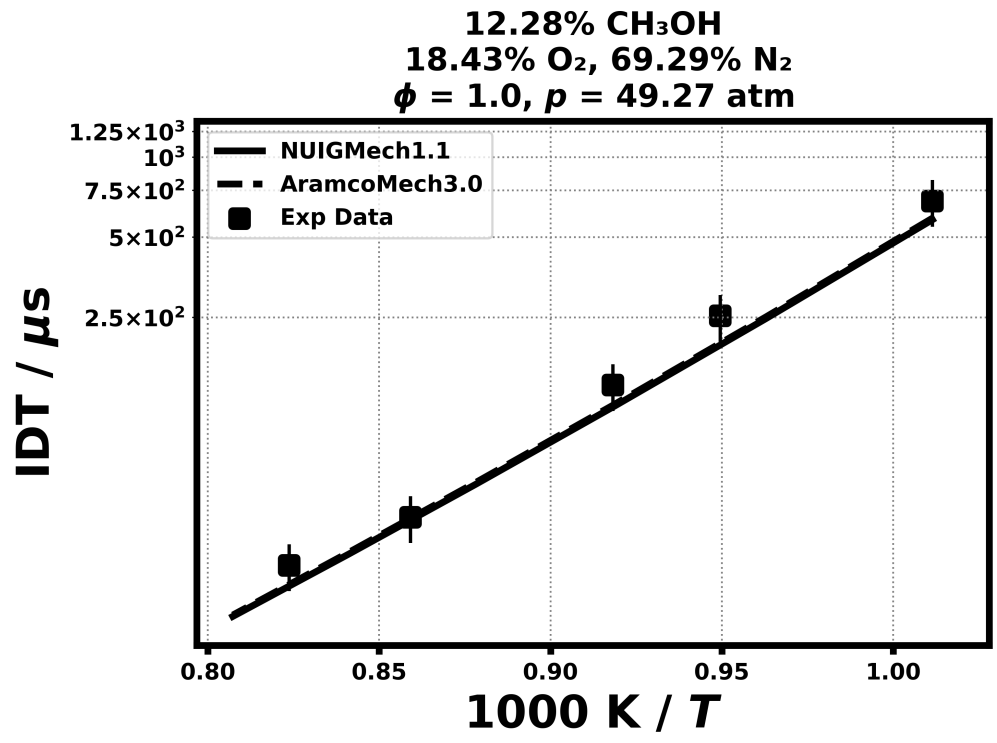


21.88% CH₃OH
16.41% O₂, 61.71% Ar
 $\phi = 2.0, 1.96 \text{ atm}$

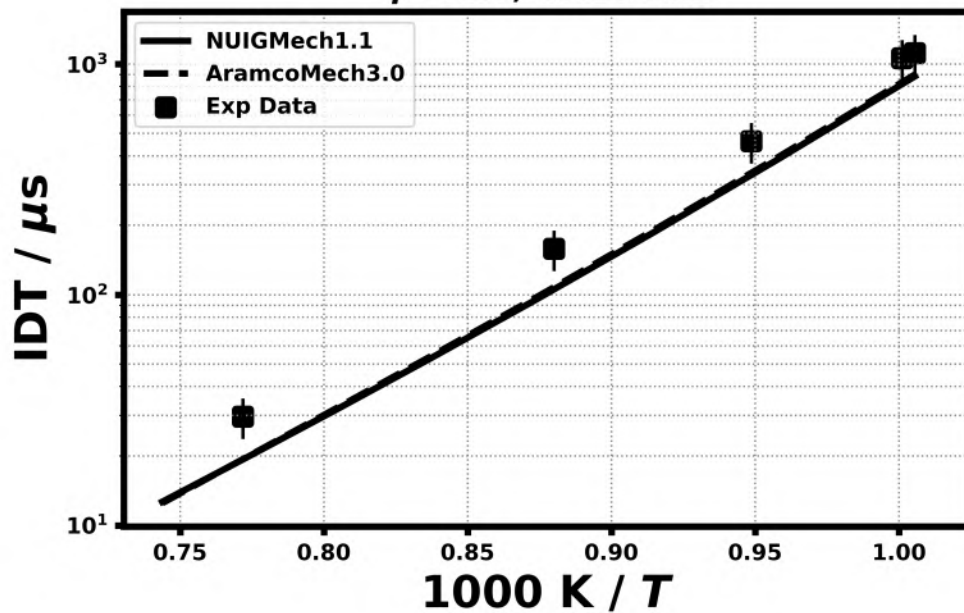


12.28% CH₃OH
18.43% O₂, 69.29% N₂
 $\phi = 1.0, 30.88 \text{ atm}$

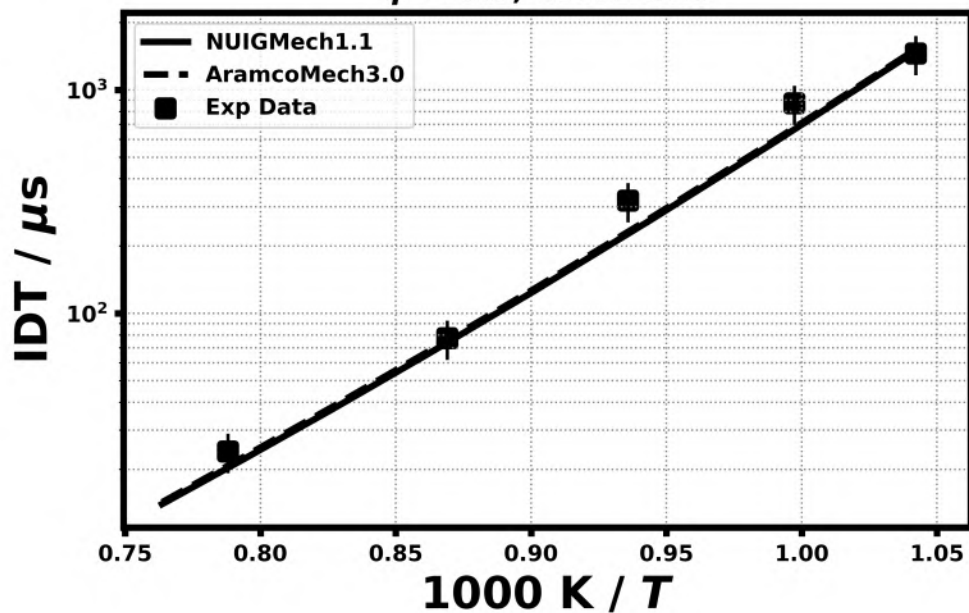




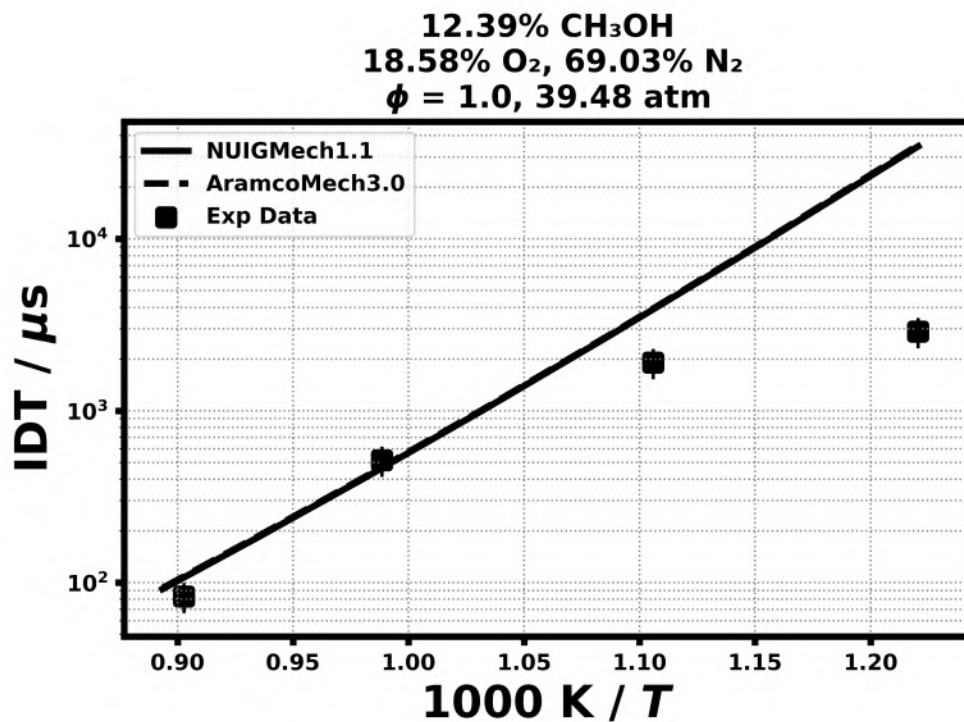
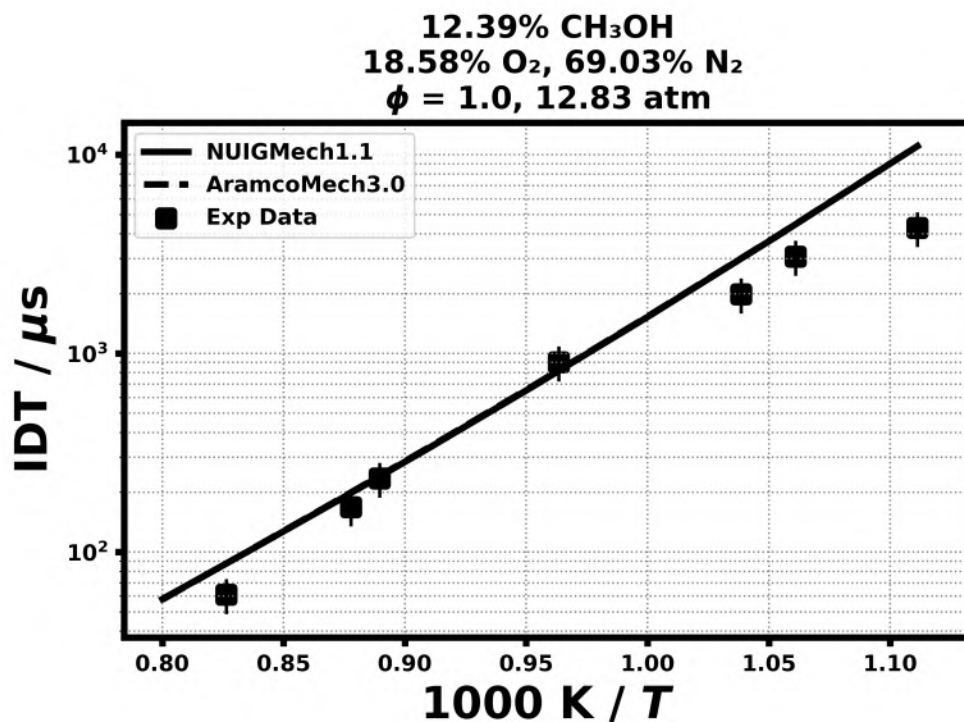
5.7% CH₃OH
8.55% O₂, 85.75% Ar
 $\phi = 1.0$, 49.18 atm



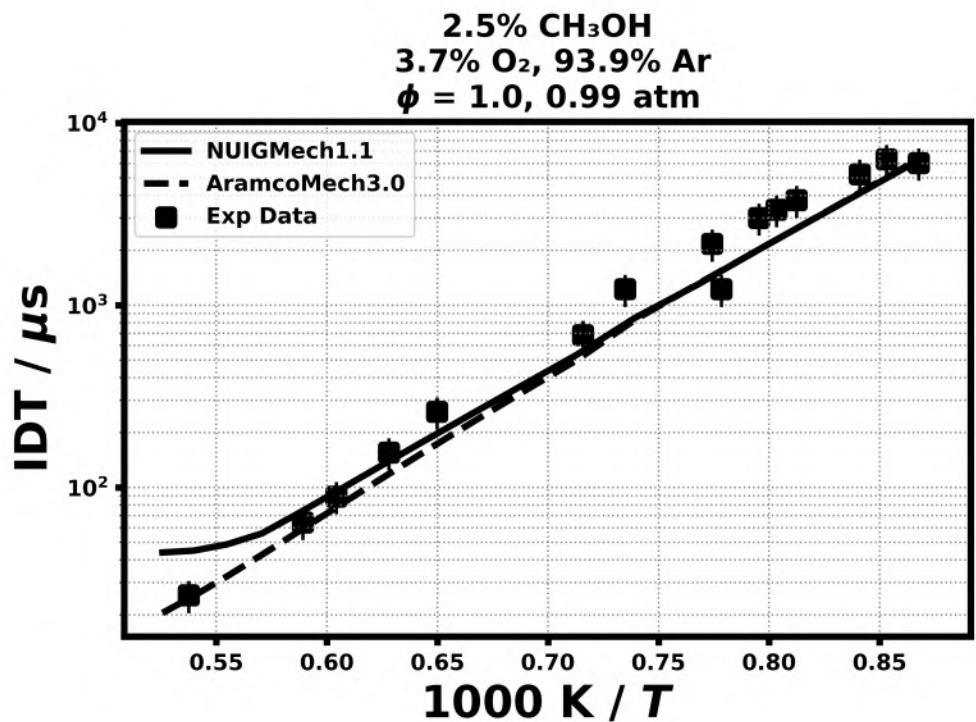
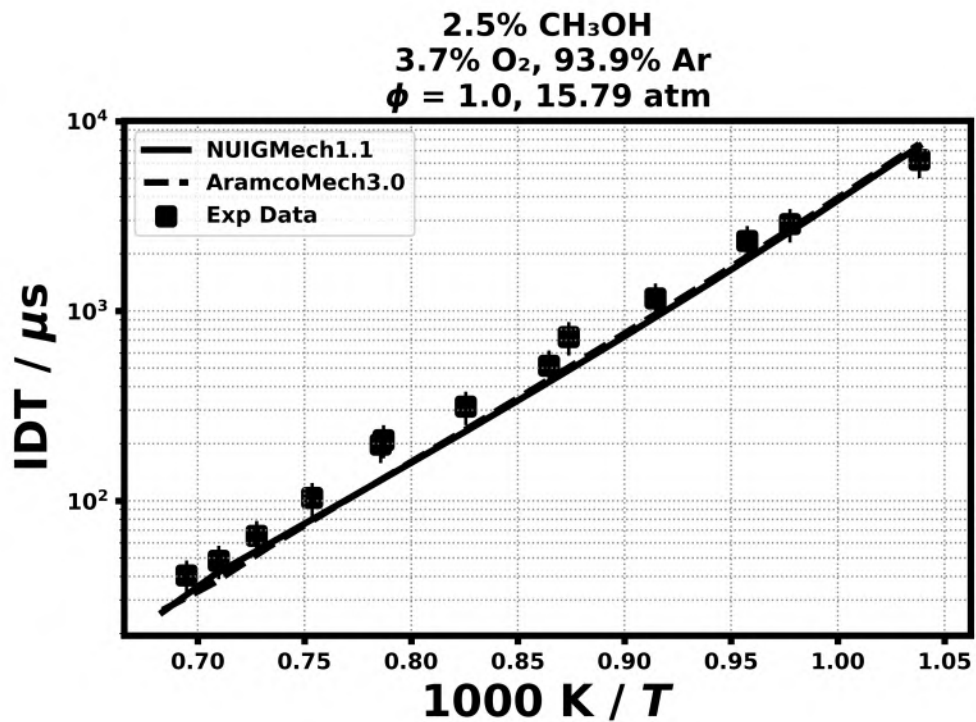
6.54% CH₃OH
19.63% O₂, 73.82% N₂
 $\phi = 0.5$, 49.61 atm

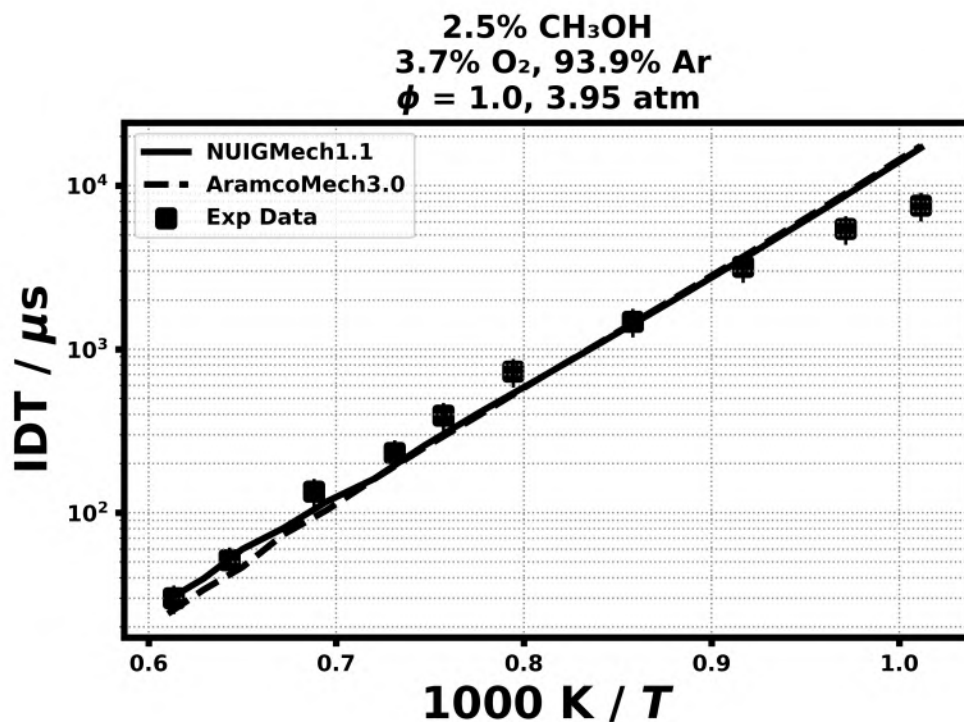


4.3) Fieweger, K., Blumenthal, R., & Adomeit, G., Combustion and Flame, 109(4) (1997) 599-619.

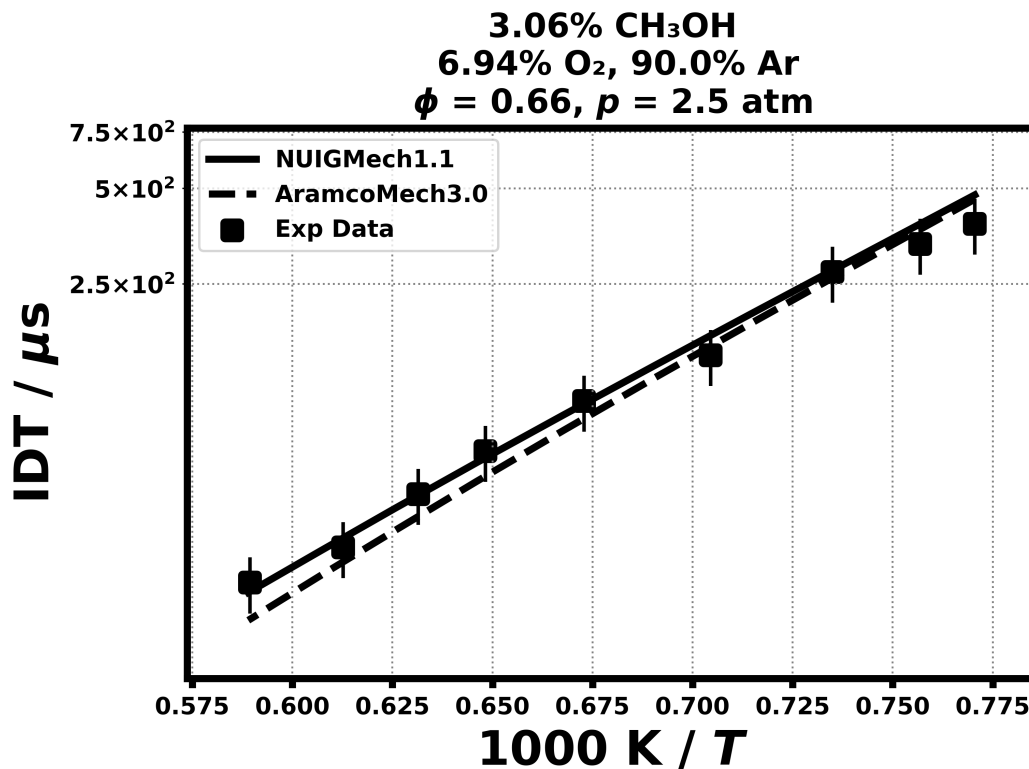


4.4) J. Herzler, European Combust. Meeting, Lund Sweden, 2013

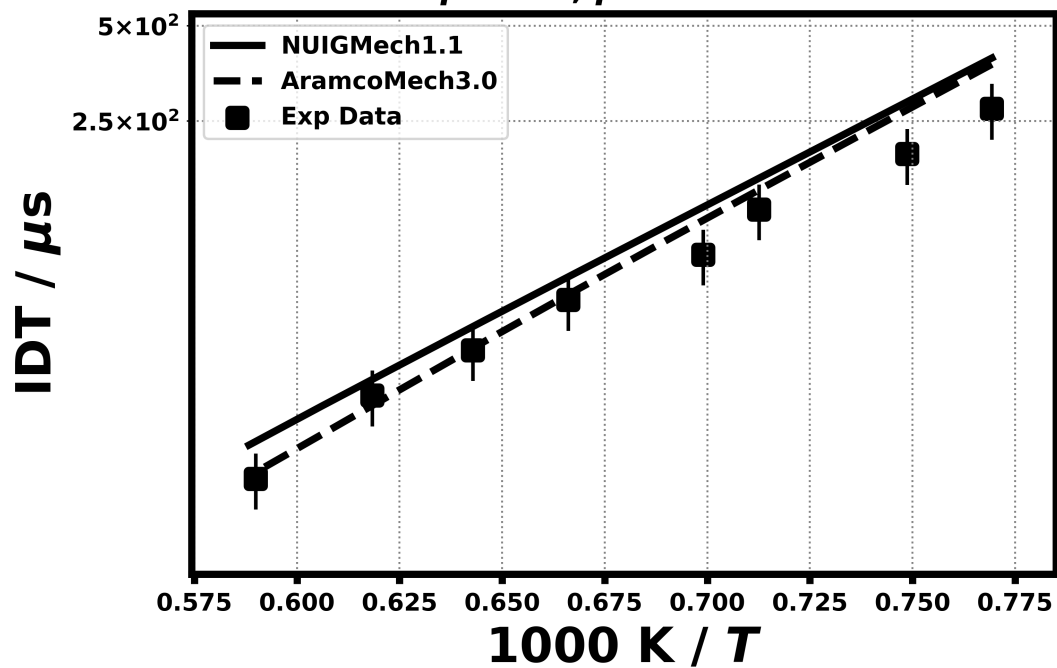




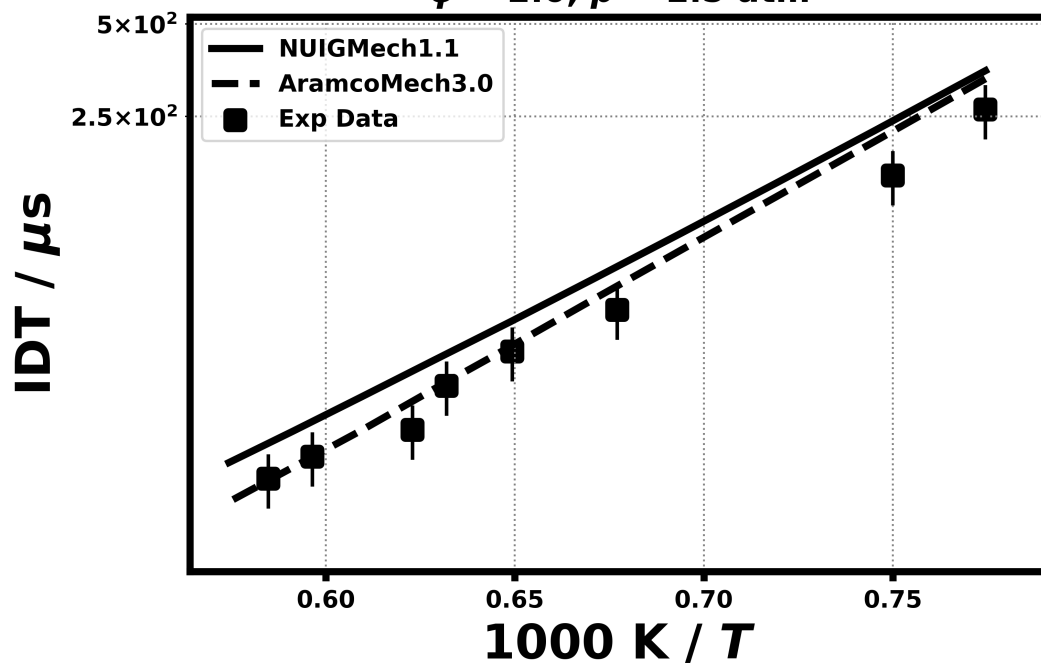
4.5) Natarajan, K., & Bhaskaran, K. A., *Combustion and Flame*, 43 (1981) 35-49.



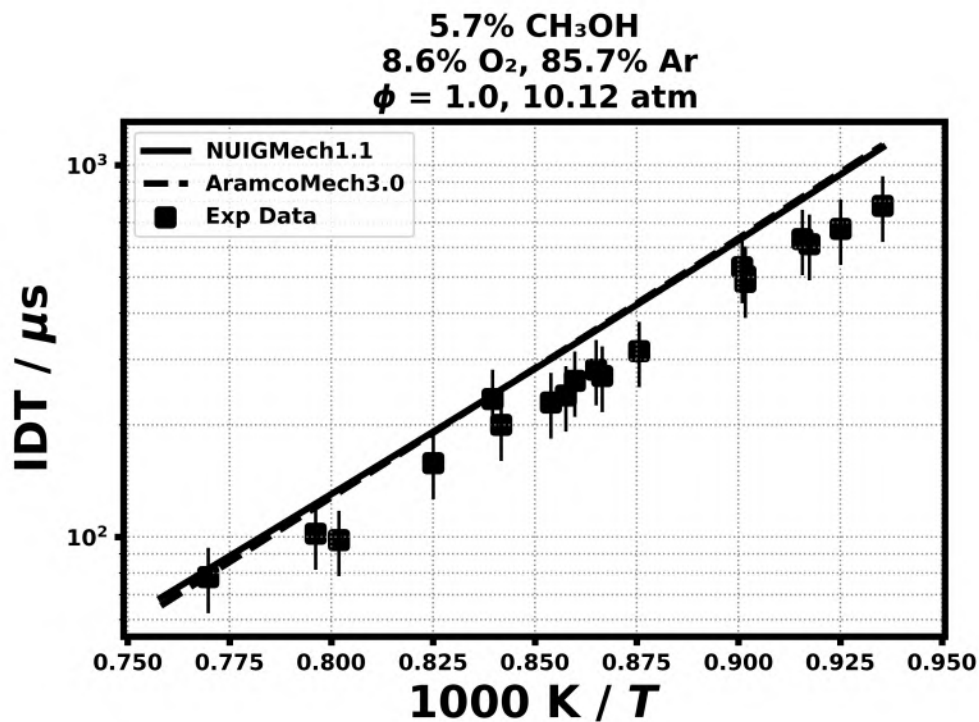
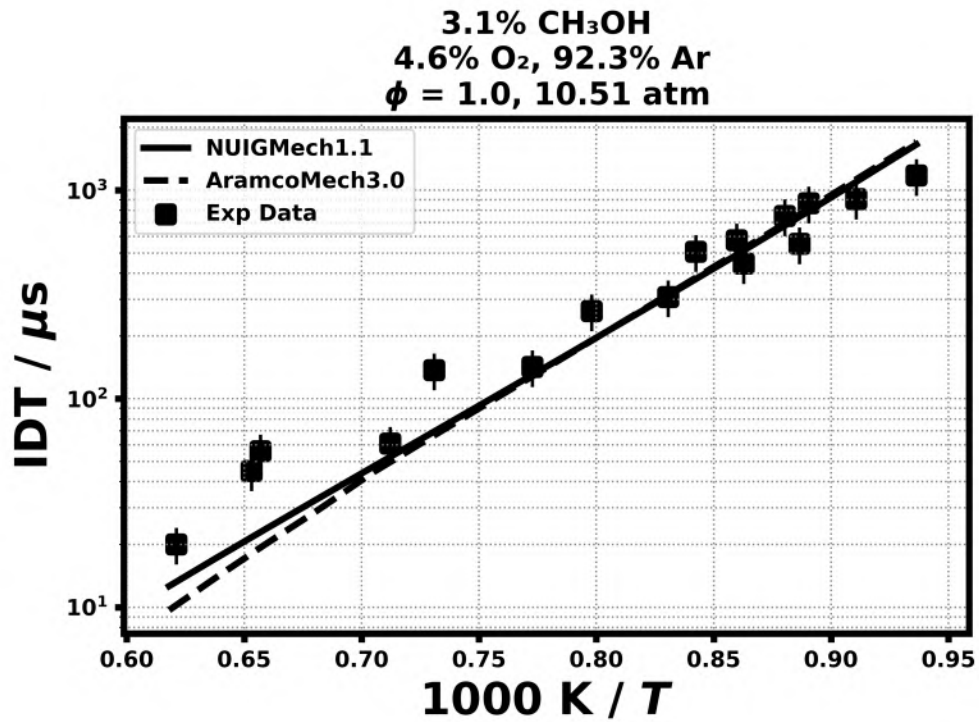
4.0% CH₃OH
6.0% O₂, 90.0% Ar
 $\phi = 1.0, p = 2.5 \text{ atm}$



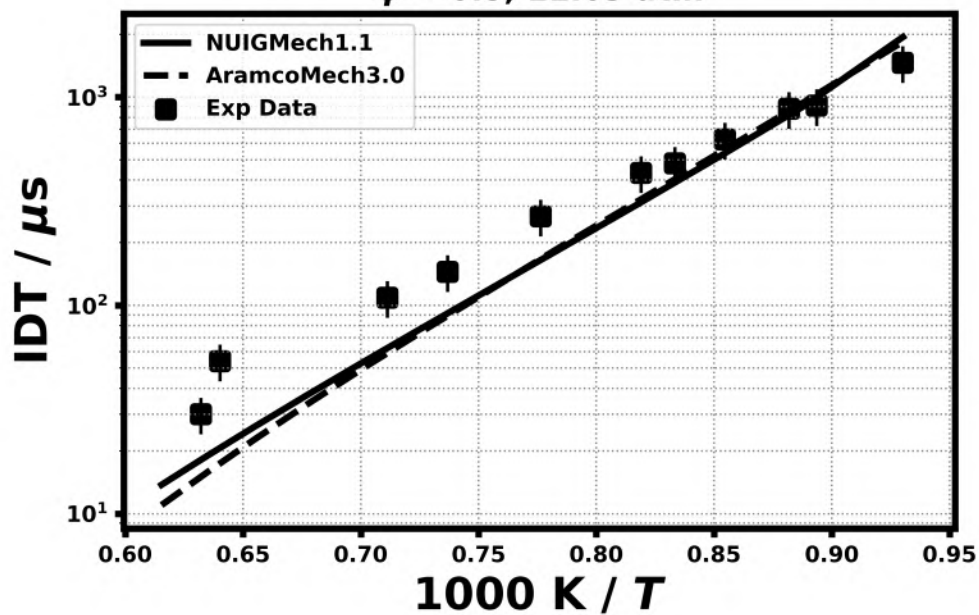
5.71% CH₃OH
4.29% O₂, 90.0% Ar
 $\phi = 2.0, p = 2.5 \text{ atm}$



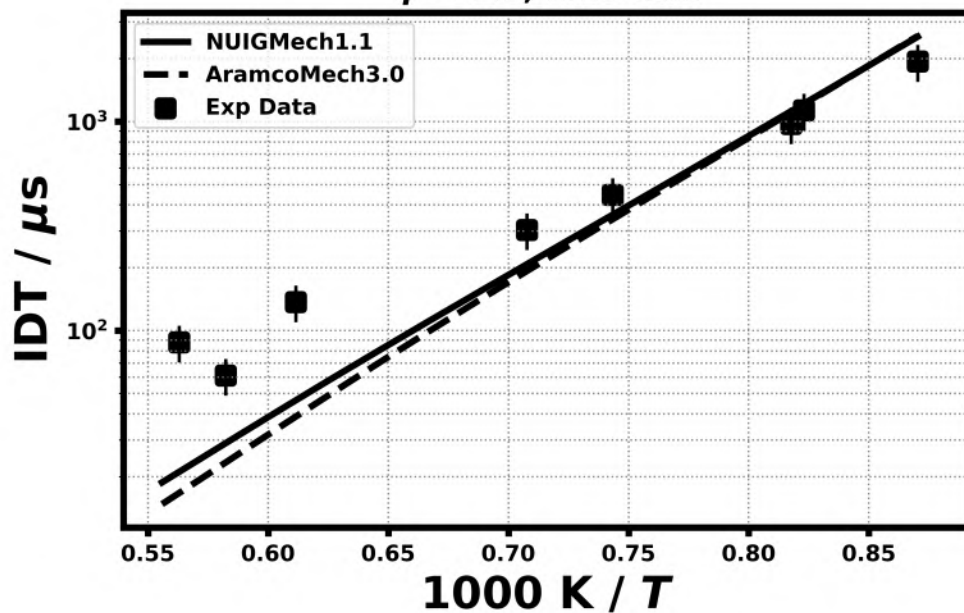
4.6) Noorani, K. E., Akih-Kumgeh, B., & Bergthorson, J. M., Energy & fuels, 24(11) (2010) 5834-5843.



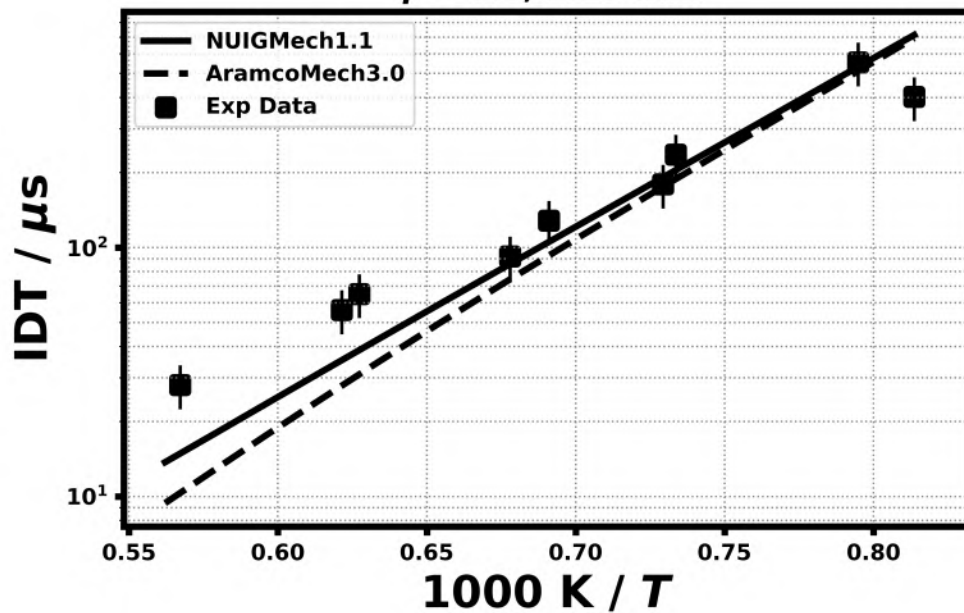
2.0% CH₃OH
6.1% O₂, 91.9% Ar
 $\phi = 0.5$, 11.63 atm



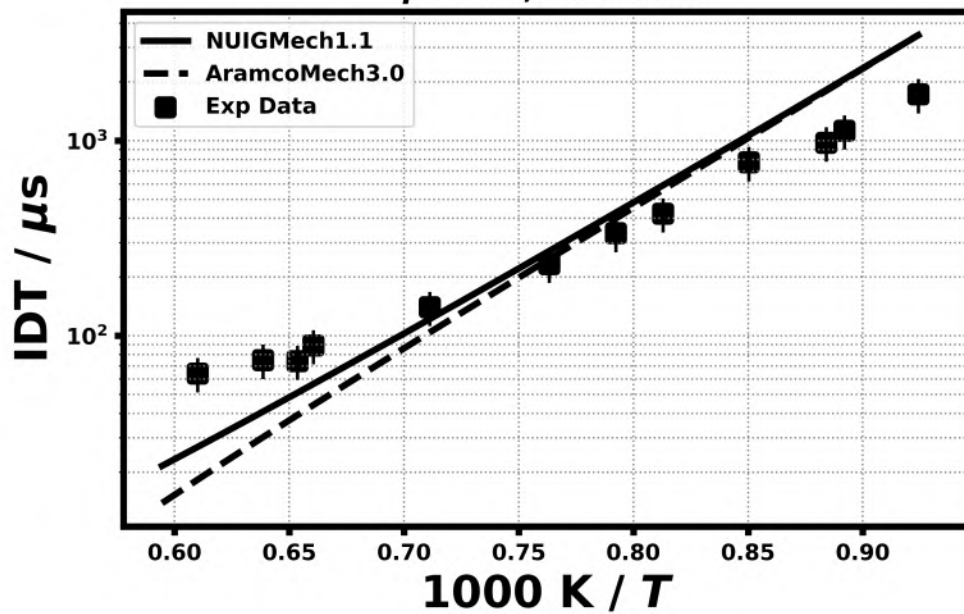
3.1% CH₃OH
4.6% O₂, 92.3% Ar
 $\phi = 1.0$, 2.23 atm



5.7% CH₃OH
8.6% O₂, 85.7% Ar
 $\phi = 1.0, 2.13 \text{ atm}$

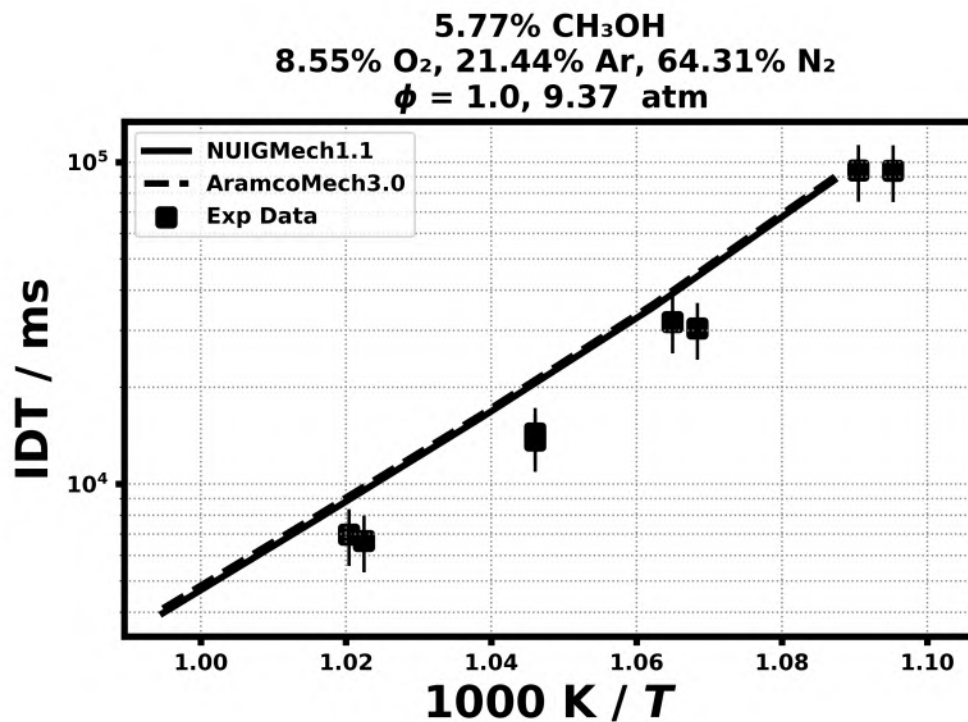
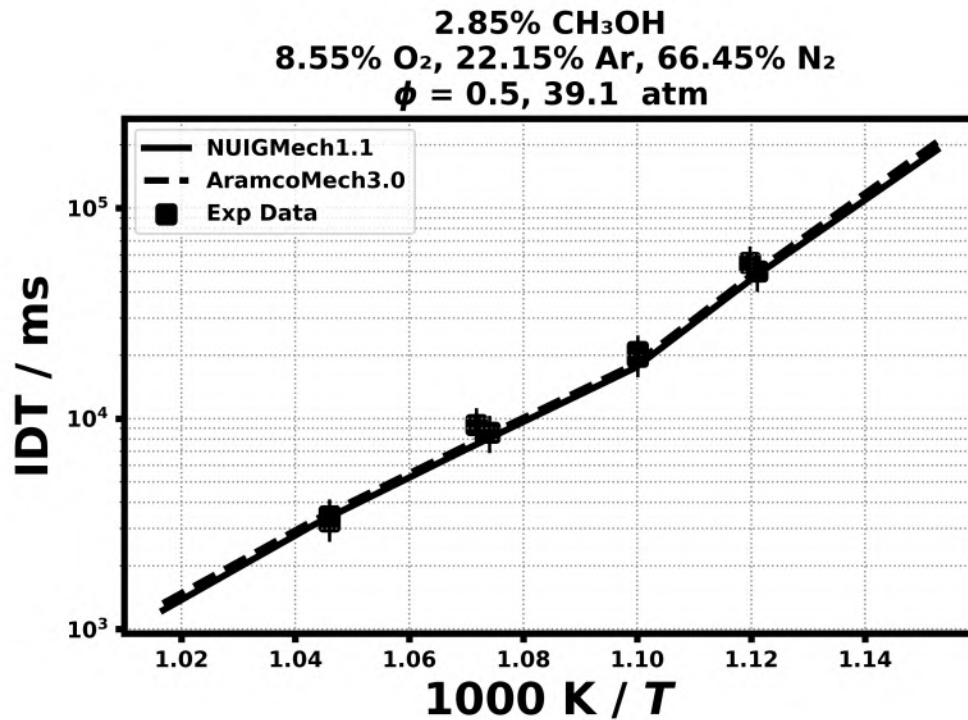


7.7% CH₃OH
5.8% O₂, 86.5% Ar
 $\phi = 2.0, 2.18 \text{ atm}$

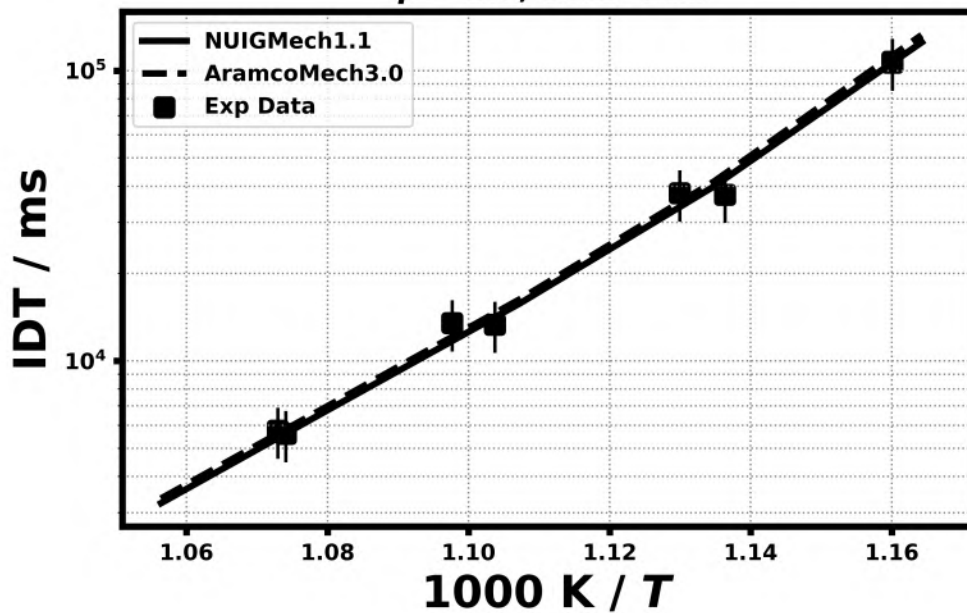


RCM Ignition delay time

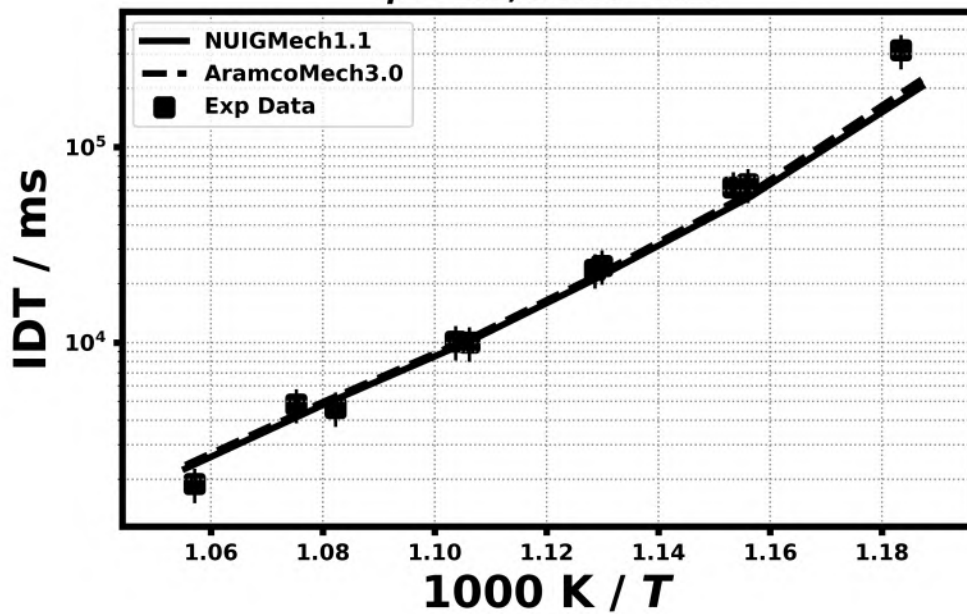
4.7) Burke, U., Metcalfe, W. K., Burke, S. M., Heufer, K. A., Dagaut, P., & Curran, H. J., Combustion and Flame, 165 (2016) 125-136.



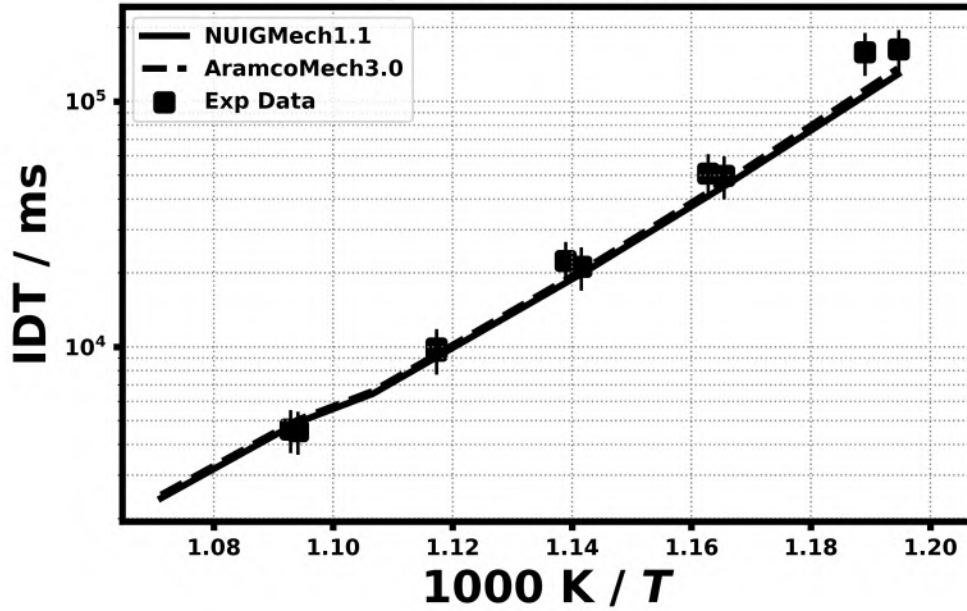
5.77% CH₃OH
8.55% O₂, 21.44% Ar, 64.31% N₂
 $\phi = 1.0, 30.1 \text{ atm}$



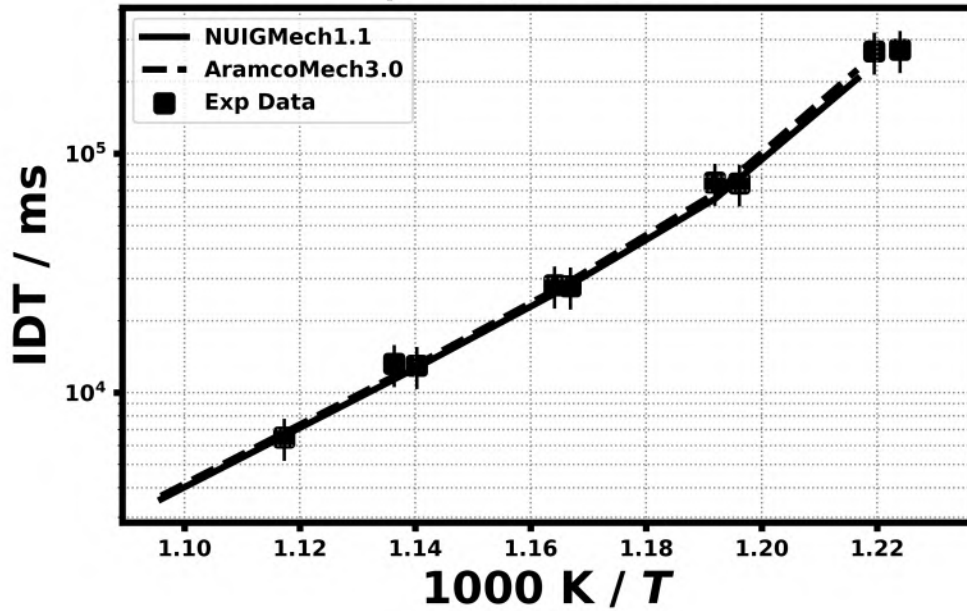
5.77% CH₃OH
8.55% O₂, 21.44% Ar, 64.31% N₂
 $\phi = 1.0, 38.65 \text{ atm}$



11.4% CH₃OH
8.55% O₂, 20.01% Ar, 60.04% N₂
 $\phi = 2.0, 29.52 \text{ atm}$

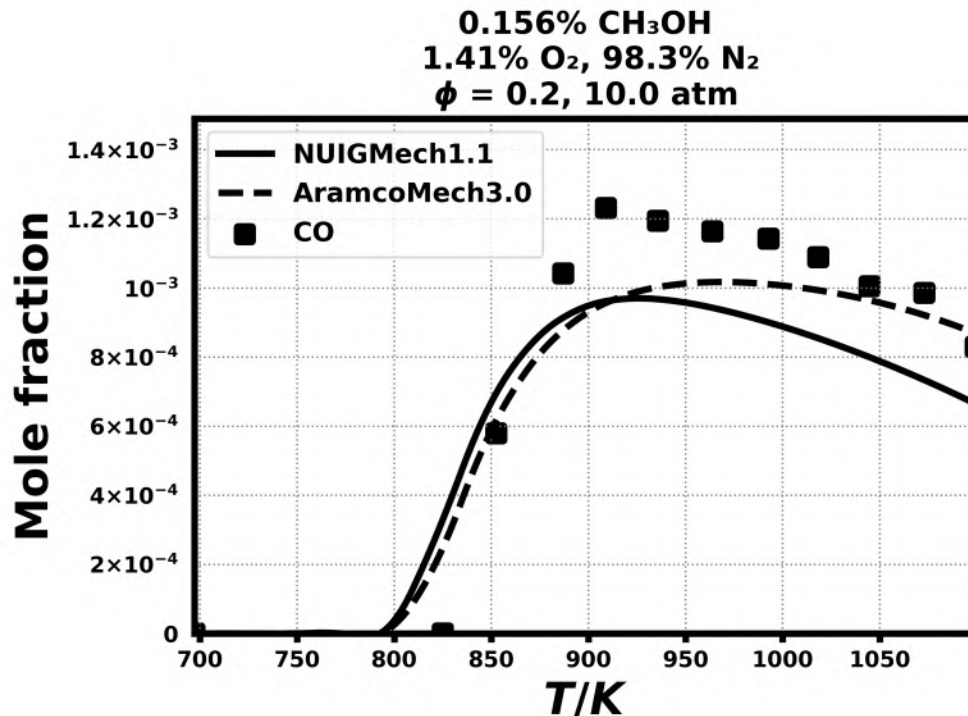
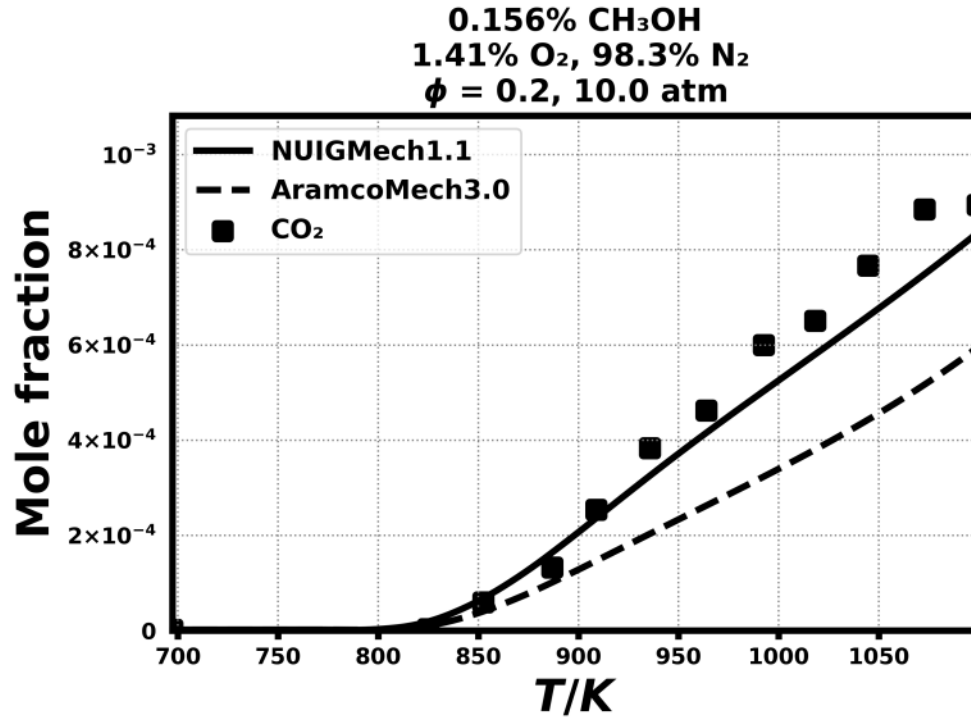


11.4% CH₃OH
8.55% O₂, 20.01% Ar, 60.04% N₂
 $\phi = 2.0, 40.04 \text{ atm}$

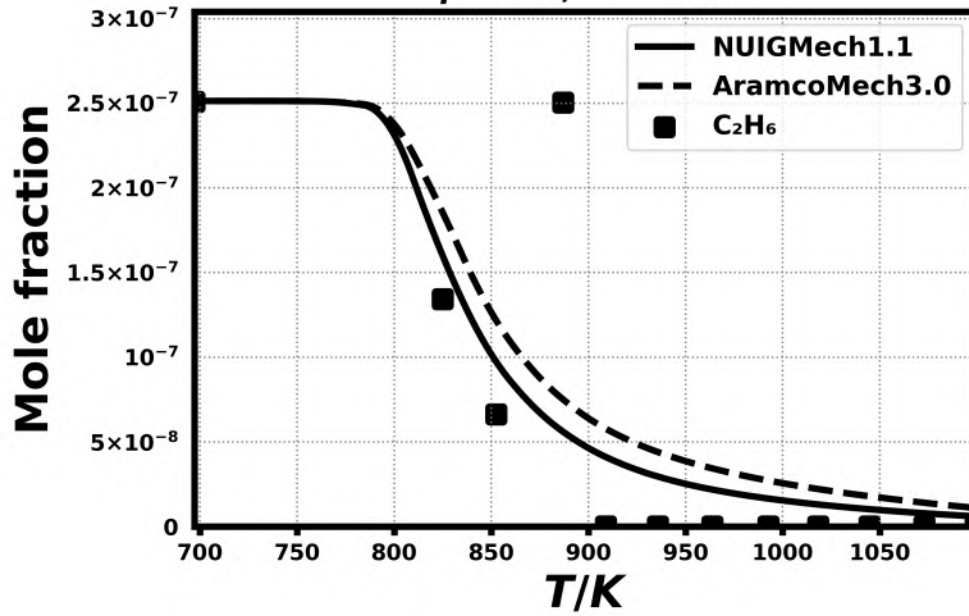


Speciation in Jet-stirred reactor

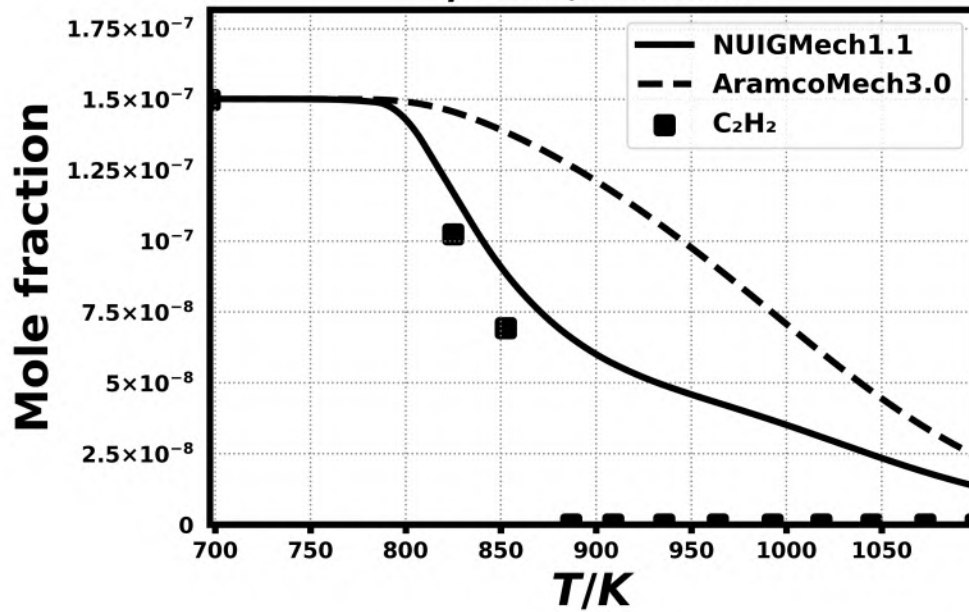
4.8) Burke, U., Metcalfe, W. K., Burke, S. M., Heufer, K. A., Dagaut, P., & Curran, H. J., *Combustion and Flame*, 165 (2016) 125-136.



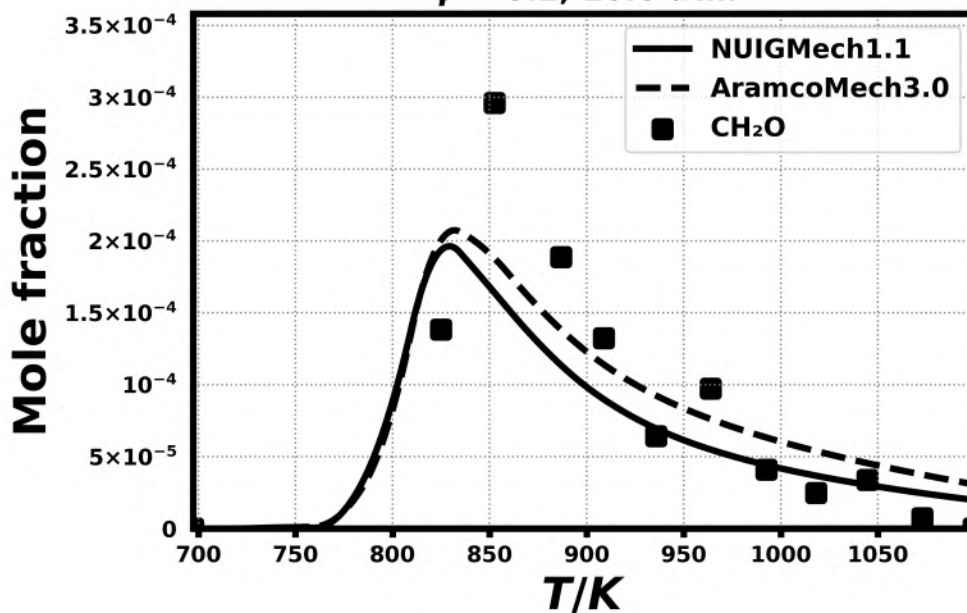
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2, 10.0 \text{ atm}$



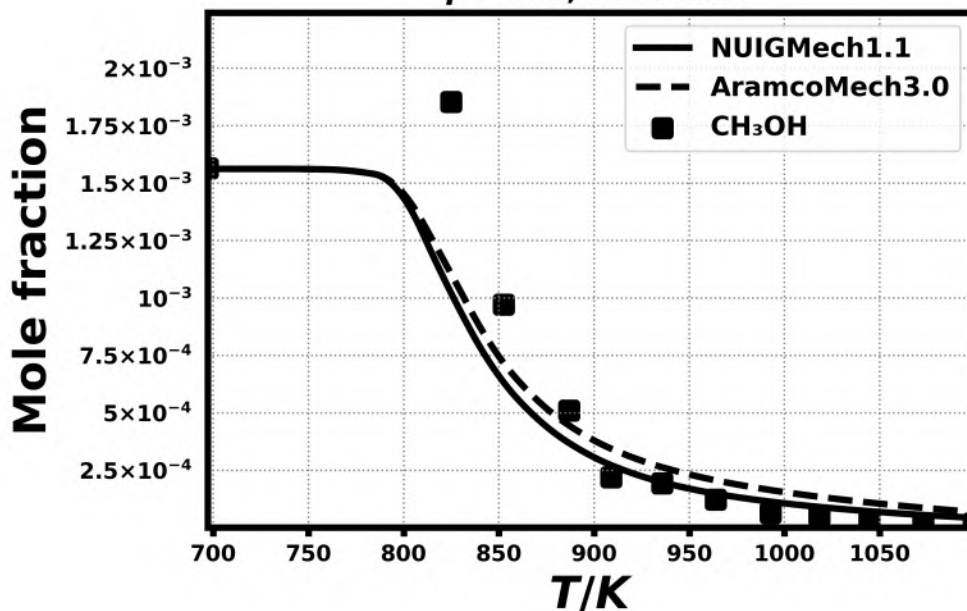
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2, 10.0 \text{ atm}$



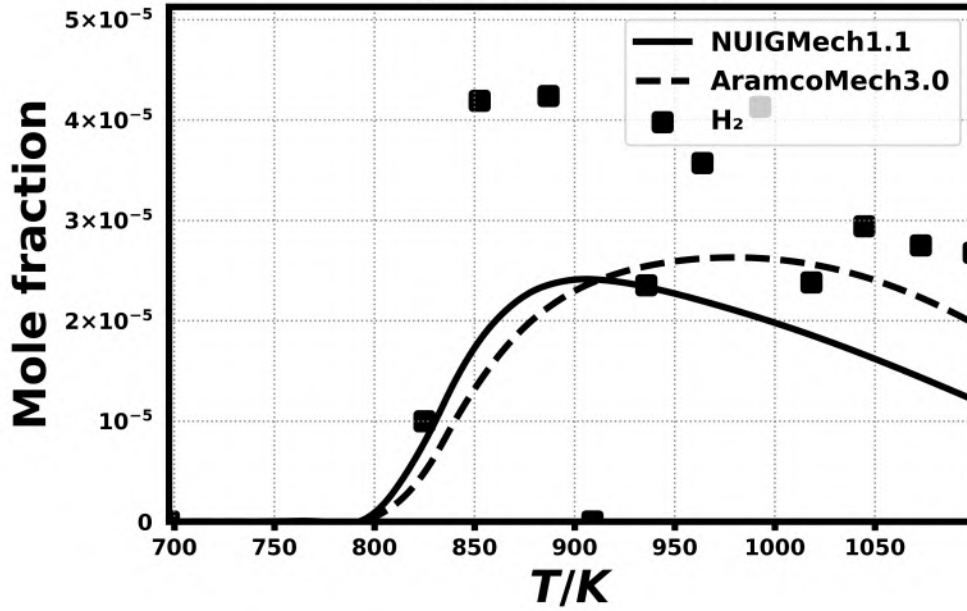
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2, 10.0 \text{ atm}$



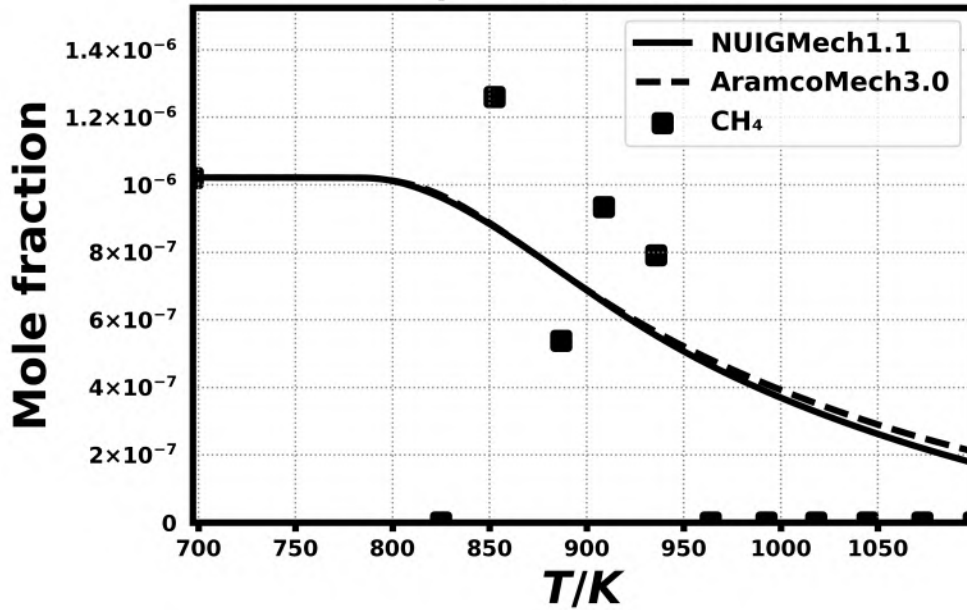
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2, 10.0 \text{ atm}$



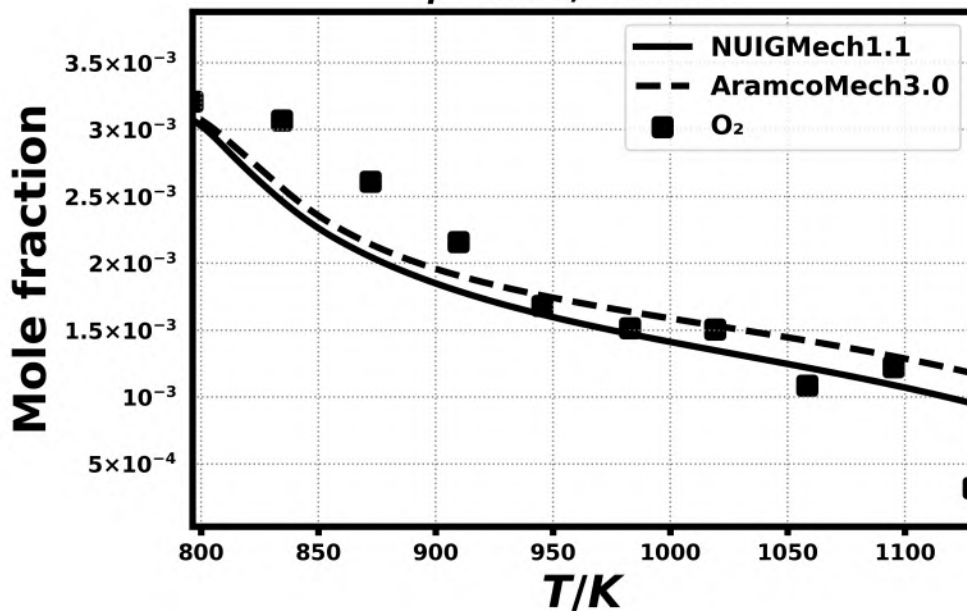
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2$, 10.0 atm



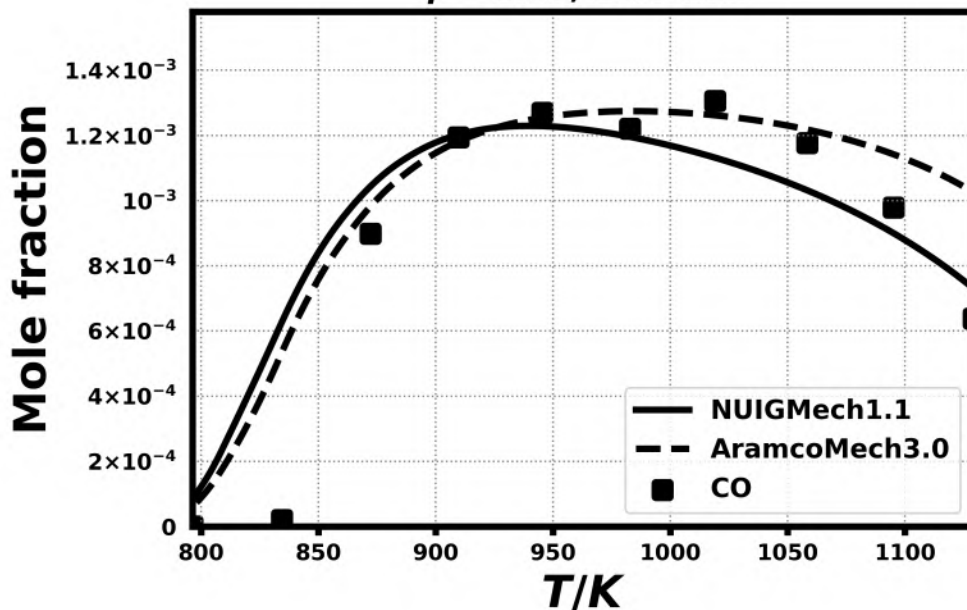
0.156% CH₃OH
1.41% O₂, 98.3% N₂
 $\phi = 0.2$, 10.0 atm



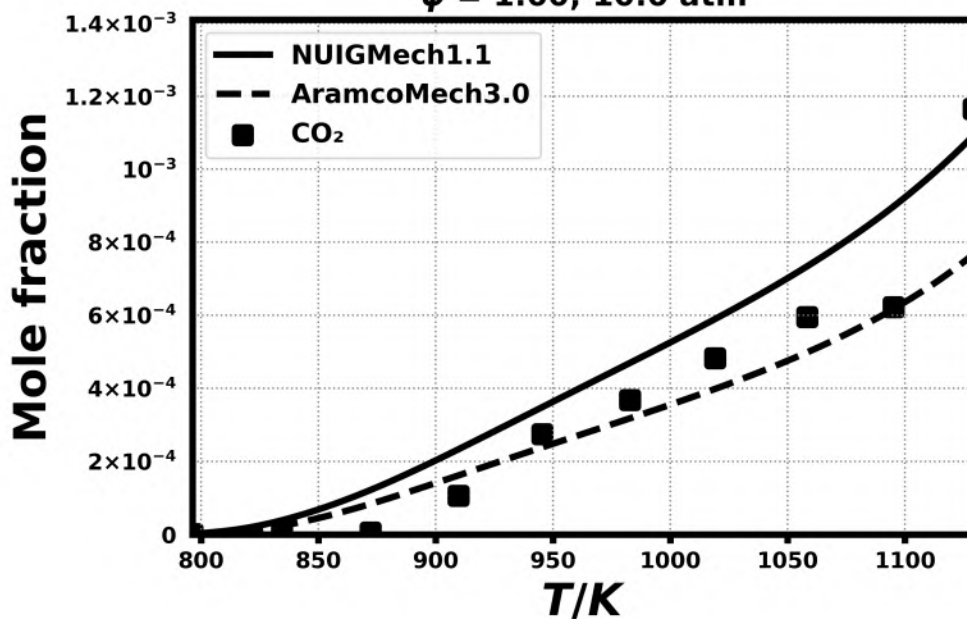
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



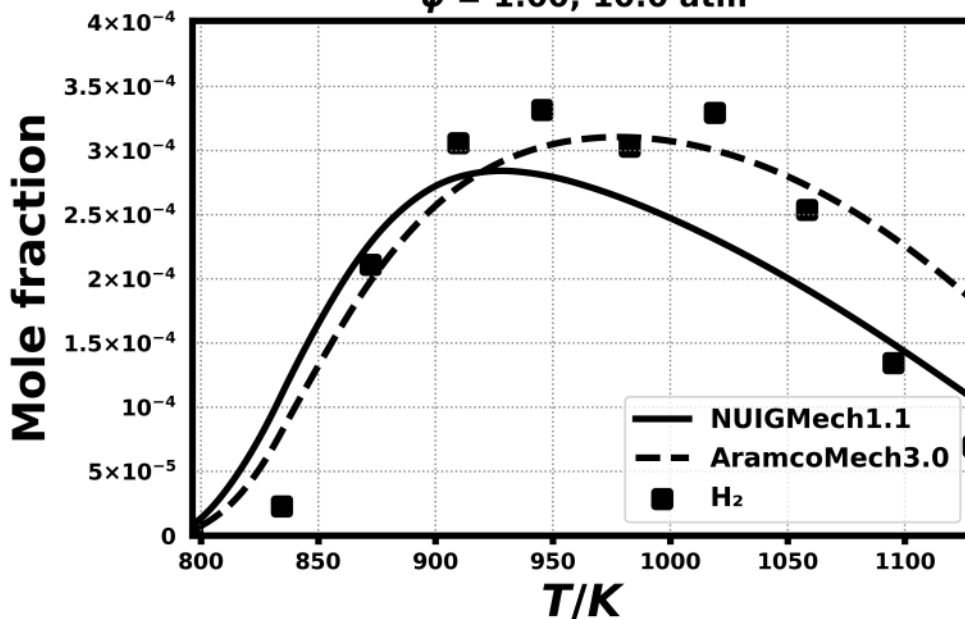
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



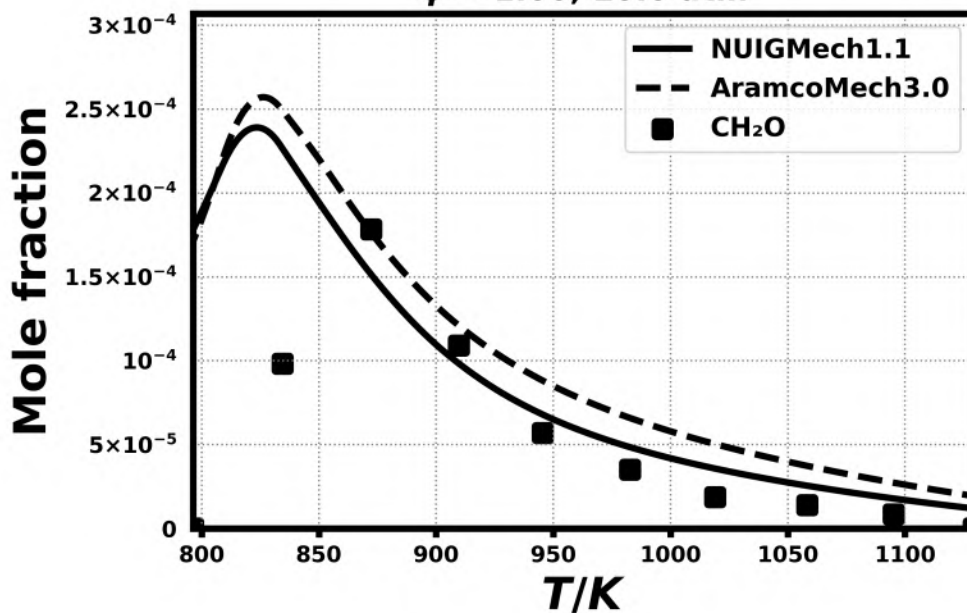
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
 0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



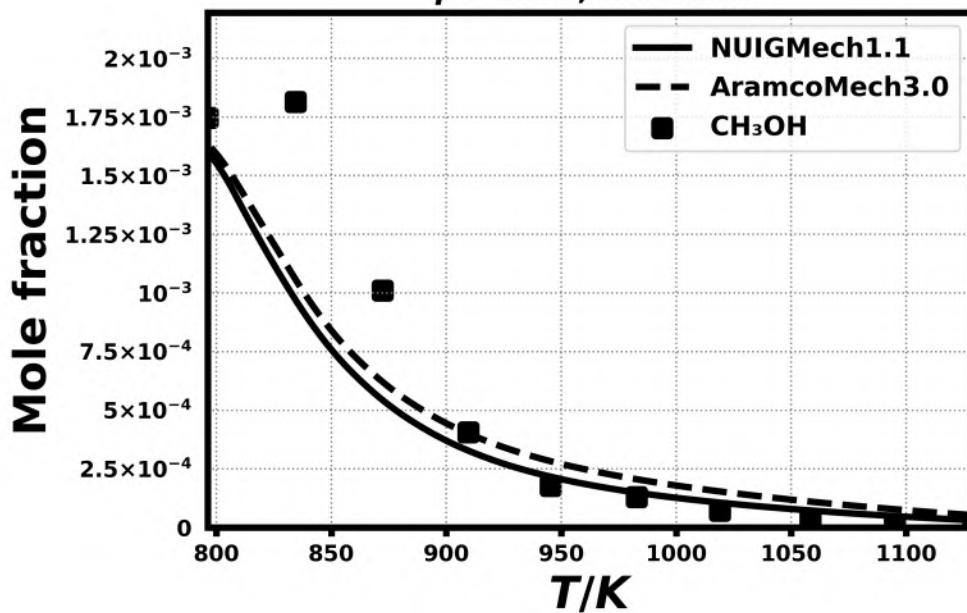
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
 0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



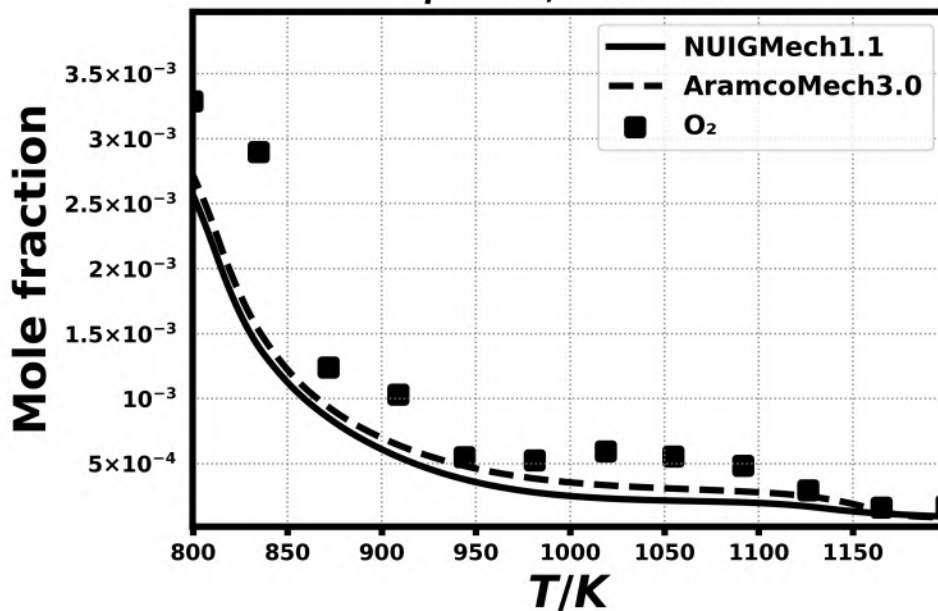
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



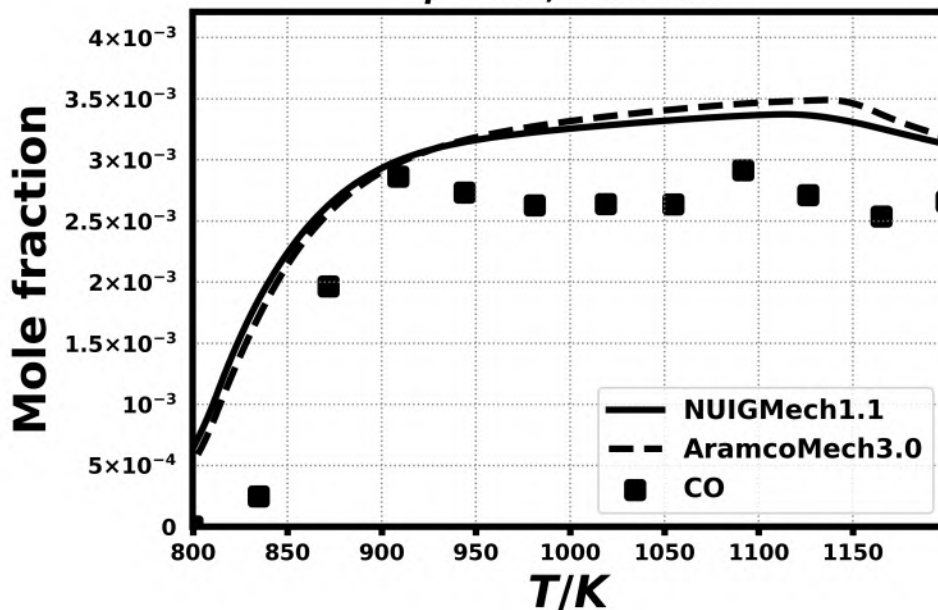
0.175% CH₃OH, 0.009822% CH₂O, 0.002% CO
0.321% O₂, 99.5% N₂
 $\phi = 1.00$, 10.0 atm



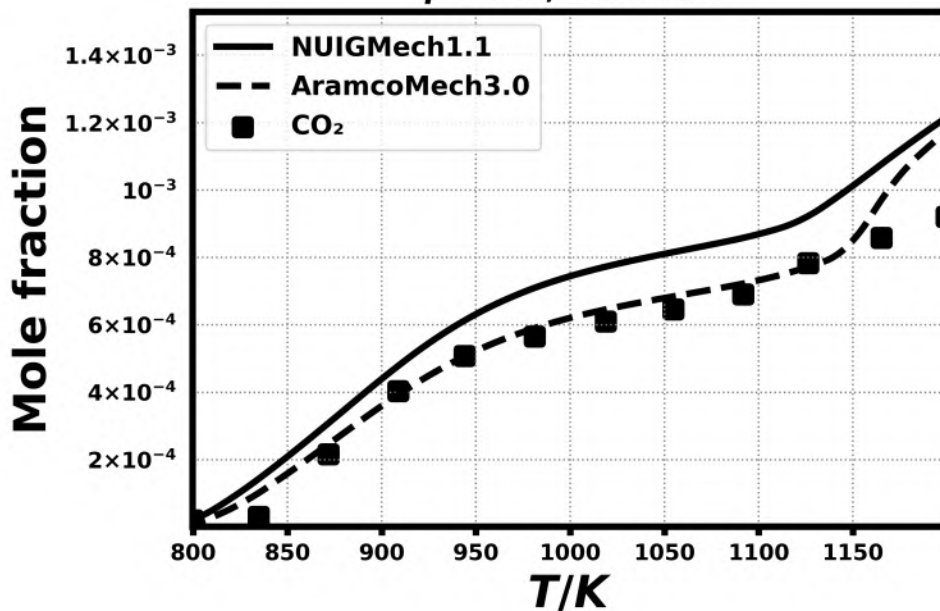
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



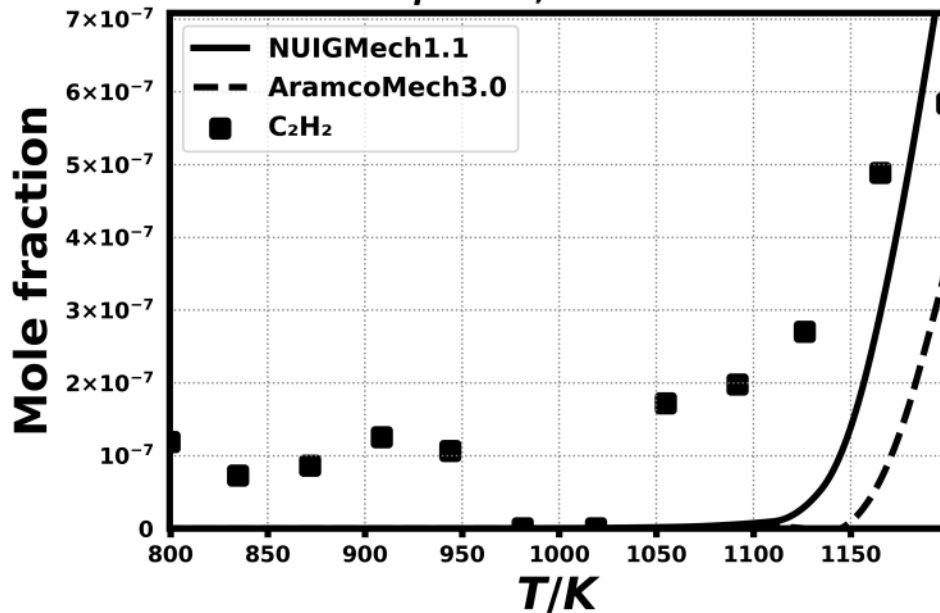
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



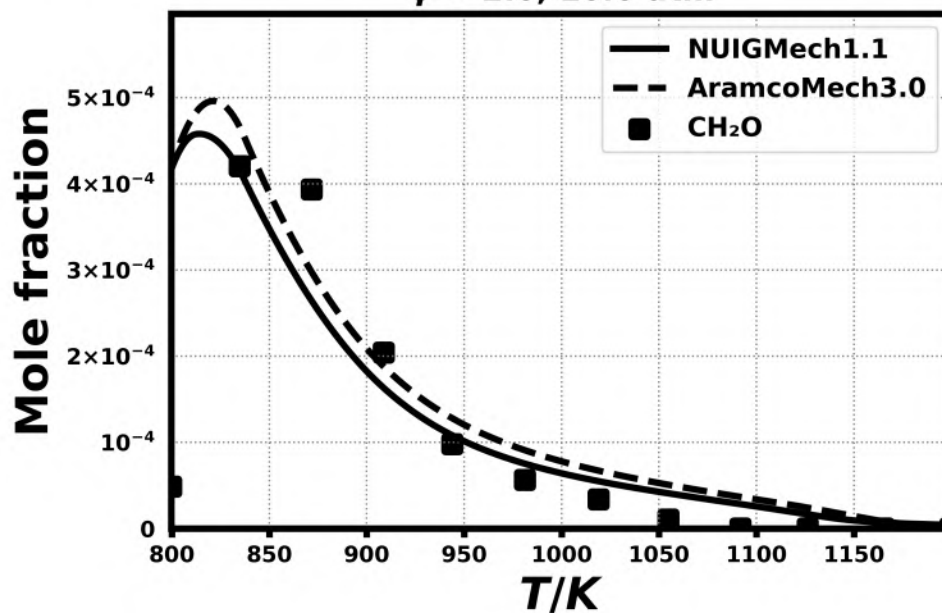
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



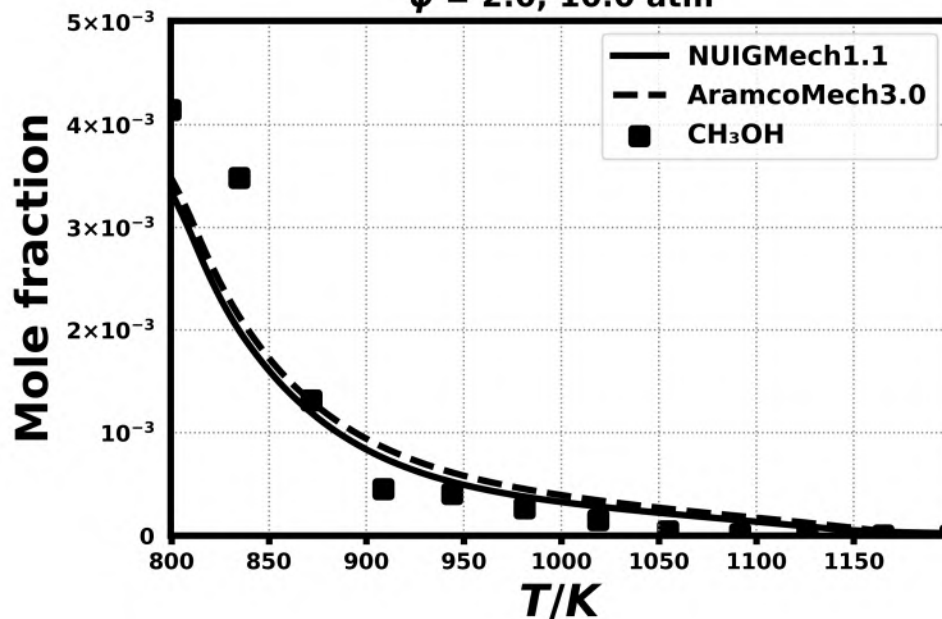
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



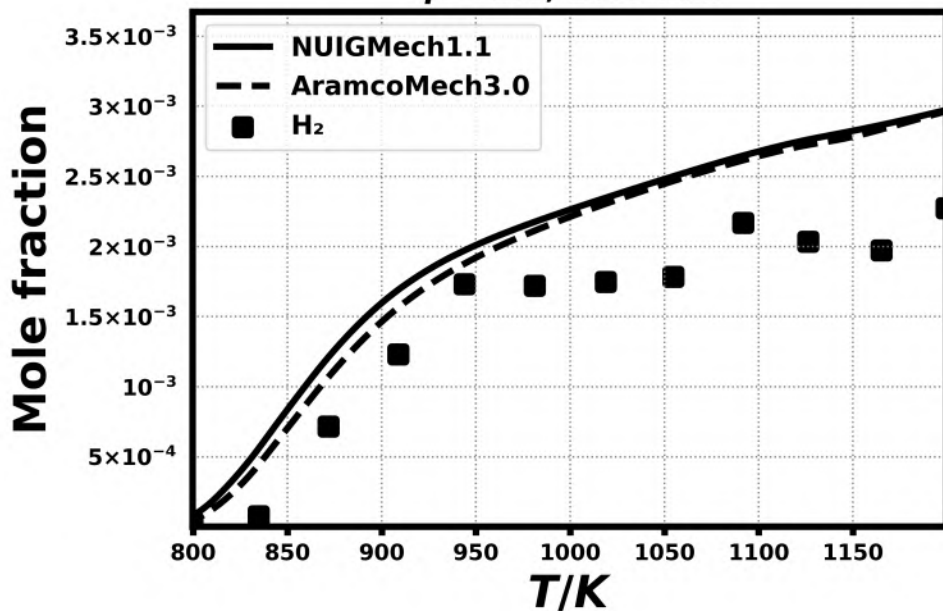
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



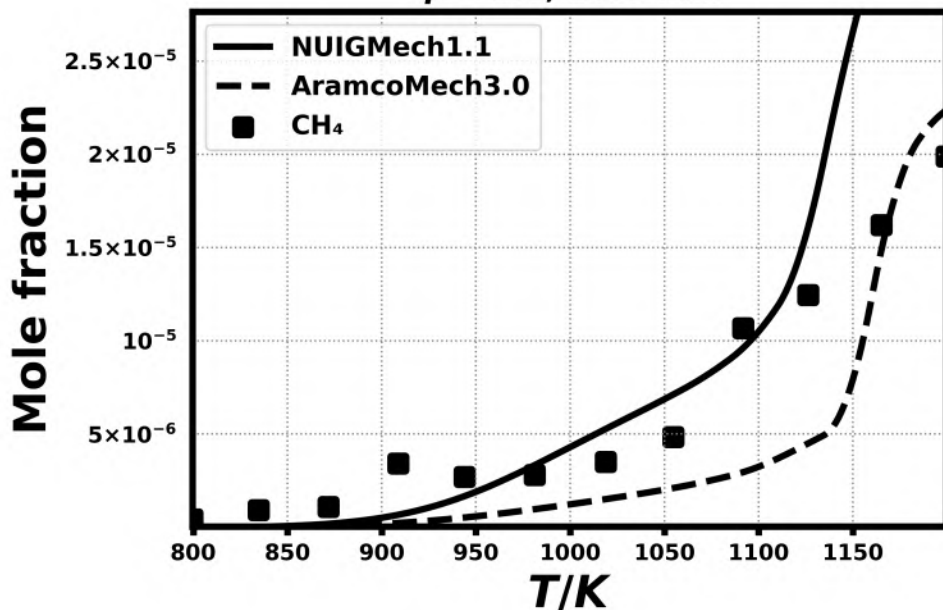
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



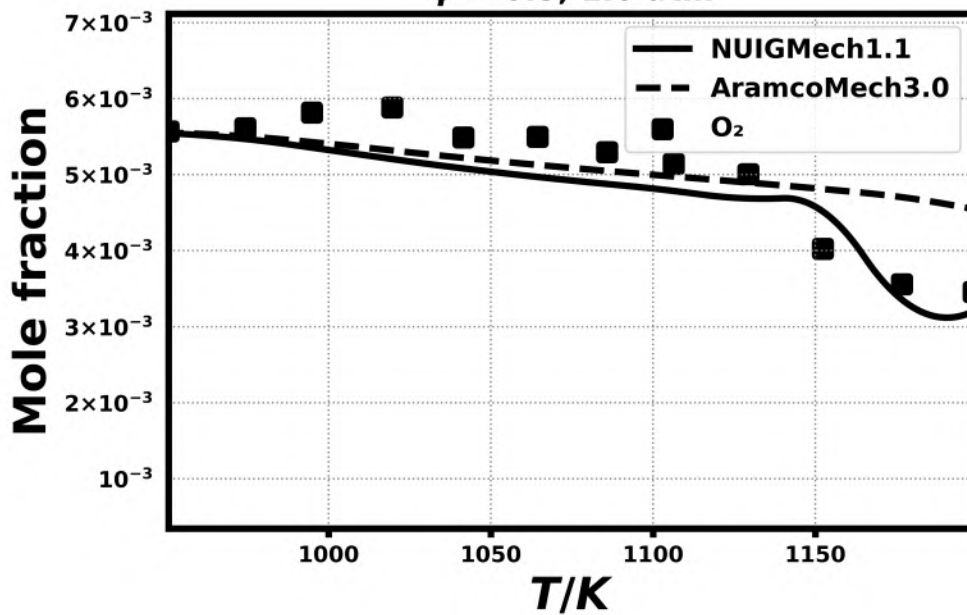
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



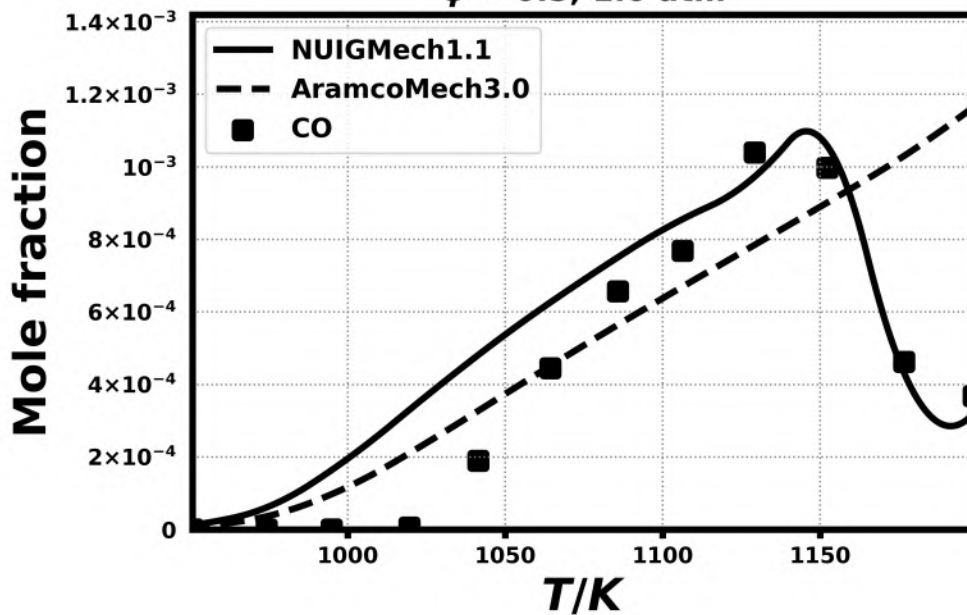
0.414% CH₃OH, 0.004866% CH₂O, 0.02464% CO
0.329% O₂, 99.3% N₂
 $\phi = 2.0, 10.0 \text{ atm}$



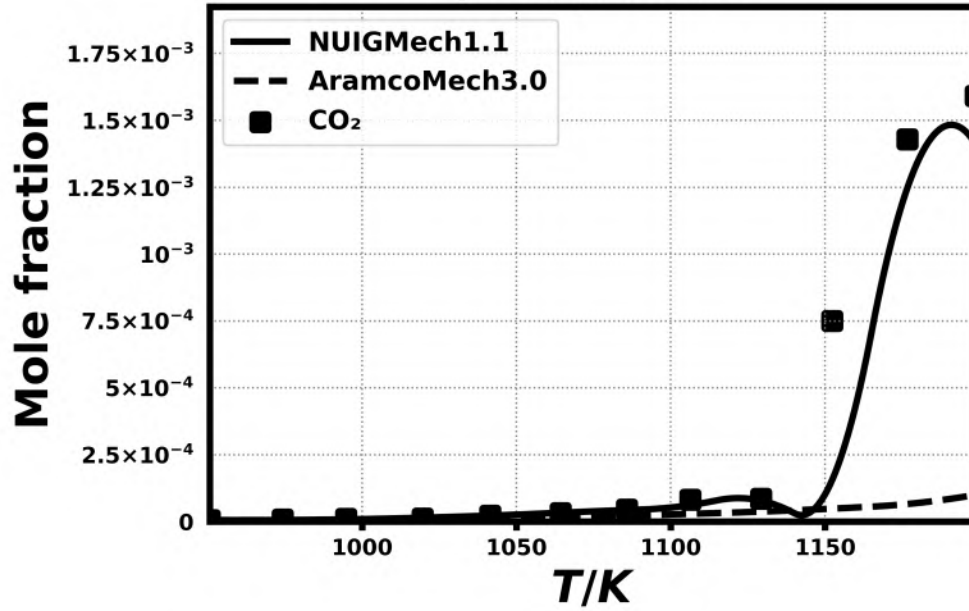
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



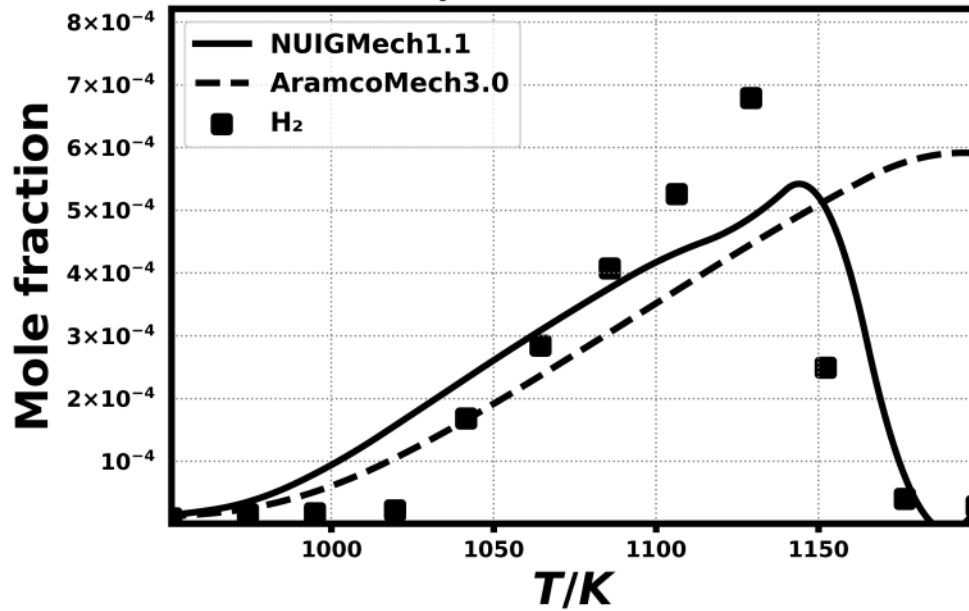
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



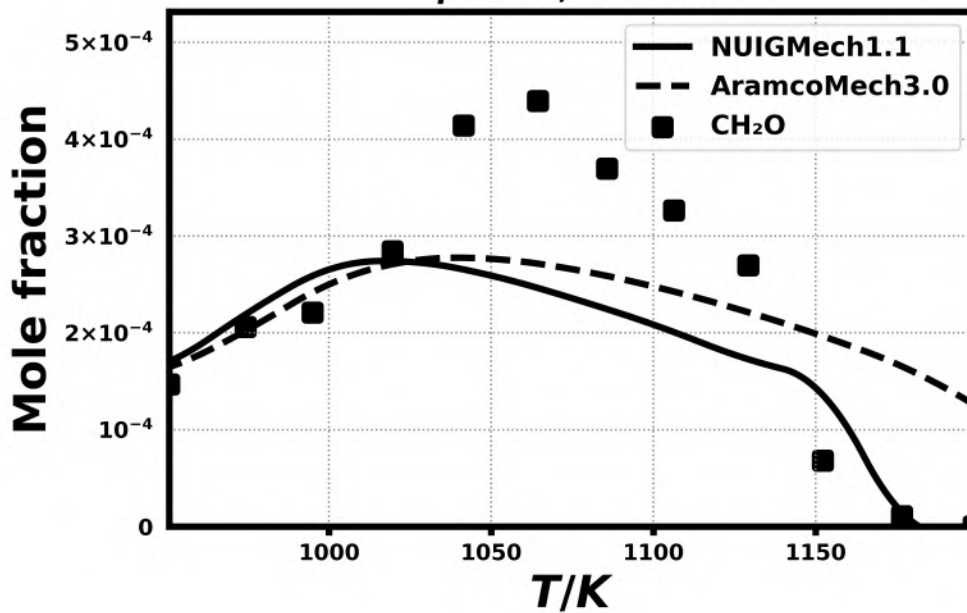
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



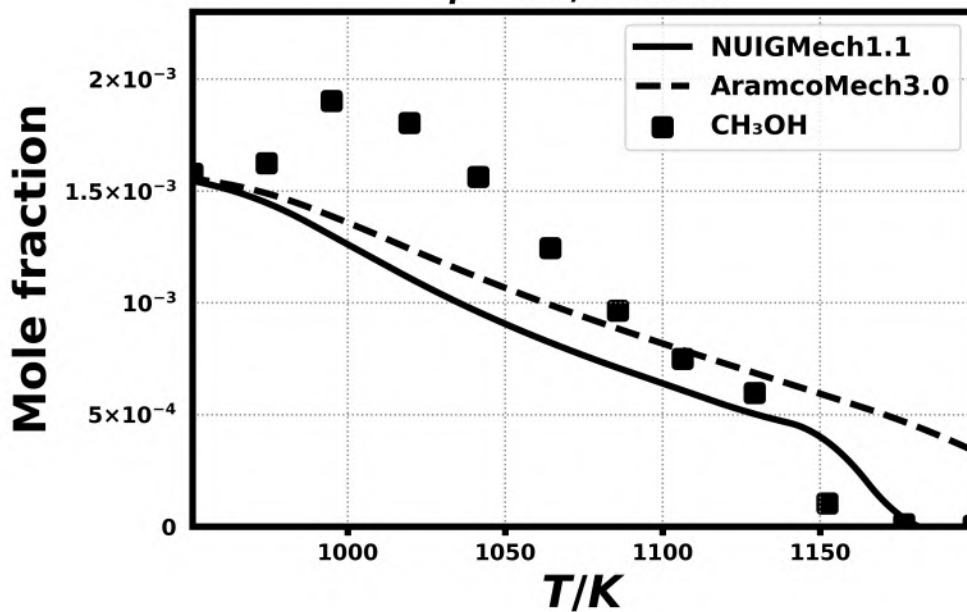
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



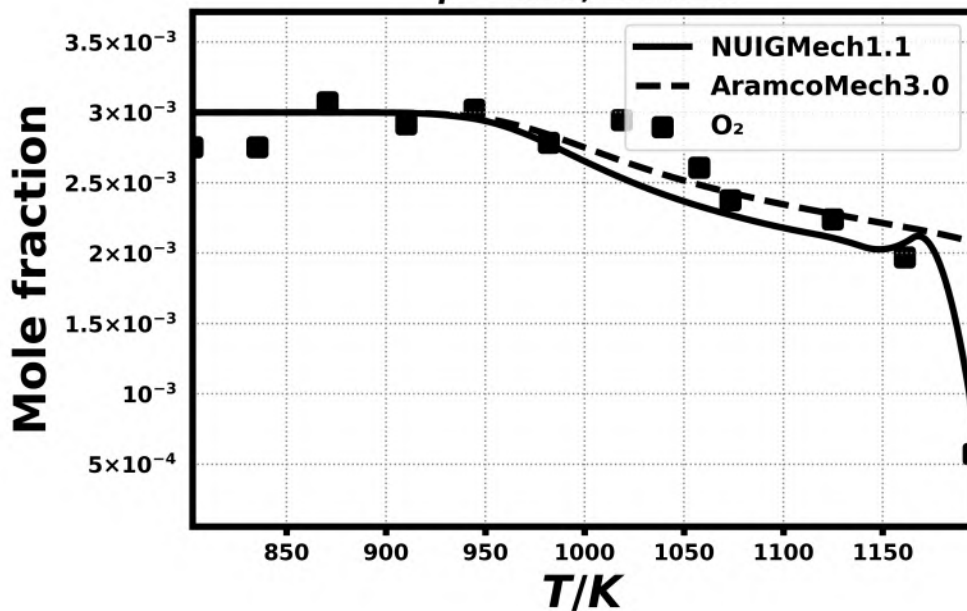
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



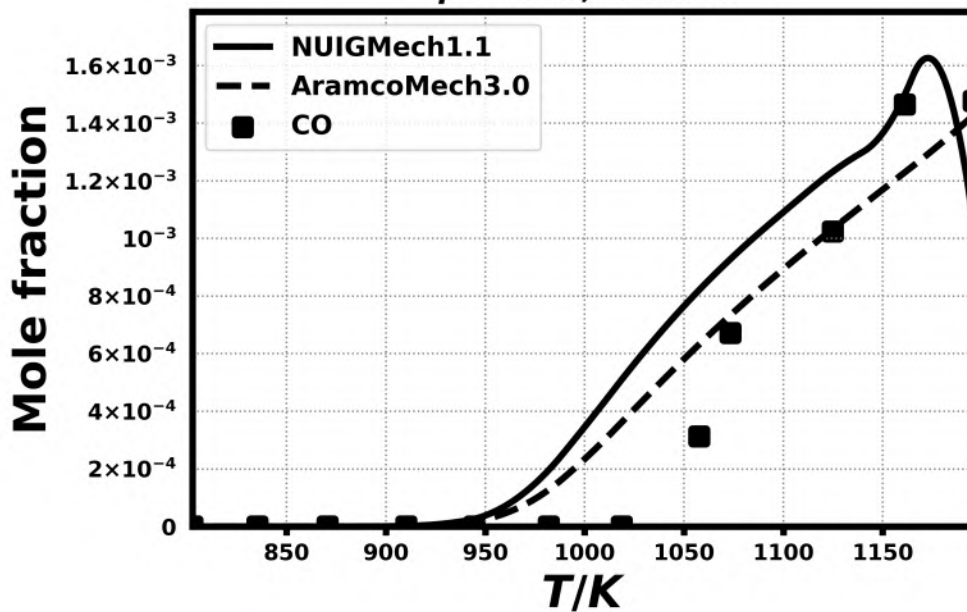
0.158% CH₃OH, 0.0147% CH₂O
0.557% O₂, 99.2% N₂
 $\phi = 0.5, 1.0 \text{ atm}$



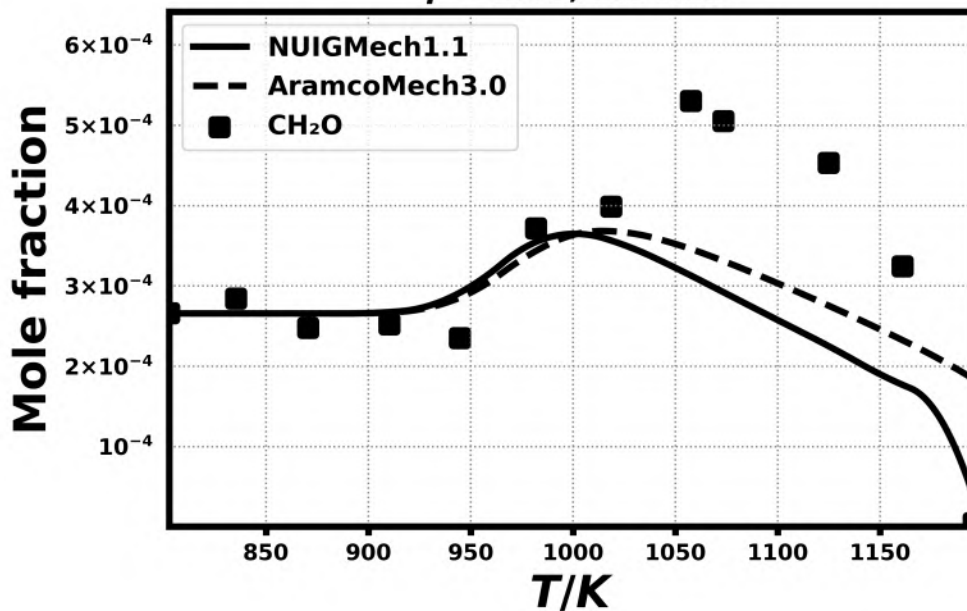
0.2% CH₃OH, 0.02658% CH₂O
0.3% O₂, 99.5% N₂
 $\phi = 1.00, 1.0 \text{ atm}$



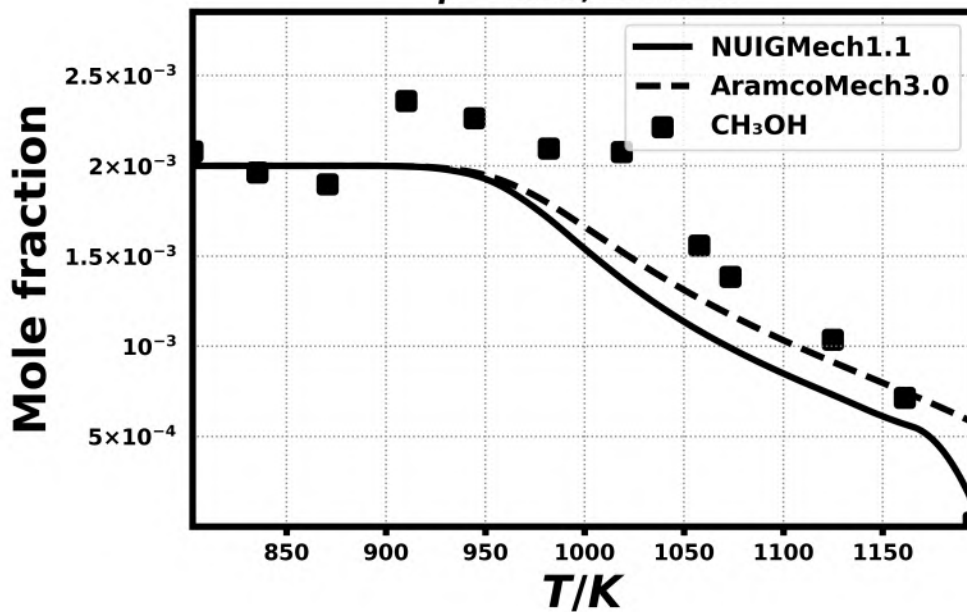
0.2% CH₃OH, 0.02658% CH₂O
0.3% O₂, 99.5% N₂
 $\phi = 1.00, 1.0 \text{ atm}$



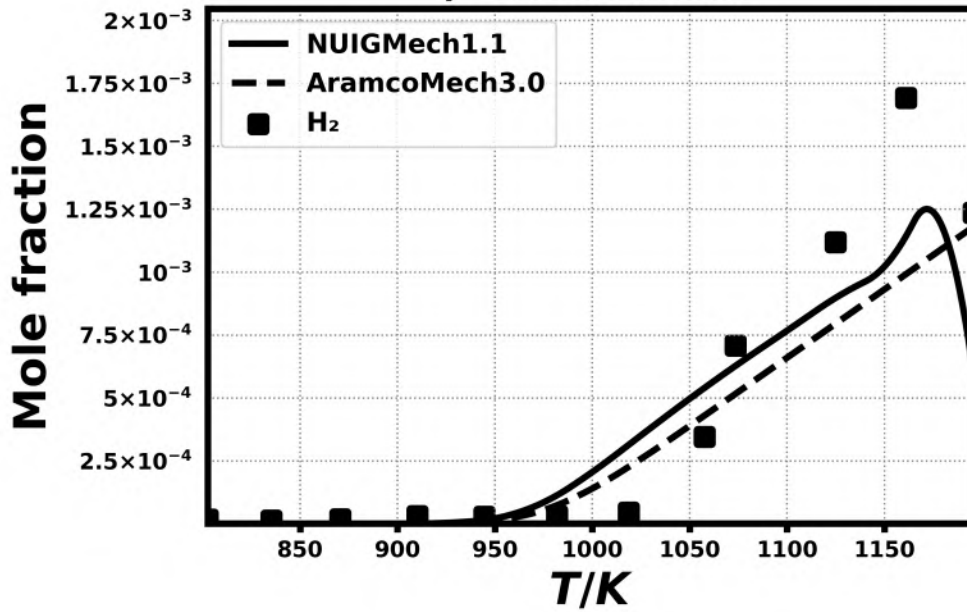
0.2% CH₃OH, 0.02658% CH₂O
0.3% O₂, 99.5% N₂
 $\phi = 1.00, 1.0 \text{ atm}$



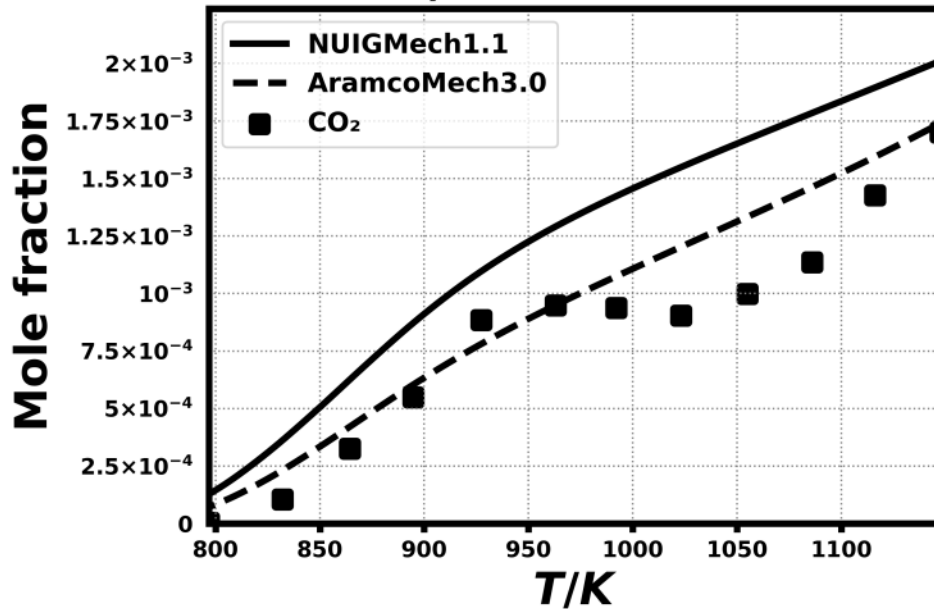
0.2% CH₃OH, 0.02658% CH₂O
0.3% O₂, 99.5% N₂
 $\phi = 1.00, 1.0 \text{ atm}$



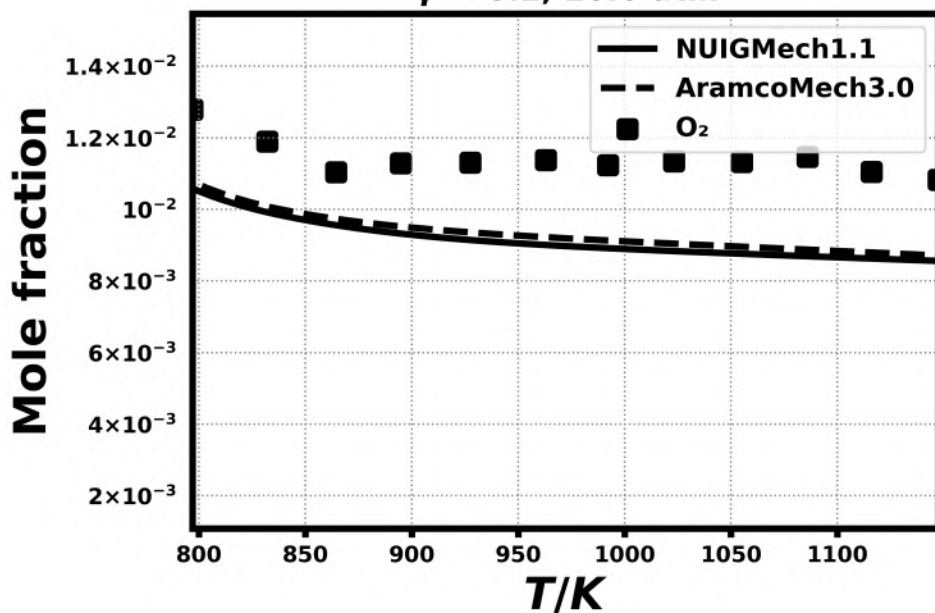
0.2% CH₃OH, 0.02658% CH₂O
0.3% O₂, 99.5% N₂
 $\phi = 1.00, 1.0 \text{ atm}$



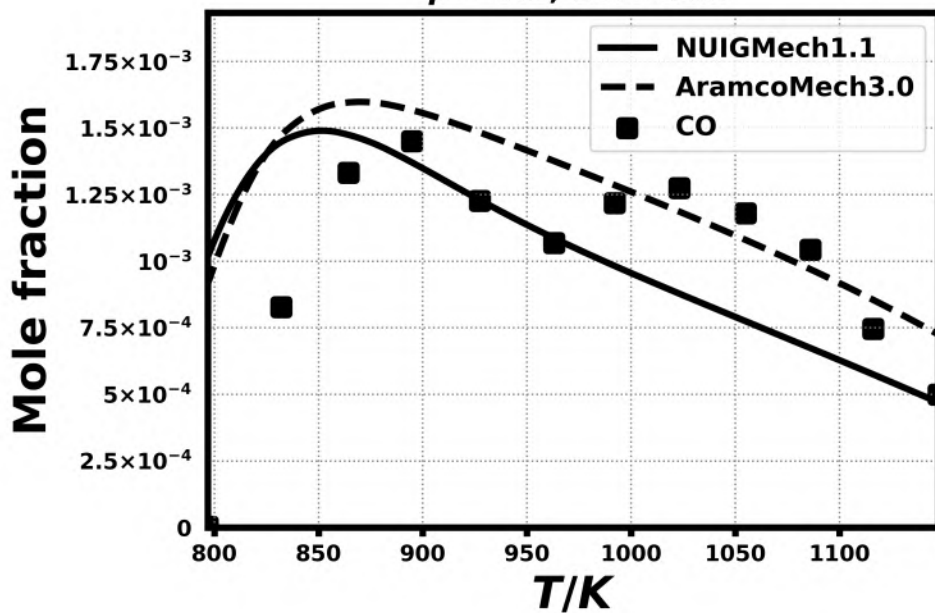
0.244% CH₃OH, 0.006384% CH₂O, 0.000324% CO
1.2% O₂, 98.3% N₂
 $\phi = 0.2, 20.0 \text{ atm}$



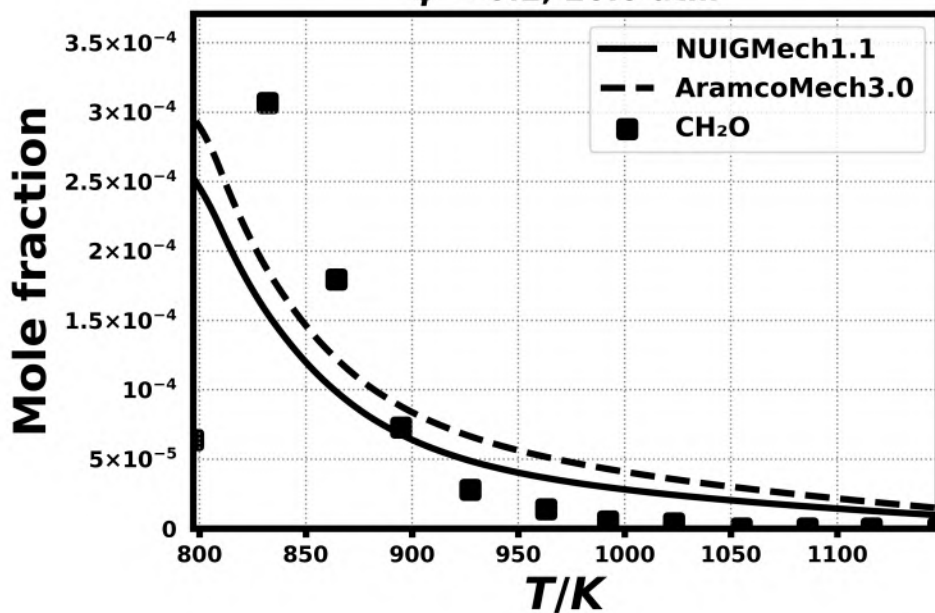
0.244% CH₃OH, 0.006384% CH₂O, 0.000324% CO
1.2% O₂, 98.3% N₂
 $\phi = 0.2, 20.0 \text{ atm}$



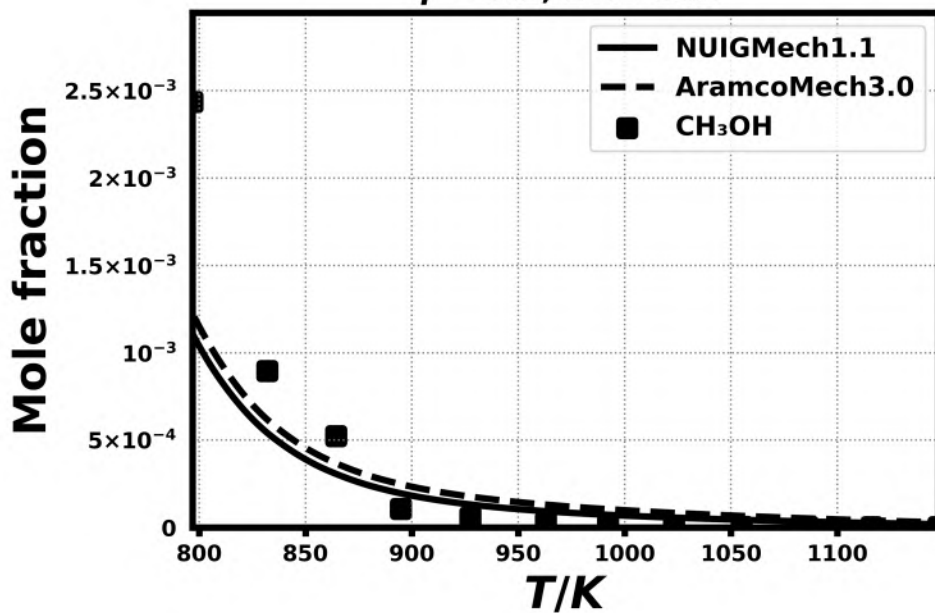
0.244% CH₃OH, 0.006384% CH₂O, 0.000324% CO
1.2% O₂, 98.3% N₂
 $\phi = 0.2, 20.0 \text{ atm}$



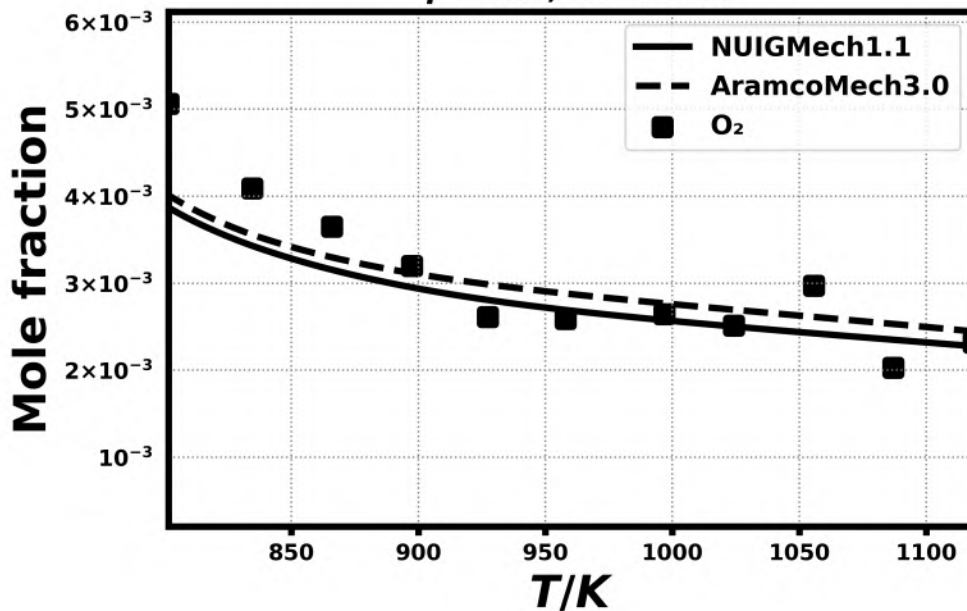
0.244% CH₃OH, 0.006384% CH₂O, 0.000324% CO
1.2% O₂, 98.3% N₂
 $\phi = 0.2$, 20.0 atm



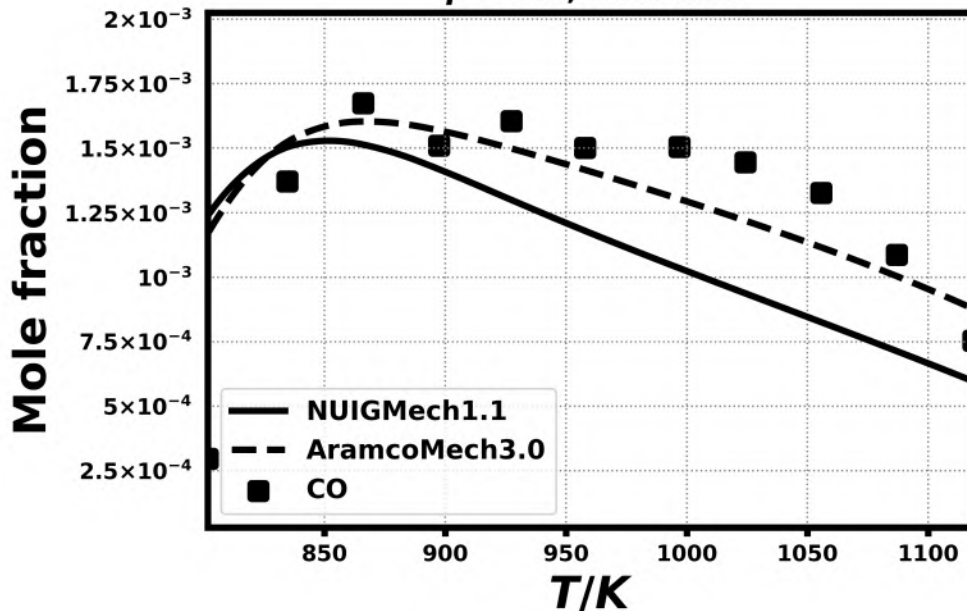
0.244% CH₃OH, 0.006384% CH₂O, 0.000324% CO
1.2% O₂, 98.3% N₂
 $\phi = 0.2$, 20.0 atm



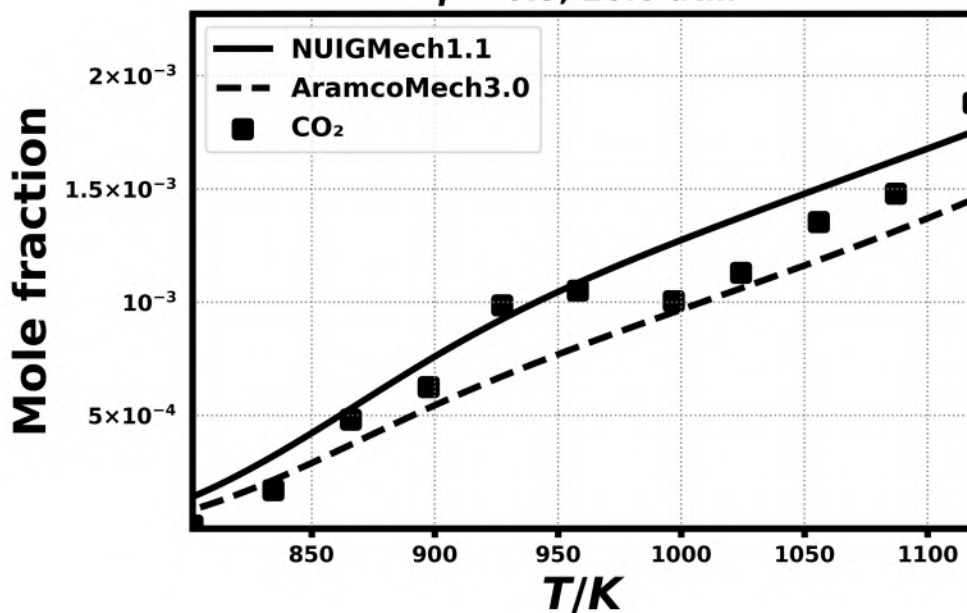
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



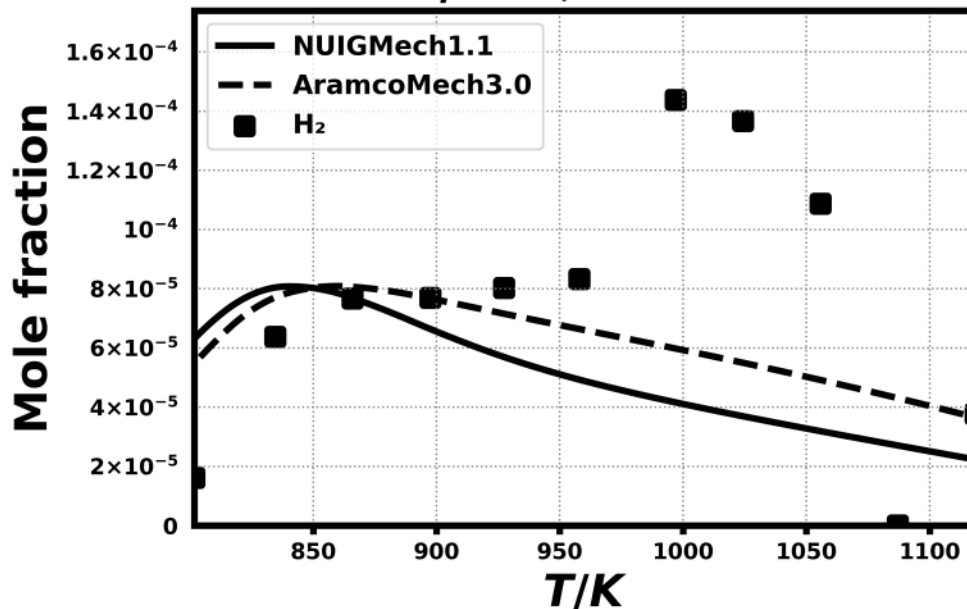
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



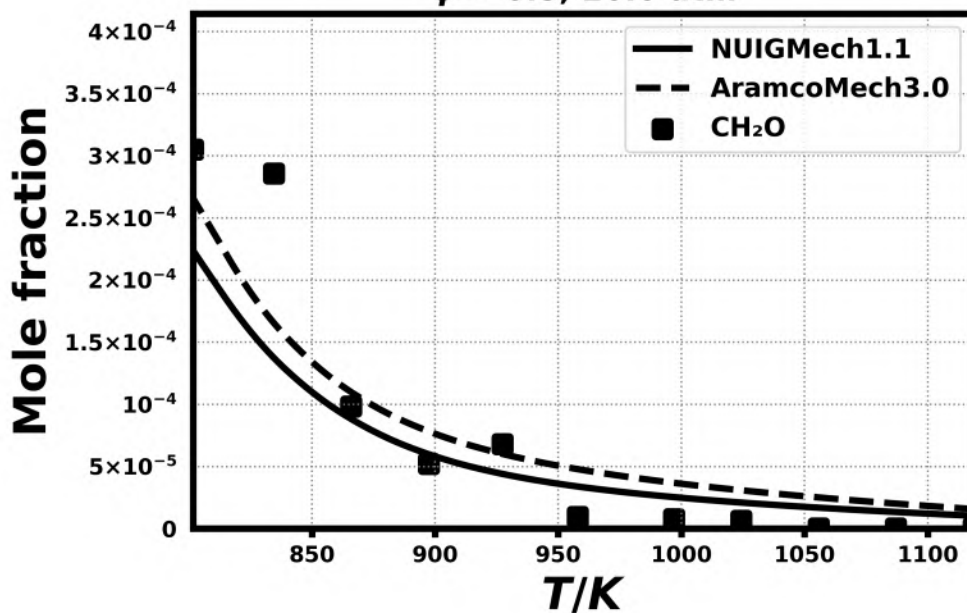
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



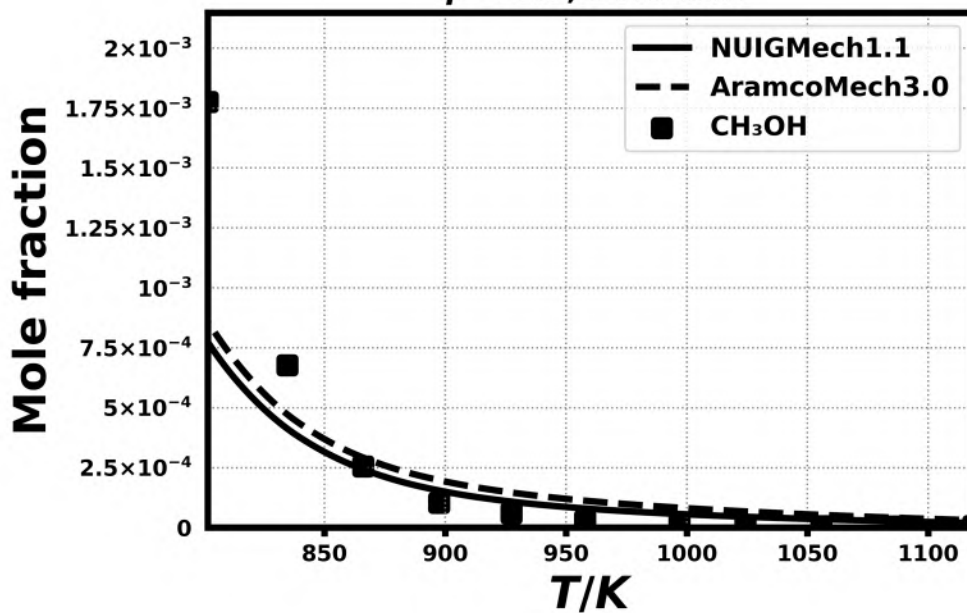
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



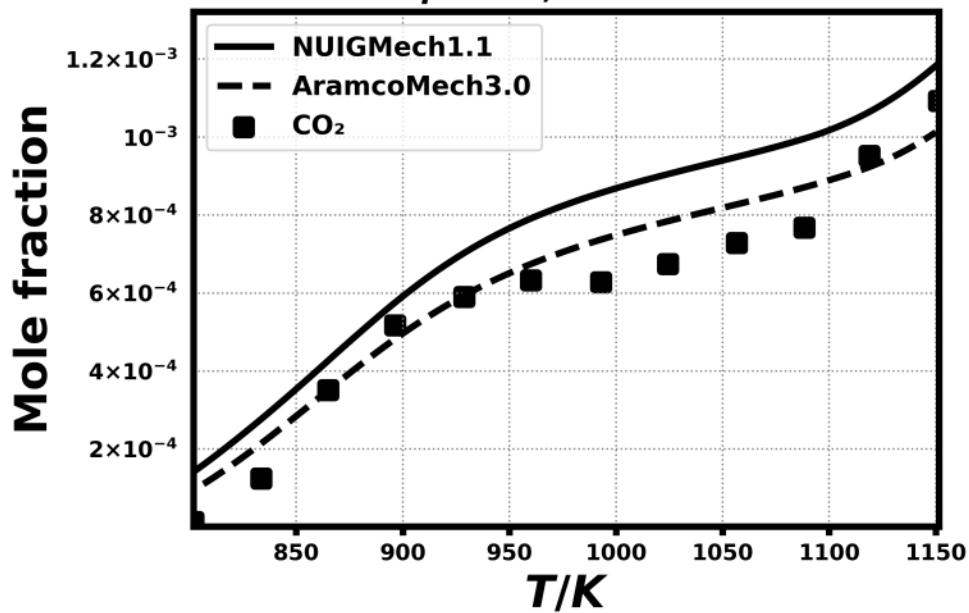
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



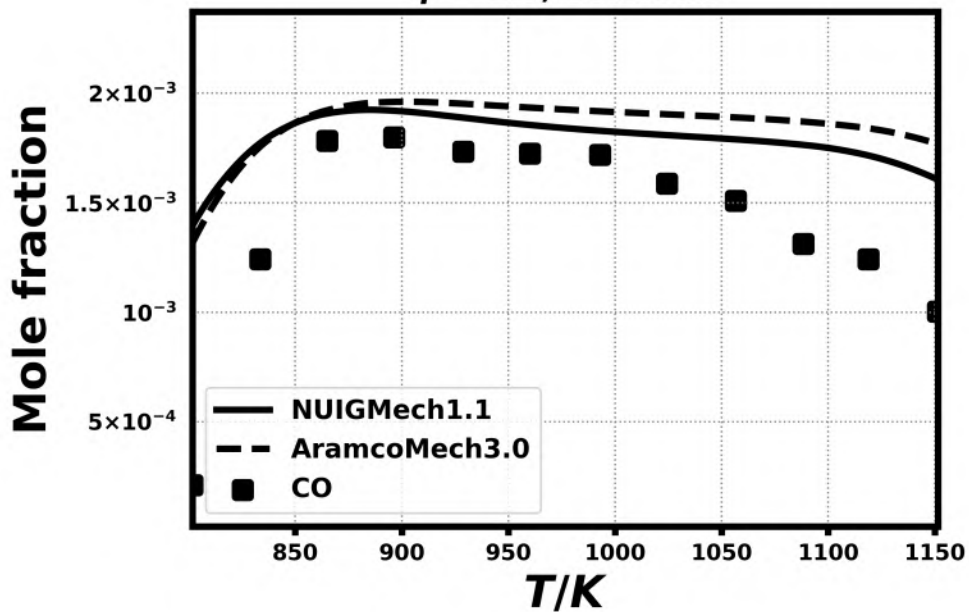
0.178% CH₃OH, 0.0305% CH₂O, 0.02959% CO
0.506% O₂, 99.2% N₂
 $\phi = 0.5, 20.0 \text{ atm}$



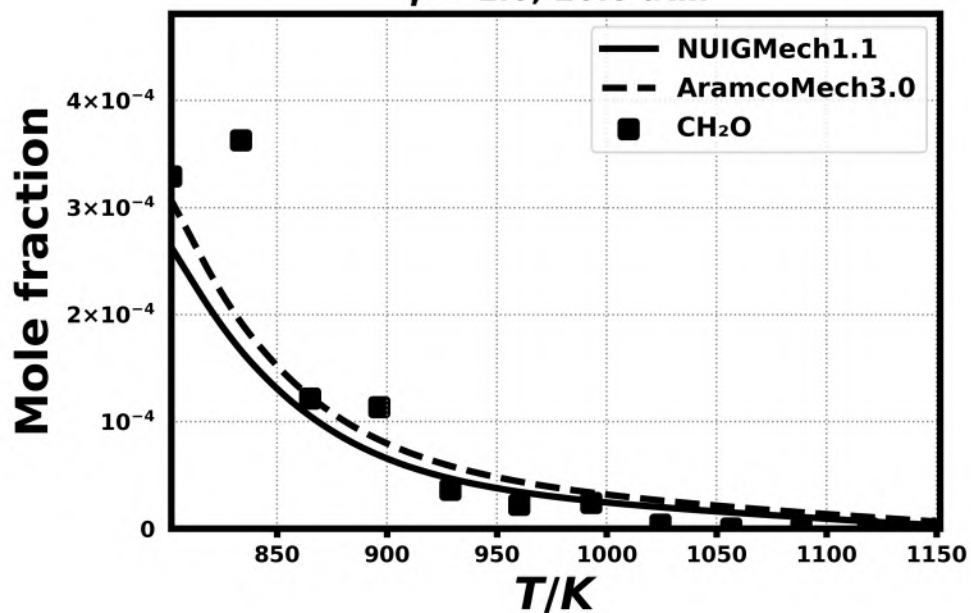
0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
0.27% O₂, 99.5% N₂
 $\phi = 1.0$, 20.0 atm



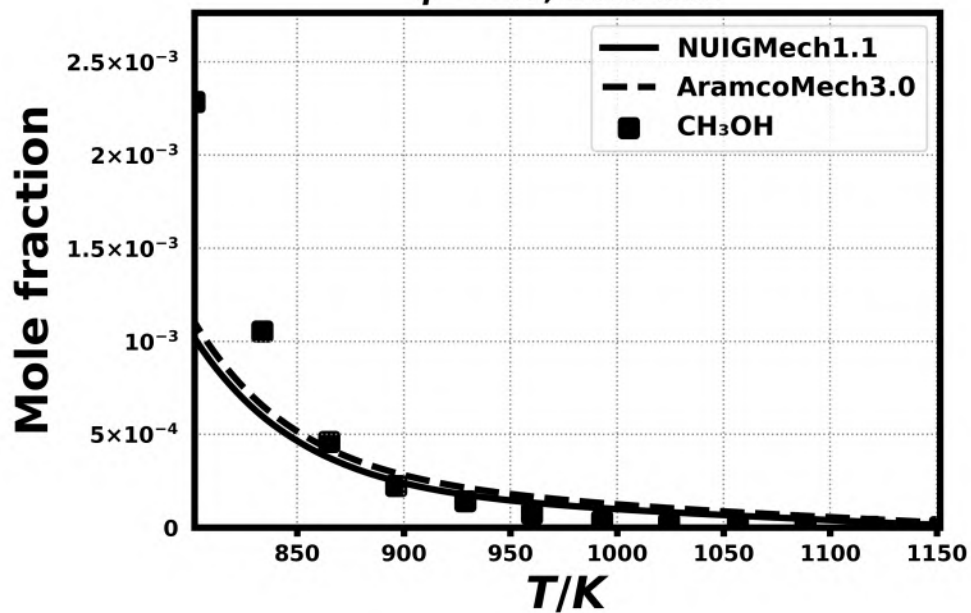
0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
0.27% O₂, 99.5% N₂
 $\phi = 1.0$, 20.0 atm



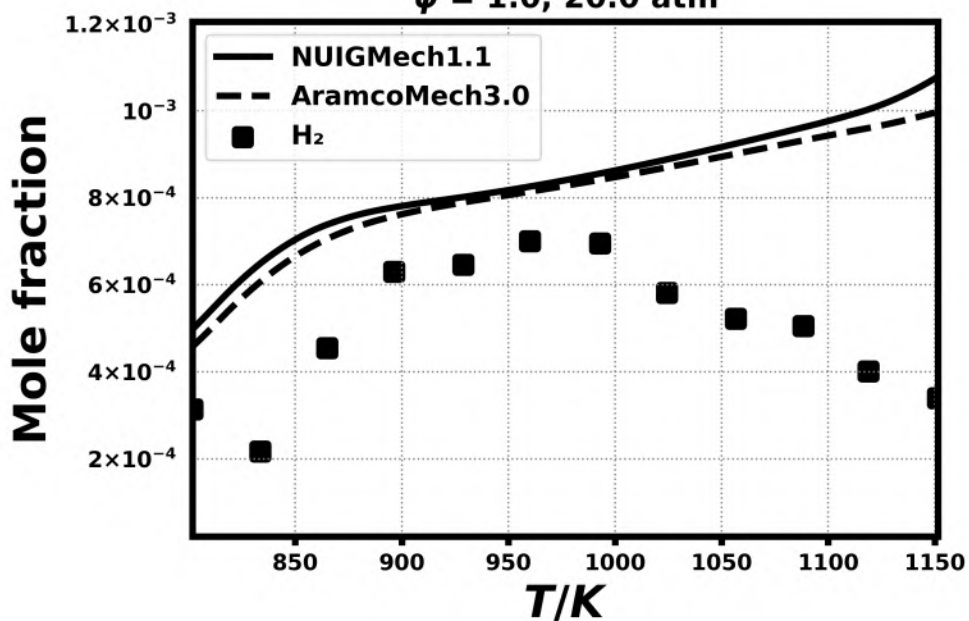
0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
0.27% O₂, 99.5% N₂
 $\phi = 1.0, 20.0 \text{ atm}$



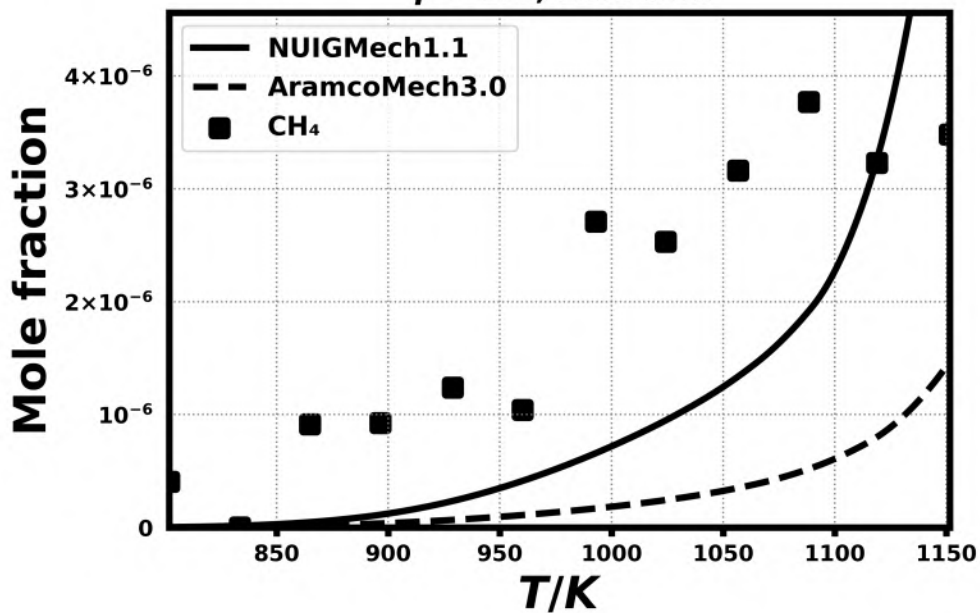
0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
0.27% O₂, 99.5% N₂
 $\phi = 1.0, 20.0 \text{ atm}$



0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
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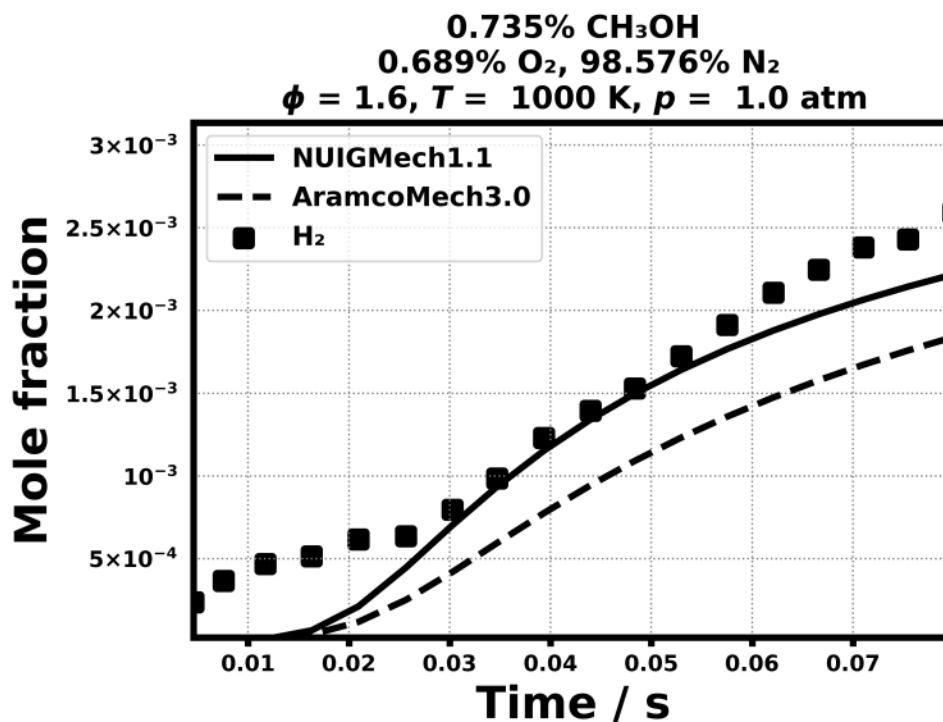
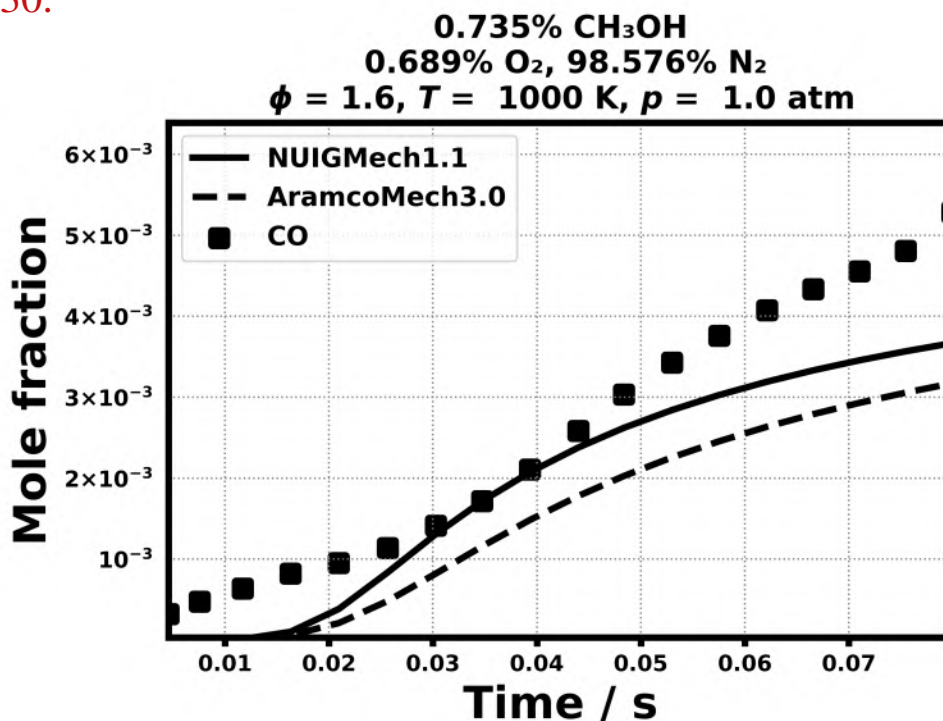


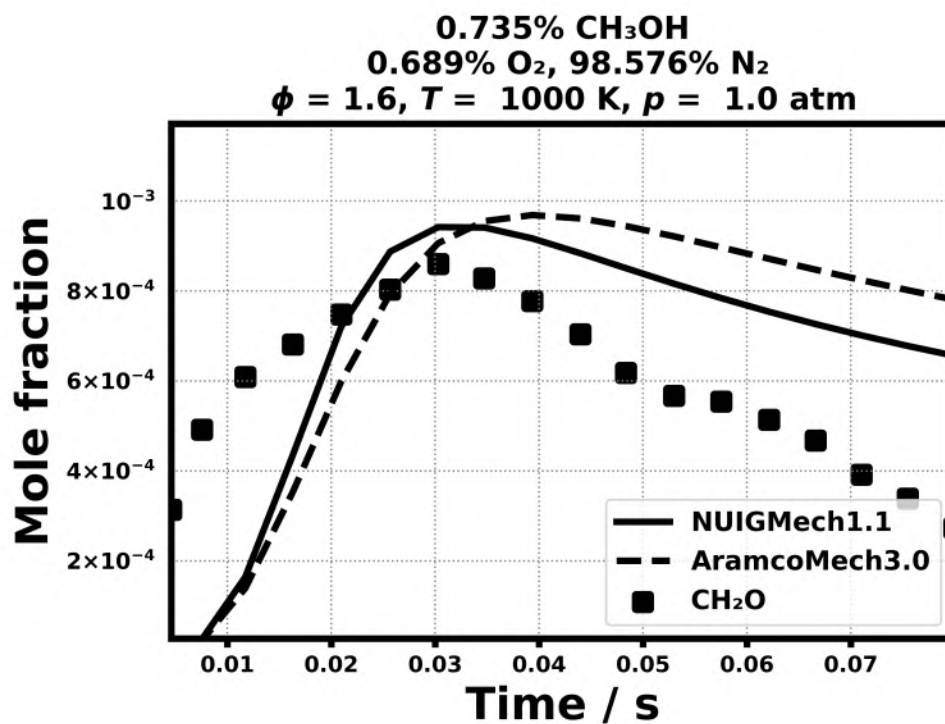
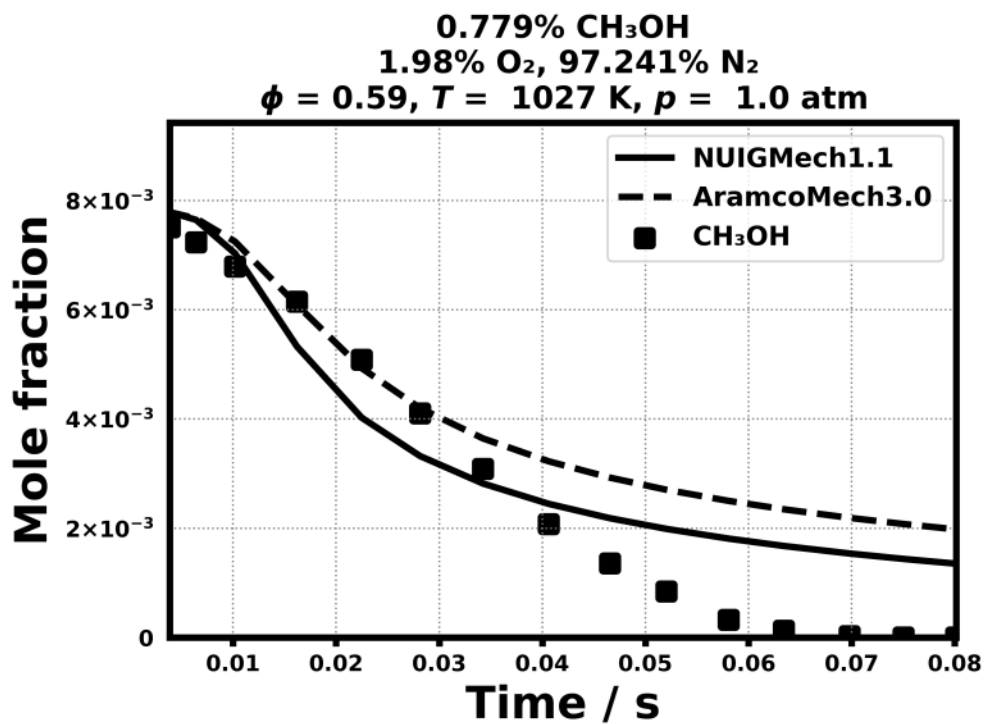
0.229% CH₃OH, 0.032918% CH₂O, 0.020953% CO
0.27% O₂, 99.5% N₂
 $\phi = 1.0, 20.0 \text{ atm}$



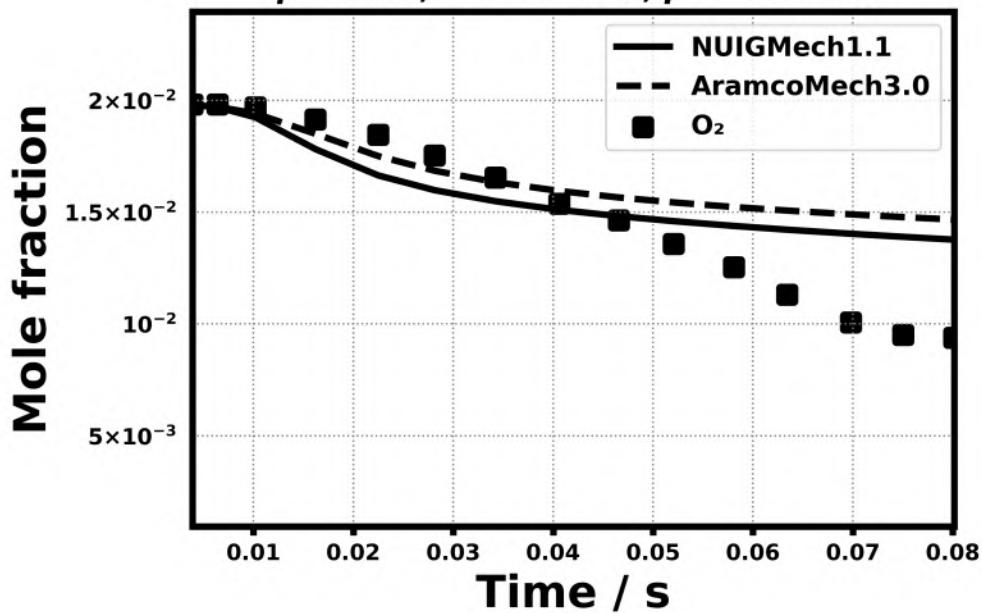
Speciation in Flow reactor

4.9) Held, T. J., & Dryer, F. L., International Journal of Chemical Kinetics, 30(11) (1998) 805-830.

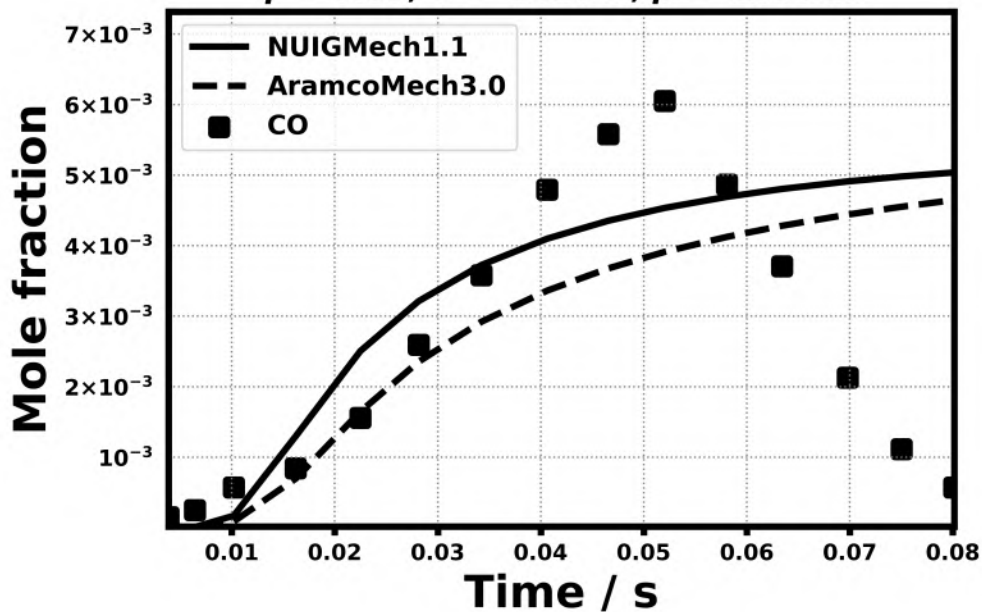




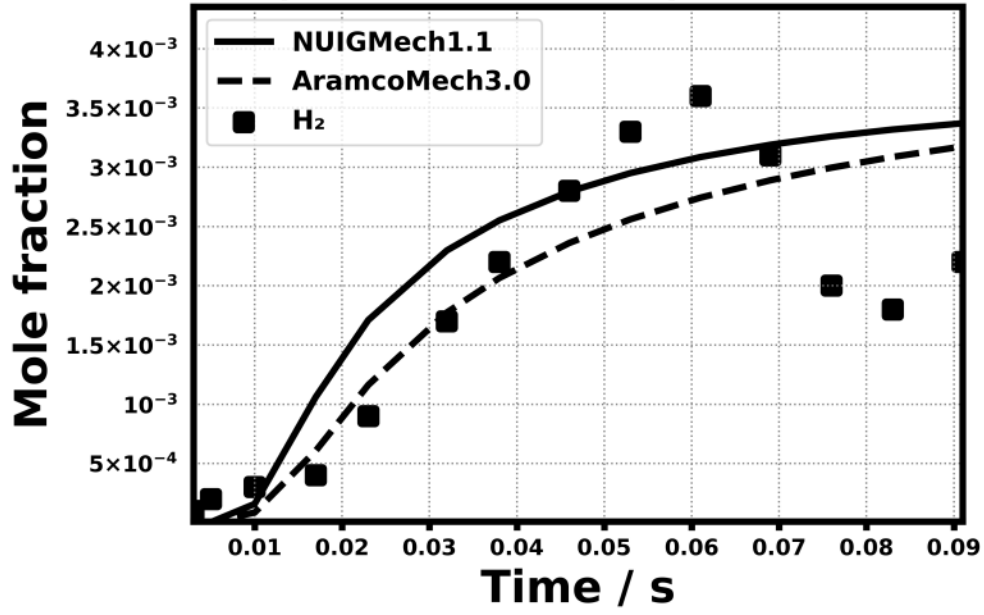
0.779% CH₃OH
1.98% O₂, 97.241% N₂
 $\phi = 0.59, T = 1027 \text{ K}, p = 1.0 \text{ atm}$



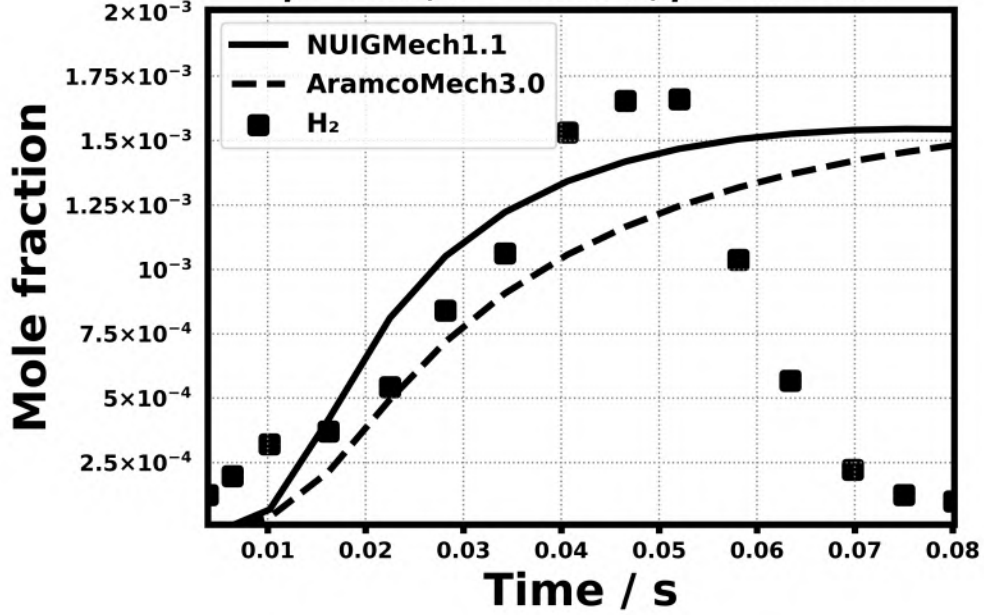
0.779% CH₃OH
1.98% O₂, 97.241% N₂
 $\phi = 0.59, T = 1027 \text{ K}, p = 1.0 \text{ atm}$



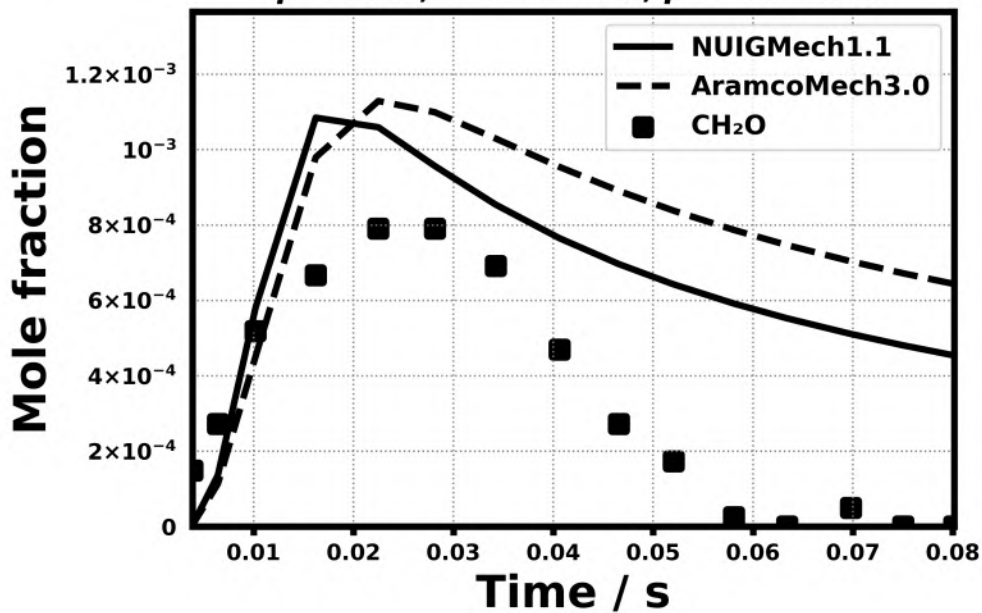
0.943% CH₃OH
1.159% O₂, 97.898% N₂
 $\phi = 1.22, T = 1030 \text{ K}, p = 1.0 \text{ atm}$



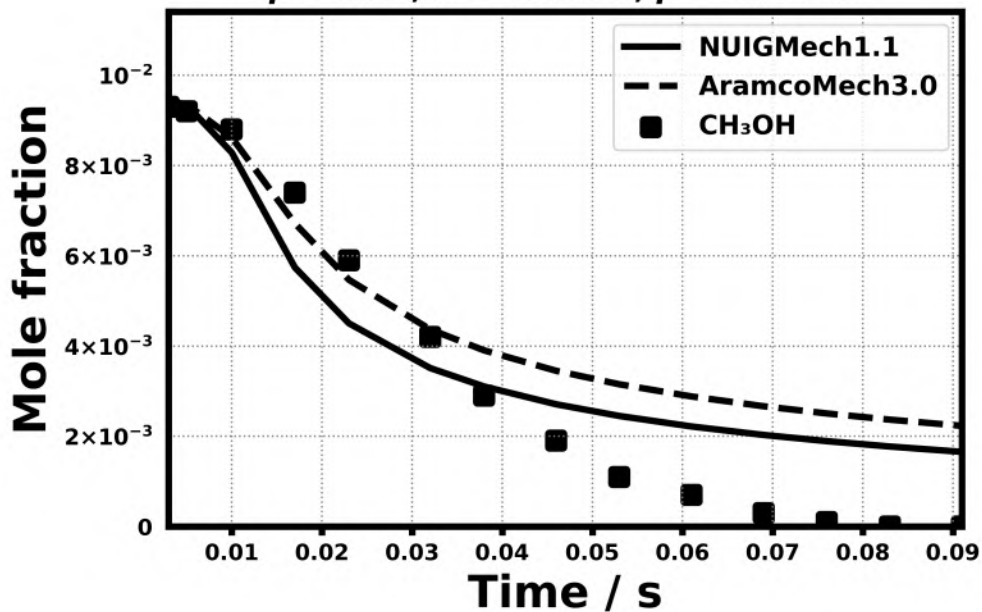
0.779% CH₃OH
1.98% O₂, 97.241% N₂
 $\phi = 0.59, T = 1027 \text{ K}, p = 1.0 \text{ atm}$

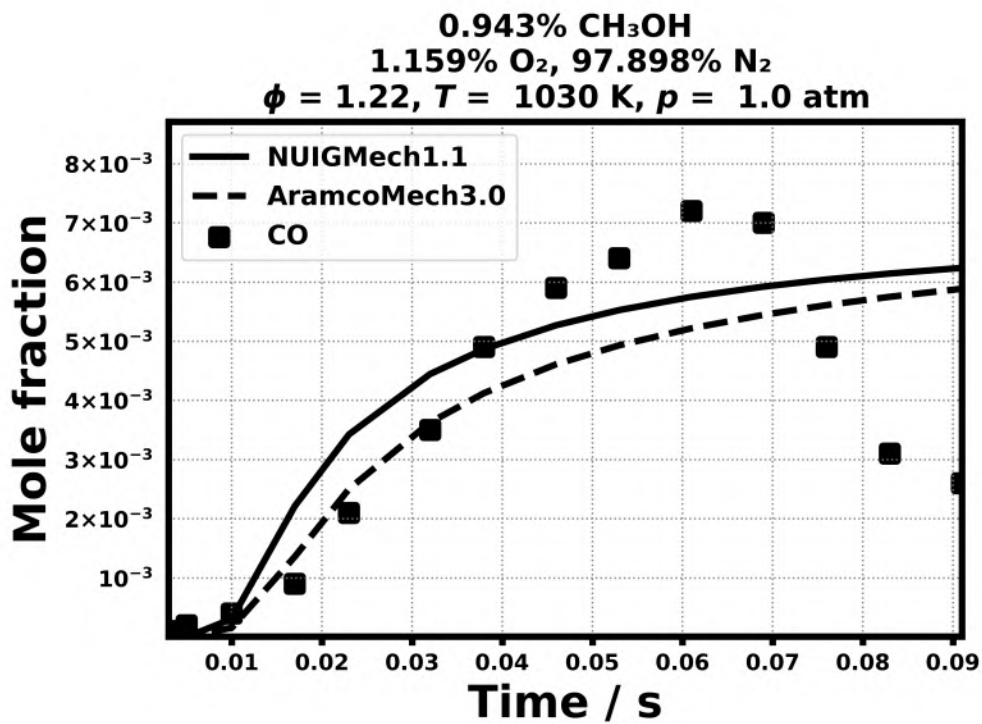
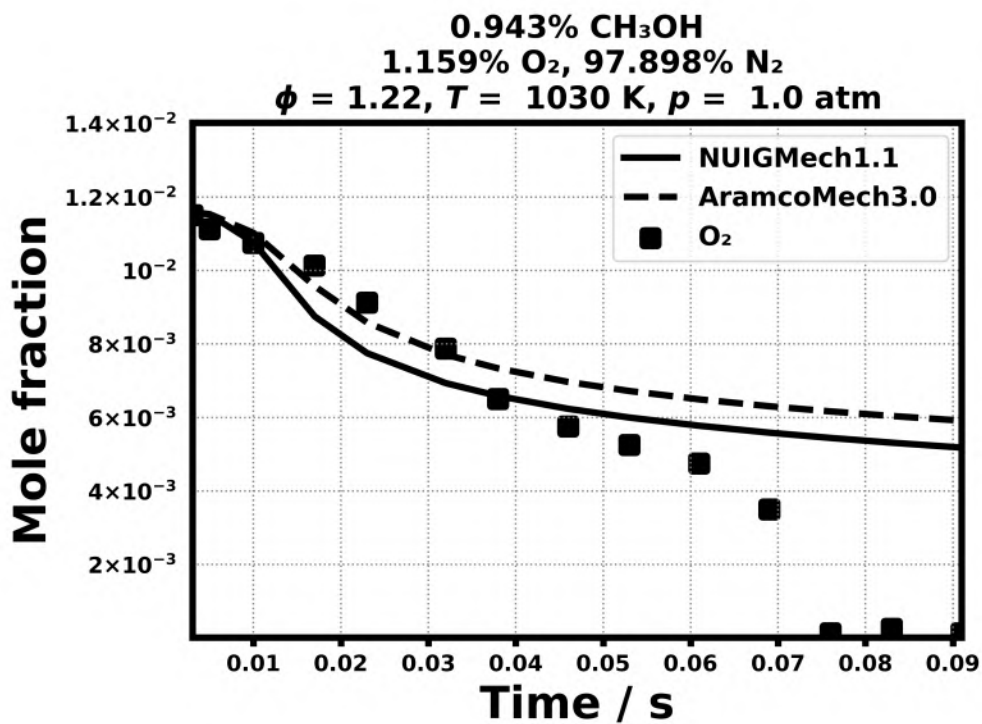


0.779% CH₃OH
1.98% O₂, 97.241% N₂
 $\phi = 0.59, T = 1027 \text{ K}, p = 1.0 \text{ atm}$

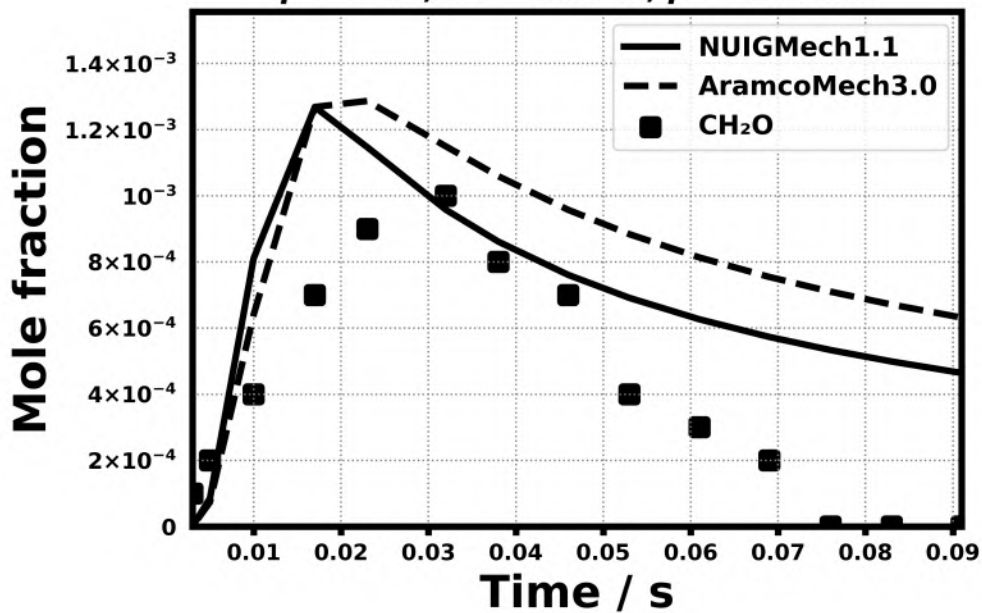


0.943% CH₃OH
1.159% O₂, 97.898% N₂
 $\phi = 1.22, T = 1030 \text{ K}, p = 1.0 \text{ atm}$

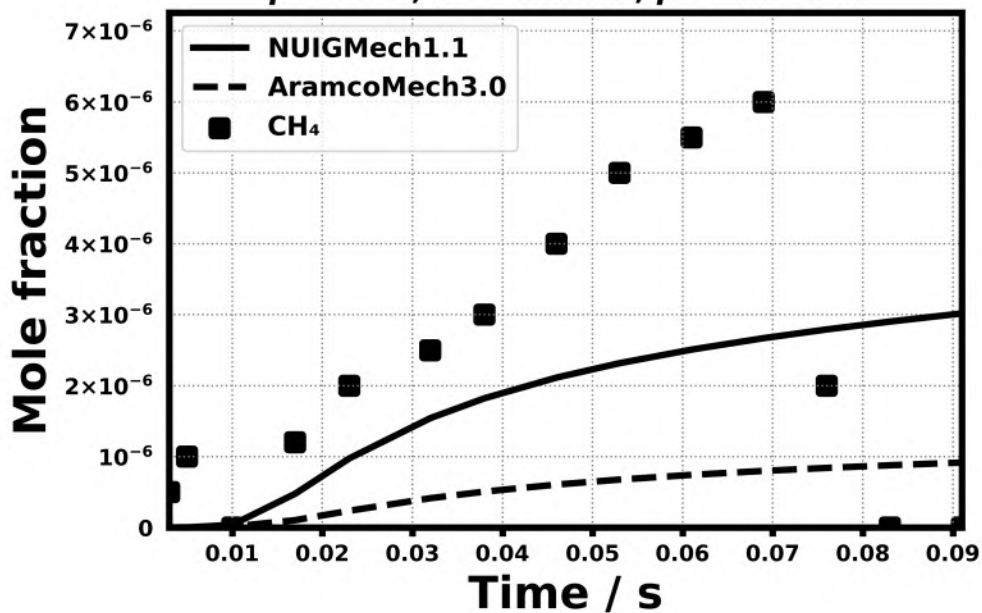


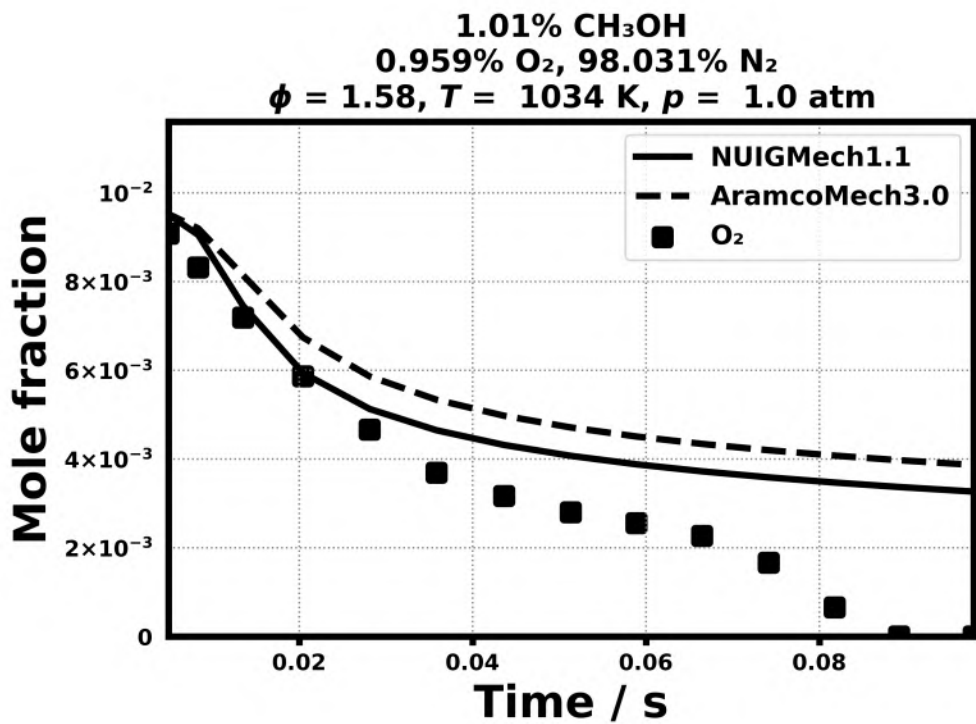
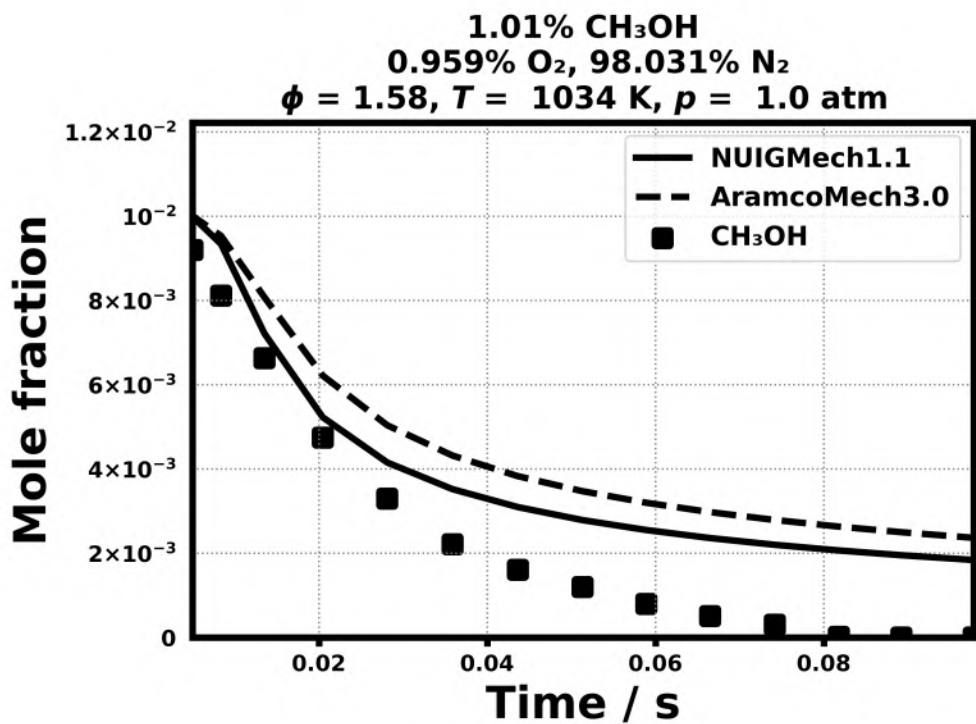


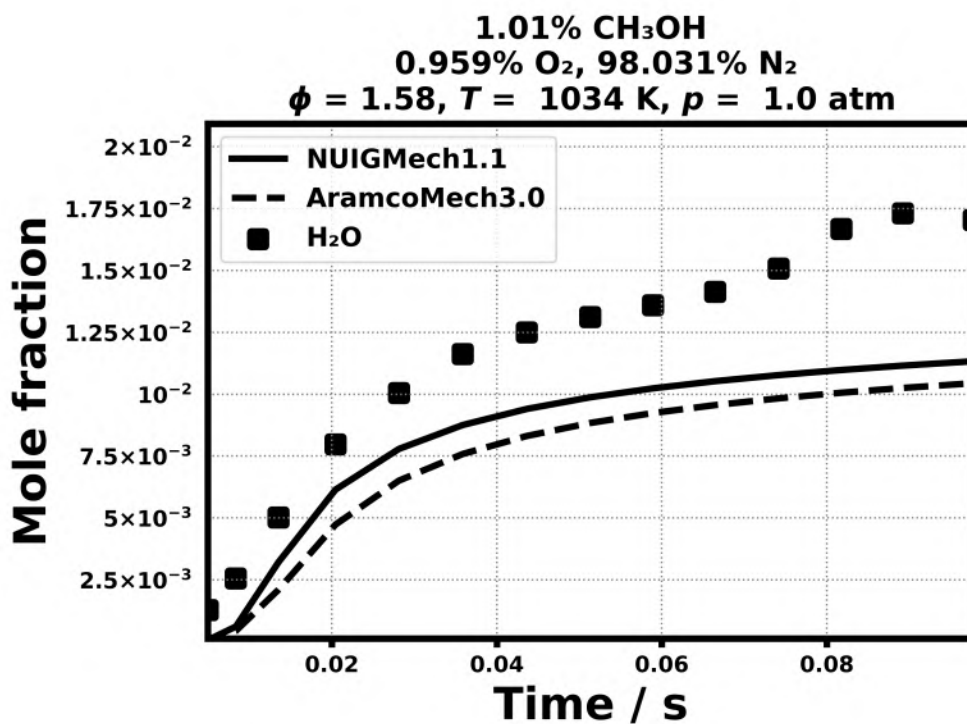
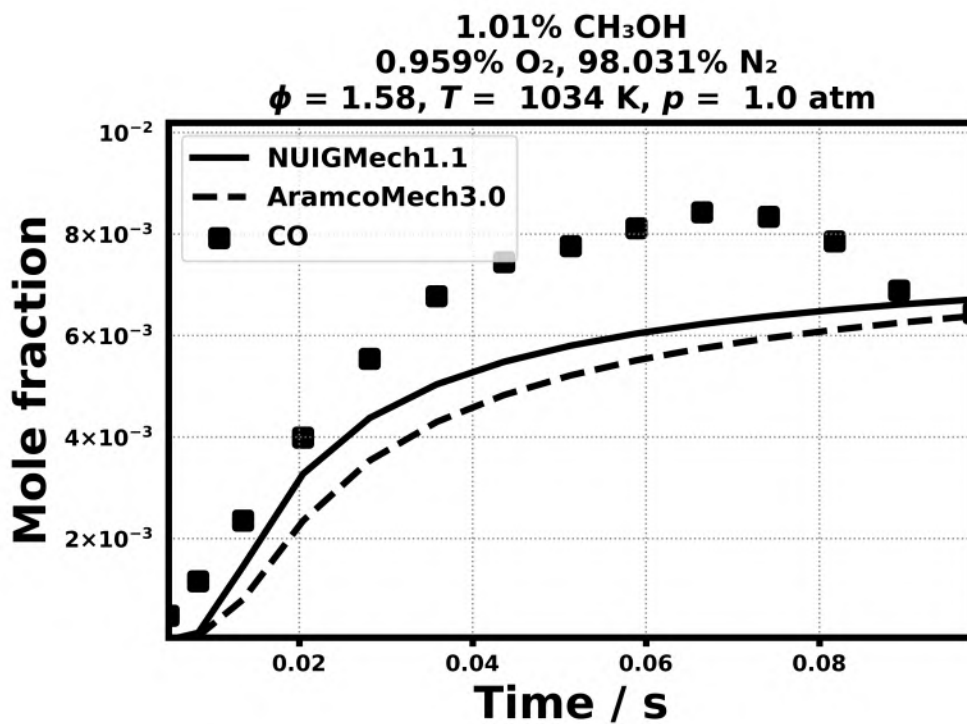
0.943% CH₃OH
1.159% O₂, 97.898% N₂
 $\phi = 1.22, T = 1030 \text{ K}, p = 1.0 \text{ atm}$



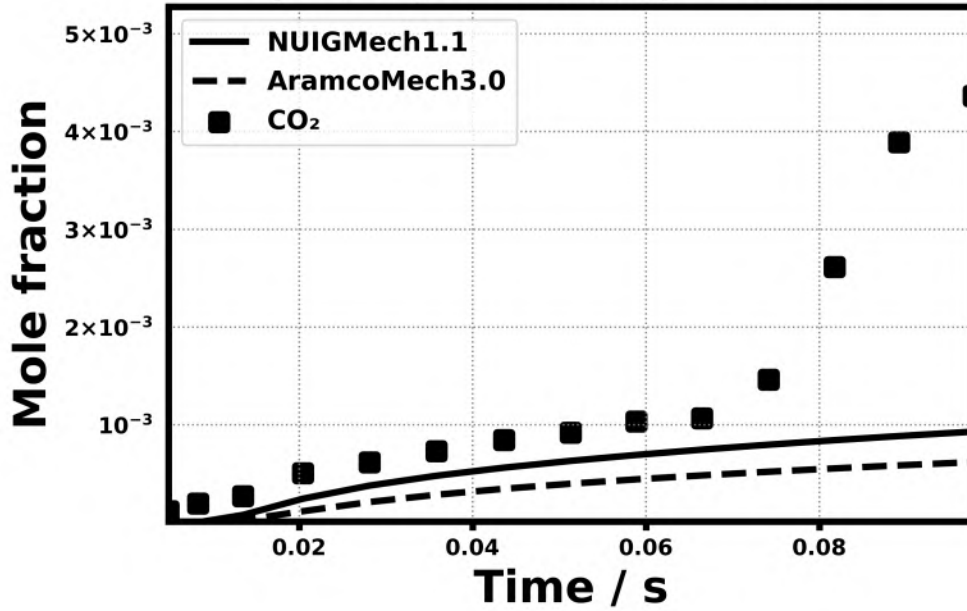
0.943% CH₃OH
1.159% O₂, 97.898% N₂
 $\phi = 1.22, T = 1030 \text{ K}, p = 1.0 \text{ atm}$



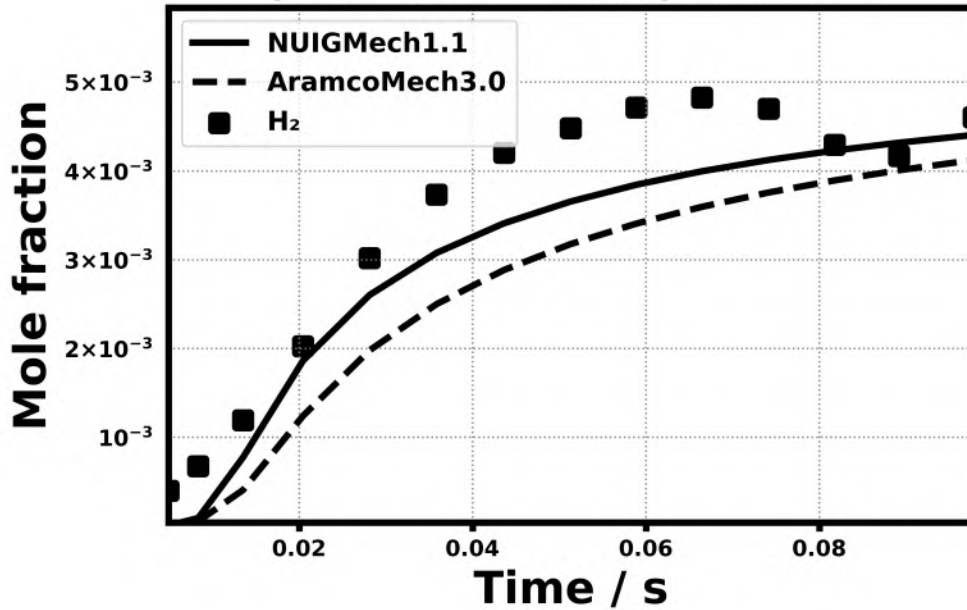


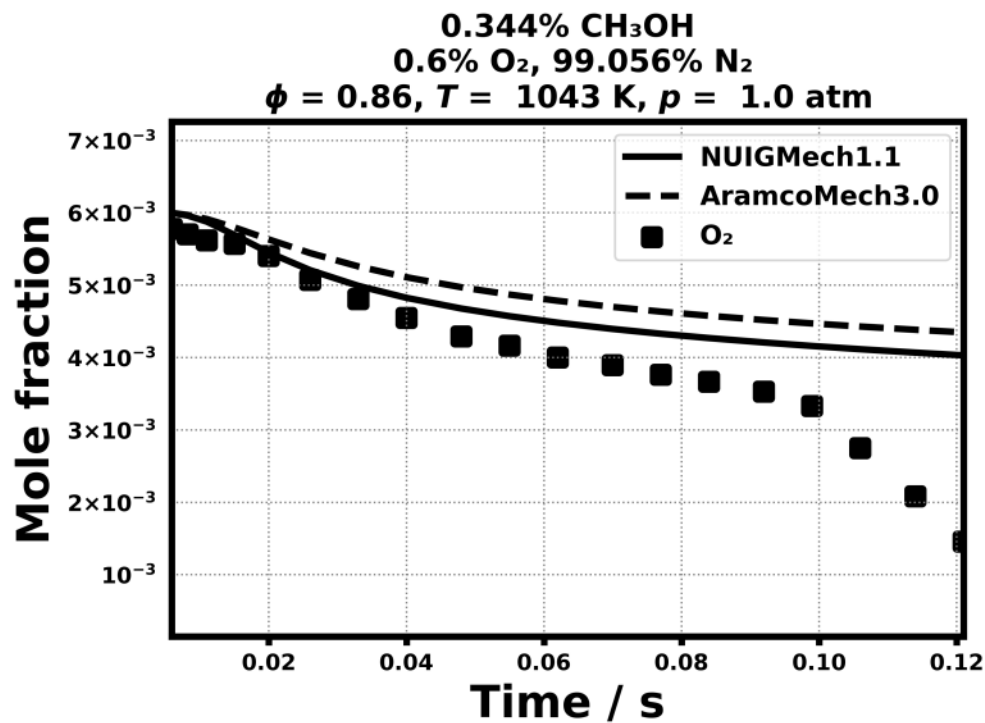
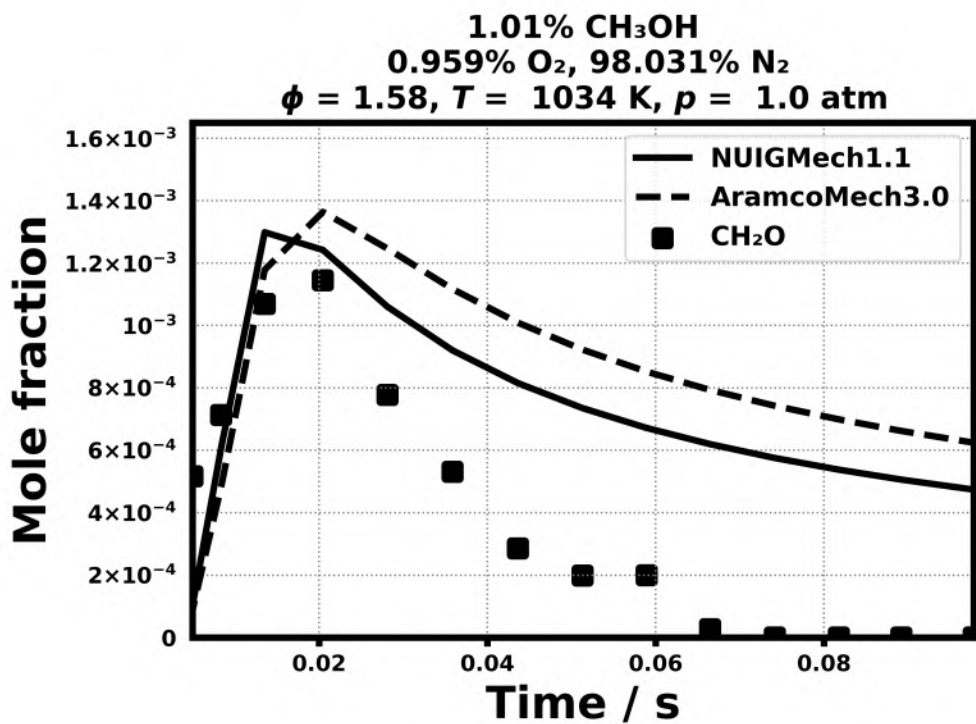


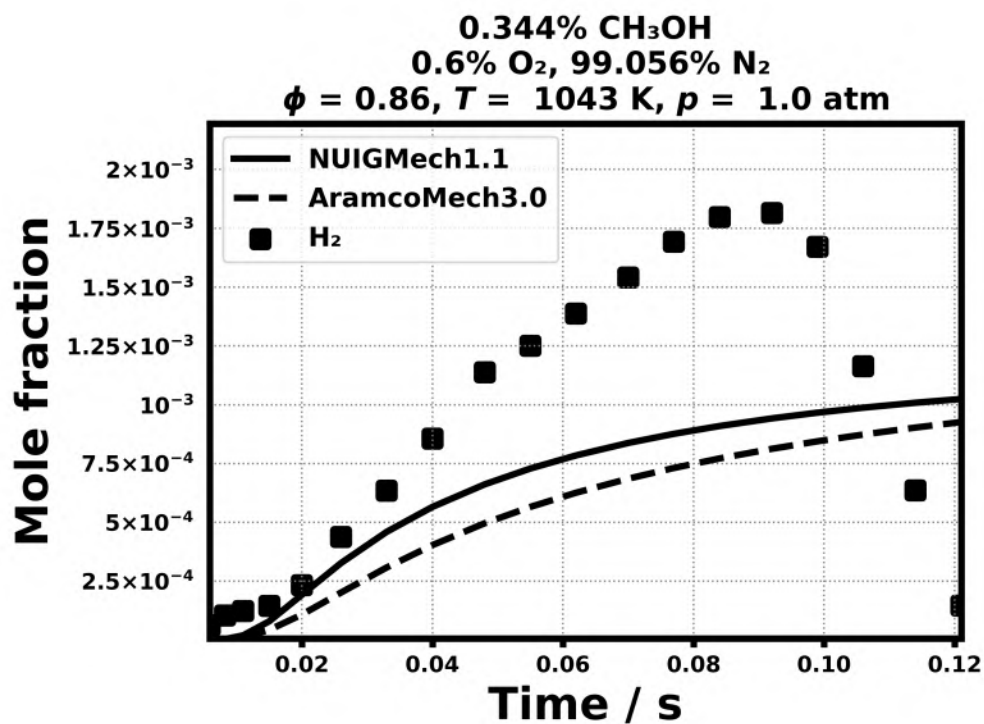
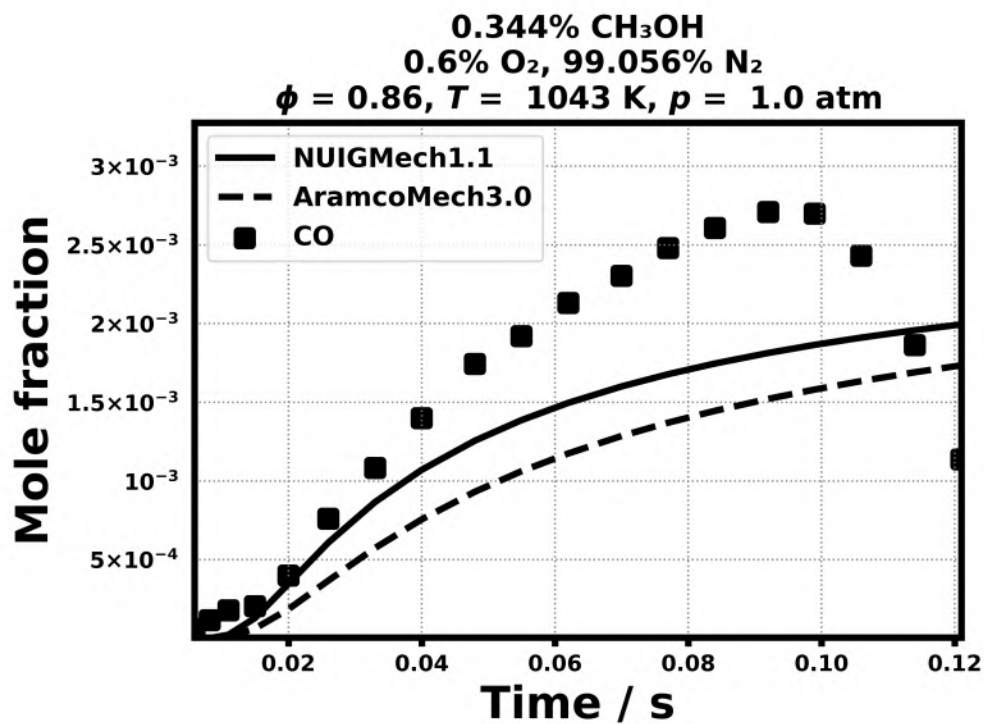
1.01% CH₃OH
0.959% O₂, 98.031% N₂
 $\phi = 1.58, T = 1034 \text{ K}, p = 1.0 \text{ atm}$

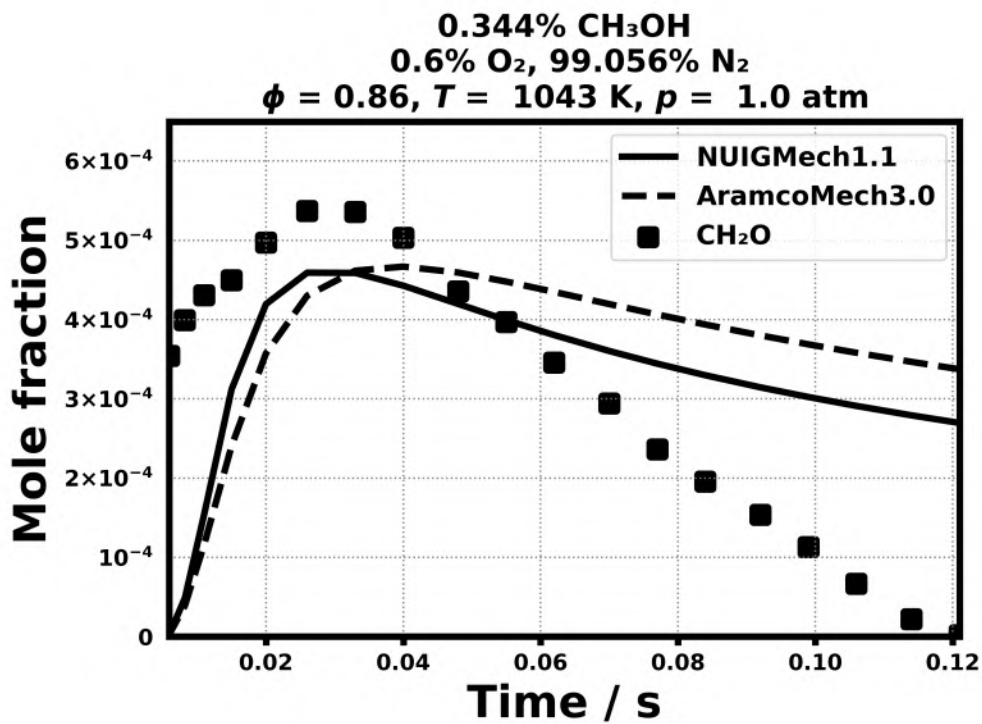
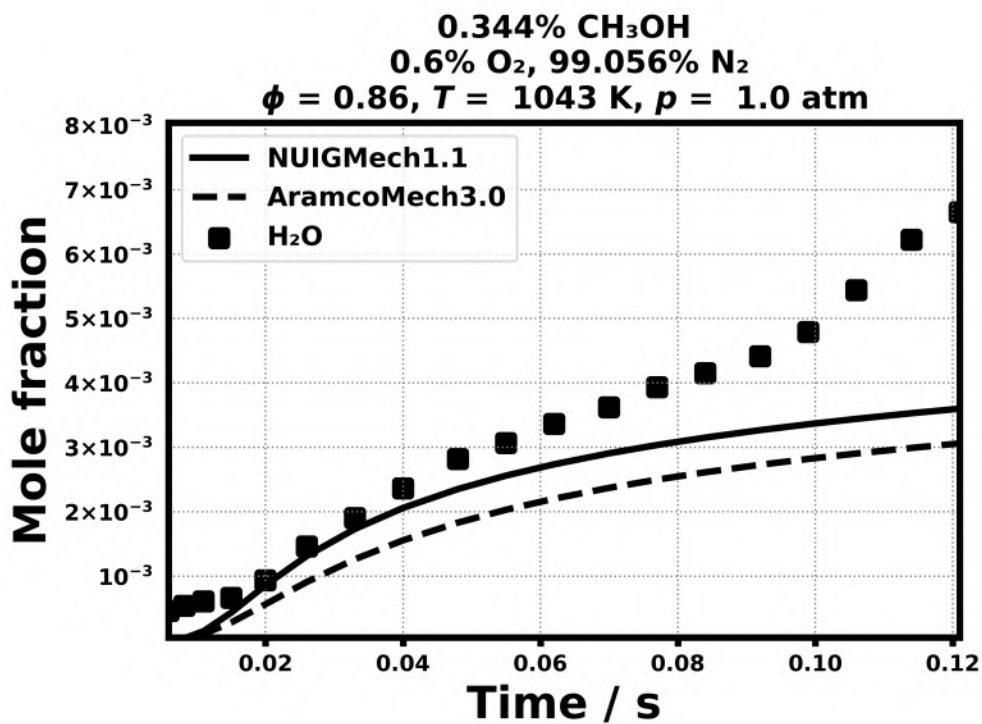


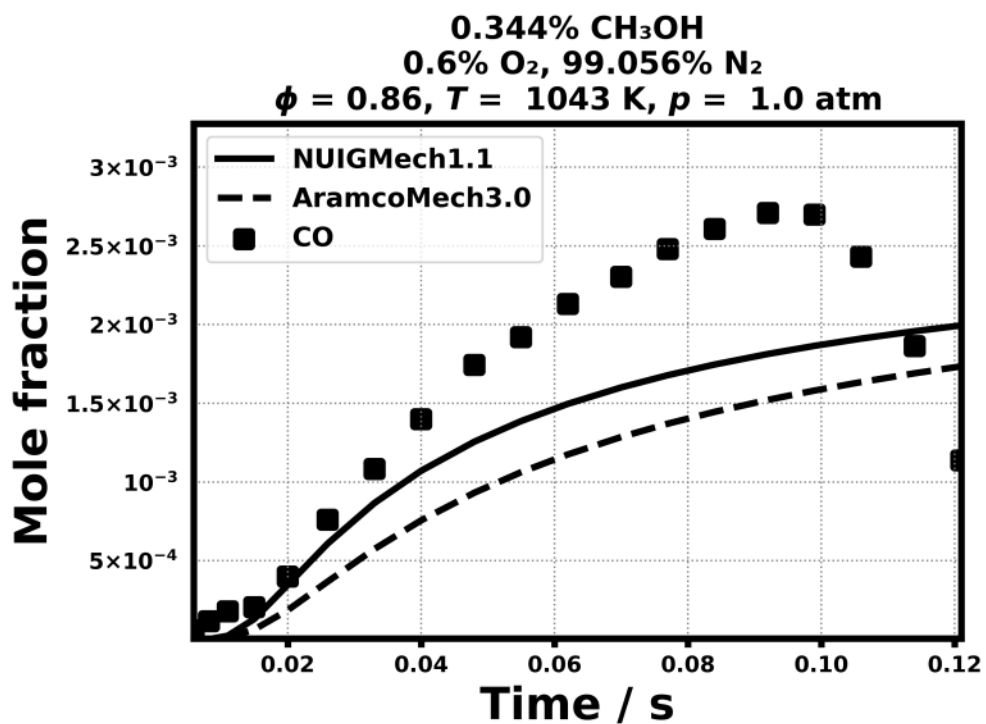
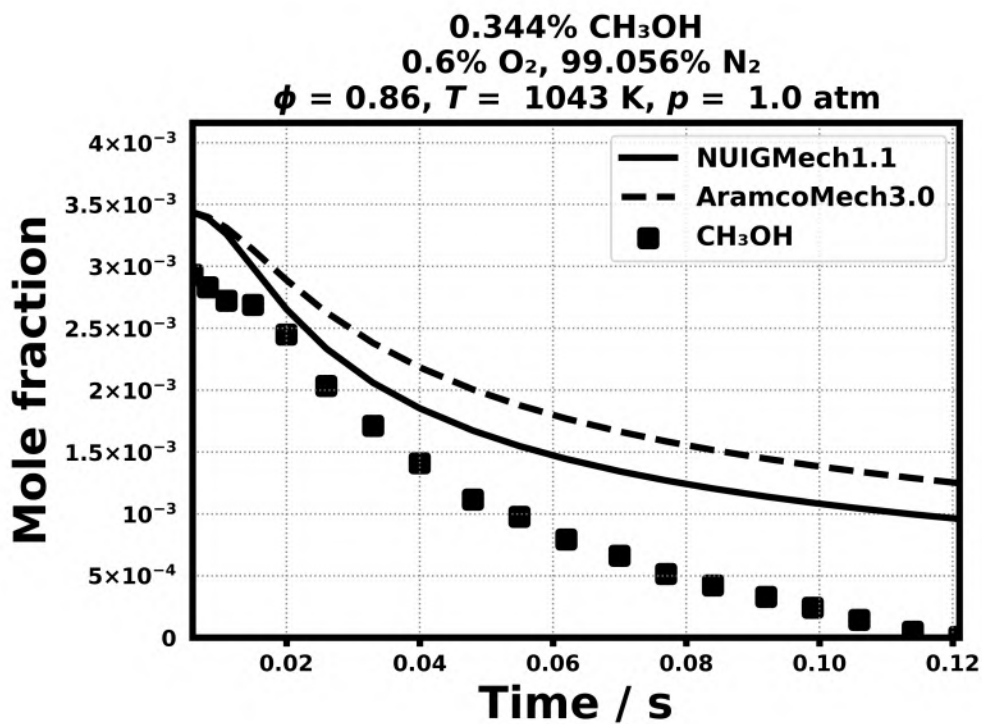
1.01% CH₃OH
0.959% O₂, 98.031% N₂
 $\phi = 1.58, T = 1034 \text{ K}, p = 1.0 \text{ atm}$

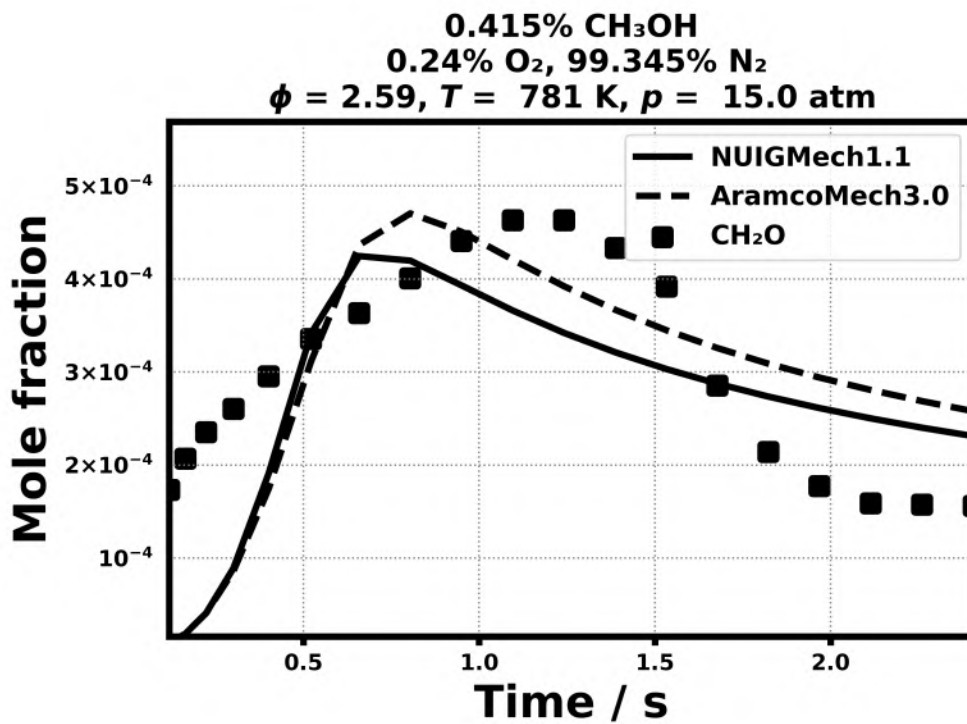
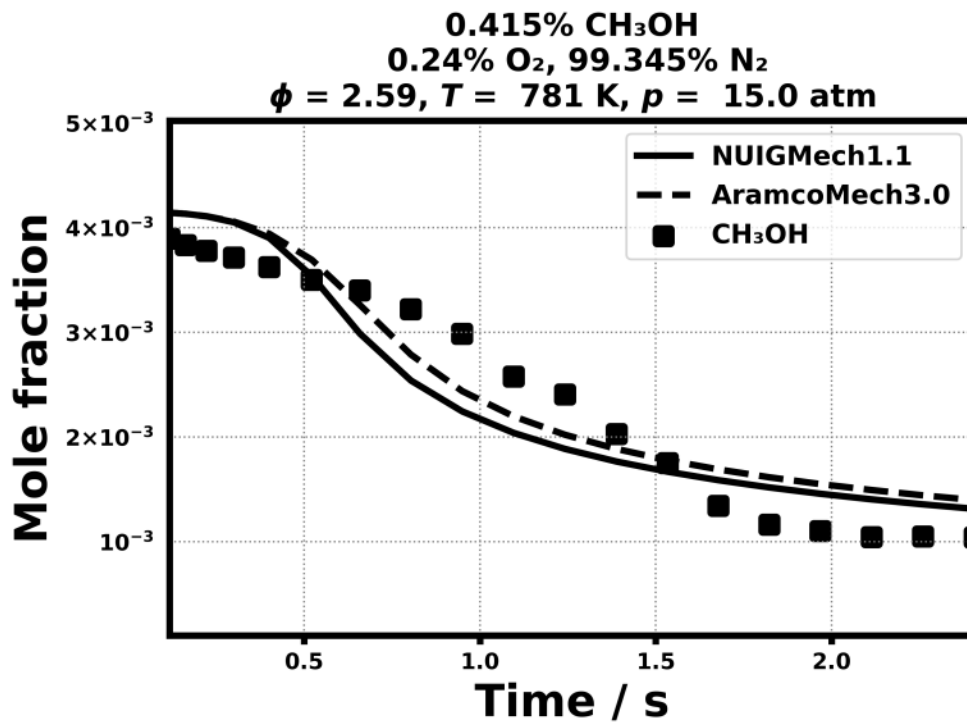




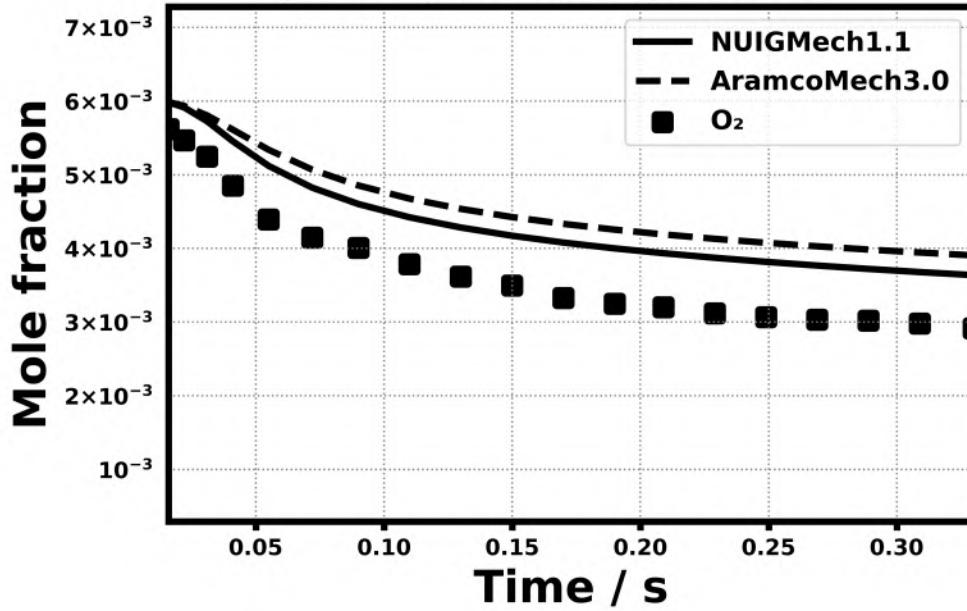




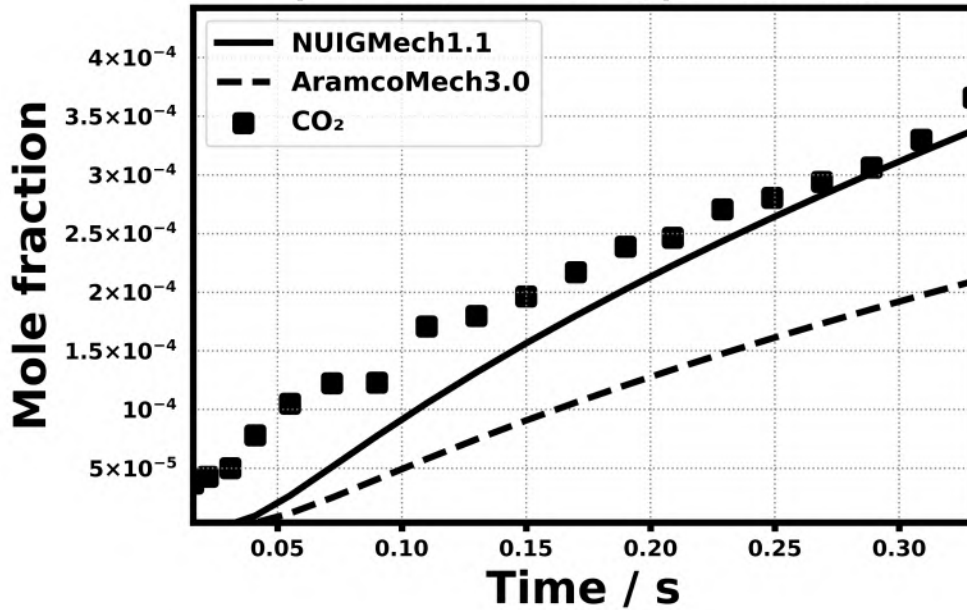




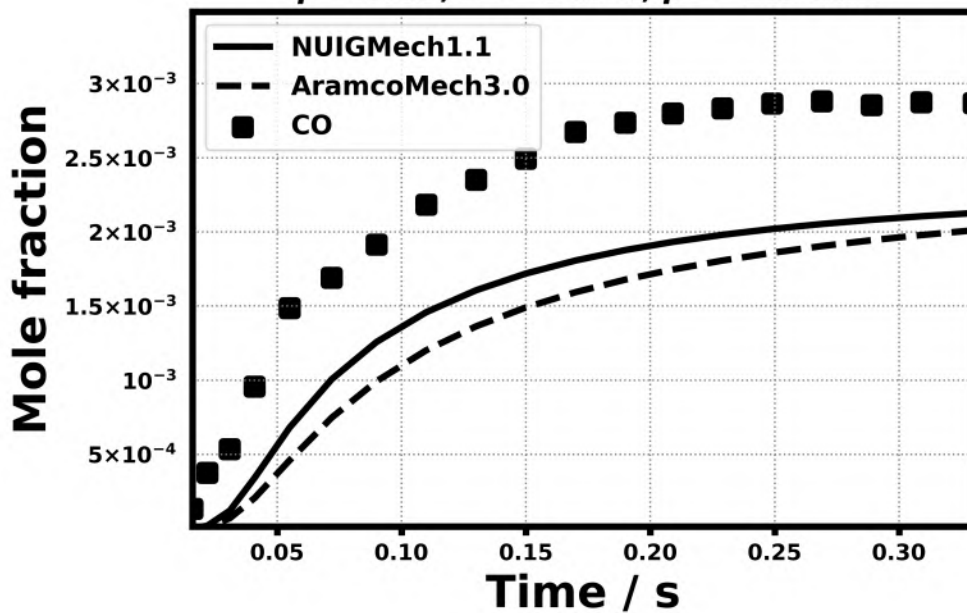
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



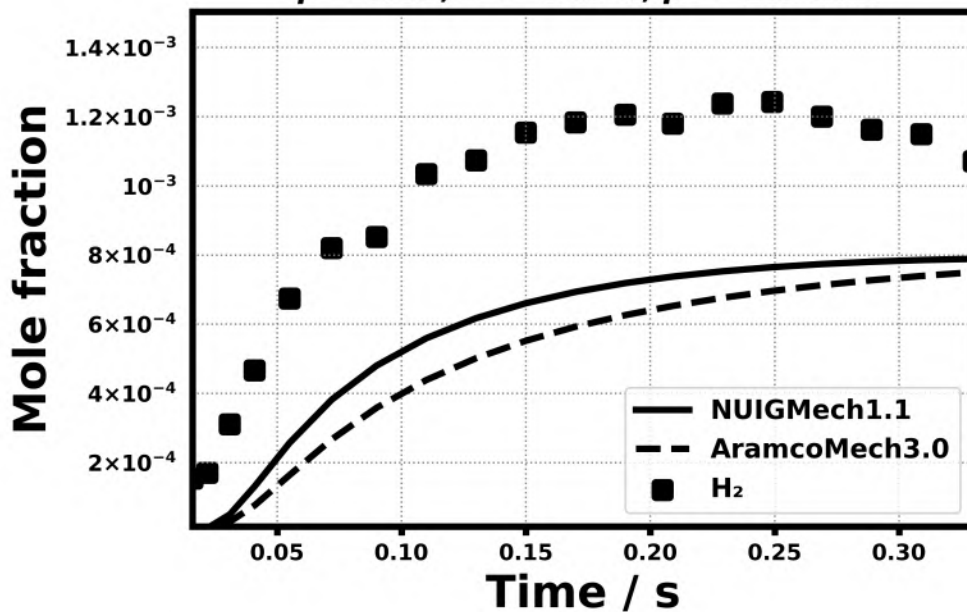
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



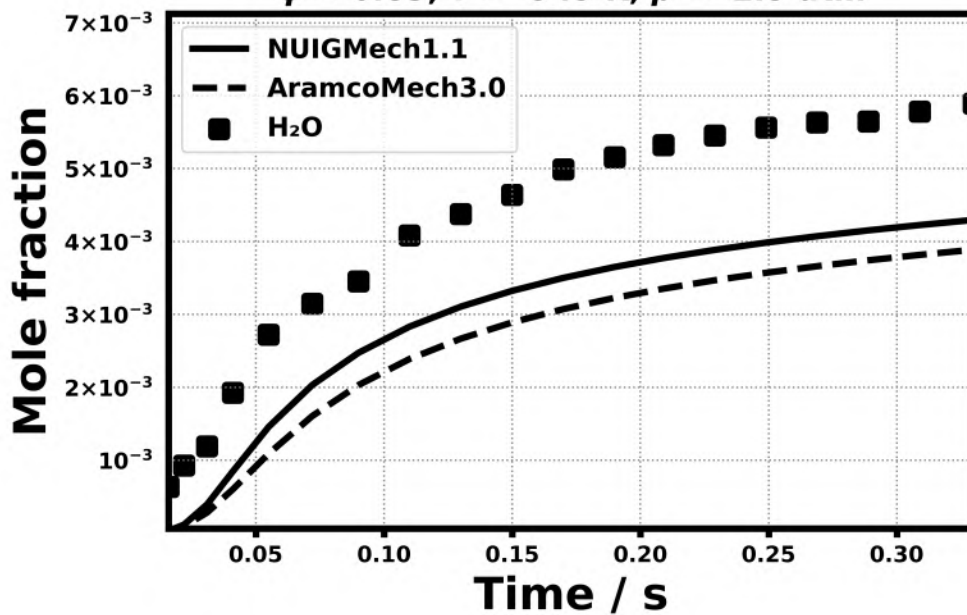
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



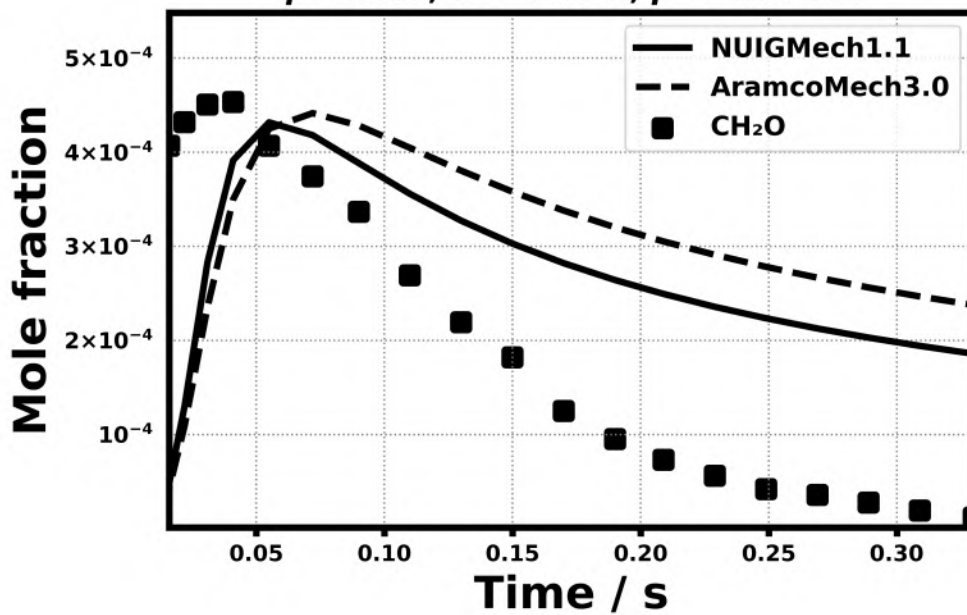
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



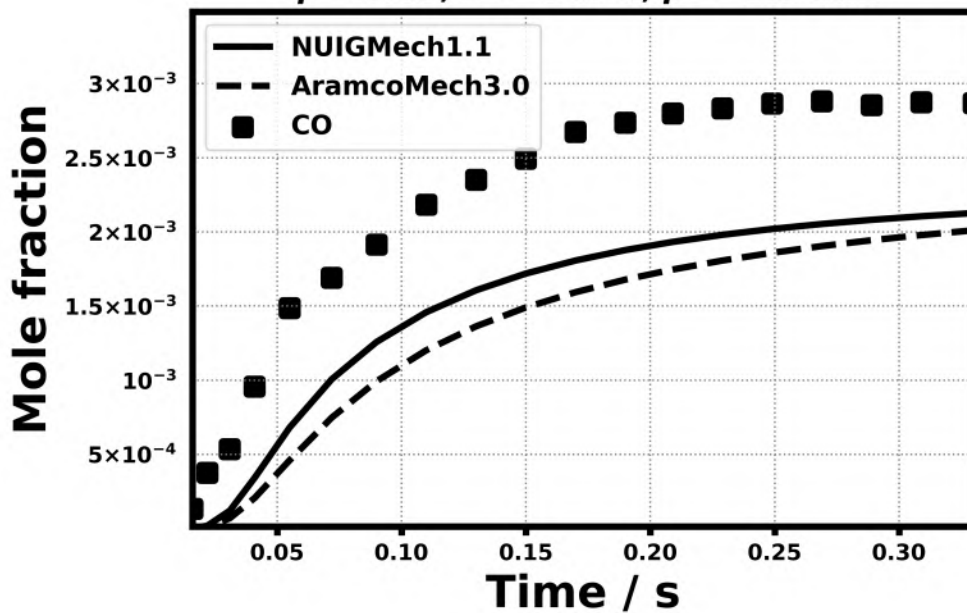
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



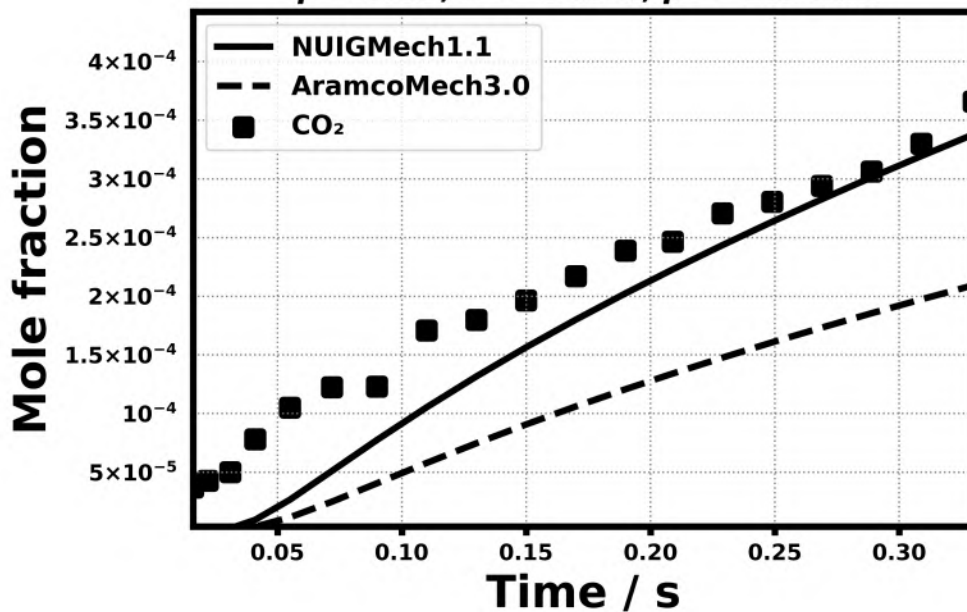
0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



0.333% CH₃OH
0.602% O₂, 99.065% N₂
 $\phi = 0.83, T = 949 \text{ K}, p = 2.5 \text{ atm}$



Laminar flame speed

- 4.10) Veloo, P. S., Wang, Y. L., Egolfopoulos, F. N., & Westbrook, C. K., *Combustion and Flame*, 157(10), (2010) 1989-2004.
- 4.11) Vancoillie, J., Christensen, M., Nilsson, E. J. K., Verhelst, S., & Konnov, A. A., *Energy & fuels*, 26 (2012) 1557-1564.
- 4.12) Sileghem, L., Alekseev, V.A., Vancoillie, J., Nilsson, E.J.K., Verhelst, S. and Konnov, A.A., *Fuel*, 115 (2014) 32-40.

